

## ***CHAPTER 2***

### **Mycotoxins with Special Reference to the Carcinogenic Mycotoxins: Aflatoxins, Ochratoxins and Fumonisin**

This chapter is an updated form of a review article that was published by Macmillan Reference in the second edition of *General and Applied Toxicology*, edited by Ballantyne, Marrs and Syversen in 1999.

#### *Contribution made by the candidate*

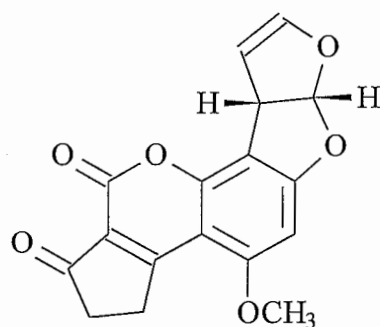
The candidate was responsible for the literature searches, research and writing of the majority of this chapter including the Ochratoxin, Aflatoxin and Fumonisin sections. Prof. P.S. Steyn co-ordinated the data compilation and wrote the sections: Mycotoxins produced by non-storage fungi and Trichothecenes.

# Mycotoxins with Special Reference to the Carcinogenic Mycotoxins: Aflatoxins, Ochratoxins and Fumonisin

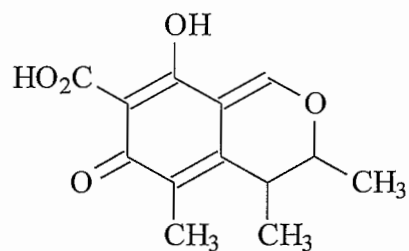
## INTRODUCTION

Naturally occurring toxicants produced by microorganisms, such as bacteria and fungi (moulds), contaminate foods and feeds; these foodborne hazards pose a serious health risk to mammals, fish and poultry. This chapter is exclusively dedicated to toxins produced by fungi, viz. mycotoxins and to diseases caused by the ingestion of mycotoxins, called mycotoxicoses: the toxin production can take place in the preharvest and/or during the postharvest stage of the crop.

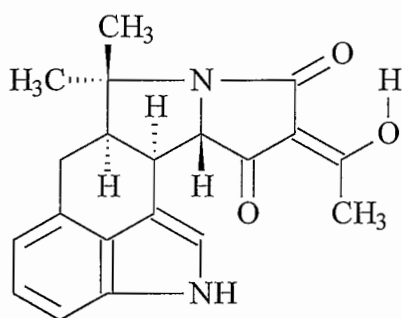
The well known mycotoxicologist, Forgacs, referred to mycotoxicoses in the early 1960s as the most neglected diseases, although many people in Russia died during World War II owing to alimentary toxic aleukia (ATA), a mycotoxicosis caused by T-2 toxin, a sesquiterpenoid mycotoxin (Ueno *et al.*, 1972; Yagen *et al.*, 1977). The resurgence of interest in mycotoxin research is directly related to the discovery of the aflatoxins during 1960, a group of structurally related hepatocarcinogens, produced on nuts and cereals by *Aspergillus flavus*, *Aspergillus parasiticus* and *Aspergillus nomius*, and their role in the aetiology of primary liver cancer in humans (Van Rensburg, 1986; Bressac *et al.*, 1991 and Groopman *et al.*, 1992). This event led to an unabated interest in mycotoxins as evidenced by the large number of monographs (Uraguchi and Yamazaki, 1978; Steyn, 1980; Lacey, 1985; Cole, 1986; Steyn, 1989; Natori, *et al.*, 1989; Smith and Henderson, 1991; Creppy *et al.*, 1993; Miller and Trenholm, 1994, and Jackson, *et al.*, 1996), numerous reviews (e.g. Steyn and Vleggaar, 1985; Pohland, *et al.*, 1992; Steyn, 1993; 1995; Beardall and Miller, 1994; Grove 1996; Bennet and Keller, 1997) and thousands of research papers. Mycotoxins are a chemically heterogeneous group (see **Figure 1** for the structures of representative members of the important mycotoxins) of low molecular weight compounds which are produced by the secondary metabolism of fungal genera such as *Aspergillus*, *Penicillium*, *Fusarium*, *Alternaria* and *Claviceps*. The mycotoxins are mostly produced by the so-called storage fungi; however, some unique mycotoxicoses such as ergotism, lupinosis and facial eczema are caused by some parasitic and saprophytic fungi (see later). It is, therefore, not surprising that mycotoxins induce powerful and dissimilar pathological effects as shown in **Table 1**.



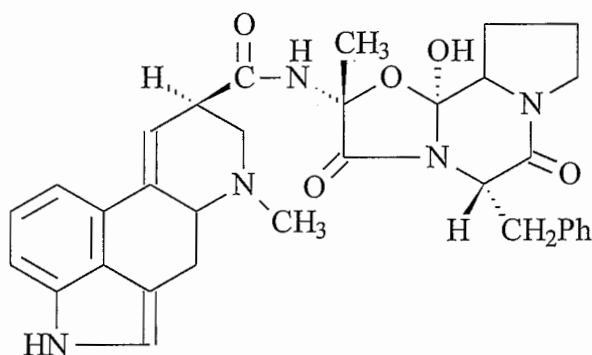
Aflatoxin B<sub>1</sub>



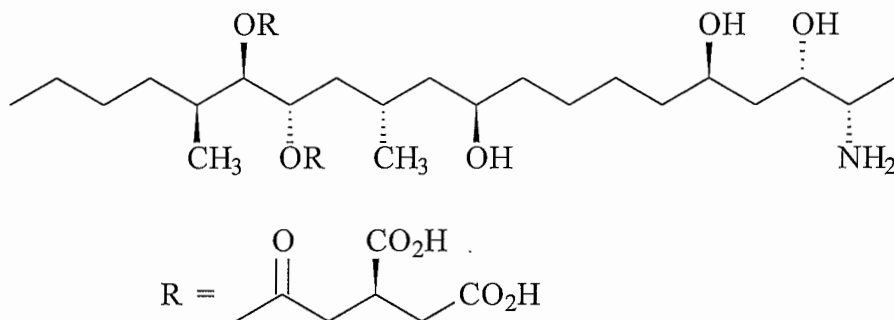
Citrinin



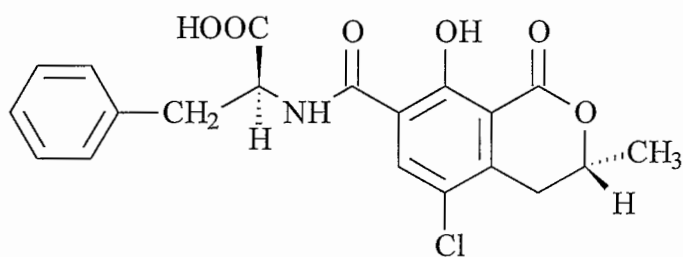
Cyclopiazonic acid



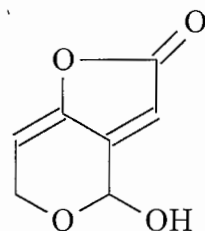
Ergotamine



Fumonisin B<sub>1</sub>

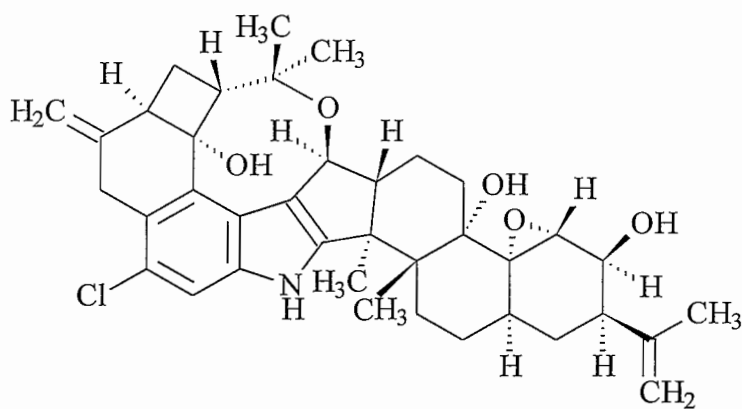


Ochratoxin A

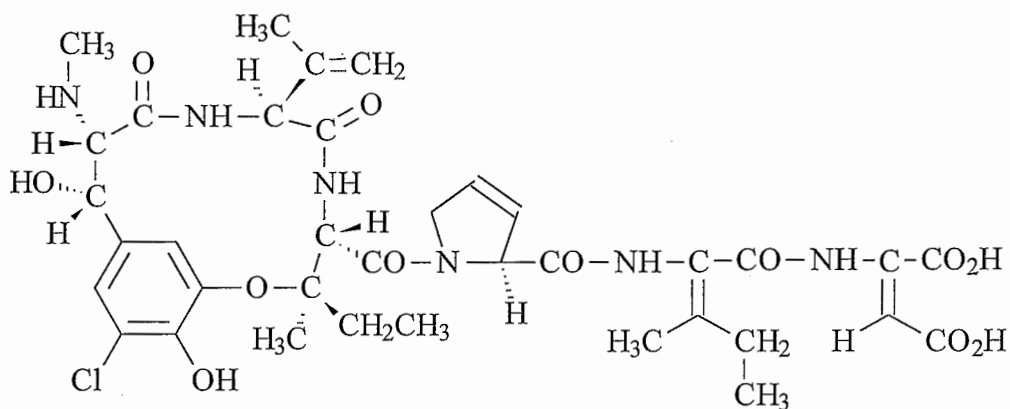


Patulin

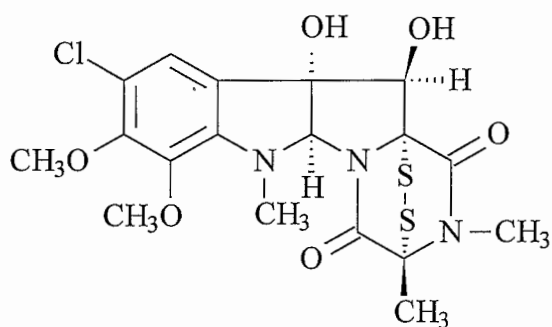
**Figure 1:** Structures of representative mycotoxins



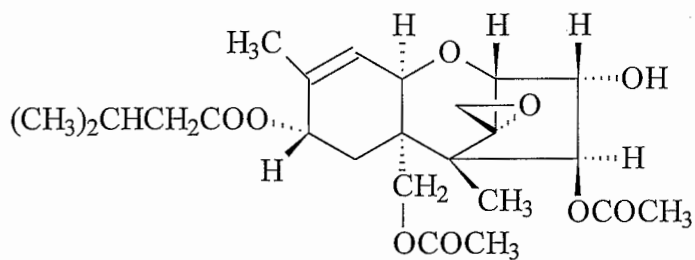
Penitrem A



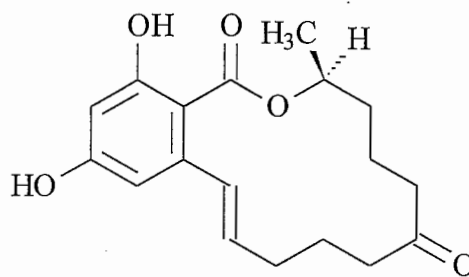
Phomopsin A



Sporidesmin



T2-toxin



Zearalenone

Figure 1: Continues

**Table 1:** Diverse biological activity displayed by some representative mycotoxins.

Mycotoxin	Biological Activity	Producing Genus	
Aflatoxin B <sub>1</sub>	Carcinogenicity, teratogenicity	<i>Aspergillus</i>	Büchi <i>et al.</i> (1966); Van Rensburg (1986), Groopman <i>et al.</i> (1992)
Citrinin	Nephrotoxicity	<i>Penicillium</i> , <i>Aspergillus</i>	Betina (1984)
$\alpha$ -Cyclopiazonic acid	Neurotoxicity	<i>Penicillium</i> , <i>Aspergillus</i>	Holzapfel (1968)
Ergotoxins (ergotamine)	Vasoconstriction, neurotoxicity	<i>Claviceps</i>	Stoll (1952); Scott <i>et al.</i> (1992)
Fumonisin B <sub>1</sub>	Carcinogenicity, neurotoxicity	<i>Fusarium</i>	Bezuidenhout <i>et al.</i> (1988); Jackson <i>et al.</i> (1996)
Ochratoxin A	Carcinogenicity, nephrotoxicity	<i>Aspergillus</i> , <i>Penicillium</i>	Van der Merwe <i>et al.</i> (1965); Pohland <i>et al.</i> , (1992); Creppy <i>et al.</i> (1993).
Patulin	Mutagenicity, antibacterial	<i>Penicillium</i>	Engel and Teuber (1984)
Penitrem A	Neurotoxicity	<i>Penicillium</i>	De Jesus <i>et al.</i> (1983); Steyn and Vleggaar (1985)
Phomopsin A	Hepatotoxicity	<i>Phomopsis</i>	Culvenor <i>et al.</i> (1989)
Sporidesmin A	Hepatotoxicity, photosensitivity	<i>Pithomyces</i>	Mortimer <i>et al.</i> (1978)
Trichothecenes (T-2 toxin)	Dermatotoxicity, hematopoietic	<i>Fusarium</i>	Wannemacher <i>et al.</i> (1991); Plattner <i>et al.</i> (1989); Grove (1996)
Zearalenone	Estrogenism, reproductive irregularities	<i>Fusarium</i>	Urry <i>et al.</i> (1966)

The mycotoxicooses are not only clinically diverse, but also often extremely difficult to diagnose owing to the numerous pharmacological effects of the causative mycotoxins. Some of the diseases associated with mycotoxins are, for example, aflatoxin [human primary liver cancer and Turkey-X disease (Van Rensburg, 1986; Bressac *et al.*, 1991)]; citreoviridin (Yellow rice disease in humans); ergotoxins [ergotism, St. Anthony's Fire in humans (van Rensburg and Altenkirk, 1974)]; fumonisins [encephalomalacia in horses, pulmonary oedema in swine (Bezuidenhout *et al.*, 1988)]; ochratoxins [nephropathy in pigs (Danish porcine nephropathy, Krogh, 1974) and poultry (Pohland *et al.*, 1992)]; phomopsin A [lupinosis in sheep (Culvenor *et al.*, 1989)]; sporidesmin A [facial eczema in sheep (Mortimer *et al.*,

1978)]; T-2 toxin and other trichothecene toxins [alimentary toxic aleukia (ATA)] and zearalenone (hyperestrogenism, vulvovaginitis and abortion in swine).

Mycotoxins are ubiquitous owing to the global distribution of toxinogenic fungi, thereby putting crops and consumers (man and animals) at risk and cause serious problems in the agricultural economies and international trade of nuts and cereals. The level of mycotoxin contamination of such commodities varies from year to year, depending on climatic conditions, commodity and location in a country. It has been estimated that one quarter of the world's food crops is at risk owing to mycotoxin contamination. Ammoniation can be effectively utilised to reduce aflatoxin levels in corn and cotton seeds by more than 99% (Lee *et al.*, 1992; Park, 1992).

Kuiper-Goodman, (1995, 1996), Kuiper-Goodman and Scott, (1989) and Kuiper-Goodman *et al.*, (1996) have made sterling contributions to the risk assessment of mycotoxins. The human health concerns depend on the amount of the mycotoxins consumed, the toxicity of the compound (extrapolation of the test species to humans), the body weight and physical condition of the individual, the presence of other mycotoxins as well as other dietary factors. In the case of animal mycotoxicoses, outbreaks vary according to the agricultural practice and climatic conditions in a region.

Hsieh, (1990) defined the criteria of a human mycotoxicosis as follows:

- Occurrence of the mycotoxin(s) in food supplies;
- Human exposure to the mycotoxins;
- Correlation between exposure and incidence;
- Reproducibility of characteristic symptoms in experimental animals;
- Similar mode of action in humans and animal models.

The extrapolation of toxicological data from animals to humans using safety factors or other methods to arrive at an estimate of safe intake is the most challenging aspect of such assessments (Kuiper-Goodman, 1995). The extrapolation of animal toxicity data to humans is complicated by species differences in metabolic disposition, such as differences in absorption, and binding to plasma and tissue constituents. Species differences in biotransformation as well as plasma and tissue half-life are also important.

The study of a specific mycotoxicosis requires the isolation and identification of the toxinogenic fungus involved, the chemical identification of the mycotoxin(s) and accurate and

reliable laboratory methods for monitoring and regulating the mycotoxin and its metabolic products in different matrices. The toxins of most of the food-and feed-borne fungi have been characterized; the research benefited from the advent of effective and mild chromatographic techniques, high-resolution mass spectroscopy, (including MS-MS, GLC-MS and HPLC-MS), high-field nuclear magnetic resonance spectroscopy and single crystal X-ray crystallography (Cole, 1986). The routine screening of food/feed samples benefited greatly from the ELISA analysis (Morgan, 1989); reliable analytical kits are now commercially available.

Some mycotoxins, (listed in **Table 1** and **Figure 1**) are associated with specific human (Beardall and Miller, 1995) and animal mycotoxicoses (Smith and Henderson, 1991). In the following section special attention will be directed to three of the unique groups of toxins, viz. ergotoxins, sporidesmins and phomopsins.

## MYCOTOXINS PRODUCED BY NON-STORAGE FUNGI

### Ergotoxins

The ergotoxins (e.g. ergotamine, ergocristine, ergocryptine, and ergocornine) are among the most pharmacologically active peptides and are the main alkaloids of *Claviceps purpurea*, the etiological agent in gangrenous and convulsive ergotism. Ergotism is probably the oldest known human mycotoxicosis, known as St Anthony's Fire or the Holy Fire in the Middle Ages in Europe, and is caused by consumption of rye flour contaminated with *C. purpurea* (Stoll, 1952; Van Rensburg and Altenkirk, 1974 and Lacey, 1991). As a human disease, ergotism has almost been eliminated (Scott *et al.*, 1992), but as an animal disease it can still occur widely, in the latter case also owing to contamination with *Claviceps paspali*. An isolated case of human ergotism occurred in Wollo, Ethiopia, when 150 people died due to the consumption of wild oats (*Avena abyssinica*) contaminated by a *Claviceps* species (King, 1979). The ergot alkaloids of *C. purpurea* are firmly associated with the plant species because the causative fungus is a specific pathogen of the rye plant.

## Sporidesmins

The sporidesmins, e.g. sporidesmin A (**Table 1** and **Figure 1**), are a group of epipolythiodioxopiperazines which cause photosensitization diseases among sheep in New Zealand (facial eczema) and in South Africa [yellow thick head disease ('geeldikkop' - Afrikaans)]. In New Zealand, the saprophytic fungus *Pithomyces chartarum* infects the senescent and dead material of rye grass pastures which are consumed by sheep grazing on the new spring growth. Facial eczema frequently caused major losses, e.g. \$40-100 million in 1981 (Smith, 1985). In the semi-arid Karroo region of South Africa, a similar photosensitization disease among sheep was associated with consumption of a common weed, *Tribulus terrestris*, infected with the perfect or teleomorph species *Leptosphaerulina* of which the *Pithomyces* is an anamorph (Roux, 1986). A severe outbreak of 'geeldikkop' incapacitated 250 000 sheep during a serious outbreak in 1949. Facial eczema is a secondary photosensitive expression of the toxic effects of sporidesmin on the liver (Mortimer *et al.*, 1978). In addition to its direct toxic effect on the liver parenchyma, excretion of the sporidesmin in the bile results in inflammation of the bile duct epithelium, followed by progressive necrosis of the duct wall and periductal concentric lamellar fibrosis and granulation causing the ducts to eventually become occluded (Mantle, 1991). Excretion of phylloerythrin, the product of hepatic degradation of chlorophyll is thus impaired by the obliterative cholangitis, and the abnormally high level of phylloerythrin in the peripheral blood thus causes photosensitivity leading to oedema and inflammation of the exposed skin of sheep (Mantle, 1991).

## Phomopsins

The phomopsins, a group of complex hexapeptides, containing  $\beta$ -dehydroamino acids, e.g. phomopsin A (**Table 1** and **Figure 1**), cause lupinosis in sheep in Australia and South Africa (Culvenor *et al.* 1989). The phomopsins are produced by *Phomopsis leptostromiformis* (Kühn). Bubak *ex* Lind, a fungus which in nature appears to be a specific pathogen and saprophyte of *Lupinus* spp. However, it can be cultivated on cereals (maize) and liquid media and still retain its toxigenicity. The liver is the major target for toxicity associated with the phomopsins; the affected liver accumulates lipids, turns yellow and becomes enlarged (Jago *et al.*, 1982). In case of long term exposure (low levels of the phomopsins), atrophy of the liver, fibrosis and bile duct proliferation develop. Phomopsin A acts as a mitotic drug both *in vivo*



and *in vitro*, and the observed symptoms of lupinosis can be related to the specific interaction of phomopsins with tubulin and microtubules (Tönsing *et al.*, 1984).

## TRICOTHECENES

This chapter is primarily dedicated to the carcinogenic mycotoxins: aflatoxins, ochratoxins and fumonisins. However, the trichothecenes are sufficiently important to warrant a brief description. The chemistry of the trichothecenes has been adequately covered by Grove (1996), who reported that a total of 182 trichothecenes, based on the trichothecane skeleton have been isolated from natural sources. They comprise 113 non-macrocyclic and 69 macrocyclic compounds.

The trichothecenes are produced by various species of *Fusarium*, *Trichoderma*, *Myrothecium*, *Verticimonisporium* and *Stachybotris* and comprise a group of closely related chemical compounds designated sesquiterpenoids. All the naturally occurring toxins contain an olefinic bond at C-9,10 and an epoxy group at C-12,13 (Grove, 1996)(See **Figure 1**). The trichothecene mycotoxins do not require metabolic activation prior to exerting their toxic effects.

The trichothecenes occur frequently in nature, and have been implicated in ATA in Russia, scabby grain toxification (Tatsuno, 1997) as well as in a number of animal diseases such as skin toxicity, bone marrow damage, haemorrhagic and ill-thrift syndromes (Wannemacher *et al.*, 1991). ATA was associated with the death of more than 10% of the population in Orenburg district, close to Siberia, during the period 1942-47. The symptoms of ATA, which include vomiting, diarrhoea, skin inflammation, leukopenia, multiple haemorrhage and exhaustion of the bone marrow, are similar to those induced by T-2 toxin. Ueno *et al.* (1972) and Yagen *et al.* (1977) concluded that T-2 toxin was the likely aetiological agent in ATA. The trichothecenes, e.g. T-2 toxin and nivalenol (NIV) induce karyorrhexis in actively dividing cells, a marked reduction in bone marrow cells and have the ability to inhibit protein and DNA synthesis and induce apoptosis in HL-60 cells (Yoshino *et al.*, 1996, 1997a, b; Sugamata *et al.*, 1997). In the *in vitro* study on T-2 toxin induced apoptosis in human peripheral blood lymphocytes, Yoshino *et al.* (1997a) observed that the toxin affected the human peripheral blood lymphocytes, and elicited apoptic cell death, causing, in part, a marked decrease in circulating white blood cells as observed in animals which received T-2 toxin. In the ultrastructural study of apoptic cellular damage induced by acute NIV toxicoses

in mice, Sugamata *et al.* (1997) observed NIV to be a potent inducer of apoptic cell death in the thymus, spleen and liver.

The trichothecenes such as T-2 toxin, diacetoxyscirpenol (DAS), vomitoxin, and 4-deoxynivalenol (DON) are frequent contaminants of agricultural commodities (Tanaka *et al.*, 1988; ApSimon *et al.*, 1990; Gilbert, 1995; Scott, 1997). The trichothecenes, e.g. T-2 toxin are optimally produced at relatively low temperatures, viz. 8-14°C; however, Rabie *et al.* (1986) reported the production of large quantities of T-2 toxin at 25°C by *Fusarium acuminatum*. The analytical methodology for the trichothecenes in animal feedstuffs is well established (Steyn *et al.*, 1991), e.g. by applying mass spectrometry and tandem mass spectrometry (Plattner *et al.*, 1989). Yeasts (*Kluyveromyces marxianus*) and bacteria (*Bacillus brevis*) are useful indicator organisms for the bioassay of several of the common mycotoxins. Madhyastha *et al.*, (1994) evaluated the relative toxicity of 16 trichothecenes and some of their interactions by using *K. marxianus*, and found toxicity to decrease in the following order: T2-toxin > DAS > DON > NIV. Similar methodology and results were obtained by Engler *et al.*, (1999) who used a colorimetric technique.

## AFLATOXINS

The deaths in 1960 in England of over 100 000 turkeys and ducklings, which consumed Brazilian peanut meal, led to the discovery of the aflatoxins, a group of hepatocarcinogenic bishydrofurano mycotoxins produced by certain strains of *A. flavus* and *A. parasiticus*. Aflatoxin B<sub>1</sub> (AFB<sub>1</sub>) has the highest level of occurrence and is the most carcinogenic of the known aflatoxins, while aflatoxin M<sub>1</sub> (AFM<sub>1</sub>) is excreted in the milk of cows and has toxic properties similar to those of AFB<sub>1</sub> (Holzapfel *et al.*, 1966).

## Chemistry and Metabolism

Aflatoxins B<sub>1</sub>, B<sub>2</sub>, G<sub>1</sub> and G<sub>2</sub> are the main metabolites of *A. flavus* and *A. parasiticus*. Büchi *et al.* (1966) elucidated the structures of the aflatoxins, and confirmed these by total synthesis. Aflatoxins may be classified into two broad groups according to their chemical structure: the difurocoumarocyclopentenone series (AFB<sub>1</sub>, AFB<sub>2</sub>, AFB<sub>2a</sub>, AFM<sub>1</sub>, AFM<sub>2</sub>, AFM<sub>2a</sub> and aflatoxicol) and the difurocoumarolactone series (AFG<sub>1</sub>, AFG<sub>2</sub>, AFG<sub>2a</sub>, AFGM<sub>1</sub>, AFGM<sub>2</sub>, AFGM<sub>2a</sub> and AFB<sub>3</sub>) (Heathcote, 1984)(see **Figure 2**). Aflatoxin B<sub>1</sub> (C<sub>17</sub>H<sub>12</sub>O<sub>6</sub>) is a pale-

white to yellow, odourless, crystalline solid with a blue fluorescence under UV light (see **Table 2** for physical and spectroscopic data).

The aflatoxins display decreasing potency in the order  $B_1 > G_1 > B_2 > G_2$ , as illustrated by their  $LD_{50}$  values for day-old ducklings (**Table 3**). Structurally, the dihydrofuran moiety, containing the double bond and the substituents linked to the coumarin moiety are of importance in producing biological effects. Demethylation of AFB<sub>1</sub> leads to a toxic derivative, AFP<sub>1</sub> and hydroxylation of the bridge carbon of the furan rings to AFM<sub>1</sub>, with similar toxic effects as AFB<sub>1</sub>. AFM<sub>1</sub> is, however, considerably less carcinogenic than AFB<sub>1</sub>.

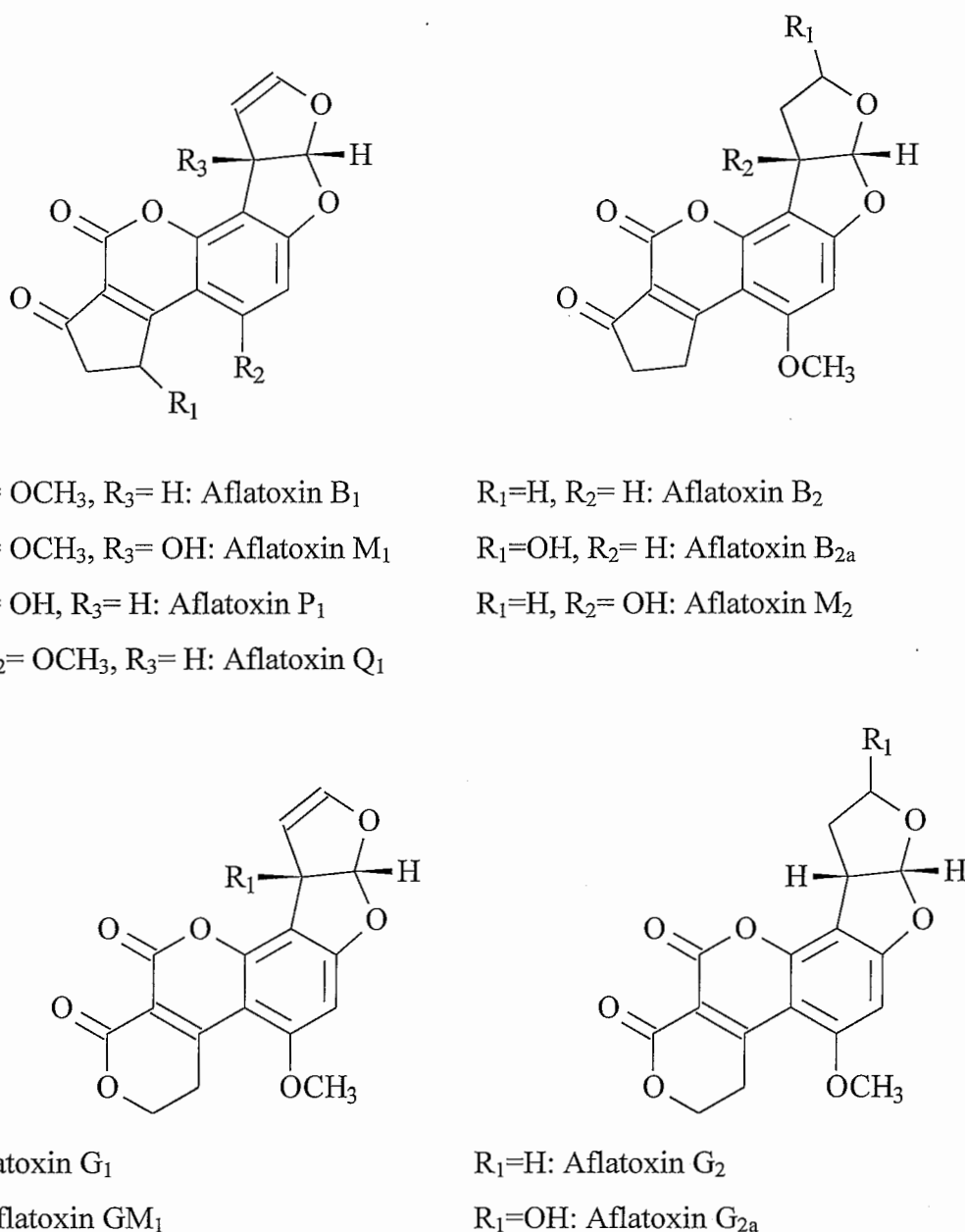
**Table 2:** Physical and spectroscopic data of aflatoxin B<sub>1</sub>

Melting point	269-271 °C
Optical rotation $[\alpha]_D$	-559 ° (concentration 625 µmol/l in chloroform)
Infrared	Strong bands at 1770, 1600, 1570, 1390 and 1310 nm.
Molecular weight	312.3

Source: Pohland *et al.* (1982).

Aflatoxin	$LD_{50}$ (µg per day-old duckling)
B <sub>1</sub>	18
G <sub>1</sub>	39
B <sub>2</sub>	84
G <sub>2</sub>	173
M <sub>1</sub>	17
M <sub>2</sub>	62

**Table 3:** Toxicities of the principal aflatoxins (Carnaghan *et al.*, 1963; Heathcote, 1984).



**Figure 2:** Structures of the important aflatoxins.

## Biosynthesis

In an effort to control aflatoxin contamination of food and feed, scientists focused on the biosynthetic pathway of aflatoxin to understand its regulation and evolution, and to eliminate the toxin from the food chain. Aflatoxin thus has one of the best studied polyketide pathways known (Trail *et al.*, 1995). The pathway involves approximately 20 enzymes: AFB<sub>1</sub> and AFB<sub>2</sub> are produced by two parallel pathways (Bennett *et al.*, 1994 and references cited). There are two possible initial steps: the first one involves condensation of acetate and nine

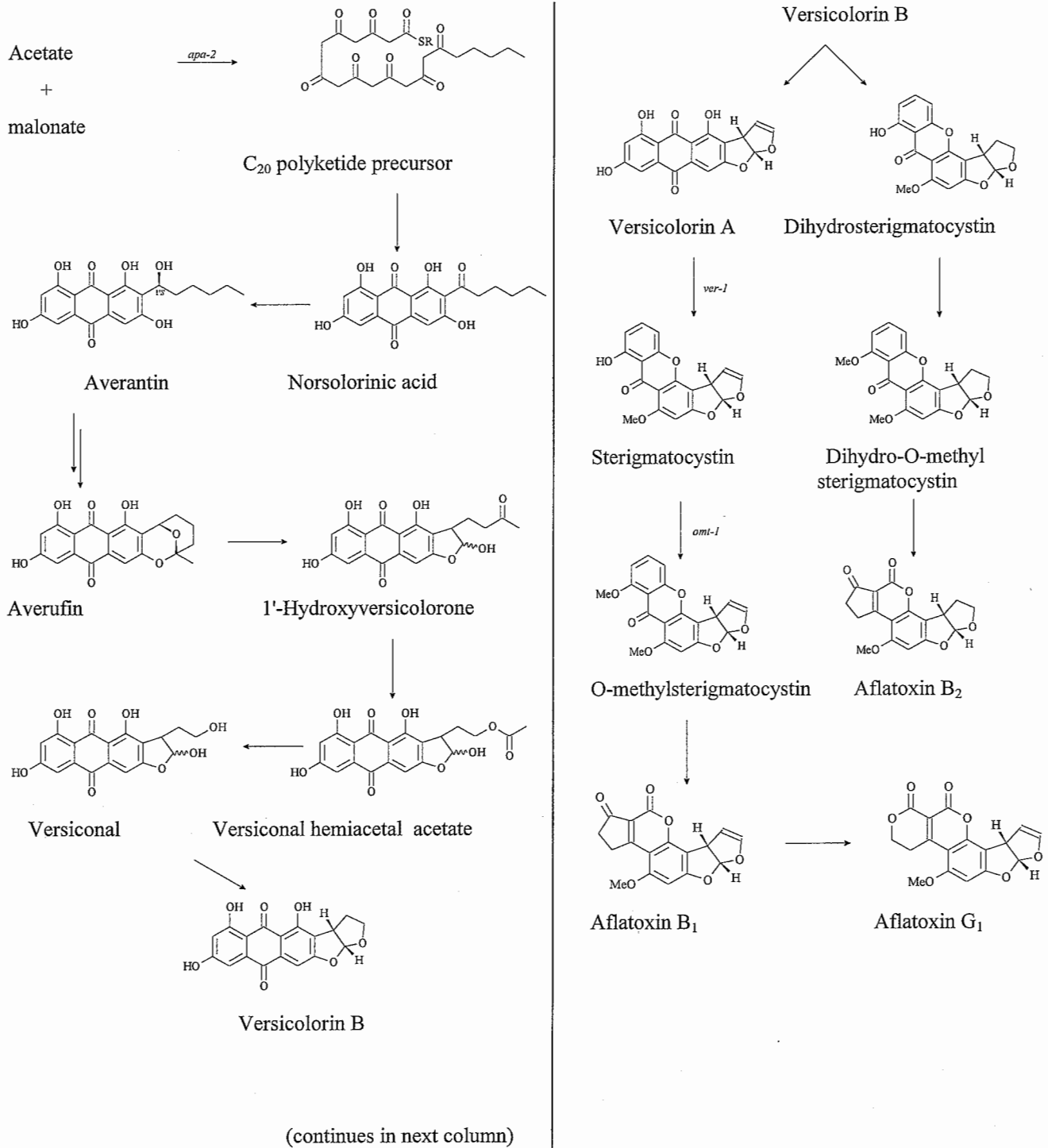
malonate units, whereas the alternative step involves the early synthesis of a 6-carbon hexanoate (Townsend, 1986) [which is then extended by a polyketide synthetase to generate a C<sub>20</sub> polyketide (Trail *et al.*, 1995 and references cited; Steyn, 1980)]. (see **Figure 3** for the proposed pathway). The final step in aflatoxin biosynthesis involves the oxidative cleavage and rearrangement of O-methyl-sterigmatocystin with loss of a C<sub>1</sub>-unit (Chatterjee and Townsend, 1994).

The detection of aflatoxigenic fungi in grains, using PCR methodology, is based on three genes from the aflatoxin biosynthetic pathway (**Figure 3**) (Shapira *et al.*, 1996). The three genes code for key enzymes involved in discrete biosynthetic steps: polyketide → norsolorinic acid (*apa-z*), versicolorin A → sterigmatocystin (*ver-1*) → O-methylsterigmatocystin (*omt-1*). The DNA sequences of the enzymes were established, and three primer pairs, each complementing the coding portion of one of the genes were generated. The PCR technology enabled Shapira *et al.* (1996) to differentiate between the aflatoxigenic strains of *A. flavus* and *A. parasiticus* and the non-aflatoxigenic *Penicillium* and *Aspergillus* species.

## Production

The aflatoxins are produced by *A. flavus*, *A. parasiticus*, *A. nomius* and *A. tamaritii* (Goto *et al.*, 1996). Watson *et al.* (1999) found the presence, but not expression, of homologues of three structural genes and a regulatory gene necessary for aflatoxin biosynthesis in *Aspergillus parasiticus* and *A. flavus* in *A. oryzae* and *A. sojae*. The latter two fungi are important fungi used in the food and ingredient manufacture.

In addition to genetic requirements for production, the yield of aflatoxin depends on the growth conditions, such as moisture, temperature (optimal conditions for *A. flavus* are 16-24% moisture at 20-38 °C), substrate, aeration (culturing moulds on a rotary shaker greatly increases yields), and other factors which affect the qualitative state of development of the mould.



**Figure 3:** Biosynthesis of aflatoxin. Source: Townsend (1986), Steyn, (1980).

The production yields of the different aflatoxins vary with different growth conditions e.g. enhanced levels of AFB<sub>1</sub> relative to the different aflatoxins vary with different growth conditions e.g. enhanced levels of AFB<sub>1</sub> relative to AFG<sub>1</sub> occur in *A. parasiticus* at elevated temperatures as a result of accelerated catabolism of AFG<sub>1</sub> (Detroy and Ciegler 1971 and references cited). Although 20-38 °C is the optimum temperature range for production, aflatoxin formation can also take place at temperatures as low as 7-12 °C, if an extended incubation period is utilised. Therefore the storage of commodities at reduced temperatures can not be used to prevent aflatoxin production. Aflatoxin production is also affected by trace metals, insecticides, herbal drugs, spices, tricarboxylic acid cycle intermediates and food preservatives and is highly dependent on nutritional factors, e.g. sources of carbohydrates like glycerol (Mateles and Adye, 1965) and glucose.

## Determination

At first, the analysis of the aflatoxins involved the grinding of the samples, Soxhlet extraction, solvent partition and clean-up by SiO<sub>2</sub> columns followed by determination by paper chromatography, which was subsequently replaced by SiO<sub>2</sub> TLC. (Shepherd *et al.*, 1987). A number of samples can be analysed simultaneously on one TLC plate and confirmation achieved by derivatisation on the plate followed by a second development. The modern techniques rely on the same principles, although ready-packed clean-up columns are used which contain silica or modified silica (especially C<sub>18</sub>-bonded phase), e.g. Sep-Pak (Waters, Milford, MA, U.S.A.) and Bond-Elut (Analytichem International, Harbor City, CA, USA); however, the introduction of HPLC is replacing TLC methods in the final quantification step (Shepherd *et al.*, 1987). The problem of fluorescence quenching can be circumvented by pre-column derivatisation with trifluoroacetic acid or post-column reaction with bromine or iodine (Shepherd and Gilbert, 1984; Kok *et al.*, 1986). The latter method has a lower detection limit of 5-30 pg of AFB<sub>1</sub>. Cepeda *et al.*, (1996) introduced cyclodextrines to increase the fluorescence of AFB<sub>1</sub> and AFG<sub>1</sub> in the postcolumn excitation of these toxins: substantial improvements (*ca* 30 fold) in detection limits of AFB<sub>1</sub> and AFG<sub>1</sub> were subsequently obtained by the use of heptakis-2,6-β-O-dimethyl-β-cyclodextrin. Gas chromatography-mass spectrometry (GC-MS) or liquid chromatography-electrospray ionisation tandem-mass spectrometry (LC-ESI-MS-MS) can be used for confirmation of aflatoxin B<sub>1</sub> (Kussak *et al.*, 1995). The quest for better and more accurate methods for the determination of the aflatoxins

is still continuing. Newer developments include the detection of the aflatoxins with an amperometric detector (Elizalde-González *et al.*, 1998). They separated the four aflatoxins by reversed phase HPLC and used a glassy carbon electrode at a constant potential of 1.4 V as amperometric detector. This method is more sensitive than conventional methods for the detection of the less toxic aflatoxin B<sub>2</sub>, which is always present in grains contaminated with aflatoxins. Methods for the determination of the aflatoxins are summarised in **Table 4**.

Substrate	Method	Recoveries, detection limits
Dust and urine	LC-MS	2 pg.mg <sup>-1</sup> , 50 pg.ml <sup>-1</sup> Kussak <i>et al.</i> (1995)
Milk	Affinity columns and HPLC	85.7 %, 50 pg.ml <sup>-1</sup> Mortimer <i>et al.</i> (1987)
Cheese	Affinity columns and HPLC	75 %, 5 ng.kg <sup>-1</sup> Sharman <i>et al.</i> (1992)
Ground peanuts	ELISA	62-84 %, 5 µg.kg <sup>-1</sup> Li <i>et al.</i> (1994)

**Table 4:** Methods for the determination of aflatoxins.

### Immunological methods

Immunological-based screening methods for aflatoxins are rapid: the columns are commercially available and the procedures can be fully automated. Holaday-Velasco minicolumns have been widely used since 1980 in the screening of aflatoxins in corn and peanuts. A disadvantage of this method is that it relies on the characteristic fluorescence of the aflatoxins, which can be very subjective (Gilbert, 1993). There are three types of immunological based assays: batchwise quantitative methods (ELISA), semiquantitative but rapid methods for single samples, and affinity column clean-up. Agri-screen is an ELISA method and Afla-20-cup, EZ-screen and Cite-probe are all commercially available kits for aflatoxin screening, based on the principles of ELISA but using absorbed antibodies in a sandwich format. Aflatest and Oxoid are commercially available affinity columns (Gilbert, 1993). The performance of these different kits has been assessed by Koeltzow and Tanner, (1990) and Dorner and Cole, (1989). Radioimmunoassays (RIA) for AFB<sub>1</sub> have already been reported as far back as 1976 but have not been commercialized (Shepherd *et al.*, 1987).

In commercial double-antibody ELISA kits a microtiter plate with aflatoxin-protein conjugate absorbed to the surface of the wells is supplied. Buffer-diluted methanol extracts of the



samples are pipetted separately into the wells of the titer plates, followed by a limited amount of anti-aflatoxin antibodies. There is competition between the bound aflatoxin and the aflatoxin in the sample for antibodies. The titer plate is then washed and a second antibody, with a colour-producing enzyme attached to it, is added, which binds to the anti-aflatoxin antibody bound to the well, and colour is produced with an intensity that is inversely proportional to the aflatoxin concentration in the sample, when the substrate for the colour producing enzyme is added. In single-antibody ELISAs the colour-producing enzyme is conjugated directly to the first anti-body. Although this assay is much quicker it uses more antibodies and is thus more expensive. Commercial ELISA kits have detection limits of about  $2 \mu\text{g.kg}^{-1}$  and are best suited for large batches of samples, since up to 93 samples can be assayed on one titer plate, but are not sufficiently reliable to be used as quantitative methods, (Shepherd *et al.*, 1987; Gilbert, 1993). However, recent developments enabled this technology to be utilised with a high level of sensitivity and precision ( $\text{ng.kg}^{-1}$ ) (Franco, 1996).

Immunoaffinity columns consist of an anti-aflatoxin antibody bound to a gel material contained in a small plastic cartridge. In practice the crude extract is forced through the column and the aflatoxin is left bound to the recognition site of the immunoglobulin. Extraneous material can be washed off the column with water or aqueous buffer, and the aflatoxin is obtained in a purified form by denaturing the protein gel by elution solvent such as methanol or acetonitrile (Gilbert, 1993). The aflatoxin in this eluate can be quantified with HPLC or with a standard UV spectrometer. The advantages of immunoaffinity columns are that the approach is the same for all matrices (peanuts, milk, nuts etc.), it is much faster and cleaner than conventional silica gel columns (there are no peak interferences when HPLC is used for quantification) and unlike ELISA, all four aflatoxins ( $\text{AFB}_1$ ,  $\text{AFB}_2$ ,  $\text{AFG}_1$  and  $\text{AFG}_2$ ) can be determined individually (HPLC analysis). The disadvantage is that it is still more time-consuming than ELISA (Gilbert, 1993).

The latest development is fiber-optic-based biosensors that uses the high fluorescence of the aflatoxins for detection. These detections require little specialized training and it is very quick to perform analysis with similar detection limits as ELISA's. Aflatoxin was detected using its native fluorescence and also in the competition mode with either FITC (fluorescein isothiocyanate)-labelled antibody or enzyme labelled antibody by Carter *et al.* (1997).

## Control and Decontamination

Ammonia has been used to destroy aflatoxins in various feedstuffs either in its gaseous form, or as an ammonium hydroxide solution (Simpson and Pemberton, 1989; Lee *et al.*, 1992; Park, 1992). Addition to animal feeds of sequestering agents, such as activated charcoal, sodium bentonite and hydrated sodium aluminosilicate, which bind to aflatoxin and decrease its bioavailability, has been proposed as a detoxification strategy (Bonna *et al.*, 1991; Harvey *et al.*, 1989; Lindemann *et al.*, 1990). Aflatoxin levels can also be reduced by the process of cooking, the dry roasting of peanuts and the popping of corn; the reduction is, however, modest (Simpson and Pemberton, 1989). Aflatoxin production can be controlled by the isolation or development of atoxigenic strains of *A. flavus* and *A. parasiticus*; these strains are able to competitively exclude toxigenic field strains from the host plant (Cole and Cotty, 1990; Cotty, 1994). Although these strains are not producing aflatoxins, they still have the genes for aflatoxin biosynthesis. Cary *et al.* (1999) isolated and characterized experimentally induced, aflatoxin biosynthetic pathway deletion mutants of *Aspergillus parasiticus* by engineering a plasmid vector (pDEL2) for the purpose of introducing a deletion within the aflatoxin biosynthetic gene cluster. The vector was constructed by PCR amplification of a region of the AF gene cluster from an *A. parasiticus* isolate that had undergone an aberrant recombinational event during transformation with a *norA-niaD* gene disruption vector. Maas *et al.* (1998) demonstrated that AFB<sub>1</sub> can be degraded up to 50 % by *Aspergillus niger* in liquid culture, the degradation product was, however, aflatoxicol which is also a carcinogenic compound. Similar studies on the ability of microorganisms to degrade aflatoxins has been investigated by several authors including Nakazato *et al.* (1990). Hoogenboom *et al.* (1998) studied, tested and compared three decontamination processes (two based on ammoniation and one based on biological degradation) in an EU project. One of these ammoniation methods, the SOCOFAG procedure was tested to see if the decontamination products are still hazardous or if they could become hazardous again being metabolized in the rumen or liver of a cow. Kuilman *et al.* (1998) found no aflatoxin metabolites in incubations of cultured bovine hepatocytes with extracts of decontaminated peanut meal.

## Occurrence

The aflatoxins are frequent contaminants of commodities such as corn (maize), peanuts, pecan nuts, Brazil nuts, cotton seeds, other energy-rich foodstuffs, and even herbs and spices (Davis

*et al.*, 1986; MacDonald and Castle, 1996; Resnik *et al.*, 1996). High levels of aflatoxin contamination are frequently associated with tropical climatic conditions, poor agricultural practices, drought stress, and insect and mechanical damages. Internationally due cognisance is usually given to the standards set by the US FDA (see **Table 5** for the permissible levels of aflatoxin contamination).

**Table 5:** Maximum levels for aflatoxin contamination set by US Food and Drug Administration.

Substrate	Maximum Level
Food for humans and feed for some animal species	20 ppb
Milk	0.5 ppb
Feed for feedlot cattle	300 ppb
Feed for market hogs	200 ppb
Feed for breeding cattle, breeding hogs and mature poultry	100 ppb

## Biological Effects and Mechanism of Action

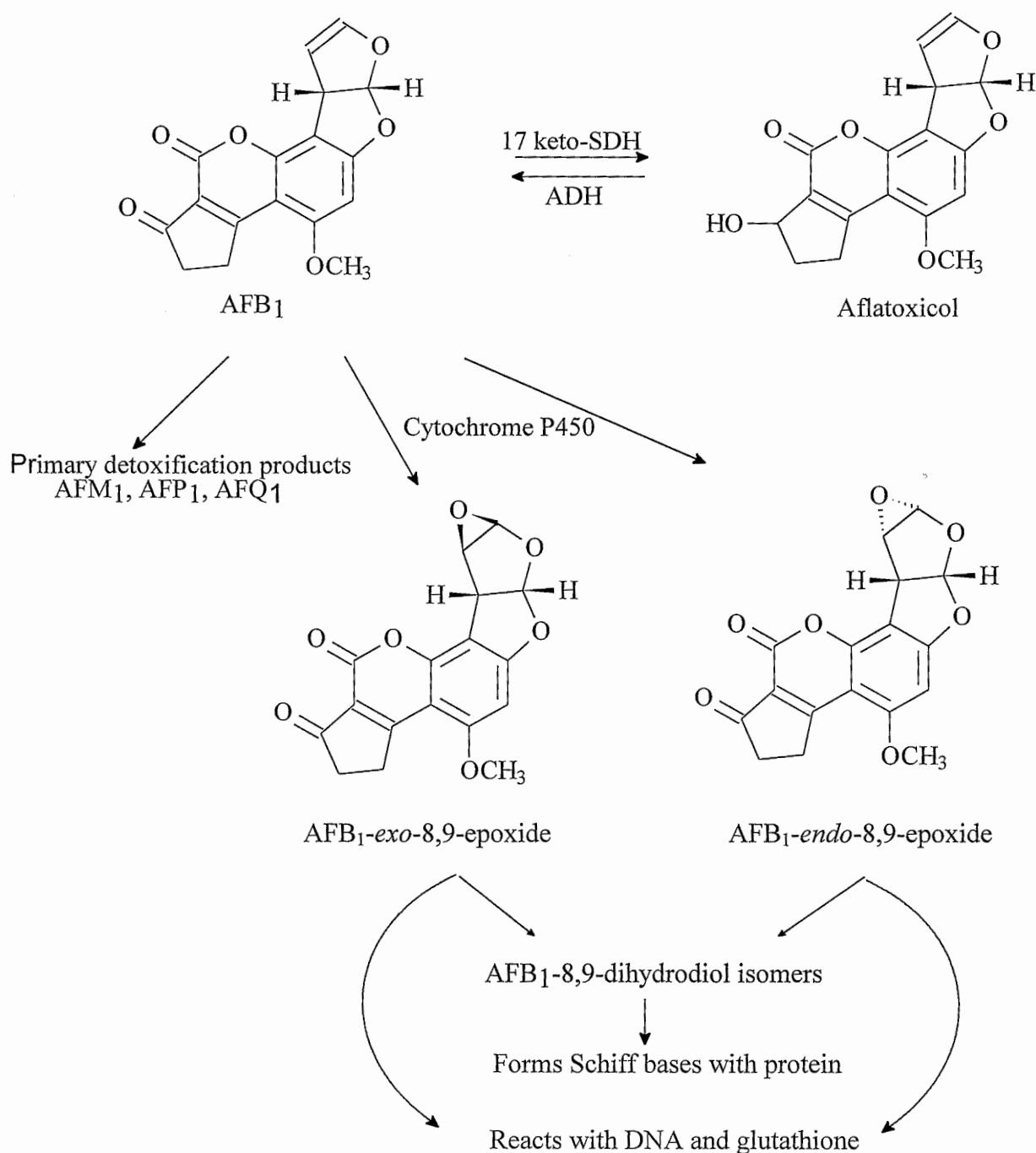
Toxicologically, the aflatoxins particularly AFB<sub>1</sub> should be regarded as a quadruple threat, i.e. as a potent toxin, a carcinogen, a teratogen and a mutagen. AFB<sub>1</sub> induces liver cancer in all animal species tested so far and has also been linked to liver cancer in humans (Wang *et al.*, 1996). Statistical correlations between contaminated food supplies and high frequencies of human hepatocellular carcinomas (HCC) in Africa and Asia have long implicated aflatoxins as risk factors in human liver cancer (Van Rensburg, 1986). The toxicity of AFB<sub>1</sub> toxicity varies from species to species and differences in susceptibility to aflatoxin are also found amongst individuals and between sexes (See **Table 6**). Numerous biochemical, toxicological and histological analyses have been performed to clarify the fundamental mechanisms of liver injury and hepatoma development. Molecular aetiology has substantiated the previous bio-statistical studies, since AFB<sub>1</sub> causes an activation of the *K1 ras* proto-oncogene and modulates the *p53* tumour suppressor gene. A molecular 'hot spot' in the *p53* gene, a G → T transversion at the third base position of codon 249 has been identified in independent studies on HCC patients from Qidong, China (Hsu *et al.*, 1991) and from sub-Saharan Africa (Bressac *et al.*, 1991). The IARC, Lyon, France classified AFB<sub>1</sub> as a human carcinogen in 1987 and confirmed the classification in 1992 and 1999.

**Table 6:** Acute oral toxicities of aflatoxins

Species	LD <sub>50</sub> (mg.kg <sup>-1</sup> )						
	B <sub>1</sub>	B <sub>2</sub>	G <sub>1</sub>	G <sub>2</sub>	M <sub>1</sub>	M <sub>2</sub>	B <sub>2a</sub>
Duckling	0.36	1.68	0.78	1.42	0.32	1.22	24
Rabbit	0.3						
Cat	0.55						
Pig	0.62						
Dog	0.5-1.0						
Sheep	1.0-2.0						
Guinea-pig	1.04						
Monkey	2.2-7.8						
Chicken	6.3						
Rat	7.2 (M)- 6 (F)						
Mouse	9.0						
Hamster	10.2						

Source: Ciegler (1975), Heathcote (1984).

The mutagenic and carcinogenic effects of AFB<sub>1</sub> have been well studied and are believed to arise from metabolic activation of the electron-rich dihydrobisfuran to the corresponding epoxide by the liver P450 isoenzyme CYP 2C and to a lesser extent CYP 1A2 in rats and CYP 3A4 in humans (Gopalkrishnan *et al.*, 1990; Ishii *et al.*, 1986; Forrester *et al.*, 1990). Raney *et al.*, (1992) demonstrated that the AFB<sub>1</sub>-epoxide occurs in an *endo* and *exo* form each with different affinities for DNA (see **Figure 4**). The *exo*-epoxide is highly electrophilic and reacts with regions in the DNA helixes which are rich in guanine to form covalent bonds at the N-7 of guanine residues, leading to depurination and strand scission events (Essigmann *et al.*, 1982). The *Salomonella* reversion assay indicates that the *exo*-epoxide is at least 500 times more potent as a mutagen than the *endo* stereoisomer (Iyer *et al.*, 1994). AFB<sub>2</sub> on the other hand has no double bond at this position and is practically inactive. AFB<sub>1</sub>-epoxide can be metabolised further to 8,9-dihydro-8,9-dihydroxy-aflatoxin B<sub>1</sub> which may bind to cellular proteins, via a Schiff base formation with primary amino groups, inducing cellular injury and eventually cell death (Fink-Gremmels, 1996). In mice virtually all the epoxidation leads to the *exo* isomer, in rats the ratio of *exo* to *endo* is 32 to 1 while in humans the proportion of *endo* to *exo* is higher than in the rat (Neal, 1995)(see **Table 6** for toxicities).



**Figure 4:** Metabolism of AFB<sub>1</sub> (Neal, 1995).

AFB<sub>1</sub>-epoxide forms a conjugate with glutathione by a glutathione-S-transferase (GST) mediated mechanism; glutathione is an alternative for aflatoxin to binding to other nucleophilic centres and is apparently the most important detoxification system (Degen and Neumann, 1978; Neal, 1995). As is evident from **Figure 4** several hydroxylations occur during the metabolism of AFB<sub>1</sub>, catalysed by cytochrome P450 enzymes, leading to its

secondary metabolism (AFM<sub>1</sub>, AFP<sub>1</sub> and AFQ<sub>1</sub>). The secondary conjugating processes involve glucuronidation, sulphation and acetylation of primary AFB<sub>1</sub> metabolites (Neal, 1995).

## OCHRATOXIN A

The ochratoxins, metabolites of *Aspergillus ochraceus* Wilh. (van der Merwe *et al.*, 1965), are the first group of mycotoxins discovered subsequent to the epoch-making discovery of the aflatoxins. Ochratoxin A (OTA) is a very important mycotoxin owing to its frequent occurrence in nature, its established role in Danish porcine nephropathy and in poultry mycotoxicoses and its implicated role in Balkan endemic nephropathy and urinary system tumours in North Africa (Achour *et al.*, 1993, Bacha *et al.*, 1993, Maaroufi *et al.*, 1995).

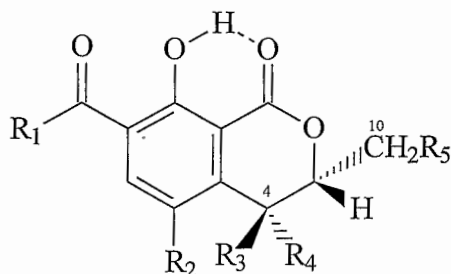
### Chemical characteristics and biosynthesis of OTA

Ochratoxin A comprises of a pentaketide-derived dihydroisocoumarin moiety linked via its 12-carboxy group by a peptide bond to *L*-β-phenylalanine (see **Figure 5**). It forms colourless crystals when recrystallized from benzene (mp. 90 °C with the loss of benzene) and it melts at 169-171 °C upon crystallization from xylene (Van der Merwe *et al.*, 1965).

The IR spectrum of OTA (CHCl<sub>3</sub>) displays bands at 1655, 1535 and 3430 cm<sup>-1</sup> (secondary amide); 1723 cm<sup>-1</sup> and broad band between 2500 and 2700 cm<sup>-1</sup> (carboxyl carbonyl group) and a band at 1678 cm<sup>-1</sup> (lactone carbonyl group). The UV absorption spectra of OTA has λ<sub>max</sub> 216 nm (ε = 31 500) and 330 nm (ε = 6 400) in MeOH / 0,0005 M H<sub>2</sub>SO<sub>4</sub>. The most abundant peaks in the mass spectrum of OTA are: *m/z* M<sup>+</sup> 403 (13%), 359 (31%), 358 (18%), 357 (14%), 258 (52%), 257 (97%), 256 (100%), 255 (86%), 242 (49%), 241 (94%), 239 (81%), 238 (99%). IR spectroscopy and X-ray crystallography have demonstrated when OTA exists in solution and in the solid state in the β form; viz. the amide NH is hydrogen-bonded to the phenolic oxygen (Bredenkamp *et al.*, 1989). <sup>13</sup>C NMR spectroscopy provided evidence for hydrogen bonding of the phenolic proton to the lactone carbonyl group (Bredenkamp *et al.*, 1989).

The biosynthetic origin of OTA was established employing radioactive precursors, *e.g.*, [1-<sup>14</sup>C]- and [2-<sup>14</sup>C]acetate, [2-<sup>14</sup>C]malonate, DL-[methyl-<sup>14</sup>C]methionine and chlorine-36 (Steyn *et al.*, 1970) and stable isotope precursors, *e.g.* sodium [<sup>13</sup>C]formate, [1-<sup>13</sup>C]- and [1,2-

$^{13}\text{C}$ ]acetate (De Jesus *et al.*, 1980). OTA is derived from combined pathways, viz., the shikimic acid pathway (phenylalanine) and the polyketide pathway (dihydroisocoumarin); the chlorine atom is probably derived through the action of a chloroperoxidase.



	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>
Ochratoxin A	Phenylalanine	Cl	H	H	H
Ochratoxin B	Phenylalanine	H	H	H	H
Ochratoxin C	Phenylalanine ethyl ester	Cl	H	H	H
Ochratoxin A methyl ester	Phenylalanine methyl ester	Cl	H	H	H
Ochratoxin B methyl ester	Phenylalanine methyl ester	H	H	H	H
Ochratoxin B ethyl ester	Phenylalanine ethyl ester	H	H	H	H
Ochratoxin α	OH	Cl	H	H	H
Ochratoxin β	OH	H	H	H	H
4 <i>S</i> -Hydroxyochratoxin A	Phenylalanine	Cl	OH	H	H
4 <i>R</i> -Hydroxyochratoxin A	Phenylalanine	Cl	H	OH	H
10-Hydroxyochratoxin A	Phenylalanine	Cl	H	H	OH
Bromo-ochratoxin B	Phenylalanine	Br	H	H	H

**Figure 5:** Structures of the ochratoxins.

## Analogues of OTA

OTB, the natural des-chloro analogue of OTA, is 10 times less toxic than OTA. OTA can be converted into OTB by catalytic dechlorination with palladium-charcoal and ammonium formate (Bredenkamp *et al.* 1989). Steyn and Holzapfel (1967) identified the methyl and ethyl esters of OTA and OTB in a culture of *A. ochraceus* on both sterilised cornmeal and liquid media. The toxicity of the esters of OTA is similar to that of OTA, whereas the OTB derivatives are, as to be expected, non-toxic.

OTA is hydrolysed to the non-toxic OTα (7-carboxy-5-chloro-3,4-dihydro-8-hydroxy-3-methylisocoumarin) in various organs in rats mostly the caecum, duodenum, ileum and the pancreas, whereas the activity in the liver and kidneys is very low or non-existent in rat hepatocytes (Suzuki *et al.*, 1977; Hansen *et al.*, 1982; Størmer *et al.*, 1983). OTA is

chemically hydrolysed by 6 N HCl and more readily by treatment with  $\alpha$ -chymotrypsin or carboxypeptidase A, yielding L- $\beta$ -phenylalanine and the optically active lactonic acid (OT $\alpha$ ). OT $\beta$  the des-chloro analogue of OT $\alpha$ , is the hydrolysis product of OTB, and was detected in culture extracts. The less toxic ochratoxin D, 4-hydroxyochratoxin A was isolated by Hutchison *et al.* (1971) from *P. viridicatum*. The (4*R*)-OH-OTA epimer is the major of the two epimers formed from OTA in human and rat liver microsomal systems under the influence of cytochrome P-450's (Størmer *et al.*, 1981, 1983) while the (4*S*)-OH-OTA epimer is more prevalent in pig liver microsomes (Moroi *et al.*, 1985). Oster *et al.* (1991) characterized four cytochrome P-450 fractions in pig liver microsomes; the two predominant forms A<sub>2</sub> and A<sub>3</sub>, both with a molecular weight of 54 kDa, and the minor form B<sub>a</sub> play an important role in the oxidation of OTA. These two epimers were also found in rat and rabbit liver (Størmer *et al.*, 1981) and rat kidney (Stein *et al.*, 1985). The 10-OH metabolite of OTA was formed from OTA with rabbit liver microsomal system (Størmer *et al.*, 1983). Hadidane *et al.*, (1992) discovered three natural analogues of OTA with the Phe-group replaced with a serine, proline and hydroxyproline group, while Xiao *et al.*, (1996b) reported the isolation of OT $\alpha$ , OT $\beta$ , (4*R*)-OH OTA, (4*R*)-OH OTB and 10-OH OTA from a culture of *A. ochraceus*.

Five analogs of OTA including the ethylamide of OTA (OE-OTA), the D-phenylalanine form of OTA (d-OTA), the decarboxylated OTA, (DC-OTA), the O-methyl ether of OTA (OM-OTA) and the methyl ester of OT $\alpha$  (M-OT $\alpha$ ) were synthesised using OTA or ochratoxin  $\alpha$  by Xiao *et al.*, (1995b). The toxicities of these analogues to HeLa cells are shown in **Table 7**. Xiao and coworkers activated OTA to the N-hydroxysuccinimide ester (OTA-NHS) and OT $\alpha$  to acyl chloride (OT $\alpha$ -Cl). They then used nucleophilic substitution reactions with primary amines, amino acids and alcohols to form corresponding amides and esters. OM-OTA was synthesised by the base-hydrolysis of O-methylochratoxin methyl ester. An open lactone form of OTA which is much less toxic than OTA is produced at high pH and is relatively stable at physiological pH (Xiao *et al.*, 1996a). Steyn and Payne (1999) synthesised the bromoanalogue of OTA by the treatment of OTB with pyridiniumhydrobromide perbromide. Steyn *et al.*, (1975) prepared 13 new analogues of OTA by substituting L-Phe for L-amino acids Trp, Ala, Tyr, Cys, Pro(4-OH), Glu, Met, Val, Pro, Ser, Asp, Thr and Leu. The typical lesions in cell culture which were associated with OTA toxicity were caused to various extents by all the compounds. In the group of compounds with a higher toxicity rating, four contained an aromatic ring.



**Table 7:** The toxicity of OTA and its analogs to HeLa cells

Analog	HeLa cell LC <sub>50</sub> (mM)
OTA	0.005
OTC	0.009
OTB	0.054
d-OTA	0.163
OT $\alpha$	0.56
OM-OTA	0.83
DC-OTA	7.6
OE-OTA	10.1

Source: Xiao *et al.*, (1995b).

## Production of OTA

OTA is produced by a number of both *Aspergillus* and *Penicillium* species, as shown in Table 8.

**Table 8:** Reported OTA-producing species

<i>Aspergillus</i>	<i>Penicillium</i>
<i>A. ochraceus</i> ( <i>A. alutaceus</i> )	<i>P. viridicatum</i> ( <i>P. verrucosum</i> )
<i>A. melleus</i> ( <i>A. quercinus</i> )	
<i>A. alliaceus</i>	
<i>A. ostianus</i>	
<i>A. sclerotiorum</i>	
<i>A. albertensis</i>	
<i>A. wentii</i>	
<i>A. auricomus</i>	
<i>A. niger</i> var. <i>niger</i>	
<i>A. awamori</i>	
<i>A. carbonarius</i>	
<i>A. foetidus</i>	
<i>A. sulphureus</i> ( <i>A. fresenii</i> )	

Source: Marquardt and Frohlich (1992); Abarca *et al.* (1994); Varga *et al.* (1996); Bragulat *et al.* (1998).

A number of other *Aspergillus* and *Penicillium* species were incorrectly reported to be producers of OTA due to the difficulty associated with the correct identification of the fungi (Pitt, 1987; Samson and Frisvad, 1991).

OTA occurs extensively in many plant and animal products; the contamination is typically associated with grain stored in the temperate climate of Europe and North America. There are substantial annual variations in the OTA content of grains because the production is

determined mainly by the temperature and water activity ( $a_w$ ) of the substrate and the type of substrate, presence of competitive microflora, strains of fungi and the quality of the seed (Marquardt and Frohlich, 1992). The minimum  $a_w$  conditions for OTA production by *e.g.* *A. ochraceus* are 0.83-0.87 and the minimum temperature is 12 °C; the optimal temperature for toxin production is 28 °C and the optimal time depends on the substrate, ranging from 7 to 14 days.

## Isolation and Purification

Mouldered substrates are extracted with hot chloroform, ethyl acetate, chloroform-methanol, acidified chloroform or by hexane followed by chloroform-methanol. The preliminary clean-up step is to transfer acidic components of the extract, including OTA and OTB, into a sodium bicarbonate solution, followed by acidification, extraction and column chromatography. The ochratoxins can be separated by chromatography *e.g.* ion-exchange chromatography, partition chromatography on formamide-impregnated cellulose powder, column chromatography on silica gel impregnated with oxalic acid, Sephadex LH20, Sephadex G-25, Florosil, Sephadex chromatography followed by silica gel chromatography, by preparative liquid chromatography or by preparative thin layer chromatography. For final purification OTA is crystallized from benzene, toluene or chloroform (Steyn, 1984). OTA crystallises from toluene and chloroform without solvent of crystallization.

## Analysis of OTA

Various methods for the analysis of OTA are based on HPLC, thin layer chromatography (TLC) and ELISA techniques.

TLC analysis is a relative inexpensive way of screening for OTA and was frequently used during the early years of mycotoxin research; it still has many applications. On TLC, OTA displays an intense blue-green fluorescence under long wavelength UV light. It is necessary to use acid modifiers (*e.g.* acetic acid) in TLC mobile phases (*e.g.* chloroform-methanol or toluene) to prevent streaking of OTA on silica gel. Problems with the rapid fading of the fluorescence intensity on the plate can be overcome by exposure of the plate to ammonia vapour, which converts the OTA to its ammonium salt, which displays a more intense blue fluorescence under UV illumination. The limits of detection for OTA on TLC are in the  $\mu\text{g.kg}^{-1}$  range. Paulsch *et al.* (1982) developed a two-dimensional TLC technique using an

acidic and alkaline developing solvent. This method in combination with exposure of the TLC plate to methanol-ammonia, leads to its effective detection. Paulsch *et al.* (1982) used a simple confirmatory test for OTA based on the formation of OTA methyl ester on the TLC plate. There is an increase in the use of reverse phase TLC (RPTLC) and high performance TLC (HPTLC) in mycotoxin analysis. The latter has a much better efficiency of separation and uses less solvent than conventional TLC plates. Frohlich *et al.*, (1988) developed a RPTLC method for sample preparation in which the OTA is extracted from the spot for quantitation by direct spectrofluorimetric testing or by subsequent HPLC analysis. This method has high levels of recovery (94%) and requires smaller amounts of solvents than the standard packed column methods.

HPLC separations are more efficient than those obtained by TLC and have been widely applied to the determination of OTA in various matrices (See **Table 9** for a few examples). Sample clean-up consists usually of extraction with organic solvents together with an acid to suppress the ionization of OTA, followed by further clean-up steps like silica gel, cyano or reversed phase-solid phase extraction (RP-SPE) cartridges, immunoaffinity columns, and liquid-liquid extractions or preparative HPLC techniques. RP columns *e.g.* octadecyl silane (ODS, C18) are usually used in quantitative HPLC separations with acidic aqueous acetonitrile or methanol as mobile phase (Van Egmond, 1991). Detection limits in the  $\mu\text{g.kg}^{-1}$  range or lower can be obtained by using an HPLC equipped with a fluorescence detector with excitation and emission wavelengths at 330 and 460 nm, respectively (Cohen and Lapointe, 1986).

Improved HPLC detectors such as photodiode-array detectors and improved computer search capabilities have made it possible to monitor the whole spectrum of compounds after separation, permitting further identification (Chu, 1992, Paterson and Kemmelmeier, 1990).

Jiao *et al.* (1991) developed a method for the identification of OTA in food samples by chemical derivatization of OTA to *O*-methyl-OTA ester and GC-MS using negative ion chemical ionization. Thermospray MS has been described by Rajakyla *et al.* (1987) as an expensive alternative to fluorescence detection.

Radioimmunoassay (RIA) is an analytical method that uses radioactivity for quantitative determination of compounds. It usually involves the incubation of a specific antibody with a solution of unknown sample or known standard at a constant amount of labelled toxin followed by the separation of the free and bound toxin and the determination of the

radioactivity in the fractions (Chu, 1992). Commercial kits are available for the determination of OTA in various feeds and foodstuffs with RIA.

**Table 9:** Methods for the determination of OTA in different matrices.

Substrate	Cleanup	Recoveries, detection limits	
Faba beans and wheat	SPE	70%, 0.7 $\mu\text{g.kg}^{-1}$	El-Banna and Scott (1984)
Wheat and barley	solvent partition	40 $\mu\text{g.kg}^{-1}$	Lepom (1986)
Animal feed, grain	2-stage SPE	90%, 5 $\mu\text{g.kg}^{-1}$	Cohen and Lapointe (1986)
Wheat, barley, oats and mixed feed	1. $\text{CHCl}_3$ extraction 2. SPE	77-96%, 0.1-0.3 $\mu\text{g.kg}^{-1}$	Langseth <i>et al.</i> (1989)
Human blood, serum, milk and some foodstuffs	immunoaffinity column	85%, 5-10 $\text{pg.g}^{-1}$	Zimmerli and Dick (1995)
Human urine	extraction, column chromatography	60-75%, 5 $\text{ng.l}^{-1}$	Castegnaro <i>et al.</i> (1990)
Beer	1. SPE ( $\text{C}_{18}$ silica) 2. immunoaffinity column	82-100%, 0,05-0,1 $\text{ng.ml}^{-1}$	Scott and Kanhere (1995)

ELISAs involve the use of antibodies generated against conjugates. These conjugates are made by linking OTA to protein (enzymes like horseradish peroxidase) through its carboxylic acid function. Direct ELISA involves the use of a OTA-enzyme conjugate, while indirect ELISA uses a protein-OTA conjugate and a secondary antibody to which an enzyme has been conjugated (Morgan *et al.*, 1986). Commercial kits like RIA are available, but the ELISA technique can be used to measure samples as low as 2.5 pg, which makes ELISA 10 to 100 times more sensitive than RIA (Chu, 1992). Although a lot of work is being done to improve the selectivity of the antisera used in ELISAs (Xiao *et al.*, 1995a), the possibility of cross-reactions cannot be fully ruled out and positive findings obtained by immunoassays need to be confirmed by other techniques.

Antibody technology is used for the clean-up of cereal and animal sample extracts by utilising immunoaffinity columns for the selective isolation of OTA (Sharman *et al.*, 1992). Nakajima *et al.*, (1990), prepared monoclonal antibody affinity columns for the determination of OTA in coffee by binding antibodies specific for OTA to Sepharose 4B. Immunoaffinity columns are commercially available and are used routinely in laboratories for sample clean-up followed by quantitation by HPLC or direct spectrofluorimetric measurement.

## Regulations for OTA

OTA contamination is widespread in cereals, coffee, pulses, feedstuffs and other plant products. Raw agricultural products, contaminated with OTA and used as feed, can also contaminate meat and meat products of non-ruminant animals such as poultry and pigs (Van Egmond and Speijers, 1994). This problem does not occur in adult ruminants because OTA is hydrolysed by protozoan and bacterial enzymes in the fore-stomachs of these animals (IPCS, 1990). The detection of OTA in human milk indicates the carry-over of OTA from contaminated food by lactating women. Data on the occurrence of OTA in food and feed are relatively abundant in European countries but scarce for other continents. The levels of ochratoxin A contamination were found at the highest incidences in cereals (corn from 10-500  $\mu\text{g.kg}^{-1}$ , wheat from 5-135  $\mu\text{g.kg}^{-1}$  and barley from 10-500  $\mu\text{g.kg}^{-1}$ ) and for foodstuffs from animal products (kidneys from pigs varying from 2-100  $\mu\text{g.kg}^{-1}$ ) (Van Egmond and Speijers, 1994). The occurrence of OTA is related to the climate and especially the harvest and post-harvest storage conditions (Pohland *et al.*, 1992). There are 77 countries that have mycotoxin regulations and only eight have specific regulations for OTA in one or more commodities (FAO, 1996). The following factors must be taken into consideration before the limits for mycotoxins can be chosen: the availability of toxicological and survey analytical data, the availability of reliable analytical methods, data on the availability on the occurrence of mycotoxins in various commodities, intercountry trade as well as the existence of sufficient food supply (FAO, 1996). The current (proposed) limits are indicated in **Table 10**. The tolerance levels for OTA have been suggested at 1  $\mu\text{g.kg}^{-1}$  for infant foods and at 5  $\mu\text{g.kg}^{-1}$  for cereals for the EU (Verardi and Rosner, 1995).

The homogeneity of OTA in products is very important, if care is not taken to ensure representative sampling, incorrect estimations of toxin concentrations will be made (FAO, 1996). Analytical quality assurance is essential to guarantee the accuracy of results and the use of reference materials (RMs) plays an important role in checking the performance of methods and proficiencies of laboratories (Van Egmond, 1996). The European Commission's Standards, Measurements and Testing Program has initiated the development of certified RMs for OTA and other mycotoxins (Wood *et al.*, 1995).

**Table 10:** Limits for Ochratoxin A in the different commodities

Commodity	Limits
Children and infant foods	0.5 - 5 $\mu\text{g.kg}^{-1}$
Foods	2 - 50 $\mu\text{g.kg}^{-1}$
Animal feeds	5 - 300 $\mu\text{g.kg}^{-1}$

Source: FAO (1996)

## Ochratoxicosis

OTA is nephrotoxic to all animal species tested so far, and induces experimental liver and kidney tumours (see **Table 11**).

**Table 11:** Acute oral toxicities of the ochratoxins in different species (mg OTA per kg body weight)

Species	OTA LD <sub>50</sub>	OTC LD <sub>50</sub>	OTB LD <sub>50</sub>	OT $\alpha$ LD <sub>50</sub>
Pigs	1.0-6.0			
Chickens	3.3	4,8	41.4	22
Dogs	0.2			
Neonatal rats	3.9			
Mature rats	20-30			
Mice	46-58			

Source: Chu (1974); Krogh (1987); Kuiper-Goodman and Scott (1989).

The kidneys are the organs most susceptible to OTA, which can cause both acute and chronic kidney diseases. The renal lesions associated with the diseases include degeneration of the proximal tubules, interstitial fibrosis in the renal cortex, hyalinization of the glomeruli and atrophy in the tubular epithelium (Krogh *et al.* 1974, 1977, Krogh 1987). Pigs fed OTA showed reduced feed intake, loss of body weight, increased water consumption followed by polyurea diarrhoea, polydipsia and dehydration (Szczzech *et al.*, 1973). Residues of OTA are greatest in the kidney, and in declining order in lean meat, liver and fat (Madsen *et al.*, 1982). Krogh *et al.* (1988) reported nephropathy in pigs fed diets containing 0,2-4  $\text{mg.kg}^{-1}$  OTA after four months of exposure; all lesions were confined to the kidney. Gross pathological examination of dogs administered 0.2 to 3  $\text{mg}$  of  $\text{OTA.kg}^{-1}$  BW alone or in two dose combinations for 14 days indicated moderate to severe mucohemorrhagic enteritis of the caecum, colon, and rectum and enlargement of the lymph nodes, which were oedematous, hyperaemic, and focally necrotic. Histopathological examination indicated that renal damage is the main feature of this toxicosis (Szczzech *et al.*, 1973). OTA inhibits cell division in these

kidney cells and causes apoptotic type morphological lesions in these cells. The nuclear lesions seen in apoptosis are associated with enhanced endonuclease activity, which are responsible for DNA cleavage. OTA causes apoptosis-associated DNA degradation in human lymphocytes (Seegers *et al.*, 1994, and references cited). OTA also causes decreased natural killer cell activity in mice (Luster *et al.*, 1987) and the inhibition of cell division in hematopoietic stem cells (Boorman *et al.*, 1984) and lymphocytes (Creppy *et al.*, 1983b).

Krogh proposed in 1974 that the endemic human interstitial nephropathy diseases in the rural areas of Bulgaria, Romania and Yugoslavia might be related to a high OTA exposure. Balkan endemic nephropathy (BEN) was first identified in the 1950s and is an invariably fatal chronic kidney disease, characterized by contracted kidneys and features changes exclusively in the renal cortex of the kidney. OTA has been found more frequently in food samples and in the serum of people taken from villages with BEN, than in areas where the disease is unknown (Krogh *et al.*, 1977; Pavlovic *et al.*, 1979, Petkova-Bocharova and Castegnaro 1985; and Petkova-Bocharova *et al.*, 1988). The blood of 95% of Tunisian people suffering from urinary system tumours (UST) are OTA positive, with blood concentrations higher than 90 ng.ml<sup>-1</sup> in several cases (Maaroufi *et al.*, 1999). OTA is also the cause of a nephropathy (Danish porcine nephropathy) affecting many pigs fed mouldy cereal feeds in Scandinavian countries. In Denmark carcasses are condemned if residue levels of OTA in the kidney exceed 25 ng.g<sup>-1</sup>.

## Genotoxicity

OTA has been considered from many years as a non-mutagenic carcinogen, since it has been generally negative in a number of gene mutation tests, based on both microorganisms and mammalian cells, and both with and without metabolic activation (Würgler *et al.*, 1991; Bendele *et al.*, 1995; Sakai *et al.*, 1992). However, OTA induced mutations in the modified Ames assay (Dirheimer, 1996), sister chromatid exchange in human peripheral lymphocytes *in vitro* and SOS DNA repair in *E. coli* (cited by Neal 1995; Hennig *et al.*, 1991). OTA-induced single-strand breaks were observed in primary rat hepatocytes (De Groene *et al.*, 1996b) and *in vivo* in rats and mice (Creppy *et al.*, 1985; Kane *et al.*, 1986). Furthermore, chromosomal aberrations could be detected after OTA exposure (Manalowa *et al.*, 1990) and DNA-adducts could be observed *in vivo* in the liver, kidney and spleen of rodents and humans (Pfohl-Leszkowicz *et al.*, 1993a, 1993b) and in cytochrome P450-expressing BEAS-2B cells (Grosse *et al.*, 1994, 1995) and monkey kidney cells (Grosse *et al.*, 1995). These adducts were

different for the different organs suggesting different routes of metabolic activation. De Groene *et al.* (1996a) recently demonstrated that OTA mutagenicity requires a cytochrome P450-dependent activation step by using the *lacZ'* gene as reporter gene for mutations in cell lines expressing selected human cytochrome P450 forms.

### **Teratogenicity and Immunotoxicity**

OTA is teratogenic to rats, mice, hamsters and chickens (Brown *et al.*, 1976; Shreeve *et al.*, 1977; Fukui *et al.*, 1987). It causes a marked increase in the number of dead and resorbed fetuses and a decrease in foetal body weight when administered to pregnant rats. Multiple gross, visceral and skeletal anomalies in pups are related to the treatment and dose of OTA administered to rats (Brown *et al.*, 1976). The subchronic exposure of Balb/c mice to OTA suppressed the antibody-production plaque forming cells, decreased thymocyte cell counts and the proportion of mature thymic lymphocyte ( $CD^{4+}$  or  $CD^{8+}$ ) cells (Creppy *et al.*, 1982, 1983b). The mitogenic responsiveness of thymocytes and splenocytes to concanavalin A (Con A) is also significantly decreased by OTA exposure. However, interleukin-2 production of Con A-stimulated lymphocytes, natural killer cell activity and humoral antibody titres to a viral antigen are not affected by OTA (Thuvander *et al.*, 1995). Thuvander and coworkers also reported immunosuppression in Balb/c mice (Thuvander *et al.*, 1997) and Sprague-Dawley rats (Thuvander *et al.*, 1996a) after prenatal exposure to OTA. OTA exposure resulted in a decrease in proliferation and antibody, thereby indicating that subchronic, oral exposure to OTA affects certain immune functions in mice but does not suppress immune functions in the offspring (Thuvander *et al.*, 1996b).

### **Pharmacokinetics of OTA**

About 40-65% of OTA orally administered to rats is absorbed in the small intestine, primarily in the proximal part of the jejunum (Kumagai, 1988). OTA has a high binding affinity for plasma constituents and binds to serum-albumin and to an as yet unidentified macromolecule(s) as soon as it reaches the circulation system. This characteristic of OTA retards its elimination by limiting the transfer of OTA from the bloodstream to the hepatic and renal cells and consequently contributes to the prolonged half-life of the toxin (Chu, 1971, Stojkovic *et al.*, 1984; Kumagai, 1985; Hagelberg *et al.*, 1989). OTA also binds more specifically than plasma albumins to a smaller molecular fraction in blood (Stojkovic *et al.*,



1984). There may be a relation between the predominant nephrotoxic affect of OTA in mammals and the binding of OTA to these molecules because such molecules can easily pass through the normal glomular membrane enabling the accumulation of OTA into the kidney (Marquardt and Frohlich, 1992). The pharmacokinetics of OTA and its metabolites in rats were recently reported by Li *et al.*, (1997). The plasma half-life of OTA depends on the degree of absorption and the degree of binding to serum-albumin and a number of other factors. OTA has a very high affinity for this unknown macromolecule (see above) in human serum and may thus have a long plasma half-life. The toxic activity of OTA in monkeys, with an elimination half-life of 35 days and slow absorption from the gastrointestinal tract, seems to be a good model for humans (Hagelberg *et al.*, 1989). Stein *et al.* (1985) reported an efficient reabsorption of OTA by the renal tubules of the kidney that also facilitates reabsorption of OTA into the plasma. This process in the renal proximal tubules which may be responsible for damages to the kidneys of various animal species (Albassam *et al.*, 1987; Szczech *et al.*, 1973; Elling, 1979) was reported by Jung and Endou (1989) to affect only the middle and terminal portions of the nephrons.

Rats excrete OT $\alpha$ , OTA and (4R)-OH-OTA mainly in the bile and urine after OTA had been administered intraperitoneally or *per os* (Støren *et al.*, 1982a, 1982b; Kane *et al.*, 1986; Xiao *et al.*, 1996b). Approximately 1-2% of the OTA was recovered as 4-OH-OTA and a total of 25-40% of the administered OTA was recovered as OT $\alpha$  and 6% as OTA. The relative quantity of OTA eliminated *via* the kidneys and the liver depends partly on the animal species, route of administration and dose, the enterohepatic recirculation and the binding of the toxins to serum macromolecules (Roth *et al.*, 1988). Data on elimination half-lives and distribution half-lives of other species are given in **Table 12**.

There are contradictory reports on the sub-chronic and chronic toxicity of OTA: some of the early reports regard OTA as the toxic agent, since its known metabolites are equally or less toxic than OTA, whereas other researchers consider the toxic effects as due to one of its metabolites, since the simultaneous feeding of phenobarbital increases the incidence of liver tumours seen after the feeding of OTA alone (Suzuki *et al.*, 1986) and other more recent findings described below. Phenobarbital is known to induce the activity of various constitutive cytochrome P450 forms in the liver (Soucek and Gut, 1992). Fink-Gremmels *et al.* (1995) provided evidence for a number of unknown metabolites produced by metabolically competent hepatocytes, whereas Malaveille *et al.* (1994) discussed the formation of an OTA phenoxide radical and a thiol derived toxic metabolite as reactive metabolites of OTA.

Distinct possibilities are that OTA might be metabolised yielding both detoxified products and other metabolites responsible for the mutagenic effect described above or OTA may induce DNA base modifications such as alkylation, m5dC and 8-hydroxyguanine due to OTA's induction of oxydative stress (El-Ghissassi *et al.*, 1995). 8-Hydroxyguanine is known to induce G to T and A to C mutations where m5dC could lead to impairment of the regulation (Cheng *et al.*, 1992).

**Table 12:** Pharmacokinetic data for OTA and some of its derivatives.

Species		OTA	OP-OTA	OT $\alpha$	OTB	OTC	OTA-OH
Rats (iv) <sup>a</sup>	t <sup>1</sup> / <sub>2<math>\alpha</math></sub> (min)	160±17	163±5	31±5	14±4	6±1.2	19±4.7
	t <sup>1</sup> / <sub>2<math>\alpha</math></sub> (min)	126					
Rabbits <sup>b</sup>	t <sup>1</sup> / <sub>2<math>\alpha</math></sub> (min)	114					
Chickens <sup>c</sup>	t <sup>1</sup> / <sub>2<math>\alpha</math></sub> (min)	30					
Cattle <sup>e</sup>	t <sup>1</sup> / <sub>2<math>\alpha</math></sub> (min)	108					
Rats (iv) <sup>a</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	103±16	50.5±2.8	9.6±2.3	4.2±1.2	0.6±0.2	6±0.9
Rats (iv) <sup>f</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	120					
Rats (p) <sup>f</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	170					
Rats (o/iv) <sup>b</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	55					
Rats (o/iv) <sup>g</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	56					
Pig <sup>c,f,h</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	72-150					
Pre-ruminant calf <sup>i</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	77					
Mouse <sup>f</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	24-48					
Rabbits <sup>c</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	8.2					
Chickens <sup>c</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	4.1					
Monkeys <sup>f</sup>	t <sup>1</sup> / <sub>2<math>\beta</math></sub> (h)	840					

t<sup>1</sup>/<sub>2 $\alpha$</sub>  = distribution half-life; t<sup>1</sup>/<sub>2 $\beta$</sub>  = elimination half-life; p = peripherally, iv = intravenous, o = orally.

Sources: Galtier *et al.*, 1979<sup>b</sup>; Galtier *et al.*, 1981<sup>c</sup>; Mortensen *et al.*, 1983<sup>h</sup>; Ballinger *et al.*, 1986<sup>g</sup>; Fukui *et al.*, 1987<sup>i</sup>; Sreemannarayana *et al.*, 1988<sup>e</sup>; Hagelberg *et al.*, 1989<sup>f</sup>; Marquardt *et al.*, 1996<sup>a</sup>;

## Prevention of Ochratoxicoses

Superoxide dismutase (SOD) and catalase are enzymes which prevent most OTA-induced nephrotoxic effects and might be used for the prevention of such renal lesions (Baudrimont *et al.*, 1994). Vitamin C significantly reduces the effects of OTA on albino Swiss mice (Bose and Sinha, 1994). Other compounds that are efficient in preventing ochratoxicosis *in vivo* are radical scavengers, vitamins, prostaglandin synthesis inhibitors (*e.g.* indomethacin and aspirin), pH modifiers and absorbant resins such as cholestyramine etc. (Madhyastha *et al.*,

1992; Baudrimont *et al.*, 1995; Creppy *et al.*, 1996). Compounds that have a high binding affinity for plasma proteins, such as piroxicam are also promising as potential antidotes for OTA (Creppy *et al.*, 1995). Aspartame, structurally related to OTA prevents OTA binding to plasma proteins and is the best candidate for preventing the OTA-induced subchronic effects (Creppy *et al.*, 1995).

A practical method to prevent ochratoxicosis is to reduce the levels of OTA contamination in foods and feedstuffs by certain cooking processes (Milanez and Leitao, 1996).

## **Mechanisms of Action of OTA**

There appears to be a number of direct and several indirect effects of OTA. The best known effects of OTA are its effect on enzymes involved in the Phenylalanine (Phe) metabolism, its effect on lipid peroxidation and its effect on mitochondrial respiration.

### ***Inhibition of Phe-tRNA Formation***

OTA inhibits protein synthesis by competition with Phe in the Phe-tRNA aminoacylation reaction catalysed by phenylalanyl-tRNA synthetase (Creppy *et al.*, 1984). This was shown by Bunge *et al.* (1978), Creppy *et al.* (1979a,b) and Konrad and Rösenthaller (1977) in both bacterial and eukaryotic systems *in vitro*. This inhibition can be reversed by the administration of Phe in hepatoma cells (Creppy *et al.*, 1979a) and *in vivo* in mice (Creppy *et al.*, 1984). In addition to the inhibition of protein synthesis, DNA and RNA synthesis are consequently also inhibited (Creppy *et al.*, 1986). Phe also provides partial prenatal protection from the teratogenic effects of OTA (Mayura *et al.*, 1984) and prevents the immunosuppressive effects of OTA in Balb/c mice (Creppy *et al.* 1983a; 1983b; Haubeck *et al.* 1981). Creppy *et al.*, (1983a) found similar inhibitory effects in the respective tRNA synthetase enzymes, when the Phe was replaced by other amino acids. Roth *et al.* (1993) reported that the inhibitory effect on Phe-tRNA synthetase alone cannot explain the inhibitory effects of OTA on growth in bacteria and that other mechanisms involving OTA-activated substances must be involved. OTA also inhibits the activity of phosphoenolpyruvate carboxykinase (PEPCK) and  $\gamma$ -glutamyl transpeptidase and abolishes the cAMP-mediated increase in the concentration of PEPCK mRNA (Thekkumkara and Patel, 1989). Removal of the Phe moiety from OTA prevents the *in vivo* inhibition of PEPCK activity and protein synthesis (Meisner and Meisner

1981). OTA also inhibits other reactions in which Phe is involved like those catalysed by Phe-hydroxylase (Creppy *et al.*, 1990).

### ***Lipid Peroxidation***

Ochratoxin A disrupts hepatic microsomal calcium homeostasis by impairment of the endoplasmic reticulum membrane, probably via lipid peroxidation (Omar *et al.*, 1991). OTA greatly enhances the rate of NADPH- or ascorbate dependent lipid peroxidation both *in vivo* (rats) and *in vitro* (liver or kidney microsomes) as measured by malondialdehyde formation. The efficiency for lipid peroxidation enhancement is related to the presence of the phenolic hydroxyl group of the different ochratoxins and correlates well with their known toxicities (Rahimtula *et al.*, 1988, 1989). OTA stimulates lipid peroxidation primarily by chelating ferric ions ( $\text{Fe}^{3+}$ ) and facilitating their reduction to ferrous ions ( $\text{Fe}^{2+}$ ); the subsequent reoxidation is accompanied by  $\text{O}_2$  consumption (Omar *et al.*, 1990). The  $\text{Fe}^{3+}$ -OTA complex produces the extremely damaging hydroxyl radical in the presence of the NADPH-cytochrome-P-450 reductase system and NADPH (Hasinoff *et al.*, 1990). The OTA- $\text{Fe}^{2+}$  complex provides thus the active species which initiates lipid peroxidation in the presence of oxygen. Once this process is initiated, it can be easily propagated in the cellular environment where polyunsaturated fatty acids and oxygen are present. The oxidation of lipids by oxygen continues in a chain of radical reactions. As a consequence of this biochemical process, a wide range of degradation compounds are formed, which are chemically very reactive and produce structural injuries (Baudrimont *et al.*, 1997). There may be a connection between nephropathy caused by OTA, citrinin, iron and lipid peroxidation. Iron is presented to the tubular lumen in proteinuric states because of the glomerular leak of transferrin. Iron would be expected to be dissociated from transferrin in the tubular fluid because of its low pH and bicarbonate content, and exists in a form that could catalyse hydroxyl radical formation. It has been suggested that if iron is available in a form capable of catalysing OH-radical formation, it could result in lipid peroxidation of tubular cell membranes (Størmer *et al.*, 1996). Lipid peroxidation caused by the free reactive oxygen species induced by OTA can be prevented in Vero cells by adding superoxide dismutase and catalase, piroxicam or aspartame to the culture medium prior to OTA addition to the medium (Baudrimont *et al.*, 1997). Pfohl-Leszkowicz *et al.*, (1993a,b) linked this ability of OTA to enhance lipid peroxidation, to the genotoxicity expressed by DNA adduct formation. It is also possible that the OTA-induced DNA single

strand breaks in mice and rats are produced by reactive oxygen species (Creppy *et al.*, 1985; Kane *et al.*, 1986).

### ***Inhibition of Mitochondrial ATP production***

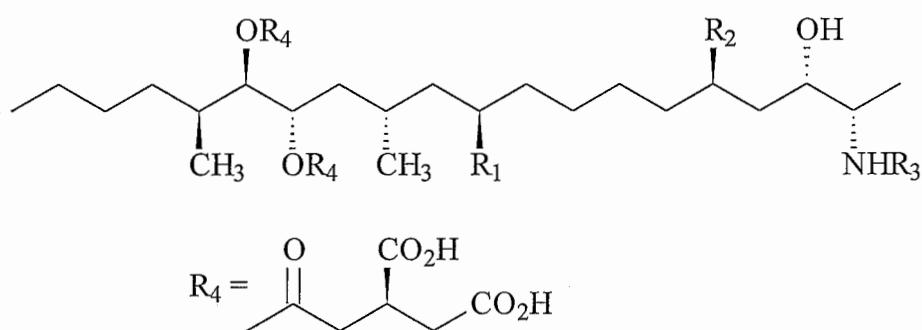
OTA inhibits mitochondrial state 3 and 4 respiration in isolated rat liver mitochondria (Moore and Truelove, 1970) by acting as a competitive inhibitor of mitochondrial transport carrier proteins located in the inner mitochondrial membrane (Meisner and Chan, 1974; Wei *et al.*, 1985; Meisner, 1976). OTA also alters the mitochondrial morphology after *in vivo* administration to rats (Suzuki *et al.*, 1975; Brown *et al.*, 1986). The mitochondrial uptake of OTA is an energy consuming process that results in the depletion of intramitochondrial ATP and the observed ATP decrease was most pronounced in the middle (S2) and the terminal (S3) segment of the proximal tubule (Jung and Endou, 1989). Aleo *et al.*, (1991) suggested that mitochondrial dysfunction is an early event during the development of OTA toxicity and that OTA toxicity to rat proximal tubules in suspension was not related to iron-mediated lipid peroxidation as measured by malondialdehyde production. The importance of the mitochondrial mechanism is not clear because OT $\alpha$ , which is non-toxic, was also able to inhibit mitochondrial respiration more effectively than OTA in rat liver mitochondria (Moore and Truelove, 1970).

## **FUMONISINS**

The fumonisins are a group of mycotoxins, consisting of a 2-amino-12,16-dimethylpolyhydroxyeicosane backbone esterified with propane-1,2,3-tricarboxylic acid side chains on C<sub>14</sub> and C<sub>15</sub> (see **Figure 6**). These toxins were first discovered by a South African group led by Marasas (Bezuidenhout *et al.*, 1988; Gelderblom *et al.*, 1988) after an investigation into the cause of equine leukoencephalomalacia (LEM) or better known as 'hole in the head disease' in horses fed feeds contaminated with the fungus *Fusarium moniliforme*. LEM is a well known disease in many countries such as Mexico, the United States of America, Egypt, and South Africa and causes the liquefactive necrosis of the white matter of the brain of horses and donkeys. Corn contaminated with *F. moniliforme* has also been associated with human oesophageal cancer in the Transkei area of South Africa (Rheeder *et al.*, 1992) and China (Yang, 1980; Chu and Li, 1994). Fumonisins have also been reported to cause pulmonary edema syndrome in pigs (Harrison *et al.*, 1990) and a nondescribed poultry

disease (ill thrift). Fumonisin A<sub>1</sub> (FA<sub>1</sub>) and fumonisin A<sub>2</sub> (FA<sub>2</sub>), the *N*-acetyl derivatives of fumonisin B<sub>1</sub> (FB<sub>1</sub>) and fumonisin B<sub>2</sub> (FB<sub>2</sub>) respectively, are produced in low yields in cultures of *F. moniliforme* and have the lowest toxicities. These two structural analogues and FB<sub>4</sub> (**Figure 6**) do not occur under natural conditions. The C series of fumonisins (the chemical structures of the C series is identical to the B series except that the C-1 terminal methyl group is missing) was recently discovered in naturally contaminated corn by Seo and Lee (1999) in the presence of the B series, but at much lower concentrations. These compounds were previously found in an isolate of *Fusarium oxysporum* during a screening of fumonisins by the same group in 1996.

FB<sub>1</sub>, the most abundant fumonisin in culture and naturally occurring in corn (Rheeder *et al.*, 1995), has been shown to promote tumour formation in rats and to inhibit ceramide synthetase in neuronal cells, an important enzyme in sphingolipid biosynthesis (Wang *et al.*, 1991).



	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>
Fumonisin A <sub>1</sub>	OH	OH	CH <sub>3</sub> CO
Fumonisin A <sub>2</sub>	H	OH	CH <sub>3</sub> CO
Fumonisin B <sub>1</sub>	OH	OH	H
Fumonisin B <sub>2</sub>	H	OH	H
Fumonisin B <sub>3</sub>	OH	H	H
Fumonisin B <sub>4</sub>	H	H	H

**Figure 6:** Structures of the fumonisins.

### Chemical characteristics of the fumonisins

FB<sub>1</sub> is a stable compound that persists through most normal food processing procedures (see **Table 13** for the physical data of FB<sub>1</sub>).

**Table 13:** Physical and spectroscopic data of FB<sub>1</sub>.

Melting point:	103-105 °C
Optical rotation [ $\alpha$ ] <sub>D</sub> :	-28° (concentration 2 mg/ml)
Infrared (KBr):	3450, 2934, 1729 and 1632 cm <sup>-1</sup>

Fumonisin is soluble in most polar solvents including water and is not soluble in non-polar solvents. There is no known detoxification process for fumonisin-contaminated foods and feeds. Bezuidenhout *et al.* (1988) elucidated the structure of the fumonisins by employing NMR and mass spectroscopic techniques, while the absolute configuration (see **Figure 6**) was determined by contributions from a number of authors including Blackwell *et al.* (1994a), ApSimon *et al.*, (1994) and Shier *et al.*, (1995) using a combination of NMR and chiral GC methods. Laurent *et al.* (1990), prepared the ammonium salt of FB<sub>1</sub>, while the acetylated and methylated derivatives of FB<sub>1</sub> were prepared by Bezuidenhout *et al.*, (1988) and Laurent *et al.*, (1990) respectively. FB<sub>1</sub> and FB<sub>2</sub> can be hydrolysed by heating with hydrochloric acid or potassium hydroxide to yield the aminopentol of FB<sub>1</sub> and the aminotetraol of FB<sub>2</sub> (Gelderblom *et al.*, 1993).

## Production of the fumonisins

*F. moniliforme* is the most important producer of the fumonisins, however, a number of other *Fusarium* species are known to be producers of these toxins as shown in **Table 14**.

**Table 14:** Fungal producers of fumonisins

<i>Fusarium moniliforme</i>	<i>Fusarium napiforme</i>
<i>Fusarium dlamini</i>	<i>Fusarium proliferatum</i>
<i>Fusarium nygamai</i>	<i>Fusarium anthophilum</i>
<i>Fusarium subglutinans</i>	

Source: Thiel *et al.* (1991); Marassas (1995) and references cited.

Optimum conditions for the production of fumonisins are: *F. moniliforme* is grown on wet sterilized corn and incubated at 20°C for 11-13 weeks (Alberts *et al.*, 1990 and Le Bars *et al.*, 1992). The mouldy material is extracted with aqueous methanol, and the fumonisin-containing extract purified by liquid-liquid partition; followed by cleanup with XAD-2, silica gel and reverse phase chromatography (Cawood *et al.*, 1991; Vesonder *et al.*, 1990). Although corn cultures are still the best way of producing large quantities of unlabeled

fumonisin, liquid media are ideal for the production of unlabelled and  $^{14}\text{C}$  labelled fumonisins (Miller, 1994). Liquid media usually consists of a carbon source like glucose or sucrose, a phosphate buffer ( $\text{pH} \pm 4$ ) and salts like  $\text{MgSO}_4$ ,  $\text{CaCl}_2$ ,  $\text{NaCl}$ ,  $\text{NH}_4\text{Cl}$ ,  $\text{Na}_2\text{SO}_4$  and  $\text{MnSO}_4$ . Alberts *et al.* (1993), developed a technique for the production of  $[^{14}\text{C}]\text{FB}_1$  by *F. moniliforme* MRC 826 in 'patty' corn cultures by using *L*-[methyl- $^{14}\text{C}$ ]methionine as the precursor (Blackwell *et al.*, 1994b).

## Determination and occurrence of the fumonisins

A wide variety of extraction and clean-up methods exist for the determination of fumonisins. In general the feeds or foods are extracted with an organic solvent mixture, cleanup by liquid-liquid partition, solid phase extraction, column chromatography or immunoaffinity columns. Quantitation is done by TLC (Shelby *et al.*, 1994), HPLC (Thiel *et al.*, 1993), post-hydrolysis gas chromatography (Sydenham *et al.*, 1990), GC-MS (Plattner *et al.*, 1990), liquid secondary ion mass spectrometry (LCIMS), FAB/MS (Korfmacher *et al.*, 1991) or MS/MS (Plattner *et al.*, 1990). In many cases fumonisins are reacted with naphthalene-2,3-dicarboxyaldehyde - potassium cyanide; *o*-phthalaldehyde - mercapto-ethanol (Shephard *et al.*, 1990); FMOC (Holcomb *et al.*, 1993) or 4-fluoro-7-nitrobenzofurazan (Scott and Lawrence, 1992) to yield fluorescent derivatives which can easily be detected by HPLC systems equipped with fluorescent detectors. Several immunochemical methods for the determination of  $\text{FB}_1$  have also been developed including enzyme-linked immunosorbent assay (ELISA) with monoclonal antibodies (Azcona-Olivera *et al.*, 1992a); polyclonal antibodies (Azcona-Olivera *et al.*, 1992b, Usleber *et al.*, 1994) and anti-idiotypic/anti-anti-idiotypic antibodies (Chu *et al.*, 1995). Cross-reaction with  $\text{FB}_2$  and  $\text{FB}_3$  is very low and immunochemical methods can thus underestimate the total concentration of fumonisins in a commodity; no cross-reaction occurs with the hydrolysed fumonisins. Schneider *et al.* (1995) developed a competitive direct dipstick enzyme immunoassay (EIA) and an enzyme-linked immunofiltration assay (ELIFA) for the detection of  $\text{FB}_1$ . A nylon membrane was coated with anti- $\text{FB}_1$  antibodies and with anti-horseradish peroxidase (HRP) antibodies. An  $\text{FB}_1$ -HRP conjugate was used both as the labelled antigen for competitive assay of  $\text{FB}_1$  and for non-competitive binding to the anti-HRP antibodies (negative control). Immunoaffinity columns which use monoclonal antibodies are commercially available (FumoniTest, Vicam, Watertown, MA) and have detection limits at



ppm levels. Detection limits are reported to be 100 ppb for TLC, 50 ppb for LC-MS and 200 ppb for ELISA (Thiel *et al.*, 1996 and references cited).

Fumonisin contamination of corn and corn-based products has been reported on all continents.

**Table 15** contains suggested safety limits for fumonisins in animal feeds (Thiel *et al.*, 1996 and references cited).

**Table 15:** Suggested safety limits for fumonisins

Species	Limit
cattle and poultry	50 000 ppb
Pigs	10 000 ppb
Horses	5 000 ppb

Source: Thiel *et al.* (1996) and references cited.

**Table 16** contains some information regarding fumonisin contamination found in different countries. Corn (maize) is a staple food of many people living in Southern Africa; FB<sub>1</sub> contamination of corn may, therefore, pose a serious threat to human health in these regions.

**Table 16:** References to fumonisin contamination found in different countries.

Country	Commodity	Level	Author
Italy	Corn and corn based foods	<5310 ppb for FB <sub>1</sub> & <1480 ppb for FB <sub>2</sub>	Doko and Visconti (1994)
Argentina	Corn	combined fumonisin levels of 1585 - 9990 ppb	Sydenham <i>et al.</i> (1993a)
Kenya	Corn	<1 ppm FB <sub>1</sub> (mostly)	Kedera <i>et al.</i> (1999)
Brazil	Corn and corn based feeds	0,2 - 38,5 ppm FB <sub>1</sub> & 0,1 - 12,0 ppm FB <sub>2</sub>	Sydenham <i>et al.</i> (1994)
U.S.A.	Corn and feeds	1,3 - 27,0 ppm FB <sub>1</sub> & 0,1 - 12,6 ppm FB <sub>2</sub>	Thiel <i>et al.</i> (1991)
India	Corn		Chatterjee and Mukherjee (1994)
India	Corn kernels	100-4740	Shetty and Bhat (1997)
Switzerland	Corn-based products		Pittet <i>et al.</i> (1992)
China	Corn	18-155 ppm FB <sub>1</sub>	Chu and Li (1994)
New Zealand	Forage grass	1 and 9 ppm of FB <sub>1</sub>	Mirocha <i>et al.</i> (1992)

## Decontamination

The fine particulate matter (< 3 mm) in corn contains the highest levels of fumonisins. By removing the 'fines' from bulk shipments of corn the fumonisin contamination is reduced significantly (Sydenham *et al.*, 1994). Sydenham *et al.*, (1993b) developed a chemical method

for the reduction of fumonisin levels in corn by treating it with a slurry of 0.1 M calcium hydroxide for 24 hours at 25 °C. The process of milling and the ammoniation treatment of contaminated corn also reduce the levels of fumonisins in corn products (Norred *et al.*, 1991). All these different treatments suggest that the fumonisins are concentrated on the outer pericarp layer of corn kernels (Sydenham *et al.*, 1993b).

### **Biological Effects and Mechanism of Action of the Fumonisin**

FB<sub>1</sub> has been established to induce completely different toxic effects in different animal species [LEM in horses, hepatotoxic and hepatocarcinogenic to rats and pulmonary oedema syndrome in pigs (Marasas *et al.*, 1988; Harrison *et al.*, 1990)]. In rats the liver is the main target for toxicity, characterised by cirrhosis and cholangiofibrosis. FB<sub>1</sub> is also foetotoxic to rats, suppressing both growth and foetal bone development (Lebepe-Mazur *et al.*, 1995). FB<sub>1</sub>, FB<sub>2</sub> and FB<sub>3</sub> also affect the kidneys of rats after prolonged exposure, but are, according to Gelderblom *et al.* (1988, 1992), not very acutely toxic to rats. Toxic response to fumonisins by rat hepatoma cell line H4TG was visible within 48 h with an IC<sub>50</sub> value of 4 µg.ml<sup>-1</sup> for FB<sub>1</sub> and 2 µg.ml<sup>-1</sup> for FB<sub>2</sub> (Shier *et al.*, 1992). IC<sub>50</sub> values were not affected by the density of the cell cultures, indicating that fumonisins are not metabolically activated to express their toxic effects. This was confirmed by the absence of metabolites of fumonisins in the urine, bile and blood of rats fed FB<sub>1</sub> and FB<sub>2</sub> (Shephard *et al.*, 1993, 1995) and in primary rat hepatocytes (Cawood *et al.*, 1994). FA<sub>1</sub> and FA<sub>2</sub> are less cytotoxic, but PA<sub>1</sub> and PA<sub>2</sub> the hydrolysis products have similar or greater toxicity than the parent compounds (Abbas *et al.*, 1993; Gelderblom *et al.*, 1993). Elimination half-times of 18 min and 40 min were found in toxicokinetic studies of FB<sub>1</sub> in blood plasma of rats and monkeys respectively (Shephard *et al.*, 1992; 1993). Fumonisin is non-mutagenic according to the *Salmonella* test and non-genotoxic according to the DNA-repair assay with *Escherichia coli* (Gelderblom *et al.*, 1996b) and do not induce unscheduled DNA synthesis in primary rat hepatocytes (Gelderblom *et al.*, 1989; Norred *et al.*, 1990). FB<sub>1</sub> mimics genotoxic carcinogens, both in cancer initiation and promotion (Gelderblom *et al.*, 1992, 1994a,b) and also with respect to the induction of resistant hepatocytes in rat liver. FB<sub>1</sub> induces gamma glutamyltranspeptidase (GGT) and the placental form of glutathione-S-transferase (GSTP). These enzymes are histological markers for putative preneoplastic lesions which are initiated by genotoxic carcinogens (Gelderblom *et al.*, 1996b). Cancer initiation is affected by the induction of 'resistant' hepatocytes, whose

multiplication can be stimulated selectively by a cell proliferation stimulus in the presence of a 2-acetylaminofluorene induced mitoinhibitory effect (Gelderblom *et al.*, 1992 and 1993). When diethylnitrosamine is used as a cancer initiator, FB<sub>1</sub> acts as a cancer promoter, as indicated by the formation of  $\gamma$ -glutamyltranspeptidase and GSTP positive foci (Gelderblom *et al.*, 1988, 1996c). Fumonisin differ from genotoxic carcinogens in the sense that their cancer initiation step requires prolonged exposure of the fumonisin while in the case of genotoxic carcinogens this step is normally completed within a few hours or days (Gelderblom *et al.*, 1992). Gelderblom and co-workers, (1994b), for example, found that administration of a dose of 30.8 mg FB<sub>1</sub> per 100 g body weight to rats over a period of 21 days initiated cancer while the administration of a similar dosage over 7 days did not initiate cancer.

Of all the possible mechanisms of action of fumonisin toxicity in mammals, the one most studied is the inhibition of sphingosine and sphinganine *N*-acyltransferase. The disruption of sphingolipid biosynthesis, by the inhibition of the conversion of sphinganine to *N*-acyl-sphinganine (dihydroceramides) which precedes the introduction of the double bond of sphingosine, is reported to be connected with the diseases associated with fumonisins (Wang *et al.*, 1991). The disruption of the mechanism of sphingolipids (free long chain bases which are important components of cell membranes) could have serious effects on cell growth, differentiation and behaviour (Merrill, 1991). This action of fumonisins has been studied in rat liver hepatocytes (Wang *et al.*, 1991), mouse cerebellar neurons *in situ* (Merrill *et al.*, 1993) and *in vivo* in ponies (Wang *et al.*, 1992) and pigs (Riley *et al.*, 1993) and leads to the accumulation of sphingoid bases, which according to Schroeder *et al.*, (1994) is more likely to cause fumonisin mitogenicity than the inhibition of complex sphingolipid biosynthesis *per se*. Mitogens often affect cell transformations and this effect may explain the carcinogenicity of fumonisins. FB<sub>1</sub> and FB<sub>2</sub> are cytotoxic to renal epithelial (LLC-PK<sub>1</sub>) cells and inhibit proliferation after a lag period of at least 24 h in which the cells appear to function normally. Inhibition of sphingolipid biosynthesis, with an EC<sub>50</sub> of 10-15  $\mu$ M for FB<sub>1</sub>, occurred before cell proliferation and cell death thus supporting the hypothesis that this inhibition is an early event in the toxicity of fumonisins (Yoo *et al.*, 1992). Yoo *et al.* (1992), also found that the sphinganine levels increased greatly after only 6 hr exposure to 35  $\mu$ M FB<sub>1</sub> in LLC-PK<sub>1</sub> cells, and that these cells are much less sensitive than primary rat hepatocytes to fumonisin inhibition of *de novo* sphingolipid biosynthesis. Riley and co-workers (1993) found a relationship between the lower toxicity of FB<sub>1</sub> to the kidneys than to the liver of Sprague-Dawley rats, and the degree of disruption of the sphingolipid metabolism of these two organs.

The significantly higher elevation of the levels of free sphingosine, free sphinganine and the free sphinganine:sphingosine ratios found in the kidney when compared to the liver were also closely reflected in the urine of the rats. Free long chain bases and lysosphingolipids modulate intracellular signalling systems [e.g. protein kinase C, enzymes of diacylglycerol and phosphatidic acid metabolism and the tyrosine kinase activity of the epidermal growth factor (EGF) receptor] are cytotoxic to some cells and affect protein translocation, ATPases and calcium homeostasis (Riley *et al.*, 1993 and references cited). Gelderblom *et al.*, (1995) demonstrated that FB<sub>1</sub> inhibits the mitogenic response of the EGF *in vitro* in primary hepatocytes. Very little is known about the mechanisms involved in the inhibition of growth-related responses in hepatocytes, although the disruption of fatty acid metabolism has been implicated as playing a role (Gelderblom *et al.*, 1996a; 1997).

Apart from the inhibition of sphingolipid biosynthesis, FB<sub>1</sub> has also been found to affect the synthesis of cellular lipids by altering the incorporation of palmitic acid (Gelderblom *et al.*, 1996a). Gelderblom and co-workers (1996a) monitored the fatty acid levels of the major phospholipids (phosphatidylethanolamine and phosphatidylcholine) and neutral lipid triacylglycerides *in vitro* in rat hepatocytes and *in vivo* in rats and found alterations in the n-6 fatty acid profiles and decreases in the free cholesterol (membrane associated) levels, which resulted in a higher phosphatidylcholine: cholesterol ratio; this suggested a more rigid membrane structure. A significant increase in the serum and the total cholesterol content of the liver was found when the highest level of 250 mg FB<sub>1</sub>.kg<sup>-1</sup> was administered to rats. Fumonisin may thus have important effects on membrane components, the fatty acid storage pool and the accumulation of long chain fatty acids within the cell which could eventually lead to the disintegration of membrane structures and eventually result in cell death (Gelderblom *et al.*, 1996b; 1997).

Cell proliferation, an important factor in cancer initiation and promotion, is controlled by long chain fatty acids *via* their control of prostaglandin levels (Cornwell and Morisaki, 1984). Prostaglandins can either inhibit or stimulate cell proliferation depending on the cell type. *In vitro* studies using Balb/c 3T3 cells, have shown that arachidonic acid metabolism is required for the mitogenic response of the EGF (Nolan *et al.*, 1988; Handler *et al.*, 1990).

Primary hepatocytes exposed to FB<sub>1</sub> showed an accumulation of polyunsaturated fatty acids. Gavino and co-workers (1981) demonstrated that increased levels of polyunsaturated fatty acids are associated with lipid peroxidation in normal and cancer cells, which implies that FB<sub>1</sub>

can indirectly cause lipid peroxidation (See Mechanisms of Action of Ochratoxin A) (Gelderblom *et al.*, 1996b, 1998).

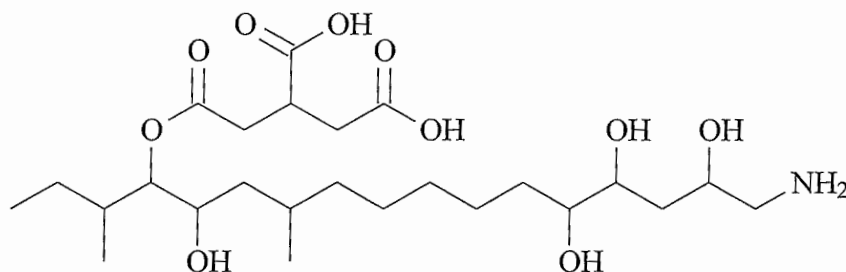
FB<sub>1</sub> causes lesions in the liver, kidneys, heart and lungs and subsequent death of broiler chicks (Javed *et al.*, 1992). Qureshi and Hagler, (1991) found that FB<sub>1</sub> can effect the macrophage dependent immune system of chickens. These findings have been confirmed by Chatterjee and Mukherjee (1994) who found a significant reduction in the viability and phagocytic potential of macrophages from chicken peritoneal exudate cells. Fumonisin consumption may thus result in a decreased immune response, which results from FB<sub>1</sub>-induced depressed macrophages and consequently leads to infections.

The short half-life of fumonisins in monkey plasma (only trace levels are left after 4 hr) indicates that the direct measurement of fumonisins in blood is not be suitable for the determination of fumonisin exposures in animals and humans. Methods have, therefore, been developed to monitor fumonisin toxicosis in monkeys by measuring the sphingamine:sphingosine ratio (Shephard *et al.*, 1993,1996).

FB<sub>1</sub> was found to be a powerful inducer of malondialdehyde (one of the secondary products formed during lipid peroxidation) and to inhibit both protein and DNA synthesis (Ennamany *et al.*, 1995; Mobio *et al.*, 1998; Abado-Becognee *et al.*, 1998).

FB<sub>1</sub> and FB<sub>2</sub> are phytotoxic and damage weed and crop cultivars like soybean and tomato. The primary site of fumonisin toxicity in jimson weed is the plasmalemma or tonoplast. It causes rapid, light-dependent cytoplasmic degeneration and chloroplast disruption (Abbas *et al.*, 1992). FB<sub>1</sub> and the TA-toxin produced by *Alternaria alternata* f.sp. *lycopersici* have similar structures and produce identical genotype-specific necrotic symptoms on detached leaves of resistant and susceptible tomato lines (Marasas *et al.*, 1996 and references cited).

FB<sub>1</sub> and TA-toxin (see **Figure 7**) are more phytotoxic to corn and tomato seedlings than FB<sub>2</sub> and FB<sub>3</sub> and cause reductions in shoot and root length (Lamprecht *et al.*, 1994). The mechanism of action in plants is not yet known.



**Figure 7:** Structure of TA-toxin.

## Conclusion

Sterling research is currently being done on mycotoxins. These efforts are focused on the molecular genetics of toxinogenic filamentous fungi (Bennett, 1994; O'Donnell, 1997; Bennet and Keller, 1997), a molecular understanding of the basic mechanism of their action; species differences in metabolism and pharmacokinetics; immunobased and physicochemical techniques for the quantification of mycotoxins; analysis of the risk involved in the exposure of man and domestic animals to mycotoxins (Kuiper-Goodman, 1989; Kuiper-Goodman and Scott, 1989; Kuiper-Goodman, 1996) and the associated regulations for the control of mycotoxin contamination (Kuiper-Goodman, 1995) and applying plant molecular biotechnological techniques to breed mycotoxin-resistant cereal- and nut-producing cultivars.

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