

# **Synthesis and *in vitro* anti-protozoan activities of nitrofuran-based azine derivatives**

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## **PREFACE**

This dissertation is submitted in an article format in accordance with the General Academic Rules (A.13.7.3) of the North-West University.

**Chapter 1: Introduction and Problem Statement**

**Chapter 2: Literature Review**

**Chapter 3: Article for submission**

**Synthesis and *in vitro* antileishmanial activities of nitrofuran-based azine derivatives**

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**Chapter 4: Summary of the study**

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## ABSTRACT

Neglected tropical diseases (NTDs) are a group of diverse infectious conditions that are closely connected to poverty and affects more than a billion people worldwide. Leishmaniasis is a vector-borne NTD that is caused by the *Leishmania* parasite. Up to one million new cases of leishmaniasis are reported annually and current treatments are limited to a few chemotherapeutic drugs with significant toxicity, administration, cost, and parasite resistance issues. Therefore, there is an urgent need to search for new, safe, and affordable antileishmanial drugs.

In this study, the antileishmanial activity of several nitrofuran-based azine derivatives was investigated. The synthesised derivatives consist of three active pharmacophores, namely the nitrofuran, hydrazone and aromatic ring. Nitrofurans possess a broad range of biological activity against various diseases, including leishmaniasis. Therefore, in this study two sub-series of compounds (nitrofuran and nitrothiophene derivatives) were synthesised using the two-step process of hydrazone formation and Schiff base reactions. The derivatives were screened for activity against *L. donovani* and *L. major* promastigotes.

The antileishmanial activity ranged from good ( $IC_{50} = 0.42 \mu M$ ) to no activity ( $IC_{50} > 100 \mu M$ ), with the nitrofuran sub-series exhibiting better activity compared to the nitrothiophene sub-series. The cytotoxicity of the compounds ranged from moderately toxic to non-toxic ( $IC_{50} 11.56 - > 100 \mu M$ ). Compounds **3a**, **5a** and **7a** had good activity against all three *Leishmania* strains and were non-toxic, thus are potential anti-promastigote hits for further investigation. These derivatives could serve as building blocks for the development of future antileishmanial agents.

**Keywords:** *Leishmania*, promastigote, cytotoxicity, nitrofuran, nitrothiophene

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## LIST OF ABBREVIATIONS

|                  |  |
|------------------|--|
| µM               | Micromolar                                 |
| AmB              | Amphotericin B                             |
| APCI             | Atmospheric pressure chemical ionisation   |
| ATP              | Adenosine triphosphate                     |
| CDC              | Centers for Disease Control and Prevention |
| CL               | Cutaneous leishmaniasis                    |
| cNFs             | Clinical Nitrofurans                       |
| DAT              | Direct agglutination test                  |
| DMEM             | Dalbecco's modified Eagle's medium         |
| DMF              | <i>N,N</i> -dimethylformamide              |
| DMSO             | Dimethyl sulfoxide                         |
| DNA              | Deoxyribonucleic acid                      |
| EDG              | Electron donating group                    |
| ELISA            | Enzyme-linked immunosorbent assays         |
| EM               | Emetine                                    |
| EtOAc            | Ethyl acetate                              |
| EtOH             | Ethanol                                    |
| EWG              | Electron withdrawing group                 |
| FBS              | Fetal bovine serum                         |
| FZD              | Furazolidone                               |
| HAT              | Human African trypanosomiasis              |
| HIV              | Human immunodeficiency virus               |
| HRMS             | High resolution mass spectrometry          |
| IC <sub>50</sub> | 50% inhibitory concentration               |
| IFA              | Indirect fluorescent antibody test         |
| IM               | Intramuscular                              |

|           |  |
|-----------|--|
| IR        | Infrared   |
| IV        | Intravenous  |
| L-AmB     | Liposomal amphotericin B                               |
| LFT       | Liver function test                                    |
| M199      | Media 199  |
| MCL       | Mucocutaneous leishmaniasis                            |
| MHz       | Megahertz  |
| m.p.      | Melting point  |
| NADH      | Nicotinamide adenine dinucleotide + hydrogen           |
| NADPH     | Nicotinamide adenine dinucleotide phosphate + hydrogen |
| NEAA      | Non-essential amino acid                               |
| NFA       | 5-Nitro-2-furaldehyde                                  |
| NFT       | Nitrofurantoin   |
| NFX       | Nifuroxazide   |
| NFZ       | Nitrofurazone  |
| NMR       | Nuclear magnetic resonance                             |
| NNN       | Novy-McNeal-Nicolle                                    |
| NTA       | 5-Nitrothiophene-2-carboxaldehyde                      |
| NTD       | Neglected tropical disease                             |
| PAHO      | Pan American Health Organization                       |
| Pen/Strep | Penicillin-streptomycin                                |
| ppm       | Parts per million                                      |
| ROS       | Reactive oxygen species                                |
| RNS       | Reactive nitrogen species                              |
| rt        | Room temperature                                       |
| SAR       | Structure activity relationships                       |
| SD        | Standard deviation                                     |

|           |                                |
|-----------|--------------------------------|
| SI        | Selectivity index              |
| spp       | Species                        |
| $t_{1/2}$ | Terminal half-life             |
| TB        | Tuberculosis                   |
| TEA       | Triethylamine                  |
| TLC       | Thin layer chromatography      |
| TPSA      | Topological polar surface area |
| UK        | United Kingdom                 |
| VL        | Visceral leishmaniasis         |
| WHO       | World Health Organization      |

# CHAPTER 1

## INTRODUCTION AND PROBLEM STATEMENT

### 1.1 Introduction

Leishmaniasis is a group of vector-borne protozoan parasitic diseases that is caused by more than twenty *Leishmania* species (Georgiadou *et al.*, 2015:44). It is transmitted to humans through the bite of an infected female phlebotomine sandfly and it can manifest as three clinical forms, namely visceral (VL), cutaneous (CL) and mucocutaneous (MCL), depending on the infecting *Leishmania* species (Salam *et al.*, 2014:1; Braga, 2019:1; WHO, 2021a).

VL is the most severe form of the disease, and it can be contributed to the *L. donovani* and/or *L. infantum* species (CDC, 2021). VL affects the internal organs with symptoms such as fever, anaemia, weight loss, enlarged organs, and pancytopenia (Braga, 2019:2; CDC, 2021; WHO, 2021a) and this disease is usually fatal unless treated (WHO, 2021a). CL on the other hand, is predominantly caused by the species *L. tropica*, *L. major*, *L. aethiopica*, *L. infantum*, *L. donovani*, and *L. mexicana* (CDC, 2021) and these infections results in the development of skin lesions that heal very slowly, leaving behind atrophic scars (Braga, 2019:1). MCL is caused by the species *L. braziliensis*, *L. panamensis* or *L. guyanensis* and this form of the disease results in partial or complete destruction of soft and hard tissues of the palate, pharynx, and nose (Braga, 2019:2; CDC, 2021).

In general, diseases become neglected for three reasons: (1) when they are isolated to certain areas and are not a global threat, (2) when there is no financial benefit in treating them, and (3) when the disease is not considered deadly. Leishmaniasis is a NTD that is prevalent in tropical and sub-tropical developing countries and mostly affects poor individuals in rural areas (Wilson *et al.*, 2020:2; WHO, 2021b). There are no financial incentives to properly address leishmaniasis (Erber *et al.*, 2020:2); although, the annual death toll is currently about 30 000 deaths per year (PAHO, 2020). Drugs currently available on their market to treat leishmaniasis have certain shortcomings, i.e., impractical intravenous administration (apart from the oral drug, miltefosine), toxicity, high cost, low efficacy and drug resistance (Georgiadou *et al.*, 2015:46; Valle *et al.*, 2019:2; Karamysheva *et al.*, 2020:7). Drug efficacy is a major problem as it varies according to the infecting *Leishmania* species and the infection region. Another issue that is commonly experienced is infection relapse and this occurs especially in human immunodeficiency virus (HIV)-coinfected patients (Boelaert & Sundar, 2014:645).

Treatment of NTDs such as leishmaniasis are mostly in the form of global or organisational aid or donations to these poverty-stricken areas (WHO, 2021c). Furthermore, the cutaneous (skin

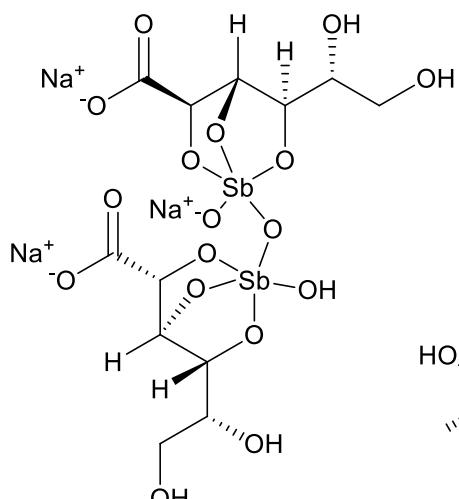
lesion) form is the most common form of leishmaniasis (CDC, 2021); hence, the disease is commonly regarded as debilitating, but not necessarily life threatening. This results in a perceived reduced need for global intervention as focus is rather placed on more deadly diseases (Erber *et al.*, 2020:2).

In the past few years, there have been various efforts, such as vector control, disease surveillance, animal reservoir host control, early diagnosis and treatment, and social mobilisation, to control the spread of leishmaniasis (WHO, 2021a). However, the currently available statistics indicate that these methods have not yet been successful. Although there was a decrease in new VL cases in 2019 (13 809 cases) compared to 2018 (17 093 cases) (WHO, 2021d), there was an increase in new CL cases (277 058 cases reported in 2019 compared to 261 285 cases in 2018) (WHO, 2021e).

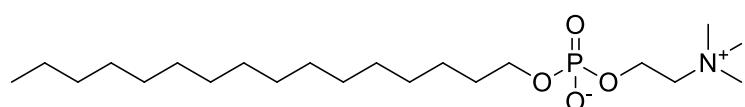
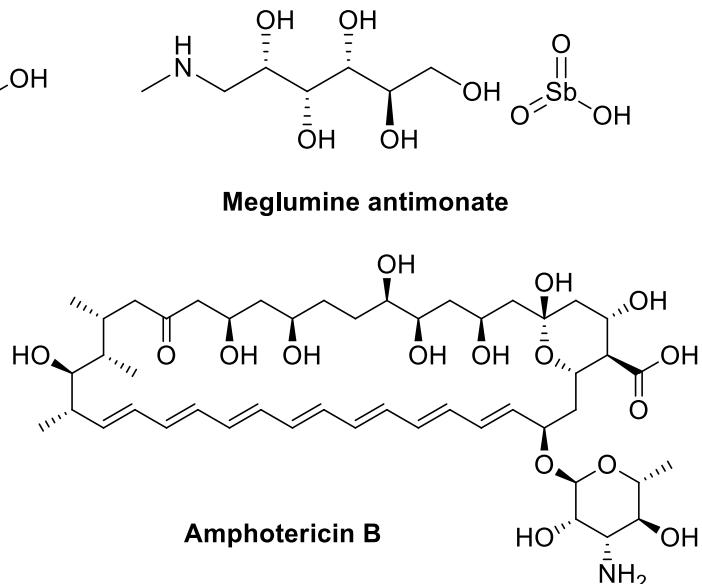
The treatment of leishmaniasis relies on a handful of diverse chemotherapeutic drugs, such as the pentavalent antimonials (sodium stibogluconate and meglumine antimonate), amphotericin B, miltefosine, pentamidine, paromomycin and the azoles (ketoconazole, itraconazole and fluconazole) (Figure 1-1) (Georgiadou *et al.*, 2015:46; Braga, 2019:2). Leishmaniasis is a difficult disease to diagnose, since it can present itself with symptoms that are similar to those of other infectious diseases such as malaria, Chagas' disease, tuberculosis (TB), schistosomiasis, and typhoid fever (Camargo & Langoni, 2006:534).

The reduction in efficacy and drug resistance of the parasites against antileishmanial treatment have led to the development and clinical implementation of combinational therapies. Combination therapy has the potential to shorten treatment duration, increase patient compliance and reduce medication dosage, thereby reducing toxicity, cost, and the risk of drug resistance development (WHO, 2010). Combination therapies that are currently in use include sodium stibogluconate with paromomycin, and miltefosine with amphotericin B (Uliana *et al.*, 2018:471).

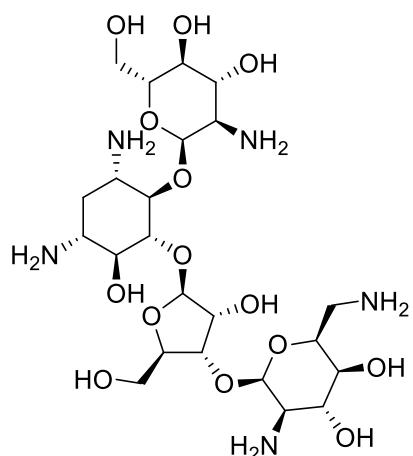
Although rare, there has been a report of resistance to combination therapy by *L. donovani* promastigotes. It was found that these parasites can develop resistance to miltefosine/paromomycin and sodium stibogluconate/paromomycin combinations (Ponte-Sucre *et al.*, 2017:14). Hence, there is an urgent need for the discovery of new, safe and effective drugs to treat this disease (Capela *et al.*, 2019:1).



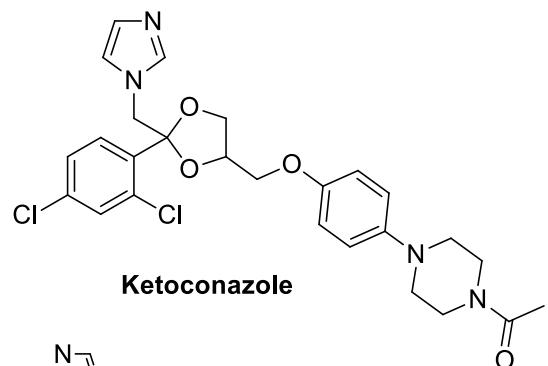
**Sodium stibogluconate**



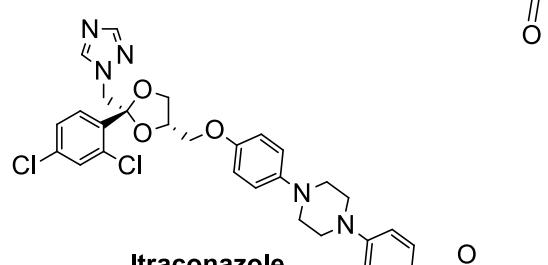
**Miltefosine**



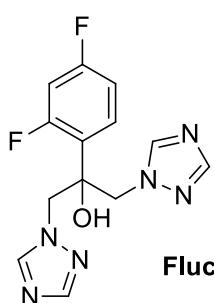
**Paromomycin**



**Ketoconazole**



**Itraconazole**



**Fluconazole**

**Figure 1-1: Current clinical antileishmanial chemotherapeutic agents**

## 1.2 Rationale for this project

Research has indicated that the nitrofuran class of drugs are biologically active against a broad spectrum of diseases, such as tuberculosis (Krasavin *et al.*, 2019:128), cancer (Peterson, 2018:1439), urinary tract infections (Huttner *et al.*, 2015:2456), and giardiasis (Ryan, 2018:1567). It has also been successfully used as pharmacophores in the treatment of Chagas disease (Sales Junior *et al.*, 2017:1289), caused by flagellated protozoan parasites of the same family (*Trypanosomatidae*) and order (*Kinetoplastida*) as *Leishmania*.

Previous studies have also indicated that the nitrofuran moiety has good antileishmanial activity. In a study conducted by Sifontes-Rodríguez *et al.* (2015), they investigated the antileishmanial activity of nitrofuran-derivatives and found that all their compounds exhibited activity against the promastigote forms of *L. amazonensis* (MHOM/77/LTB0016 strain), *L. braziliensis* (MHOM/BR/75/M2 903 strain), and *L. infantum* (MHOM/FR/78/LEM75 strain). The activity IC<sub>50</sub> values ranged between 0.8 μM to 4.7 μM against all three strains (Sifontes-Rodríguez *et al.*, 2015:169).

Another study reported four antileishmanial promising nitrofuran-derivatives against the promastigote and amastigote forms of *L. major* with IC<sub>50</sub> values in the 0.08 – 2.00 μM range (Tahghighi *et al.*, 2011:2605).

Thus, the nitrofuran scaffold seems to be a good starting block to use in the discovery of new antileishmanial drugs. In this study, the antileishmanial activity of nitrofuran-based azine derivatives was investigated in an effort to discover potentially new anti-infective compounds. This study introduced three pharmacophores in the targeted compounds (Figure 1-2). The nitrofuran moiety (green) was used as a starting block to enhance the drugs' anti-parasitic activity. The second pharmacophore, the azine bond (blue), is the linker between the nitrofuran and the aromatic ring and this group also contributes to the activity of the compound as it also possess' a broad spectrum of activity. The aromatic ring (red), pharmacophore three, not only brings stability to the compound but also contributes to its overall activity.

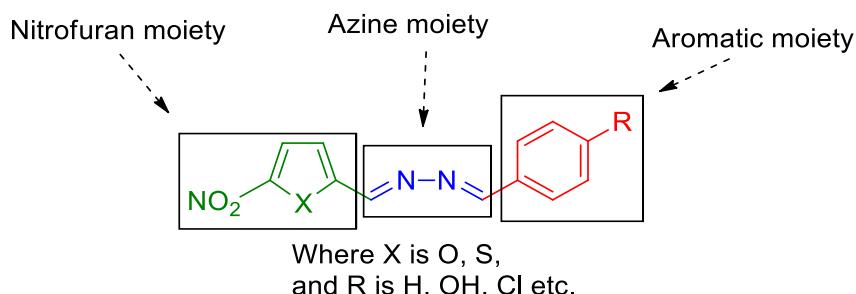


Figure 1-2: General structure of target nitrofuran-based azine derivatives

### **1.3 Aim and objectives**

The aim of this study is to investigate the antileishmanial activity of nitrofuran-based azine derivatives.

#### **Objectives of this study are:**

- To synthesise a series of 18 nitrofuran-based azine derivatives with the general structure as depicted in Figure 1-2.
- To characterise the synthesised derivatives using routine techniques such as nuclear magnetic resonance (NMR), high resolution mass spectrometry (HRMS), and infrared (IR) spectroscopy.
- To assess *in vitro* antileishmanial activity of the synthesised compounds.
- To assess *in vitro* cytotoxicity of the active compounds using mammalian cell lines.

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## CHAPTER 2:

# LITERATURE REVIEW

### 2. Introduction

NTDs are a group of diverse infectious conditions that differ in their epidemiology, distribution, impact and control (Kuper, 2019:838). NTDs are closely connected to poverty and affects more than a billion people worldwide (CDC, 2018; Engels & Zhou, 2020:2). There are more than a hundred and forty-nine countries and territories affected by at least one NTD. However, low-income countries are generally affected the most and can be affected by at least five NTDs simultaneously (CDC, 2018). These diseases are mainly caused by a variety of pathogens such as bacteria, viruses and parasites, which are common in tropical and sub-tropical climates (Engels & Zhou, 2020:1; Hotez *et al.*, 2020:2; Nweze *et al.*, 2021:2).

Parasites are organisms that live in or on another organism and depend on their host for survival. Parasites that cause diseases in humans can be divided into three main categories: protozoa, helminths, and ectoparasites (CDC, 2020a). *Leishmania* is a vector-borne neglected tropical protozoan (single-celled eukaryotic) parasite that possess a flagellum for motility (Paget, 2011:86). They are also referred to as *haemoflagellates* because they are parasitic in the host's blood and require haematin obtained from it. *Leishmania* is part of the *Trypanosomatidae* family (order *Kinetoplastida*) and possess a unique organelle called the kinetoplastid, which appears to be a DNA rich part of the mitochondrion (Paget, 2011:86).

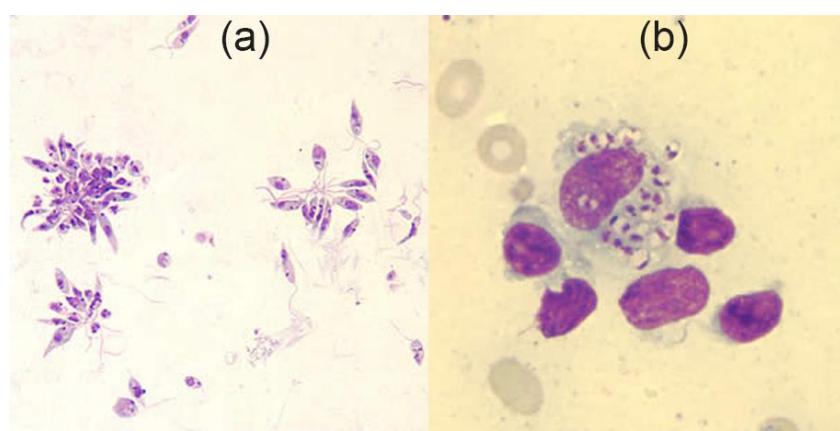


Figure 2-1: Promastigote (a) and amastigote (b) forms of *Leishmania* spp (CDC, 2017).

There are more than twenty species belonging to the *Leishmania* genus that can infect humans, resulting in the development of the disease called leishmaniasis (Braga, 2019:1; Valle *et al.*, 2019:2). *Leishmania* is transmitted to humans during hematophagy of an infected female sandfly of the *Phlebotomus* genus (WHO, 2021a). It can manifest itself in three different clinical forms

namely visceral leishmaniasis (VL), cutaneous leishmaniasis (CL) and mucocutaneous leishmaniasis (MCL) (Salam *et al.*, 2014:1). VL is the most dangerous form, causing excessive internal damage with a mortality rate of 95% if left untreated (WHO, 2021b). Of all of the species, *L. donovani* is responsible for the majority VL and *L. major* is responsible for the majority of CL (Güran, 2018:2). Leishmaniasis is ranked as one of the top ten most neglected diseases in the world (WHO, 2017). This disease is known to mostly affect individuals in developing countries and of low socioeconomic level (Georgiadou *et al.*, 2015:44).

There are more than a billion individuals living in *Leishmania* endemic areas. It is estimated that about 90 000 new cases of VL and a million new cases of CL are reported annually worldwide (WHO, 2021a). Although leishmaniasis can cause significant mortality, there are very few efforts from the global community and pharmaceutical industry to invest in the research and development of novel therapeutics because of insufficient incentives (Salam *et al.*, 2014:1).

The current drugs available for the treatment of leishmaniasis have several limitations including toxicity and intravenous administration, with the latter being a limiting factor in eradicating the disease in poverty stricken endemic areas (Freitas-Junior *et al.*, 2012:12). The eradication of leishmaniasis is further complicated by the incompatibility of the currently available drugs for pregnant women or individuals with hepatic or renal failure (Camargo & Langoni, 2006:536). Treatment recommendations varies in different regions due to the variation in response to drug therapy and the parasite species involved (Chakravarty & Sundar, 2010:167; WHO, 2021b). Therefore, there is an urgent need for new and effective antileishmanial drugs.

This chapter will focus on discussing current statistics, treatment and parasite resistance, as well as the challenges faced by leishmaniasis chemotherapy.

## 2.1 History of leishmaniasis

A study by Zink *et al.* (2006:1616) confirmed the presence of *L. donovani* found in DNA samples of Egyptian mummies of the Middle Kingdom in Thebes West dating back 4000 years ago. This disease is also mentioned in the Ebers Papyrus (a collection of ancient Egyptian medical documents dating back to 1500 BCE) as a skin condition named “Nile Pimple”, which was later identified as CL (Maspero, 1910; Steverding, 2016:4).

Scottish doctor, David Douglas Cunningham first observed *Leishmania* amastigotes in skin lesions of patients from India in 1885, but suggested that they were members of the fungi group *Mycetozoa* (Cunningham, 1885; Kobets *et al.*, 2012:1443; Steverding, 2016:5).

In 1897, a Scottish pathologist, Lieutenant General Sir William Boog Leishman worked as an assistant to Ronald Ross at Netley, where he studied patients that was sent home from India for

suspected cases of malaria. At that time, he noticed a number of soldiers with remittent fever, muscular atrophy, anaemia and spleen enlargement. He considered the possibility that the symptoms indicated a specific disease that he named 'Dum-Dum fever' (Leishman, 1903:1252; Vincent, 2017:1584). In 1900, Leishman examined samples of a deceased patient who exhibited these symptoms before his death. He stained spleen samples to highlight the chromatin, where he found two chromatin masses in the bodies from the blood cells but he could not identify them (Leishman, 1903:1253; Vincent, 2017:1584). Three years later he found similar bodies in a rat that died of trypanosomiasis that led him to suggest that 'Dum-Dum fever' was caused by trypanosomes (Leishman, 1903:1253; Vincent, 2017:1584). Later in that same year, Charles Donovan reported that he found the same bodies as previously reported by Leishman. Ross examined Donovan's preparations and agreed that they matched Leishman's; hence, they concluded that Leishman had discovered a novel organism. This newly discovered species was then named *Leishmania donovani* (Ross, 1903:1262; Vincent, 2017:1584).

## 2.2 Epidemiology

Leishmaniasis is endemic to more than 98 countries and is among the most neglected and devastating tropical diseases in the world (Güran, 2018:2). This disease is the third biggest cause of parasite-related deaths in the world, with malaria being the first and schistosomiasis the second (WHO, 2017).

Leishmaniasis is prevalent in Africa, Latin America and Asia, and is usually correlated with malnutrition, poor living conditions, weakened immune systems, and shortage of resources (Georgiadou *et al.*, 2015:44). There are about 12 to 15 million individuals in the world infected and about 1 billion people at risk of contracting the disease (WHO, 2021a). Leishmaniasis can spread to non-endemic areas through tourism, migration, and military movement (Inceboz, 2019:2). Environmental changes such as climate change, has the potential to extend the geographic range of the sandflies and therefore the areas in which leishmaniasis occurs (CDC, 2020b).

In 2019, the WHO reported 13 809 new cases of VL (20% of which occurred in India), a decrease compared to 2018 where 16 970 new cases were reported (WHO, 2021d). The total number of new CL cases reported in 2019 was 277 058 (26% of which occurred in the Syrian Arab Republic), an increase compared to the 244 005 cases reported the preceding year (WHO, 2021e). Since leishmaniasis is a NTD, there are very limited statistics available on MCL. Leishmaniasis causes about 30 000 deaths annually (PAHO, 2020).

VL, also known as 'kala-azar', is caused by *L. donovani* and *L. infantum* (also known as *L. chagasi*) (McGwire & Satoskar, 2014:9). VL is endemic to Brazil, South-East Asia, and East

Africa (Figure 2-2). Approximately 90% of these reported cases occurs in Brazil, India, Ethiopia, Kenya, Sudan, South Sudan and Somalia (Braga, 2019:2).

### Endemicity status of visceral leishmaniasis worldwide, 2019

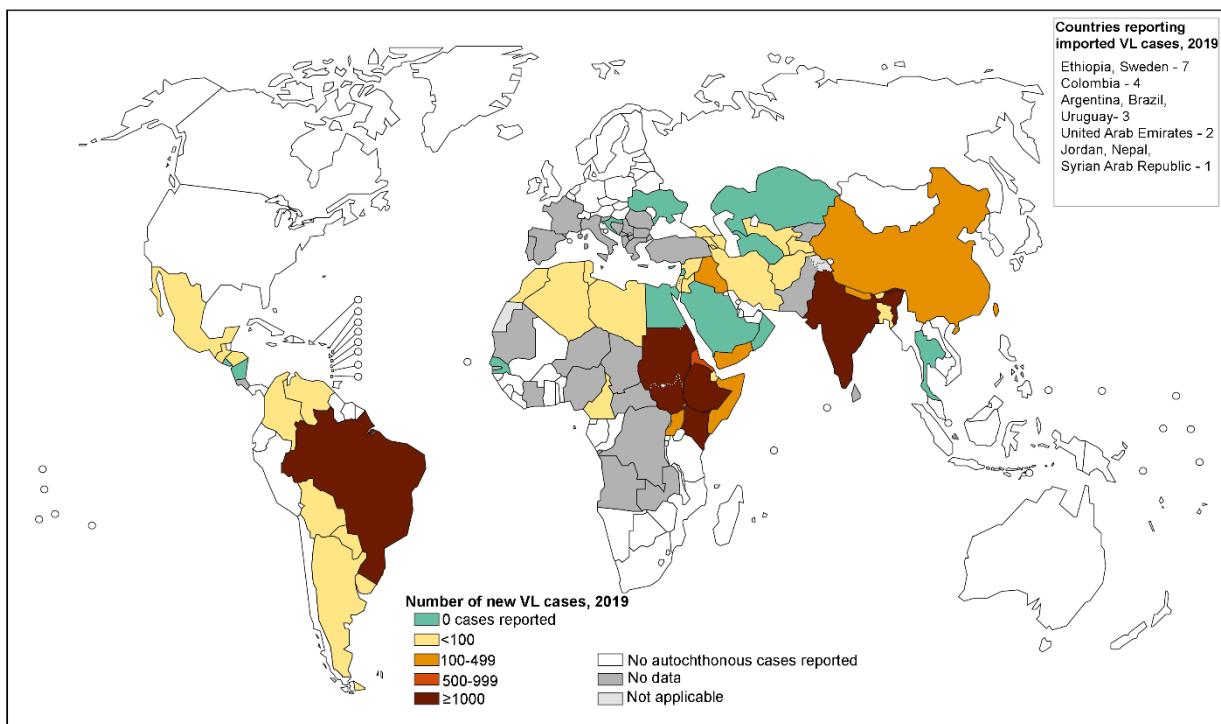


Figure 2-2: Distribution of visceral leishmaniasis (VL) around the world. Adapted from WHO (2019a).

CL is caused by *L. major*, *L. infantum*, *L. tropica*, *L. Mexicana*, *L. amazonensis*, *L. guyanensis*, *L. panamensis*, *L. peruviana*, and *L. braziliensis* (Eiras et al., 2015:53). CL is endemic to regions such as the Mediterranean basin, Latin America, and western Asia from the Middle East to central Asia (Aoun & Bouratbine, 2014:2) (Figure 2-3).

## Endemicity status of cutaneous leishmaniasis worldwide, 2019

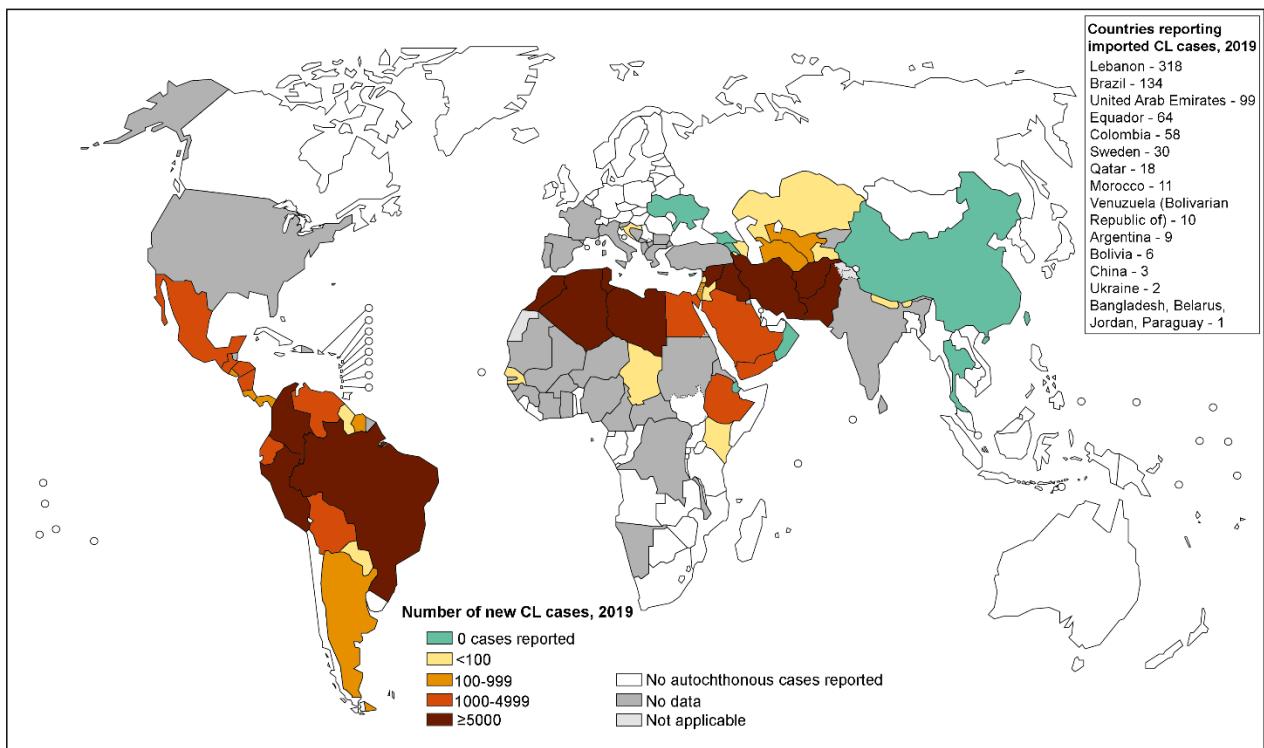


Figure 2-3: Distribution of cutaneous leishmaniasis (CL) around the world. Adapted from WHO (2019b).

MCL is caused by *L. braziliensis* and it is found to occur mainly in South American countries. The majority of cases occurs in Peru, Brazil and Bolivia, but the disease is also found in lesser degrees in Ecuador, Colombia, Venezuela and Paraguay (McGwire & Satoskar, 2014:9).

### 2.3 Transmission and life cycle of *Leishmania*

*Leishmania* parasites are transmitted through vectors from host to host. Transmission can be anthroponotic (human to vector to human) or zoonotic (human to vector to animal or vice versa) and occurs in urban and rural environments, respectively (Avila-García *et al.*, 2014:27). Transmission of the parasite from the vector to the host depends on how infective the host is, how infectious the sandfly's single bite is, the biting rate on average, and the number of sandflies that are present (Torres-Guerrero *et al.*, 2017:6). The saliva of the vector contains several components that play an important role in the establishment of a *Leishmania* infection such as vasodilatory and anticoagulant factors to aid blood flow, and immunomodulatory factors to inhibit healing of the wound and to stop the immune response (Torres-Guerrero *et al.*, 2017:6).

Although not common, in 1960 in the United Kingdom (UK), there was a report of sexual transmission of leishmaniasis in which a woman developed a genital papule containing *Leishmania donovani* bodies, despite never visiting a *Leishmania*-endemic region. Her husband

was diagnosed with VL several years before, and although at the time of his wife's lesion he had no symptoms, it was assumed that he had transmitted the disease to his wife (Turchetti *et al.*, 2014:403; Guedes *et al.*, 2020:950).

The life cycle of *Leishmania* (Figure 2-4) is divided into two stages; the motile flagellated promastigote stage (present in the vector), and the intracellular non-flagellated amastigote stage (present in mammalian host cells) (McGwire & Satoskar, 2014:7). When a sandfly bites an infected host (e.g. animal), it becomes infected and ingests the amastigote form from its host. The parasites in the fly's gut then transform from the amastigote form to the proliferative procyclic promastigote form, followed by the infectious metacyclic promastigote form. Thereafter, when the infected sandfly bites a new healthy host, the vector inoculates it *via* regurgitation of the metacyclic promastigotes (Torres-Guerrero *et al.*, 2017:6). The infection is initiated through receptor-mediated binding of the infective promastigotes to phagocytising host cells, resulting in the encapsulation of the parasites in parasitophorous vacuoles within the host cells (McGwire & Satoskar, 2014:7). Parasitophorous vacuoles containing parasites fuse with lysosomes to form phagolysosomes. In these phagolysosomes, promastigotes are transformed into and replicate as amastigotes (McGwire & Satoskar, 2014:8).

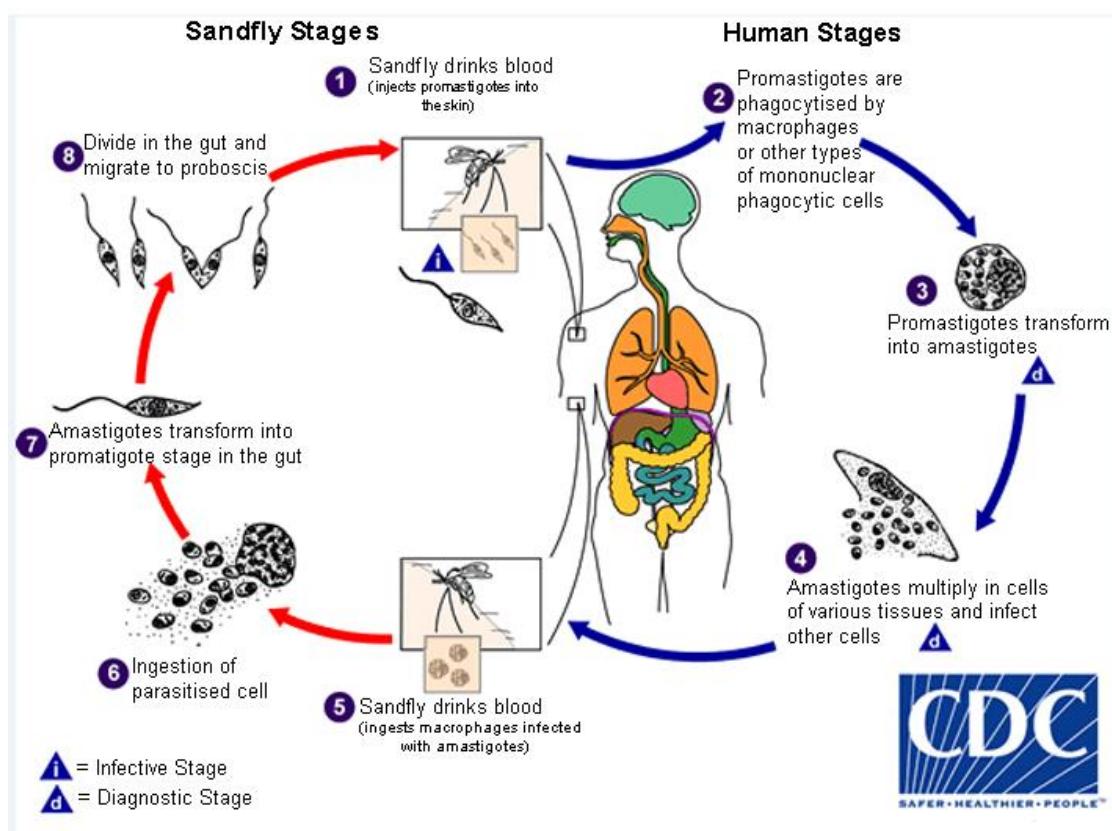


Figure 2-4: Life cycle of leishmaniasis depicting the different stages that occurs. Adapted from CDC (2020c).

When the parasites enter the host, the host's immune system activates the complement system (a large number of distinct plasma proteins that helps fight infection by reacting with one another to opsonise pathogens and induce inflammatory responses (Janeway *et al.*, 2001)). However, in humans, the parasite can evade this immune response *via* inhibition of complement-mediated lysis, which takes place in the phagolysosomes of macrophages. This effect is brought on by the membrane protease gp63, or leishmanolysin, which is responsible for the inhibition of immune attacks against the parasite cell membrane by attaching to complement components (Avila-García *et al.*, 2014:28). The parasite burden increases until the infected macrophages eventually gets disrupted and spreads extracellular amastigotes to surrounding tissue, infecting previously uninfected macrophages (McGwire & Satoskar, 2014:8).

## **2.4 Clinical features**

The clinical features of leishmaniasis are determined by the three main human forms, namely VL, CL and MCL. The outcomes of each form are determined by the infecting parasite species and the host's genetic susceptibility (McGwire & Satoskar, 2014:8). However, most people infected with the *Leishmania* parasite, do not develop any symptoms at all. Thus, leishmaniasis refers to showing symptoms of *Leishmania* infection, rather than just being infected with the parasite (WHO, 2021a).

### **2.4.1 Visceral leishmaniasis (VL)**

VL is a fatal form of leishmaniasis and has a mortality rate of 75-95% (Güran, 2018:7). VL has an incubation period of 3-8 months with preschool children, with the undernourished and immunocompromised individuals being the most at risk (Torres-Guerrero *et al.*, 2017:9).

VL is most commonly characterised by irregular fever, lymphadenopathy, progressive pallor, splenomegaly, hepatomegaly, anorexia, weight loss, anaemia, thrombocytopenia, leukopenia, hypoalbuminemia, increased globulin, night sweats, oedema, asthenia, and cutaneous pigmentation (Pedrosa, 2017:17; Torres-Guerrero *et al.*, 2017:9). Children infected with this disease can also experience chronic diarrhoea and growth retardation (Torres-Guerrero *et al.*, 2017:9). Treatment varies for VL caused by *L. donovani*, but pentavalent antimonials and amphotericin B are the two preferred therapies in use (discussed in section 2.7).

### **2.4.2 Cutaneous leishmaniasis (CL)**

CL is the most common form of leishmaniasis. It is a milder form of the disease and a sandfly bite causes skin lesions, ulcers or nodules on the host leaving behind lifelong scars and/or a serious disability (Güran, 2018:7). The incubation period of CL is 1-4 weeks but can proceed for several years (Torres-Guerrero *et al.*, 2017:7).

CL can either be local (commonly presented with painless ulcerated lesions with raised edges), disseminated (presenting with less ulcerated lesions that can occur on all parts of the body), or diffused (a chronic and rare form that can compromise the skin of the patient, with frequent recurrence) (Camargo & Langoni, 2006:533). Diffused CL can cause lymphedema, lymphadenopathy, fever and poor health (Torres-Guerrero *et al.*, 2017:8). CL is self-curing and nonlife-threatening, but accumulation of CL can often lead to permanent, disfiguring formations on the skin (Güran, 2018:7). The healing of cutaneous lesions can be sped up by treatment (McGwire & Satoskar, 2014:9). Treatment for CL consists of pentavalent antimonials, liposomal amphotericin B, miltefosine, fluconazole and ketoconazole (Eiras *et al.*, 2015:52) (discussed in section 2.7).

#### **2.4.3 Mucocutaneous leishmaniasis (MCL)**

MCL, in contrast to CL, is a potentially life-threatening form of leishmaniasis that requires treatment, however it is a less common form (David & Craft, 2009:497). MCL can occur when there is dissemination of CL lesions *via* the lymphatic or hematogenic route, or the CL lesions were not properly treated. This is due to the formation of secondary cutaneous lesions and it can occur months to years after the primary lesions have healed (Camargo & Langoni, 2006:533; McGwire & Satoskar, 2014:9).

Generally, lesions begin in the nasal mucosa and spread to the pharynx and oral mucosa, the larynx, and skin of the nose and lip (Torres-Guerrero *et al.*, 2017:9). In the early stages of MCL infection, symptoms of erythema and ulceration of the nostrils, as well as anterior and posterior nasal granulomas, lymphadenopathy, fever, and hepatomegaly can occur (David & Craft, 2009:498). During the mid-to-late stages of MCL infection, patients can present with oedema and erythema of the nostrils, ulceration of the hard or soft palate, perforation of the nasal septum, gingival oedema, and periodontitis (David & Craft, 2009:498). This disease is terribly disfiguring and causes chronic local destruction of tissue in the nose, oro- and naso-pharynx, mouth, and eyelids and it can progress to the point where respiratory functions are affected and nutrition is hindered (McGwire & Satoskar, 2014:9). MCL can usually be treated with pentavalent antimonials, pentamidine, miltefosine, amphotericin B, paromomycin or the azoles (fluconazole, ketoconazole and itraconazole) (discussed in section 2.7).

### **2.5 Diagnoses**

Leishmaniasis is diagnosed based on its clinical features, endemicity and laboratory testing such as parasitological, immunological and serological testing (Salam *et al.*, 2014:2; Georgiadou *et al.*, 2015:44).

### **2.5.1 Clinical diagnosis**

The infecting *Leishmania* species can be identified and confirmed by firstly examining patients for symptoms (previously discussed in section 2.4) and thereafter, further testing can be done in the laboratory by screening samples for either a parasitological or serological diagnosis (Torres-Guerrero *et al.*, 2017:10). This is explained in detail below.

### **2.5.2 Parasitological diagnosis**

Parasitological diagnosis is used to detect the parasites' amastigote forms. A sample is taken by performing a biopsy of lesions or by exploratory puncture of the liver, spleen, bone marrow and lymph nodes. The collected samples are then stained using Giemsa and cytologically examined by impressing them into the slides (Camargo & Langoni, 2006:534). In some cases, when it is possible, the samples are cultured in Novy-McNeal-Nicolle (NNN) or another appropriate medium so that the tissue amastigotes are converted to motile promastigotes and can multiply *in vitro* (Georgiadou *et al.*, 2015:45). The cultures are checked weekly for about 4 weeks, with growth taking 2 weeks on average. The culture sensitivity depends on the parasite load but is estimated to be between 60-85% (Georgiadou *et al.*, 2015:45).

### **2.5.3 Serological diagnosis**

Several different techniques can be used for serological diagnosis such as indirect fluorescent antibody tests (IFA), enzyme-linked immunosorbent assays (ELISAs) and direct agglutination test (DAT). However, all of them differ in specificity, sensitivity, field conditions, practical application and availability of reagents (Camargo & Langoni, 2006:535).

IFA consists of an antigen-antibody reaction that is visualised through the use of a fluorescent conjugate. The results of the test are expressed in dilutions (Rodrigues *et al.*, 2021:112). A study conducted on immunodiagnosis of CL showed that IFA has a sensitivity rating of 91.6% and a specificity rating of 81.0% (Sarkari *et al.*, 2014:3).

ELISA tests allow for the detection of low antibody titers (Rodrigues *et al.*, 2021:113). ELISAs are based on the bond between the antibody and the antigen. It is identified through the use of a conjugate that is labelled with an enzyme. The enzyme acts on a substrate and this reaction causes a colour change of the chromogen reagent (Rodrigues *et al.*, 2021:113). The sensitivity of ELISA for VL was found to be 83.6%, while the specificity was found to be 90.5% (Mikaeili *et al.*, 2007:116).

DAT is a semi-quantitative test that is usually used to diagnose VL and can be done in either serum, blood or urine. It involves a reaction between the antibodies and the antigens (dead promastigotes), where a visible agglutination reaction is generated after an incubation period

(Rodrigues *et al.*, 2021:112). This test is easily readable and requires less equipment than ELISA, and can therefore be applied in low-resource settings (Georgiadou *et al.*, 2015:45). The sensitivity of DAT for VL is 70.5%, while the specificity is 100.0% (Mikaeili *et al.*, 2007:116).

These tests are not definitive proof of active disease as patients can test positive for serum antibodies against *Leishmania* spp. for months or years after recovery. These results must, therefore, be interpreted with caution and both clinical and epidemiological information must also be taken into account (Georgiadou *et al.*, 2015:45).

## 2.6 Prevention

The most economical strategy for the prevention of infectious diseases is vaccination. Vaccines with induction immune responses can move toward the immunity pathway to control infections by producing memory lymphocytes, while also stimulating cellular and humoral immunity (Ghorbani & Farhoudi, 2018:30). To date, there are no existing vaccines for leishmaniasis; however, there are several undergoing studies in an effort to find one (U.S. National Library of Medicine, 2021).

In the meantime, vector control remains a highly effective preventative measure when thoroughly applied and sustained (Wilson *et al.*, 2020:3). Eradicating the vector as a means of prevention can be done by using insecticide-impregnated thick clothes with long sleeves, curtains and/or bed nets, eliminating stagnant water, and using insect repellents (Eiras *et al.*, 2015:53; Torres-Guerrero *et al.*, 2017:11). Another preventative measure is to avoid outdoor activities in endemic areas, especially at night (Eiras *et al.*, 2015:53; Torres-Guerrero *et al.*, 2017:11).

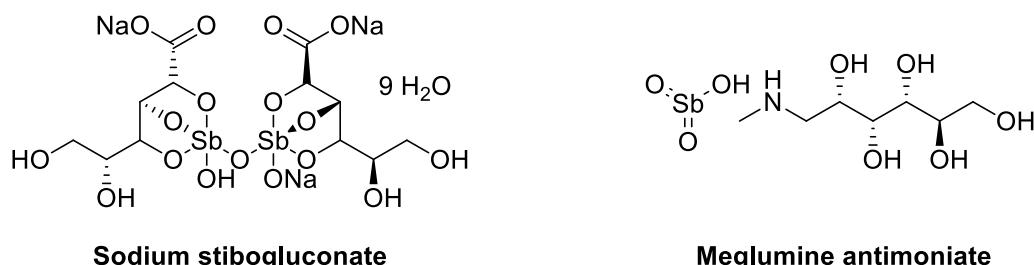
## 2.7 Treatment of leishmaniasis

The currently available treatment for leishmaniasis includes pentavalent antimonials (sodium stibogluconate and meglumine antimonate), amphotericin B, miltefosine, pentamidine, paromomycin and the azoles (ketoconazole, itraconazole and fluconazole) (Georgiadou *et al.*, 2015:46; Braga, 2019:2). None of these drugs are ideal treatments for leishmaniasis as they require intravenous administration, are highly expensive, toxic, and are experiencing parasitic resistance (Freitas-Junior *et al.*, 2012:12). These drugs are discussed more in detail below.

### 2.7.1 Pentavalent antimonials

Pentavalent antimonials are antimony-carbohydrate complexes (Figure 2-5) that have been used for more than half a century to treat leishmaniasis (Pund & Joshi, 2017:449). There is evidence to suggest that pentavalent antimonials are prodrugs, and that they are converted within the organism to trivalent antimonials and become biologically active once in the bloodstream. However, the exact reduction site and the underlying reduction mechanisms is unknown (Domingues Passero *et al.*, 2013:96). Pentavalent antimonials can be used to treat all forms of

leishmaniasis and are administered at 20 mg/kg/day for 28-30 days IV/IM for the VL form; 10-20 days for the CL form and 30 days for the MCL form (McGwire & Satoskar, 2014:11; Eiras *et al.*, 2015:55).



**Figure 2-5:** Sodium stibogluconate and meglumine antimoniate for the treatment of leishmaniasis

The antimonials have a short half-life of 1-2 hours and a terminal elimination phase of 1-3 days where they are excreted in the urine (Aronson, 2016:619; De Aguiar *et al.*, 2018:1114). General side effects of the pentavalent antimonials include nausea, vomiting, anorexia, abdominal pain, and headaches (WHO, 2010; Aronson, 2016:619). In the case of sodium stibogluconate, additional side effects such as serum transaminase elevations, pancreatitis, T-wave inversion or flattening, and rarely, myocardial, hepatic, and renal damage, may also be observed (Ryan, 2018:1568).

### 2.7.2 Amphotericin B

Amphotericin B (AmB) is a polyene macrolide antibiotic isolated from the *Streptomyces nodosus* strain (Pund & Joshi, 2017:449) (Figure 2-6). This drug is used to treat patients with HIV-leishmaniasis coinfection or when parasites are resistant to pentavalent antimonials (Braga, 2019:3). AmB forms complexes with 24-substituted sterols, such as ergosterol, on the biological membrane. The formed complexes cause pores to open, which then alter the ion balance and kill the cell (Roberts *et al.*, 2003:137; Freitas-Junior *et al.*, 2012:12). AmB deoxycholate is used to treat all forms of leishmaniasis and is administered intravenously at doses of 0.75-1.00 mg/kg/day for 15-20 alternate days for the treatment of VL caused by *L. donovani*, and for 30 days for VL caused by *L. infantum* (McGwire & Satoskar, 2014:11). For CL caused by *L. braziliensis*, the therapy length is 25-30 days, and 45 days for MCL (McGwire & Satoskar, 2014:11).

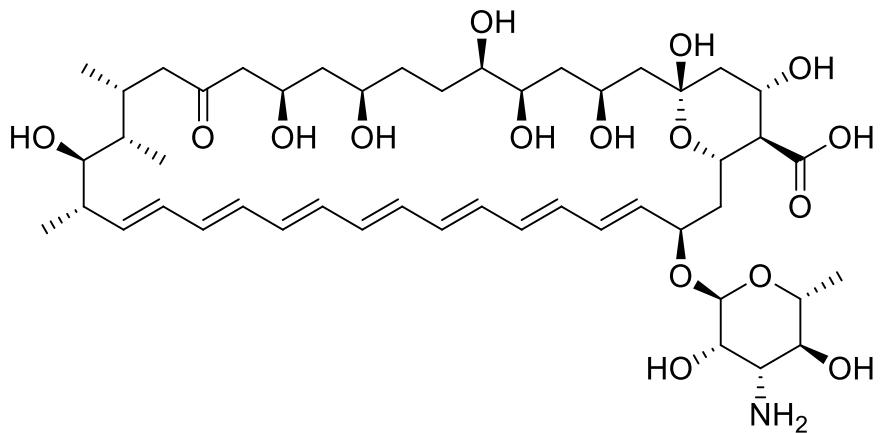


Figure 2-6: Amphotericin B for the treatment of leishmaniasis

The terminal half-life of AmB is 127 hours ( $t_{1/2} \sim 5$  days) in healthy subjects, with the drugs being extensively excreted unchanged in the urine and faeces (Bekersky *et al.*, 2002:833). AmB is generally well tolerated, although adverse effects may include infusion-related chills, high fever, and rigor, and in some cases, it can also cause thrombophlebitis of the injected vein, nephrotoxicity, and electrolyte abnormalities. Hypokalaemia and myocarditis are also uncommon, but can occur in extreme circumstances (WHO, 2010; McGwire & Satoskar, 2014:11).

In order to improve the bioavailability and pharmacokinetic properties of AmB, lipid formulations were developed. Liposomal AmB (L-AmB) has reduced side effects (mild fever, rigor, chills, and back pain, as well as transient nephrotoxicity and thrombocytopenia (WHO, 2010)) and the smaller liposomes stay in the bloodstream longer than free AmB. Therefore, it has a better half-life of 152 hours ( $t_{1/2} \sim 6$  days) and a high efficacy (Bekersky *et al.*, 2002:828; Freitas-Junior *et al.*, 2012:12). L-AmB is administered at 3 mg/kg/day IV on days 1-5, 14 and 21 (Georgiadou *et al.*, 2015:46).

### 2.7.3 Miltefosine

Miltefosine is the only oral drug currently available and is used to treat all forms of leishmaniasis. It is a phosphocholine derivative (Figure 2-7) that has been shown to act as an inhibitor of lipid biosynthesis in parasites of the *Kinetoplastida* order (Pinto-Martinez *et al.*, 2017:2). Indirectly, miltefosine causes cell death *via* a mechanism similar to apoptosis (Braga, 2019:4). It reduces the function of the overall replication mechanism of parasite cells and targets the homeostasis of  $\text{Ca}^{2+}$  in *Leishmania* parasites. This contributes to parasitic cell death through the eventual over accumulation of intracellular  $\text{Ca}^{2+}$  (Braga, 2019:4). Miltefosine is able to inhibit cytochrome-c oxidase in the mitochondria, causing an apoptosis-like cell death. Miltefosine has been reported to possess immunomodulatory properties that is believed to be another contributing factor to its antileishmanial activity (Dorlo *et al.*, 2012:2583). For VL, miltefosine is given orally at

2.5 mg/kg/day for children ≤12 years old, and at 50.0 mg/kg/day for individuals >12 years and <25kg. For a weight of 25-50 kg, 100.0 mg/kg/day is substantial, and when the weight is >50 kg, 150.0 mg/kg/day is given, all for a total of 28 days (McGwire & Satoskar, 2014:12). To treat CL caused by *L. panamensis*, miltefosine is given at 2.0 mg/kg/day for 28 days, and for MCL caused by *L. braziliensis*, it is administered at 2.5-3.3 mg/kg/day for 4 weeks (WHO, 2010).

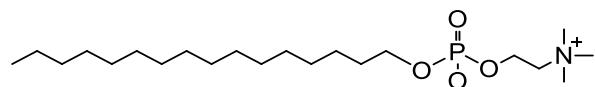


Figure 2-7: Miltefosine for the treatment of leishmaniasis

Preclinical *in vitro* studies have shown no oxidative metabolisms by isoenzyme cytochrome P450, thus the main metabolic pathway of miltefosine appears to be exertion through phospholipases (Ghorbani & Farhoudi, 2018:30). Miltefosine has a long terminal elimination phase and a half-life of 31 days (Kip *et al.*, 2018:166). Miltefosine is well tolerated but is teratogenic (Georgiadou *et al.*, 2015:47) and causes symptoms such as headaches, nausea, vomiting, and/or diarrhoea during the first week of treatment (Eiras *et al.*, 2015:56).

#### 2.7.4 Pentamidine

Pentamidine is an aromatic diamidine (Figure 2-8) that was previously used as a second-line treatment for VL in India, but due to a reduction in efficacy caused by parasitic resistance, it is no longer used to treat VL. It is, however, still an effective treatment for CL caused by *L. guyanensis* and *L. panamensis* in the Northern Amazon basin and it is administered as short courses, with four doses of 4 mg/kg IM or slow IV injections being given on alternate days (Boelaert & Sundar, 2014:647). The mechanism of action of pentamidine is poorly understood, but it is believed to undergo rapid active uptake into the cell of the protozoan parasite, where it binds to the transfer RNA. This binding inhibits the ribosomal protein synthesis process and the synthesis of nucleic acids and phospholipids (Rex & Stevens, 2015:492; Waller & Sampson, 2018:615). The accumulation of the drug in high concentrations via several transporter systems may contribute to the selectivity of pentamidine to the protozoan parasites. However, the loss of uptake of pentamidine into parasite cells can also contribute to resistance (Barrett & Croft, 2012:187).

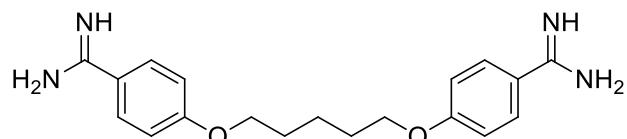


Figure 2-8: Pentamidine used for the treatment of *T.b. gambiense*

Pentamidine is distributed mainly to the liver and kidneys, where it is excreted slowly without undergoing biotransformation and remains unchanged in the faeces and urine (Rex & Stevens, 2015:492). The half-life of pentamidine is between 10 and 14 days (Waller & Sampson, 2018:615). Pentamidine is relatively well tolerated in the body; however, it still has several adverse effects that include hypotension, hypoglycaemia or hyperglycaemia, nephrotoxicity, prolongation of the QT interval, and gastrointestinal effects. In rare occasions, more severe side effects such as highly abnormal hepatic and pancreatic functions, and Stevens-Johnson syndrome, may also be observed (Babokhov *et al.*, 2013:244; Kennedy, 2013:191).

### **2.7.5 Paromomycin**

Paromomycin (previously known as aminosidine) is a broad-spectrum antibiotic belonging to the aminoglycosides class (Boelaert & Sundar, 2014:647) (Figure 2-9). Its antileishmanial activity is believed to be mediated by inhibiting parasite metabolism and mitochondrial respiration (McCarthy *et al.*, 2015:512). It is also suggested that paromomycin inhibits the protein synthesis of protozoa by binding to the 30S ribosomal subunit, which causes accumulation of abnormal 30S-50S ribosomal complexes, eventually causing cell death (Kip *et al.*, 2018:164). Paromomycin is administered at 15 mg/kg/day for 21 days as a monotherapy for Indian VL. It can also be administered as a combinational therapy with L-AmB (McGwire & Satoskar, 2014:11).

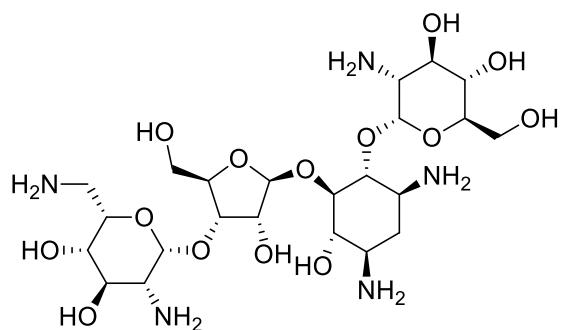


Figure 2-9: Paromomycin for the treatment of VL and CL

Paromomycin is primarily excreted unchanged by the kidneys through glomerular filtration and is therefore not metabolised. The half-life of paromomycin is 2-3 hours (Kip *et al.*, 2018:164). Side-effects of paromomycin include nephrotoxicity (renal injury), ototoxicity, and hepatotoxicity (LFTs increase) (Freitas-Junior *et al.*, 2012:12; Georgiadou *et al.*, 2015:47).

### **2.7.6 Azoles**

Ketoconazole, fluconazole and itraconazole are derived from the family of azoles and can be used to treat all forms of leishmaniasis (Figure 2-10). These drugs all share the same molecular target, the enzyme lanosterol 14- $\alpha$ -demethylase, which is involved in the biosynthesis of the main cell membrane sterol, ergosterol, in yeasts (Braga, 2019:2). *Leishmania* spp. cells have ergosterol as

the main sterol component in their membranes, thus the depletion of ergosterol by these azoles causes the parasite's cells to undergo a number of structural changes that results in the inhibition of cell growth (Braga, 2019:3). An oral dose of 200 mg/day of fluconazole is recommended by the WHO for a total of 6 weeks for treating VL caused by *L. major* (WHO, 2020). WHO also recommends 600 mg/day of ketoconazole for 28 days for treating VL caused by *L. mexicana* (WHO, 2020). Itraconazole was found to be effective to treat CL at a dose of 200 mg/day for 3 months, after which the dose was reduced to 100 mg/day for another month (Consigli *et al.*, 2006:46). A study was conducted by Amato *et al.* (2000) where they tested the efficacy of itraconazole in MCL. In this study, ten patients were given 4 mg/kg/day (up to 400 mg/day) for 6 weeks. Six out of the ten patients had healed lesions 3 months after treatment, thus concluding that itraconazole can be effective against MCL (Amato *et al.*, 2000:153).

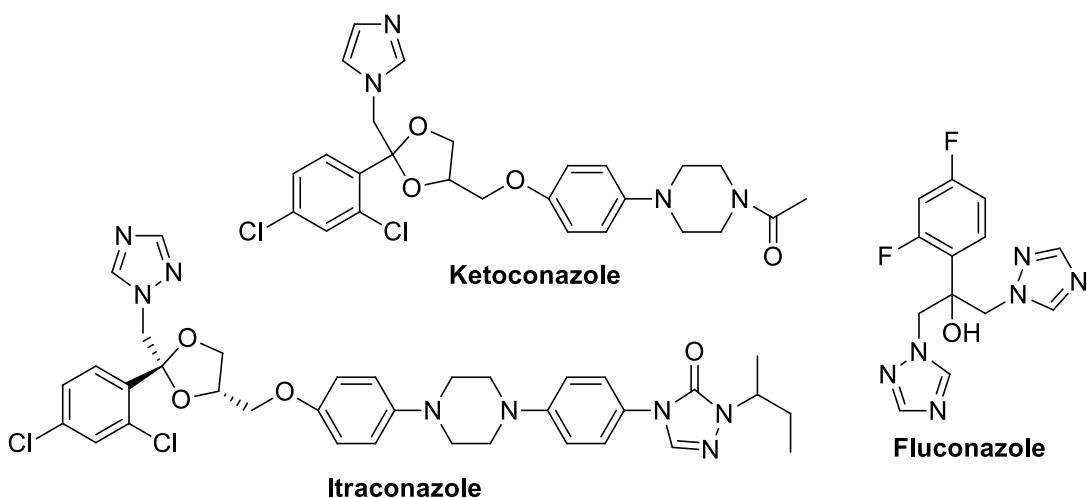


Figure 2-10: Ketoconazole, itraconazole, and fluconazole for the treatment of leishmaniasis

The elimination half-life of fluconazole is 31.6 hours (Debruyne & Ryckelynck, 1993:10), 7.5-7.9 hours for ketoconazole (Huang *et al.*, 1986:206), and 34.0 hours for itraconazole (Prentice & Glasmacher, 2005:i18). Common side-effects of the azoles include gastrointestinal effects and hepatotoxicity (Eiras *et al.*, 2015:57).

## 2.8 Drug resistance

The interaction of drugs with one or more targets is the primary effect in cell destruction. Therefore, for a parasite's survival, the altering of intracellular drug levels or the drug's ability to affect targets are of great importance. A variety of mechanisms can be used to lower drug levels at the target action site, including increased export, decreased uptake, and inactivation by sequestration or metabolism. Inhibition of primary targets often trigger cell damage and death (Croft *et al.*, 2006:114). Many anti-parasitic drugs undergo redox reactions, which produce reactive oxygen species (ROS), which in turn can result in peroxidative damage of membrane

proteins, lipids or DNA and ultimately parasite death. Therefore, overexpressed repair systems can contribute to drug resistance *via* multiple mechanisms (Croft *et al.*, 2006:114). Discussed below are the most common resistance mechanisms of *Leishmania* parasites to the previously discussed antileishmanial drugs.

### **2.8.1 Pentavalent antimonials**

Drug resistance to pentavalent antimonials was first indicated in the early 1980s in North Bihar, India, when 30% of patients were found to not respond to the administered drugs (Chakravarty & Sundar, 2010:168). This number has drastically increased over the years and today, up to 60% of patients do not respond to the pentavalent antimony drugs in North Bihar (India) (Maltezou, 2010:2). Despite several studies being done on *in vitro* parasite mutants, resistance mechanisms in field parasites are not yet fully understood. This is because of the differences between *in vitro* and field parasite antimicrobial resistance mechanisms, and also because *in vitro* unresponsiveness does not necessarily mean clinical resistance (Maltezou, 2010:2). Antimony is a heavy metal that shares characteristics with the heavy metal arsenic, and thus arsenic resistance renders cross-resistance with antimonials. This offers a possible reason for antimony resistance in North Eastern India, as high levels of arsenic in drinking water may have led to parasites becoming resistant to arsenic and therefore to antimonials (Ponte-Sucre *et al.*, 2017:6). Another setback for antimony treatment is HIV-VL coinfecting patients, as antimonials need an immune system that is intact to be effective and these patients tend to relapse. Thus, HIV-VL coinfecting patients serve as a potential source to increase drug resistance (Chakravarty & Sundar, 2010:169). *Leishmania* strains that have shown resistance to antimonials include *L. donovani* (Mukhopadhyay *et al.*, 2011:1311; Zheng *et al.*, 2020:11), *L. infantum* (Brotherton *et al.*, 2013:1; Seblova *et al.*, 2014:6274), *L. braziliensis* (Rojas *et al.*, 2006:1378), *L. panamensis* (Rojas *et al.*, 2006:1378) and *L. tropica* (Hadighi *et al.*, 2007:1319).

### **2.8.2 Amphotericin B**

Clinical isolates of *L. donovani* have been found to be resistant to AmB. *Leishmania* strains that are AmB-resistant have shown significant reduction in ROS accumulation, leading to an increase in the tryparedoxin cascade that could play a key role in kinetoplastides' antioxidative defence against ROS (Ghorbani & Farhoudi, 2018:29). The AmB-R strain of *L. infantum* can upregulate members of the tryparedoxin cascade, and their resistance to AmB is connected to ATP depletion, followed by increased ion leakage. Thus, by preventing ion leakage, ATP depletion could be inhibited, and AmB-R strains could possibly be controlled (Ghorbani & Farhoudi, 2018:29).

Previously, it was assumed that clinical resistance of AmB was rare, but the overuse of this drug has now caused it to experience parasitic resistance. This was confirmed in a recent study by Kariyawasam *et al.* (2019) where they found resistance of AmB at 0.12 µg/mL by *L. donovani*,

*L. infantum*, *L. tropica*, *L. major*, *L. amazonensis*, *L. chagasi*, *L. mexicana*, *L. braziliensis* and *L. guyanensis*.

### **2.8.3 Miltefosine**

Promastigotes that are resistant to miltefosine can be created in a laboratory by chemically mutating the parasites' proteins (Coelho *et al.*, 2012:1). Mutations in the MT gene causes imperfections in drug internalisation, which in turn is considered to be the mechanism of resistance (Ghorbani & Farhoudi, 2018:30). The reduction in internalisation is mainly due to reduced uptake or increased efflux of miltefosine. The addition of inactivating mutations or deletions in the translocation apparatus LMT and/or LRos3 in miltefosine showed drastic increase in miltefosine resistance in both *in vitro* and *in vivo* *L. donovani* assays (Ponte-Sucre *et al.*, 2017:12).

Resistance to miltefosine has been reported in *L. donovani* (Srivastava *et al.*, 2017:9) and *L. infantum* (Van Bockstal *et al.*, 2020:16).

## **2.9 Rationale for new antileishmanial drug development**

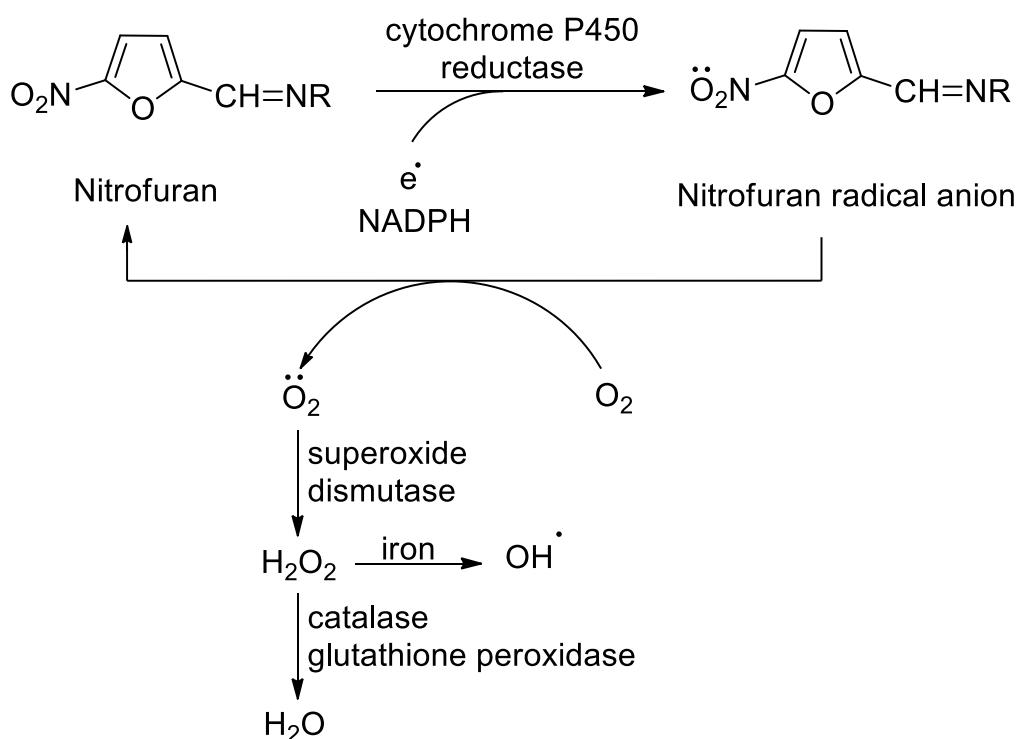
The pathogens of infectious diseases are able to survive in human hosts by multiplying and adapting. This leads to the generation of new strains of parasites, thus causing existing drugs to become less effective for the treatment of these diseases.

In the search for new, safe and effective antileishmanial drugs, extensive research on several scaffolds such as azoles (Mathur *et al.*, 2020:2), triazoles (Teixeira de Macedo Silva *et al.*, 2018:2360), nitrofurans (Sifontes-Rodríguez *et al.*, 2015:166), flavonoids (Boniface & Ferreira, 2019:2473), and various others have been explored. The nitrofuran moiety is of particular interest of all these scaffolds, as it is redox-active, and there are several reports indicating its broad spectrum of activity (Wyllie *et al.*, 2016:1; Foroumadi *et al.*, 2017:904). Also, nifurtimox (a nitrofuran drug) and fexinidazole (a nitroimidazole drug) are currently used for treating human African trypanosomiasis (HAT) (a disease caused by a taxonomically related protozoan parasite) and both these drugs are orally available, hence it may be a suitable backbone to use in the search for new drugs.

### **2.9.1 Nitrofuran**

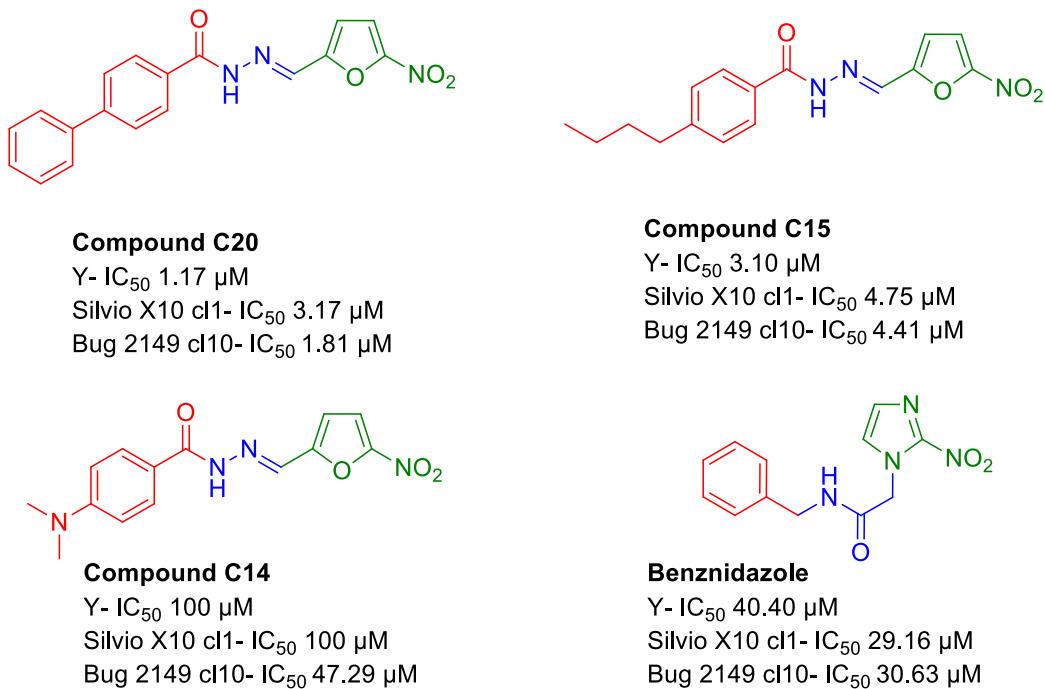
Nitrofurans are a class of redox-active anti-infective drugs that are used for the treatment of various infectious diseases. Structurally, nitrofurans contain a furan ring with a nitro group. These drugs have a broad range of biological activity against various diseases such as cancer (Peterson, 2018:1439), tuberculosis (Krasavin *et al.*, 2019:130), giardiasis (Ryan, 2018:1567) and urinary tract infections (Huttner *et al.*, 2015:2456). Drugs that contain the nitro group are very active

because the nitro group itself produces free radical species, reactive oxygen species (ROS) and reactive nitrogen species (RNS), which react with pathogen cell wall enzymes, thereby becoming lethal to these microorganisms (Pal & Bandyopadhyay, 2012:556; Kim *et al.*, 2019:203). The generation of ROS (such as  $O_2^-$ ,  $H_2O_2$  and OH) are induced by nitrofurans, and the reduction of the nitro group is catalysed by a flavin containing nitroreductase, such as NADPH or NADH (Sharma & Anand, 1997:432; Le *et al.*, 2019:2). Furthermore, nitrofurans possess several other mechanisms of antimicrobial action, which may contribute to the lack of pathogenic resistance against such therapeutic agents.



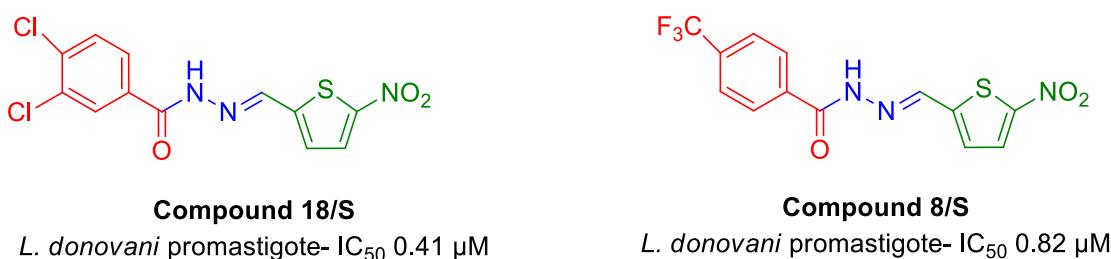
Scheme 2-1: Mechanism of action of nitrofuran drugs. Adapted from Wang *et al.* (2008)

There have been extensive studies reported on nitrofuran drugs as anti-infective drugs. In a recent study by Palace-Berl *et al.* (2015), a series of 5-nitro-2-furfurilidene derivatives were synthesised and tested for its anti-*T. cruzi* activity. Their results have indicated that all synthesised compounds, with the exception of compound C14, showed similar or better activity than the reference drug benznidazole. Compounds C20 and C15 (Figure 2-11) were the most potent against the epimastigote form of three *T. cruzi* strains. Compound C20 showed activity of 1.17  $\mu M$  (Y-strain), 3.17  $\mu M$  (Silvio X10 cl1 strain), and 1.81  $\mu M$  (Bug 2149 cl10 strain). Compound C15 showed activity of 3.10  $\mu M$  (Y-strain), 4.75  $\mu M$  (Silvio X10 cl1 strain), and 4.41  $\mu M$  (Bug 2149 cl10 strain).



**Figure 2-11:** The two most active compounds, compound C20 and C15, as well as the least active compound, compound C14, and reference drug, benznidazole (Palace-Berl *et al.*, 2015)

In another study, the antileishmanial activity of 5-nitro-2-heterocyclic benzylidene hydrazides was investigated. It has been found that compounds 2/S, 6/S, 8/S, 18/S, 18/O, and 28/O showed better activity than the drugs pentamidine (IC<sub>50</sub> 1.06 µM) and AmB (IC<sub>50</sub> 1.19 µM). The most potent compounds of this series were compounds 18/S and 8/S (Figure 2-12) showing IC<sub>50</sub> values of 0.41 µM and 0.82 µM, respectively, against *L. donovani* promastigotes (Rando *et al.*, 2008).



**Figure 2-12:** Two analogues, 18/S and 8/S, with promising anti-promastigote activity (Rando *et al.*, 2008)

### 2.9.2 Hydrazone and azine

In addition to the nitrofuran moiety, clinical nitrofurans have a hydrazone moiety (=C-NH-NH<sub>2</sub>) that possesses diverse biological and pharmacological properties, such as anti-cancer (Iliev *et al.*, 2019:756), anti-inflammatory (Kumar *et al.*, 2015:2580), anti-protozoal (Rillas & Küçükgüzel,

2007:1919), antibacterial (Popiołek *et al.*, 2020:260), anti-trypanosomal (Glinma *et al.*, 2015:001), and antileishmanial (Vargas *et al.*, 2018:1) activity. Research has indicated that the combination of hydrazones with other biologically active functional groups results in compounds with unique chemical and physical character (Verma *et al.*, 2014:69).

The exact mechanism of action of hydrazones are not yet fully understood. However, it is theorised that hydrazones are inhibitors of parasite cysteine proteases (Vargas *et al.*, 2018:2), which are hydrolase enzymes that degrade proteins. In parasitic diseases, cysteine proteases are involved in hemoglobin degradation, surface protein processing, and parasite egress (Verma *et al.*, 2016:1). Therefore, this makes promising targets for drug development.

A study conducted by Coimbra *et al.* (2018) investigated the antileishmanial activity of hydrazone-containing compounds against the promastigote and amastigote forms of several *Leishmania* strains. Their study found a few active compounds, with their best two compounds (figure 2-13) showing better activity against the amastigote form of *L. amazonensis* (IFLA/BR/1967/PH8 strain) than their reference drug, miltefosine ( $IC_{50}$  12.74  $\mu$ M). Compounds 2s and 2c showed  $IC_{50}$  values of 11.09  $\mu$ M and 11.98  $\mu$ M, respectively (Coimbra *et al.*, 2018).

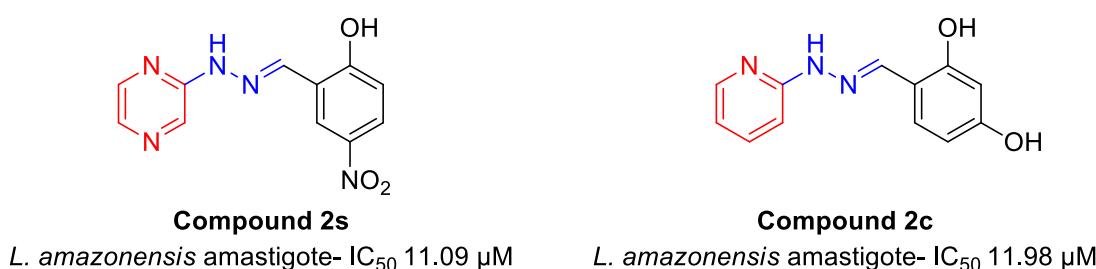


Figure 2-13: Compounds 2s and 2c, with the best antileishmanial activity (Coimbra *et al.*, 2018)

Another study conducted by Coimbra *et al.* (2019) also investigated the antileishmanial activity of hydrazone-containing compounds against the promastigote and amastigote forms of *L. amazonensis* (IFLA/BR/67/PH8 strain). Their study showed two compounds (figure 2-14) with fairly good anti-amastigote activity. Compounds 6i and 6h showed  $IC_{50}$  values of 8.41  $\mu$ M and 10.8  $\mu$ M, respectively (Coimbra *et al.*, 2019).

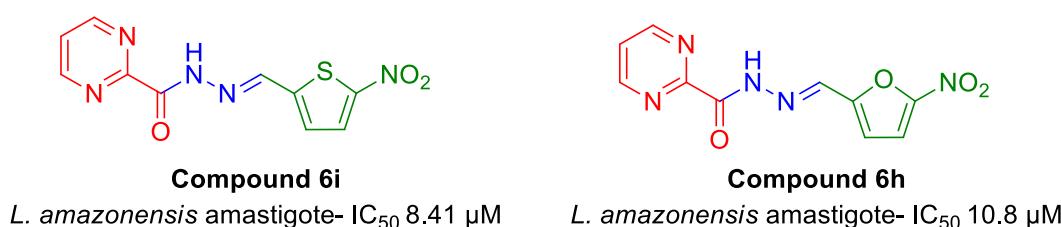


Figure 2-14: Compounds 6i and 6h, with the best anti-amastigote activity (Coimbra *et al.*, 2019)

Azines are organic molecules which bear the C=N–N=C functional unit. The azine moiety consists of two imine bonds groups joined by a N–N single bond, that are polar acceptor groups orientated in opposite directions. Azine compounds have been receiving increasing attention because of their chemical, physical, and biological properties (Singh & Pandey, 2009:57). They are thermally and chemically more stable than their hydrazone tautomers. Some azines are also redox-active (Chourasiya *et al.*, 2019:8487). The attachment of the azine moiety to the nitrofuran pharmacophore may lead to derivatives with enhanced hydrophilicity, chemical and thermally stability, and possibly increased biological activity.

### 2.9.3 Benzene ring

Aromatic rings are extensively used in drug design due to their well-known synthetic and modification paths (Aldeghi *et al.*, 2014:450). A benzene ring introduces chemical stability, diverse reactivity paths and possible enhancement of biological activity. In this study, the aromatic moiety (a benzene ring) will be incorporated into nitrofurylazine through the use of aromatic aldehydes. Aldehydes contain a carbonyl group which is a mild electrophile with increased reaction rate that is of more interest to use in biological systems upon interaction with strong nucleophiles such as hydrazine. This causes an equilibrium shift that favours the formation of hydrazone (Fernandes *et al.*, 2018:905).

In this study, nitrofuran-based azine derivatives will be synthesised, characterised and evaluated for its antileishmanial activity. The synthesised derivatives will consist of three active pharmacophores, namely the nitrofuran, the hydrazone and the aromatic ring (figure 2-15). As already discussed (section 2.9.1), nitrofuran possesses diverse biological and pharmacological properties and it is redox-active. The addition of the azine may improve solubility, thermal and chemical stability, as well as redox activity, and the addition of a benzene ring may further enhance biological activity. Thus, the combination of these three pharmacophores may form biologically active anti-parasitic agent with good stability, efficacy and long half-life.

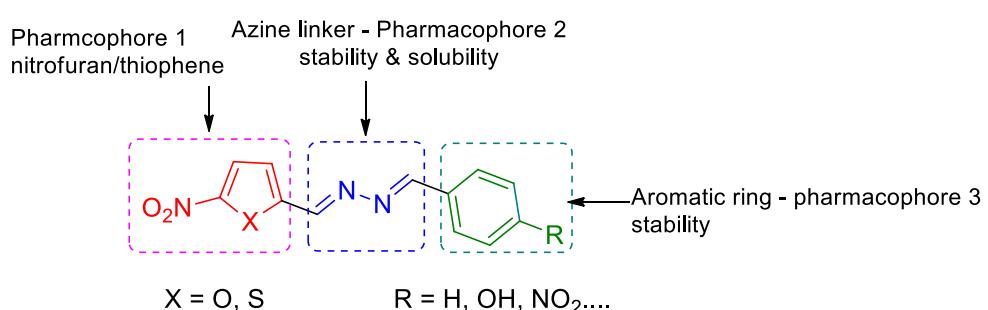


Figure 2-15: Compound design model of the study

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## **CHAPTER 3**

### **ARTICLE FOR SUBMISSION**

Chapter 3 contains the article entitled “Synthesis and *in vitro* antileishmanial activities of nitrofuran-based azine derivatives” which will be submitted to European Journal of Pharmaceutical Sciences. This article contains the biological findings of the synthesised antileishmanial compounds of the study.

The manuscript was prepared according to the Author Guidelines of the journal (see Annexure C).

# **Synthesis and *in vitro* antileishmanial activities of nitrofuran-based azine derivatives**

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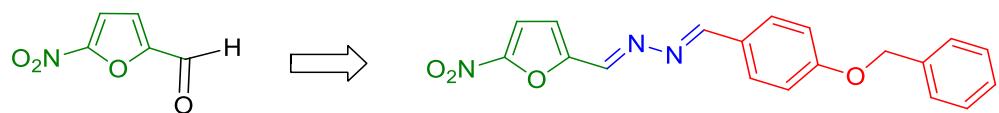
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***The synthesis was conducted by Ms M Viljoen under the guidance of Dr C Kannigadu and Prof DD N'Da. The cytotoxicity and antileishmanial activity properties were determined by Ms M Viljoen and Dr J Aucamp.***

## HIGHLIGHTS

- Synthesis of 2 sub-series of nitrofuran-based azine derivatives
- *In vitro* evaluation of antileishmanial activity and cytotoxicity
- Overall good activity of derivatives against various *Leishmania* promastigotes
- Moderate to non-cytotoxic derivatives against Vero cells
- Activity with IC<sub>50</sub> values 0.34 µM - 0.70 µM were reported for the most active compound **(7a)** against all parasite strains

## GRAPHICAL ABSTRACT



5-Nitro-2-furaldehyde, NFA

*L. donovani* 1S IC<sub>50</sub> 11.06 μM  
*L. donovani* 9515 IC<sub>50</sub> 13.71 μM  
*L. major* IR-173 IC<sub>50</sub> 7.94 μM  
LogP<sub>o/w</sub> 0.36  
Vero IC<sub>50</sub> 17.82 μM

Compound 7a

*L. donovani* 1S IC<sub>50</sub> 0.49 μM  
*L. donovani* 9515 IC<sub>50</sub> 1.61 μM  
*L. major* IR-173 IC<sub>50</sub> 0.45 μM  
LogP<sub>o/w</sub> 3.43  
Vero IC<sub>50</sub> >100 μM

## ABSTRACT

Leishmaniasis is a vector-borne infectious disease affecting both humans and animals. This neglected tropical disease can be fatal if not treated. Hundreds of thousands of new leishmaniasis cases are being reported by the WHO every year, and currently available treatments are unsatisfactory. Severe adverse effects, short half-lives and increased pathogen resistance against current clinical treatments emphasise the urgent need for the development of new antileishmanial drugs. In search for such drugs, we investigate a series of nitrofuran-based azine derivatives. Herein, we report the design, synthesis and biological activity of the derivatives against promastigotes of *Leishmania donovani* (1S and 9515), and *L. major* IR-173 strains. Compounds **3a**, **5a** and **7a** featuring chloro, methyl and benzyl substituents, respectively, were the most active against the tested *Leishmania* strains, possessing nanomolar activities up to 18-fold higher than the parent compound 5-nitro-2-furaldehyde. These three compounds were identified as anti-promastigote hits for further investigation into novel antileishmanial agents.

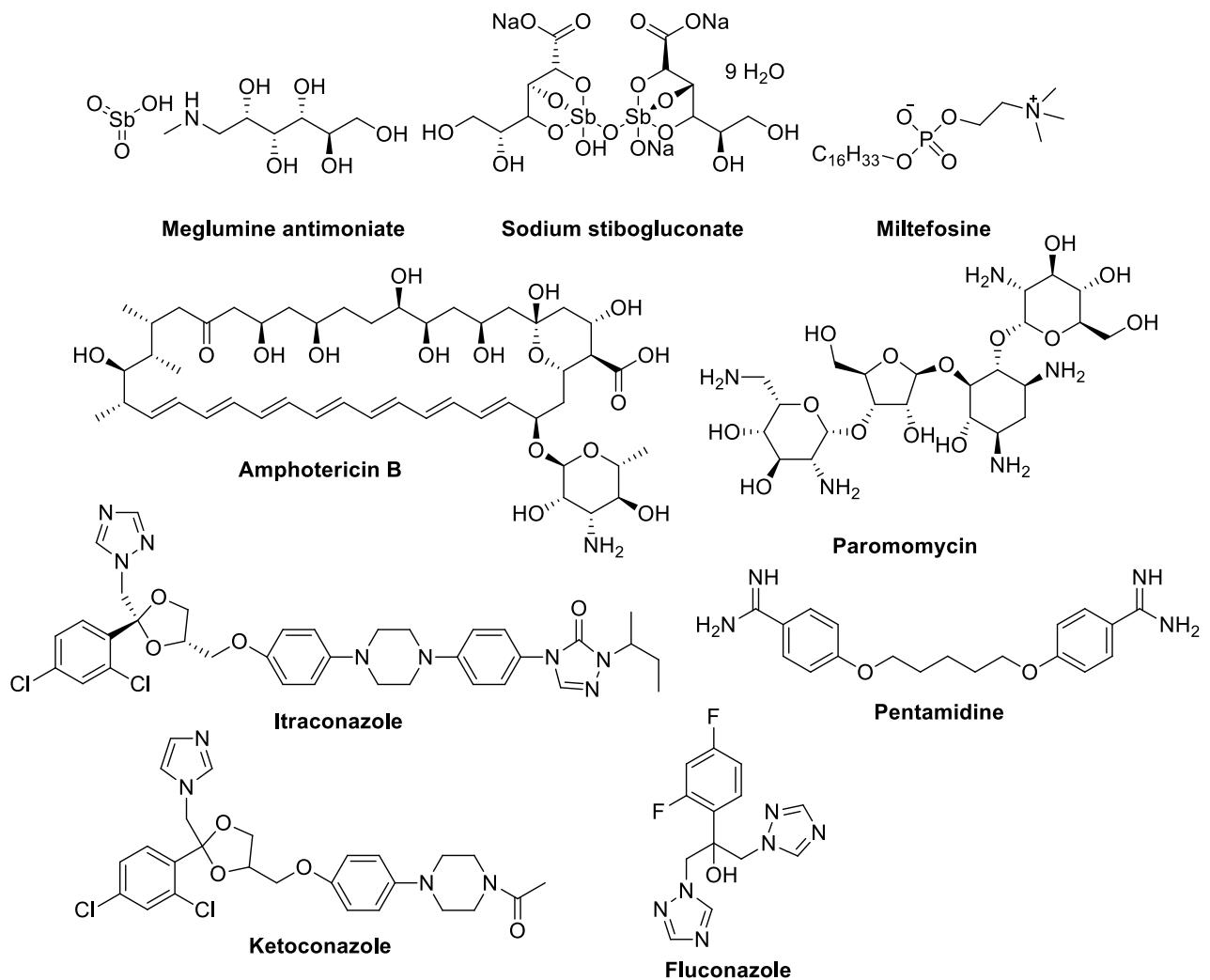
**Keywords:** *Leishmania*, promastigote, nitrofuran, thiophene, hydrazone, azine

## 1. Introduction

Leishmaniasis is a neglected tropical disease (NTD) caused by the *Leishmania* parasite and it is transmitted to humans during hematophagy of an infected female *Phlebotomus* sandfly (WHO, 2021a). There are over twenty species that can affect humans and three clinical forms in which the disease can manifest itself in (Salam et al., 2014). These forms are cutaneous (CL), which can be identified by lesions on the skin; mucocutaneous (MCL), a differentiated form with lesions on the mouth and skin; and visceral leishmaniasis (VL), identified by skin lesions and a systemic infection that results in an immune deficiency (Torres-Guerrero et al., 2017). VL is the most dangerous form, causing excessive internal damage with a mortality rate of 95% if left untreated (WHO, 2021b).

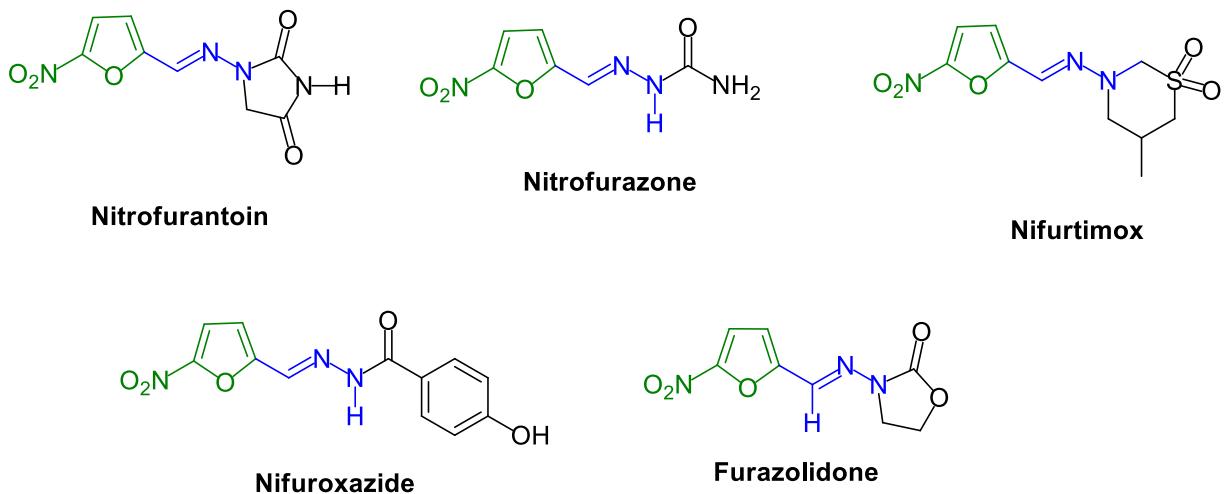
Leishmaniasis is closely associated with poor living conditions, malnutrition, weak immune systems and the lack of finances (WHO, 2021b). In 2019, the World Health Organisation (WHO) estimated 50,000 to 90,000 new cases of VL (mostly in Brazil, East Africa, and India) and 600,000 to 1 million new cases of CL (mostly in the Americas, the Mediterranean basin, the Middle East, and Central Asia) to have occurred worldwide (WHO, 2021b).

There is no vaccine for protection against this infection; however, various chemotherapeutic drugs, such as the pentavalent antimonials (sodium stibogluconate and meglumine antimonate), amphotericin B, miltefosine, pentamidine, paromomycin and the azoles (ketoconazole, itraconazole and fluconazole) (**Fig. 1**), are available for the treatment of leishmaniasis (Georgiadou et al., 2015; Ghorbani and Farhoudi, 2018; Braga, 2019). These drugs have several limitations, such as impractical administration, toxicity, cost, and parasite resistance (Freitas-Junior et al., 2012). Fight against the latter is an ongoing battle due to the misuse of drugs, financial constraints, and the unavailability of new drugs (Maltezou, 2010).



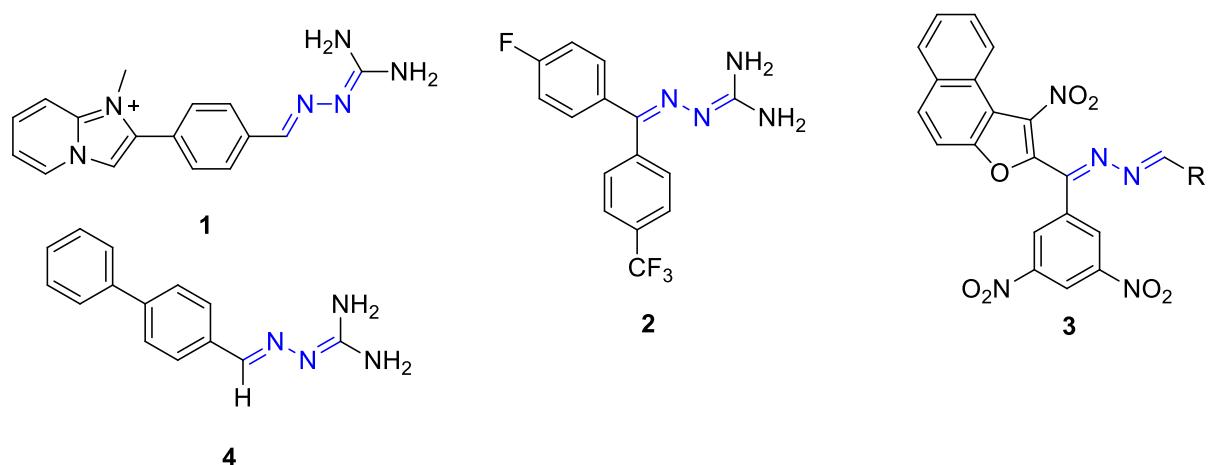
**Fig. 1:** Clinical drugs used for the treatment of leishmaniasis

The use of clinical nitrofurans (cNFs) to treat various infectious diseases has been well established over the recent years (Patterson and Wyllie, 2014; Zuma et al., 2019). The diverse biological activities of nitrofurans have been attributed to its nitro group and its ability to form reactive oxygen species (ROS) upon Fenton's reactions with enzymes (Pal and Bandyopadhyay, 2012; Patel et al., 2018). Additionally, cNFs (**Fig. 2**) also have a hydrazone moiety (blue in **Fig. 2**) that possess anti-infective activity therefore making them good candidates for drug repurposing for the treatment of infectious diseases such as leishmaniasis.



**Fig. 2:** Clinical nitrofuran drugs

The hydrazone moiety possesses diverse biological and pharmacological properties, such as anti-cancer (Sreenivasulu et al., 2019), anti-microbial (Manikandan et al., 2017), anti-trypanosomal (Glinma et al., 2015), anti-mycobacterial (Angelova et al., 2017), anti-convulsive (Dehestani et al., 2018), anti-inflammatory (Kajal et al., 2014) and analgesic (Navidpour et al., 2014) properties. Research has indicated that azines bearing the C=N—N=C functional unit are bactericidal and anti-parasitic (**Fig. 3**). These compounds are highly basic and form salts very easily, thus they are endowed with increased water solubility and become equivalent to hydrazones in the protonated state. They are also thermally and chemically more stable than their hydrazone tautomer's (Singh and Pandey, 2009; Chourasiya et al., 2019).



**Fig. 3:** Structures of biologically active azines where (1) trypanocidal agent (2) antimarial agent (3) and (4) antibacterial agent (Chourasiya et al., 2019)

Furthermore, the use of aromatic rings in drug design is a well-known strategy as they offer numerous unique and strong interaction modes with target proteins, which include classical

arene-arene ( $\pi$ -stacking) interactions, as well as arene-H bonding (edge-to-face interactions), and other interactions, such as p-cation stabilisation and sulphur-arene interactions (Ward and Beswick, 2014). Thus, the linking of nitrofuran, azines and aromatic rings may result in compounds that are more active and stable.

Based on the evidence, nitrofuran-based azine derivatives and their antileishmanial activities were examined *in vitro*. We herein report the synthesis and the biological activities of these analogues.

## 2. Materials and methods

### 2.1. Materials

Reagents hydrazine monohydrate ( $N_2H_4$ ), triethylamine ( $C_6H_{15}N$ ), benzaldehyde ( $C_7H_6O$ ), 4-nitrobenzaldehyde ( $C_7H_5NO_3$ ), 4-chlorobenzaldehyde ( $C_7H_5ClO$ ), 4-bromobenzaldehyde ( $C_7H_5BrO$ ), 4-methylbenzaldehyde ( $C_8H_8O$ ), 4-methoxybenzaldehyde ( $C_8H_8O_2$ ), 4-fluorobenzaldehyde ( $C_7H_5FO$ ), 4-hydroxybenzaldehyde ( $C_7H_6O_2$ ), 4-(benzyloxy)benzaldehyde ( $C_{14}H_{12}O_2$ ), 5-nitro-2-furaldehyde ( $C_5H_3NO_4$ ), 5-nitro-2-thiophenecarboxaldehyde ( $C_5H_3NO_3S$ ), p-Toluenesulfonic acid (PTSA), and sulphuric acid ( $H_2SO_4$ ), were obtained from Sigma-Aldrich (Johannesburg, South Africa). *N,N*-dimethylformamide (DMF) were purchased from Merck (Johannesburg, South Africa). All other solvents; ethanol (EtOH), dichloromethane (DCM), ethyl acetate (EtOAc), and hexane, as well as potassium carbonate ( $K_2CO_3$ ) and magnesium sulphate ( $MgSO_4$ ), were purchased from Associated Chemical Enterprises, ACE (South Africa). All chemicals and reagents were of analytical grade and no further purification was needed.

### 2.2. General procedures

The  $^1H$  and  $^{13}C$  nuclear magnetic resonance (NMR) spectra was recorded on a Bruker Avance<sup>TM</sup> III 600 spectrometer at a frequency of 600 and 151 MHz, respectively, in deuterated dimethyl sulfoxide ( $DMSO-d_6$ ). The chemical shifts were reported in parts per million (ppm) with the residual protons of the solvent as a reference. The abbreviations of the splitting patterns were as follows: singlet (s), doublet (d), doublet of doublet (dd), doublet of triplets (dt), triplet (t), triplet of doublets (td), triplet of triplets (tt), quartet of doublets (qd), and multiplet (m). High resolution mass spectrometry (HRMS) was recorded on a Bruker MicroTOF Q II mass spectrometer. It was equipped with an atmospheric pressure chemical ionisation (APCI) source set at 200 °C or 180 °C and analysed using a Bruker Compass Data Analysis 4.0 software. A full scan that ranged from 50 – 1500 m/z at a capillary voltage of 4500 V, an end plate offset voltage of -500 V, with the nebuliser set at 1.6 and 0.4 Bar, respectively, and a collision cell RF voltage of 100 Vpp. Infrared (FTIR) spectra were recorded on a Bruker Alpha-P FTIR instrument. A BÜCHI melting point B-545 instrument was used to determine melting points (mp) and were uncorrected. Thin

layer chromatography (TLC) was performed on silica gel plates (60F<sub>254</sub>) that was acquired from Merck (South Africa).

### 2.3. Syntheses

#### 2.3.1 Synthesis of phenylhydrazone intermediates (**1-8**)

Substituted phenylaldehydes (10 mmol, 1 eq.) were dissolved in DCM (10 mL), triethylamine (1.5 eq.) was added, and the reaction was left to stir at room temperature for 10 minutes. Thereafter, excess hydrazine monohydrate (60 mmol, 4 eq.) was added, and the reaction was left to stir at room temperature for 12 hours while monitored by TLC. Upon completion, the reaction mixture was extracted with DCM (2 × 30 mL) and water (2 × 40 mL). The organic phase was dried over anhydrous magnesium sulphate, filtered and concentrated in vacuum to yield a crude product, which was then recrystallised with ethyl acetate to form the phenylhydrazone compounds **1-8**.

#### (E)-Benzylidenehydrazine (**1**)

Dark yellow powder; yield: 47%; m.p. 88-90 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 2921 (N-H), 2851 (C-H), 1649 (C=C); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.72 (s, 1H, H-2), 8.89 (dd, *J* = 7.6, 1.8 Hz, 2H, H-4), 7.53 (dd, *J* = 13.1, 5.8 Hz, 3H, H-5/6), 6.62 (s, 2H, H-1). <sup>13</sup>C NMR (151 MHz, DMSO) δ 161.21 (C-2), 133.73 (C-3), 131.23 (C-6), 128.80 (C-4), 128.25 (C-5).

#### (E)-(4-Fluorobenzylidene)hydrazine (**2**)

Light yellow powder; yield: 62%; m.p. 181-183 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3324 (N-H), 2924 (C-H), 1596 (C=C), 1220 (C-F); <sup>1</sup>H NMR (600 MHz, DMSO) δ 7.70 (s, 1H, H-2), 7.50 (dd, *J*<sub>H-F</sub> = 8.6, 5.8 Hz, 2H, H-4), 7.14 (t, *J*<sub>H-F</sub> = 8.6 Hz, 2H, H-5), 6.70 (s, 2H, H-1). <sup>13</sup>C NMR (151 MHz, DMSO) δ 161.44 (d, <sup>1</sup>*J*<sub>C-F</sub> = 244.0 Hz, C-6), 137.11 (C-2), 132.92 (d, <sup>2</sup>*J*<sub>C-F</sub> = 3.0 Hz, C-3), 126.81 (d, <sup>3</sup>*J*<sub>C-F</sub> = 8.0 Hz, C-4), 115.26 (d, <sup>4</sup>*J*<sub>C-F</sub> = 21.7 Hz, C-5).

#### (E)-(4-Chlorobenzylidene)hydrazine (**3**)

Yellow powder; yield: 65%; m.p. 196-198 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3315 (N-H), 2933 (C-H), 1590 (C=C), 815 (C-Cl); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.72 (s, 1H, H-2), 7.90 (d, *J* = 8.5 Hz, 2H, H-4), 7.59 (d, *J* = 8.5 Hz, 2H, H-5), 6.62 (s, 2H, H-1). <sup>13</sup>C NMR (151 MHz, DMSO) δ 160.56 (C-2), 136.01 (C-6), 132.59 (C-3), 130.01 (C-4), 129.08 (C-5).

#### (E)-(4-Bromobenzylidene)hydrazine (**4**)

White crystal-like powder; yield: 61%; m.p. 72-74 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3353 (N-H), 2912 (C-H), 1588 (C=C), 507 (C-Br); <sup>1</sup>H NMR (600 MHz, DMSO) δ 7.65 (s, 1H, H-2), 7.50 (d, *J* = 8.5 Hz, 2H, H-4), 7.41 (d, *J* = 8.5 Hz, 2H, H-5), 6.91 (s, 2H, H-1). <sup>13</sup>C NMR (151 MHz, DMSO) δ 136.53 (C-2), 135.76 (C-3), 131.36 (C-5), 126.91 (C-4), 119.97 (C-6).

**(E)-(4-Methylbenzylidene)hydrazine (5)**

Yellow powder; yield: 85%; m.p. 150-152 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3316 (N-H), 2915 (C-H), 1612 (C=C); <sup>1</sup>H NMR (600 MHz, DMSO) δ 7.66 (s, 1H, H-2), 7.36 (d, *J* = 8.0 Hz, 2H, H-4), 7.13 (d, *J* = 8.0 Hz, 2H, H-5), 6.62 (s, 2H, H-1), 2.28 (s, 3H, H-7). <sup>13</sup>C NMR (151 MHz, DMSO) δ 138.56 (C-2), 136.61 (C-6), 133.59 (C-3), 128.96 (C-5), 125.06 (C-4), 20.72 (C-7).

**(E)-(4-Methoxybenzylidene)hydrazine (6)**

Light yellow powder; yield: 67%; m.p. 170-172 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 2927 (N-H), 2836 (C-H), 1595 (C=C); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.63 (s, 1H, H-2), 7.81 (d, *J* = 8.7 Hz, 2H, H-4), 7.05 (d, *J* = 8.7 Hz, 2H, H-5), 6.62 (s, 2H, H-1), 3.83 (s, 3H, H-7). <sup>13</sup>C NMR (151 MHz, DMSO) δ 161.65 (C-6), 160.45 (C-2), 129.95 (C-4), 126.55 (C-3), 114.38 (C-5), 55.37 (C-7).

**(E)-[4-(Benzylxy)benzylidene]hydrazine (7)**

White powder; yield: 84%; m.p. 125-127 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3314 (N-H), 3031 (C-H), 1597 (C=C), 1236 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 7.66 (s, 1H, H-2), 7.44 (d, *J* = 8.7 Hz, 2H, H-4), 7.41 (d, *J* = 8.5 Hz, 2H, H-9), 7.38 (d, *J* = 8.5 Hz, 2H, H-10), 7.32 (t, *J* = 7.3 Hz, 1H, H-11), 6.98 (d, *J* = 8.7 Hz, 2H, H-5), 6.48 (s, 2H, H-1), 5.10 (s, 2H, H-7). <sup>13</sup>C NMR (151 MHz, DMSO) δ 157.91 (C-6), 138.58 (C-2), 136.98 (C-8), 129.26 (C-4), 128.32 (C-10), 127.71 (C-3), 127.55 (C-9), 126.42 (C-11), 114.84 (C-5), 69.20 (C-7).

**(E)-4-(Hydrazonomethyl)phenol (8)**

Yellow powder; yield: 71%; m.p. 275-277 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3315 (O-H), 2939 (N-H), 2750 (C-H), 1585 (C=C); <sup>1</sup>H NMR (600 MHz, DMSO) δ 9.99 (s, 1H, H-7), 8.54 (s, 1H, H-2), 7.68 (d, *J* = 8.5 Hz, 2H, H-4), 6.86 (d, *J* = 8.5 Hz, 2H, H-5), 6.62 (s, 2H, H-1). <sup>13</sup>C NMR (151 MHz, DMSO) δ 160.16 (C-6), 129.96 (C-4), 126.56 (C-3), 125.05 (C-2), 115.66 (C-5).

### 2.3.2 Synthesis of phenylhydrazone intermediate (**9**)

4-Nitrobenzaldehyde (10 mmol, 1 eq.) was dissolved in ethanol (20 mL) and then hydrazine monohydrate (70 mmol, 5 eq.) and PTSA (1 mmol, 0.08 eq.) were added consecutively. The reaction was refluxed for 60 minutes and monitored by TLC. Upon completion, it was cooled down to 5-10 °C and the precipitate formed was filtered, washed with cold water and dried under vacuum to yield intermediate **9**. No further purification was done.

**(E)-(4-nitrobenzylidene)hydrazine (9)**

Bright yellow powder; yield: 91%; m.p. 134-136 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3419 (N-H), 3064 (C-H), 1572 (C=C), 1499 (N-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.16 (d, *J* = 8.9 Hz, 2H, H-5), 7.73 (s, 1H, H-2), 7.67 (d, *J* = 8.9 Hz, 2H, H-4), 7.55 (s, 2H, H-1). <sup>13</sup>C NMR (151 MHz, DMSO) δ 145.62 (C-6), 143.48 (C-2), 134.15 (C-3), 125.29 (C-4), 123.98 (C-5).

### 2.3.3 Synthesis of azine derivatives (**1a-9a** and **1b-9b**)

5-Nitro-2-furaldehyde or 5-nitro-2-thiophenecarboxaldehyde (3 mmol, 1 eq.) was dissolved in anhydrous ethanol (10 mL) and then substituted phenylhydrazone (3 mmol, 1 eq.) and 1 drop of catalytic sulphuric acid was added to the reaction. The reaction was left to stir for 12 hours and monitored by TLC. Upon completion, the reaction was quenched with water (50 mL) and the precipitate formed was filtered and recrystallised with ethyl acetate to yield the desired azines (**1a-9a**, **1b-9b**). Compounds **6a** and **7b** were further purified by column chromatography on silica gel eluting EtOAc/hexane (2:8, v/v) to yield the pure azine products.

In the case of compounds **5a**, **9a**, **5b**, **6b** and **9b**, the acid method was not suitable as the product could not be formed. For these specific derivatives, the previously used sulphuric acid was substituted for anhydrous potassium carbonate (6 mmol, 2 eq.).

#### (*1E,2E*)-1-Benzylidene-2-[(5-nitrofuran-2-yl)methylene]hydrazine (**1a**)

Brown powder; yield: 98%; m.p. 169-171 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 1629 (C=N), 1565 (C=C), 1471 (N-O), 1246 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.78 (s, 1H, H-1'), 8.64 (s, 1H, H-6), 7.90 (d, *J* = 6.9 Hz, 2H, H-3'), 7.81 (d, *J* = 3.8 Hz, 1H, H-4), 7.56 (t, *J* = 7.1 Hz, 1H, H-5'), 7.53 (d, *J* = 6.9 Hz, 2H, H-4'), 7.40 (d, *J* = 3.8 Hz, 1H, H-3). <sup>13</sup>C NMR (151 MHz, DMSO) δ 163.77 (C-1'), 152.52 (C-5), 150.68 (C-2), 149.56 (C-6), 133.30 (C-5'), 132.02 (C-2'), 129.02 (C-3'), 128.77 (C-4'), 118.54 (C-3), 114.13 (C-4). HRMS-APCI *m/z*: 244.0717 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>9</sub>N<sub>3</sub>O<sub>3</sub><sup>+</sup>, 244.0722).

#### (*1E,2E*)-1-(4-Fluorobenzylidene)-2-[(5-nitrofuran-2-yl)methylene]hydrazine (**2a**)

Yellow powder; yield: 66%; m.p. 191-193 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3113 (C-H), 1631 (C=N), 1597 (C=C), 1516 (N-O), 1342 (C-F), 1247 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.81 (s, 1H, H-1'), 8.65 (s, 1H, H-6), 7.97 (dd, *J*<sub>H-F</sub> = 8.8, 5.7 Hz, 2H, H-3'), 7.82 (d, *J* = 3.9 Hz, 1H, H-4), 7.41 (d, *J* = 3.9 Hz, 1H, H-3), 7.38 (t, *J*<sub>H-F</sub> = 8.8 Hz, 2H, H-4'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 164.24 (d, <sup>1</sup>*J*<sub>C-F</sub> = 250.3 Hz, C-5'), 162.64 (C-1'), 151.39 (C-5), 150.64 (C-2), 149.64 (C-6), 131.19 (d, <sup>2</sup>*J*<sub>C-F</sub> = 9.1 Hz, C-3'), 129.96 (d, <sup>3</sup>*J*<sub>C-F</sub> = 2.9 Hz, C-2'), 118.55 (C-3), 116.21 (d, <sup>4</sup>*J*<sub>C-F</sub> = 22.0 Hz, C-4'), 114.11 (C-4). HRMS-APCI *m/z*: 262.0635 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>8</sub>FN<sub>3</sub>O<sub>3</sub><sup>+</sup>, 262.0628).

#### (*1E,2E*)-1-(4-Chlorobenzylidene)-2-[(5-nitrofuran-2-yl)methylene]hydrazine (**3a**)

Yellow powder; yield: 60%; m.p. 175-177 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3092 (C-H), 1630 (C=N), 1562 (C=C), 1352 (N-O), 1246 (C-O), 820 (C-Cl); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.81 (s, 1H, H-1'), 8.65 (s, 1H, H-6), 7.92 (d, *J* = 8.5 Hz, 2H, H-3'), 7.82 (d, *J* = 3.9 Hz, 1H, H-4), 7.61 (d, *J* = 8.5 Hz, 2H, H-4'), 7.42 (d, *J* = 3.9 Hz, 1H, H-3). <sup>13</sup>C NMR (151 MHz, DMSO) δ 162.58 (C-1'), 152.53 (C-5), 150.56 (C-2), 149.79 (C-6), 136.56 (C-5'), 132.19 (C-2'), 130.34 (C-3'), 129.16 (C-4'), 118.68 (C-3), 114.08 (C-4). HRMS-APCI *m/z*: 278.0334 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>8</sub>ClN<sub>3</sub>O<sub>3</sub><sup>+</sup>, 278.0332).

**(1E,2E)-1-(4-Bromobenzylidene)-2-[(5-nitrofuran-2-yl)methylene]hydrazine (4a)**

Yellow powder; yield: 78%; m.p. 196-198 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3092 (C-H), 1629 (C=N), 1563 (C=C), 1475 (N-O), 1243 (C-O), 513 (C-Br); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.79 (s, 1H, H-1'), 8.65 (s, 1H, H-6), 7.84 (d, *J* = 8.5 Hz, 2H, H-3'), 7.82 (d, *J* = 3.9 Hz, 1H, H-4), 7.75 (d, *J* = 8.5 Hz, 2H, H-4'), 7.42 (d, *J* = 3.9 Hz, 1H, H-3). <sup>13</sup>C NMR (151 MHz, DMSO) δ 162.72 (C-1'), 152.55 (C-5), 150.56 (C-2), 149.81 (C-6), 132.52 (C-5'), 132.10 (C-3'), 130.51 (C-4'), 125.57 (C-2'), 118.73 (C-3), 114.10 (C-4). HRMS-APCI *m/z*: 321.9824 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>8</sub>BrN<sub>3</sub>O<sub>3</sub><sup>+</sup>, 321.9827).

**(1E,2E)-1-(4-Methylbenzylidene)-2-[(5-nitrofuran-2-yl)methylene]hydrazine (5a)**

Yellow powder; yield: 13%; m.p. 167-169 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3118 (C-H), 1626 (C=N), 1571 (C=C), 1336 (N-O), 1241 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.76 (s, 1H, H-1'), 8.64 (s, 1H, H-6), 7.82 (d, *J* = 3.9 Hz, 1H, H-4), 7.79 (d, *J* = 8.1 Hz, 2H, H-3'), 7.39 (d, *J* = 3.9 Hz, 1H, H-3), 7.34 (d, *J* = 8.1 Hz, 2H, H-4'), 2.38 (s, 3H, H-6'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 163.90 (C-1'), 152.46 (C-5), 150.79 (C-2), 149.28 (C-6), 142.27 (C-5'), 130.63 (C-2'), 129.62 (C-3'), 128.80 (C-4'), 118.31 (C-3), 114.14 (C-4), 21.23 (C-6'). HRMS-APCI *m/z*: 258.0880 [M+H]<sup>+</sup> (calculated for C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub><sup>+</sup>, 258.0879).

**(1E,2E)-1-(4-Methoxybenzylidene)-2-[(5-nitrofuran-2-yl)methylene]hydrazine (6a)**

Orange powder; yield: 27%; m.p. 169-171 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 2921 (C-H), 1706 (C=N), 1600 (C=C), 1347 (N-O), 1239 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.75 (s, 1H, H-1'), 8.62 (s, 1H, H-6), 7.86 (d, *J* = 8.8 Hz, 2H, H-3'), 7.82 (d, *J* = 3.9 Hz, 1H, H-4), 7.38 (d, *J* = 3.9 Hz, 1H, H-3), 7.09 (d, *J* = 8.8 Hz, 2H, H-4'), 3.84 (s, 3H, H-6'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 163.78 (C-1'), 162.39 (C-5), 150.97 (C-2), 148.86 (C-6), 130.73 (C-3'), 125.86 (C-5'), 118.03 (C-2'), 114.56 (C-4'), 114.56 (C-4), 114.20 (C-3), 55.48 (C-6'). HRMS-APCI *m/z*: 274.0840 [M+H]<sup>+</sup> (calculated for C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>O<sub>4</sub><sup>+</sup>, 274.0828).

**(1E,2E)-1-[4-(Benzylxy)benzylidene]-2-[(5-nitrofuran-2-yl)methylene]hydrazine (7a)**

Yellow powder; yield: 75%; m.p. 175-177 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3102 (C-H), 1618 (C=N), 1573 (C=C), 1352 (N-O), 1246 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.74 (s, 1H, H-1'), 8.61 (s, 1H, H-6), 7.86 (d, *J* = 8.8 Hz, 2H, H-3'), 7.80 (d, *J* = 3.9 Hz, 1H, H-4), 7.47 (d, *J* = 7.5 Hz, 2H, H-8'), 7.41 (t, *J* = 7.5 Hz, 2H, H-9'), 7.37 (d, *J* = 3.9 Hz, 1H, H-3), 7.35 (t, *J* = 7.5 Hz, 1H, H-10'), 7.16 (d, *J* = 8.8 Hz, 2H, H-4'), 5.21 (s, 2H, H-6'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 163.50 (C-1'), 161.41 (C-5'), 152.33 (C-5), 150.92 (C-2), 148.72 (C-6), 136.48 (C-7'), 130.63 (C-3'), 128.38 (C-9'), 127.89 (C-2'), 127.68 (C-8'), 126.02 (C-10'), 117.86 (C-3), 115.32 (C-4'), 114.09 (C-4), 69.46 (C-6'). HRMS-APCI *m/z*: 350.1151 [M+H]<sup>+</sup> (calculated for C<sub>19</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub><sup>+</sup>, 350.1141).

**4-(E)-(E)-[(5-Nitrofuran-2-yl)methylene]hydrazonomethylphenol (8a)**

Red powder; yield: 14%; m.p. 216-218 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3426 (O-H), 3112 (C-H), 1598 (C=N), 1565 (C=C), 1347 (N-O), 1211 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 10.27 (s, 1H, H-6'), 8.69 (s, 1H, H-1'), 8.59 (s, 1H, H-6), 7.81 (d,  $J$  = 3.9 Hz, 1H, H-4), 7.75 (d,  $J$  = 8.6 Hz, 2H, H-3'), 7.35 (d,  $J$  = 3.9 Hz, 1H, H-3), 6.89 (d,  $J$  = 8.6 Hz, 2H, H-4'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 164.15 (C-1'), 161.26 (C-5'), 152.35 (C-5), 151.12 (C-2), 148.44 (C-6), 130.99 (C-3'), 124.31 (C-2'), 117.77 (C-4), 115.94 (C-4'), 114.23 (C-3). HRMS-APCI  $m/z$ : 260.0686 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>9</sub>N<sub>3</sub>O<sub>4</sub><sup>+</sup>, 260.0671).

**(1E,2E)-1-(4-Nitrobenzylidene)-2-[(5-nitrofuran-2-yl)methylene]hydrazine (9a)**

Light brown powder; yield: 62%; m.p. 170-172 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3111 (C-H), 1633 (C=N), 1572 (C=C), 1526 (N-O), 1342 (N-O), 1251 (C-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.92 (s, 1H, H-1'), 8.69 (s, 1H, H-6), 8.37 (d,  $J$  = 8.7 Hz, 2H, H-4'), 8.15 (d,  $J$  = 8.7 Hz, 2H, H-3'), 7.83 (\*d,  $J$  = 3.9 Hz, 1H, H-4), 7.46 (\*d,  $J$  = 3.9 Hz, 1H, H-3). <sup>13</sup>C NMR (151 MHz, DMSO) δ 161.43 (C-1'), 150.45 (C-5), 150.27 (C-2), 149.05 (C-6), 139.19 (C-5'), 137.53 (C-2'), 129.72 (C-4'), 124.13 (C-3'), 119.30 (C-3), 114.04 (C-4). HRMS-APCI  $m/z$ : 289.0577 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>8</sub>N<sub>4</sub>O<sub>5</sub><sup>+</sup>, 289.0573).

\*d is a coalesced doublet

**(1E,2E)-1-Benzylidene-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (1b)**

Yellow powder; yield: 56%; m.p. 166-168 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3096 (C-H), 1610 (C=N), 1531 (C=C), 1495 (N-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.94 (s, 1H, H-1'), 8.75 (s, 1H, H-6), 8.19 (d,  $J$  = 4.3 Hz, 1H, H-4), 7.90 (d,  $J$  = 7.5 Hz, 2H, H-3'), 7.72 (d,  $J$  = 4.3 Hz, 1H, H-3), 7.56 (d,  $J$  = 7.5 Hz, 1H, H-5'), 7.52 (d,  $J$  = 7.5 Hz, 2H, H-4'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 163.60 (C-1'), 155.28 (C-5), 152.25 (C-2), 144.95 (C-6), 133.31 (C-5'), 132.42 (C-3), 132.00 (C-2'), 130.36 (C-4), 129.01 (C-3'), 128.78 (C-4'). HRMS-APCI  $m/z$ : 260.0512 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>9</sub>N<sub>3</sub>O<sub>2</sub>S<sup>+</sup>, 260.0494).

**(1E,2E)-1-(4-Fluorobenzylidene)-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (2b)**

Yellow powder; yield: 55%; m.p. 187-189 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3082 (C-H), 1613 (C=N), 1505 (C=C), 1335 (N-O), 1221 (C-F); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.93 (s, 1H, H-1'), 8.76 (s, 1H, H-6), 8.19 (d,  $J$  = 4.3 Hz, 1H, H-4), 7.97 (dd,  $J_{\text{H-F}} = 8.8, 5.7$  Hz, 2H, H-3'), 7.72 (d,  $J$  = 4.3 Hz, 1H, H-3), 7.37 (t,  $J_{\text{H-F}} = 8.8$  Hz, 2H, H-4'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 164.24 (d,  $^1J_{\text{C-F}} = 250.4$  Hz, C-5'), 162.48 (C-1'), 157.29 (C-5), 155.36 (C-2), 152.27 (C-6), 144.90 (C-2'), 132.44 (C-3), 131.17 (d,  $^2J_{\text{C-F}} = 9.0$  Hz, C-3'), 130.35 (C-4), 116.19 (d,  $^3J_{\text{C-F}} = 22.0$  Hz, C-4'). HRMS-APCI  $m/z$ : 278.0404 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>8</sub>FN<sub>3</sub>O<sub>2</sub>S<sup>+</sup>, 278.0400).

**(1E,2E)-1-(4-Chlorobenzylidene)-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (3b)**

Yellow powder; yield: 27%; m.p. 179-181 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3094 (C-H), 1604 (C=N), 1522 (C=C), 1321 (N-O), 815 (C-Cl); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.77 (s, 1H, H-1'), 8.57 (s, 1H, H-6), 8.17

(d,  $J = 4.4$  Hz, 1H, H-4), 8.02 (d,  $J = 8.5$  Hz, 2H, H-3'), 7.74 (d,  $J = 4.4$  Hz, 1H, H-3), 7.70 (d,  $J = 8.5$  Hz, 2H, H-4').  $^{13}\text{C}$  NMR (151 MHz, DMSO)  $\delta$  161.84 (C-1'), 155.96 (C-5), 150.88 (C-2), 137.13 (C-6), 136.35 (C-5'), 133.65 (C-3), 131.99 (C-2'), 130.88 (C-3'), 129.47 (C-4'), 129.12 (C-4). HRMS-APCI  $m/z$ : 294.0097 [M+H]<sup>+</sup> (calculated for  $\text{C}_{12}\text{H}_8\text{ClN}_3\text{O}_2\text{S}^+$ , 294.0104).

**(1E,2E)-1-(4-Bromobenzylidene)-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (4b)**

Yellow powder; yield: 41%; m.p. 193-195 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 3083 (C-H), 1610 (C=N), 1550 (C=C), 1332 (N-O), 536 (C-Br);  $^1\text{H}$  NMR (600 MHz, DMSO)  $\delta$  8.77 (s, 1H, H-1'), 8.59 (s, 1H, H-6), 8.18 (d,  $J = 4.4$  Hz, 1H, H-4), 7.95 (d,  $J = 8.4$  Hz, 2H, H-3'), 7.85 (d,  $J = 8.4$  Hz, 2H, H-4'), 7.75 (d,  $J = 4.4$  Hz, 1H, H-3).  $^{13}\text{C}$  NMR (151 MHz, DMSO)  $\delta$  161.93 (C-1'), 155.97 (C-5), 150.83 (C-2), 136.34 (C-6), 133.65 (C-3), 132.40 (C-3'), 132.31 (C-5'), 131.00 (C-4'), 129.12 (C-4), 126.18 (C-2'). HRMS-APCI  $m/z$ : 337.9592 [M+H]<sup>+</sup> (calculated for  $\text{C}_{12}\text{H}_8\text{BrN}_3\text{O}_2\text{S}^+$ , 337.9599).

**(1E,2E)-1-(4-Methylbenzylidene)-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (5b)**

Yellow powder; yield: 24%; m.p. 166-168 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 2920 (C-H), 1609 (C=N), 1532 (C=C), 1327 (N-O);  $^1\text{H}$  NMR (600 MHz, DMSO)  $\delta$  8.92 (s, 1H, H-1'), 8.71 (s, 1H, H-6), 8.18 (d,  $J = 4.1$  Hz, 1H, H-4), 7.80 (d,  $J = 8.0$  Hz, 2H, H-3'), 7.70 (d,  $J = 4.1$  Hz, 1H, H-3), 7.34 (d,  $J = 8.0$  Hz, 2H, H-4'), 2.38 (s, 3H, H-6').  $^{13}\text{C}$  NMR (151 MHz, DMSO)  $\delta$  163.67 (C-1'), 154.96 (C-5), 152.15 (C-2), 145.10 (C-6), 142.26 (C-5'), 132.24 (C-3), 130.64 (C-2'), 130.36 (C-4), 129.62 (C-3'), 128.79 (C-4'), 21.23 (C-6'). HRMS-APCI  $m/z$ : 274.0644 [M+H]<sup>+</sup> (calculated for  $\text{C}_{13}\text{H}_{11}\text{N}_3\text{O}_2\text{S}^+$ , 274.0650).

**(1E,2E)-1-(4-Methoxybenzylidene)-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (6b)**

Orange powder; yield: 46%; m.p. 171-173 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 2924 (C-H), 1608 (C=N), 1529 (C=C), 1329 (N-O);  $^1\text{H}$  NMR (600 MHz, DMSO)  $\delta$  8.90 (s, 1H, H-1'), 8.69 (s, 1H, H-6), 8.17 (d,  $J = 4.3$  Hz, 1H, H-4), 7.86 (d,  $J = 8.7$  Hz, 2H, H-3'), 7.68 (d,  $J = 4.3$  Hz, 1H, H-3), 7.08 (d,  $J = 8.7$  Hz, 2H, H-4'), 3.84 (s, 3H, H-6').  $^{13}\text{C}$  NMR (151 MHz, DMSO)  $\delta$  163.48 (C-1'), 162.40 (C-5), 154.46 (C-2), 151.97 (C-6), 145.37 (C-5'), 131.96 (C-2'), 130.72 (C-3'), 130.37 (C-3), 125.88 (C-4), 114.56 (C-4'), 55.47 (C-6'). HRMS-APCI  $m/z$ : 290.0595 [M+H]<sup>+</sup> (calculated for  $\text{C}_{13}\text{H}_{11}\text{N}_3\text{O}_3\text{S}^+$ , 290.0599).

**(1E,2E)-1-[4-(Benzylxy)benzylidene]-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (7b)**

Yellow powder; yield: 13%; m.p. 136-138 °C; IR  $\nu_{\text{max}}$  (cm<sup>-1</sup>): 2921 (C-H), 1599 (C=N), 1547 (C=C), 1328 (N-O), 1158 (C-O);  $^1\text{H}$  NMR (600 MHz, DMSO)  $\delta$  8.72 (s, 1H, H-1'), 8.52 (s, 1H, H-6), 8.16 (d,  $J = 4.4$  Hz, 1H, H-4), 7.98 (d,  $J = 8.8$  Hz, 2H, H-3'), 7.71 (d,  $J = 4.4$  Hz, 1H, H-3), 7.49 (d,  $J = 7.5$  Hz, 2H, H-8'), 7.42 (t,  $J = 7.5$  Hz, 2H, H-9'), 7.36 (t,  $J = 7.5$  Hz, 1H, H-10'), 7.26 (d,  $J = 8.8$  Hz, 2H, H-4'), 5.24 (s, 2H, H-6').  $^{13}\text{C}$  NMR (151 MHz, DMSO)  $\delta$  162.90 (C-1'), 161.87 (C-5'), 155.47 (C-5), 150.13 (C-2), 136.64 (C-6), 136.48 (C-7'), 132.99 (C-3), 131.39 (C-3'), 129.07 (C-4), 128.49 (C-9'), 128.02 (C-2'), 127.85 (C-8'), 125.80 (C-10'), 115.67 (C-4'), 69.58 (C-6'). HRMS-APCI  $m/z$ : 366.0916 [M+H]<sup>+</sup> (calculated for  $\text{C}_{19}\text{H}_{15}\text{N}_3\text{O}_3\text{S}^+$ , 366.0912).

**4-(E)-(E)-[(5-Nitrothiophen-2-yl)methylene]hydrazonomethylphenol (**8b**)**

Orange powder; yield: 42%; m.p. 243-245 °C; IR  $\nu_{max}$  (cm<sup>-1</sup>): 3453 (O-H), 3089 (C-H), 1602 (C=N), 1492 (C=C), 1332 (N-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 10.22 (s, 1H, H-6'), 8.88 (s, 1H, H-1'), 8.63 (s, 1H, H-6), 8.17 (d, *J* = 4.4 Hz, 1H, H-4), 7.75 (d, *J* = 8.6 Hz, 2H, H-3'), 7.67 (d, *J* = 4.4 Hz, 1H, H-3), 6.88 (d, *J* = 8.6 Hz, 2H, H-4'). <sup>13</sup>C NMR (151 MHz, DMSO) δ 163.80 (C-1'), 161.26 (C-5'), 154.01 (C-5), 151.83 (C-2), 145.55 (C-6), 131.75 (C-4), 130.97 (C-3'), 130.39 (C-3), 124.32 (C-2'), 115.94 (C-4'). HRMS-APCI *m/z*: 276.0434 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>9</sub>N<sub>3</sub>O<sub>3</sub>S<sup>+</sup>, 276.0443).

**(1*E*,2*E*)-1-(4-Nitrobenzylidene)-2-[(5-nitrothiophen-2-yl)methylene]hydrazine (**9b**)**

Yellow powder; yield: 66%; m.p. 239-241 °C; IR  $\nu_{max}$  (cm<sup>-1</sup>): 3108 (C-H), 1624 (C=N), 1595 (C=C), 1505 (N-O), 1336 (N-O); <sup>1</sup>H NMR (600 MHz, DMSO) δ 8.98 (s, 1H, H-1'), 8.87 (s, 1H, H-6), 8.35 (d, *J* = 8.7 Hz, 2H, H-4'), 8.19 (d, *J* = 4.3 Hz, 1H, H-4), 8.14 (d, *J* = 8.7 Hz, 2H, H-3'), 7.76 (d, *J* = 4.3 Hz, 1H, H-3). <sup>13</sup>C NMR (151 MHz, DMSO) δ 161.46 (C-1'), 156.45 (C-5), 152.71 (C-2), 149.05 (C-6), 144.40 (C-5'), 139.19 (C-2'), 133.09 (C-3), 130.33 (C-4), 129.71 (C-4'), 124.10 (C-3'). HRMS-APCI *m/z*: 305.0335 [M+H]<sup>+</sup> (calculated for C<sub>12</sub>H<sub>8</sub>N<sub>4</sub>O<sub>4</sub>S<sup>+</sup>, 305.0345).

### 3. Biological evaluation

#### 3.1. Cytotoxicity assay

The cytotoxicity of synthesised compounds was evaluated using the resazurin assay. The assay involved the irreversible enzymatic reduction of oxidised blue resazurin dye to pink, highly fluorescent resorufin by viable cells (Czekanska, 2011). This non-toxic reagent serves as an effective tool for assessing cell proliferation and drug toxicity.

Vero cells were cultured in Hyclone Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal bovine serum (FBS; Thermo Fisher, South Africa) and 1% L-glutamine, non-essential amino acids (NEAA) and penicillin-streptomycin (Pen/Strep) solution (Lonza). The cells were maintained in a humidified atmosphere at 37 °C and 5% CO<sub>2</sub>. For the resazurin assay, 96 well plates were prepared with 100 µL of 60 000 cells/mL Vero cell suspension. The Vero cell plates were incubated for 24 hours, after which the growth medium was replaced with 100 µL medium containing: (1) emetine dihydrochloride (Sigma Aldrich) solution diluted with growth medium to the necessary concentrations (positive control); (2) growth medium and solvent (negative control to compensate for possible solvent effects); (3) growth medium and experimental compound solutions. Blanks contained growth medium without cells. The treated plates were incubated for 48 hours. After incubation, 50 µL of sterile-filtered resazurin sodium salt (Sigma Aldrich) solution (0.01% in PBS) was added, and the plates were incubated for 2 hours. The Thermo Fisher Scientific GO Multiscan plate reader was used at 570 and 600 nm to measure absorbance. SkanIt 4.0 Research Edition software was used to perform data analysis for each

biological replicate. Background absorbance (600 nm) was subtracted from absorbance values (570 nm), the mean absorbance calculated, and the percentage cell viability was determined by the following equation:

$$\text{Cell viability \%} = (\Delta \text{Abs sample} - \Delta \text{Abs blank}) / (\Delta \text{Abs neg control} - \Delta \text{Abs blank}) \times 100$$

The IC<sub>50</sub> and Z-score were determined for each compound's biological replicate using GraphPad Prism 5. For the final IC<sub>50</sub> of each compound, the mean IC<sub>50</sub> of the biological replicates were calculated with standard deviation (SD).

### 3.2. *In vitro antileishmanial assay*

The anti-promastigote activity of synthesised compounds was evaluated using the resazurin assay on three *Leishmania* strains. A modified method of Kulshrestha et al. (2013) and Siqueira-Neto et al. (2010) was used.

Promastigotes of *L. donovani* (strains 1S (MHOM/SD/62/1S) and 9515 (MHOM/IN/95/9515)) and *L. major* (strain IR-173 (MHOM/IR/-173)) were cultured in Media 199 (M199) growth medium with Hank's salts and 0.68 mM L-glutamine, supplemented with 25 mM Hepes, 4.2 mM sodium bicarbonate, 0.0005% hemin, 0.0001% biotin, 0.1 mM adenine (Sigma Aldrich), 10% FBS and 50 U/mL Pen/Strep solution, the pH adjusted to 7.3 – 7.4. The promastigotes were maintained at 25 °C. For the resazurin assay, logarithmic phase promastigotes (1.25 x 10<sup>5</sup> cells/well, final volume 100 µL/well) were seeded in 96 well plates (Nunc, ThermoFisher Scientific) in the presence of 7 two-fold dilution concentrations of compounds for IC<sub>50</sub> determination. The standard drug was amphotericin B (10 µM), while growth medium without parasites served as the blank. The plates were incubated for 48 hours at 25 °C in humidified atmosphere. After the incubation period, 50 µL of resazurin solution (0.01% in phosphor-buffered saline) was added to each well and the plates were further incubated in the dark at 25 °C for 24 hours. The ThermoFisher Scientific GO Multiscan plate reader was used at 570 nm and 600 nm to measure absorbance. SkanIt 4.0 Research Edition software was used to perform data analysis for each biological replicate. Background absorbance of resazurin (600 nm) was subtracted from the absorbance values of resorufin (570 nm). The mean absorbance calculated, and the cell viability was determined by the following equation:

$$\text{Cell viability \%} = (\Delta \text{Abs sample} - \Delta \text{Abs blank}) / (\Delta \text{Abs neg control} - \Delta \text{Abs blank}) \times 100$$

The IC<sub>50</sub> and Z-score were determined for each compound's three biological replicates using the cell viability % values and GraphPad Prism 5. The mean IC<sub>50</sub> of the biological replicates served as the final IC<sub>50</sub> of each compound.

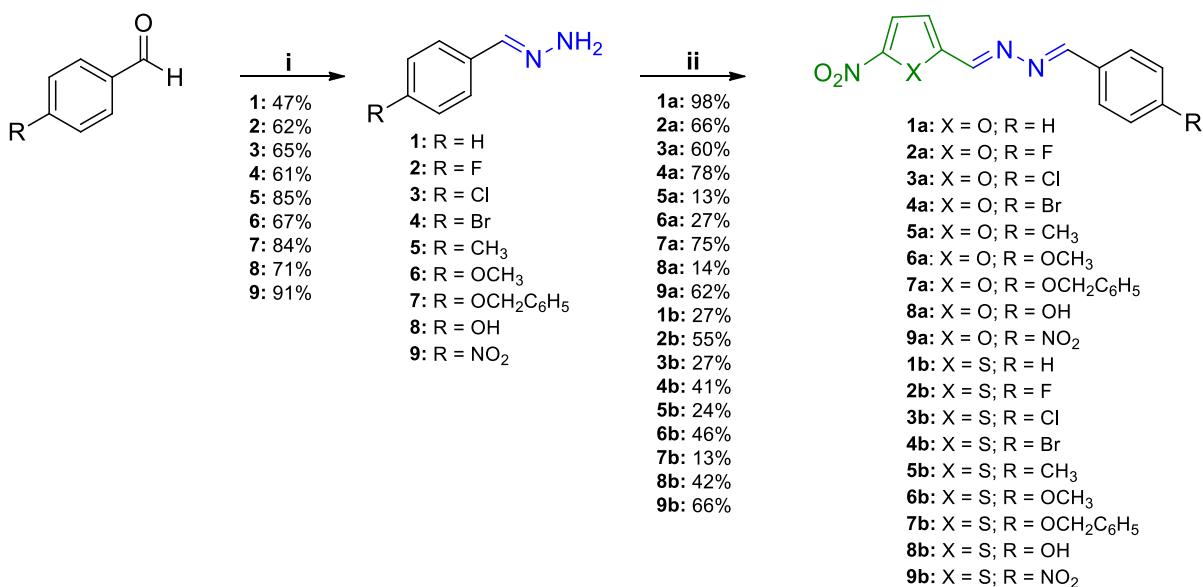
### 3.3. Statistical analysis

*In vitro* antileishmanial activities and cytotoxicity, indicated as IC<sub>50</sub> values, were derived from non-linear regression analysis. Results were represented as the mean  $\pm$  SD from the triplicate biological experiments. Statistical analysis was performed, using SkanIt 4.0 Research Edition software (Thermofisher Scientific) and Prism V5 software (GraphPad). All reported data was significant at  $p < 0.05$ .

## 4. Results and discussion

### 4.1. Chemistry

A series of nitrofuran-based azine derivatives (**1a-9a**, **1b-9b**) was synthesised *via* a two-step process (Scheme 3-1), starting from commercially available 5-nitro-2-furaldehyde (NFA) or 5-nitrothiophene-2-carboxaldehyde (NTA). Firstly, hydrazone formation occurs by the reaction of substituted aldehydes with excess hydrazine hydrate (**1-9**). This reaction is then followed by a Schiff base reaction of the hydrazones (**1-9**) with NFA or NTA, and upon elimination of water, the targeted nitrofuran-based azine derivatives (**1a-9a**, **1b-9b**) are formed in poor to excellent yields of 13-98% after recrystallisation with ethyl acetate or column chromatography.



Scheme 3-1: Two-step reaction for the formation of nitrofuran-based azine derivatives

Reagents and conditions: i. hydrazine hydrate (4 eq.), TEA (1.5 eq.), EtOH (10 mL), rt, 12 h; ii. Phenylhydrazone (**1-9**) (1 eq.), EtOH (10 mL), rt, 12 h.

The structures of all compounds were confirmed using routine analysis techniques, such as NMR, HRMS and IR. The formation of the hydrazone compounds (**1-9**) was confirmed on <sup>1</sup>H NMR by the disappearance of the characteristic aldehyde proton at *ca* δ 10.0 ppm and the appearance of a distinct singlet at *ca* δ 7.65 ppm indicative of the C-H proton. This is further collaborated by the

appearance of the characteristic singlet at *ca*  $\delta$  6.60, indicating the presence of the NH<sub>2</sub> proton of the hydrazone moiety. The disappearance of this NH<sub>2</sub> peak from the spectra **1a-9a** and **1b-9b** confirmed the success of the Schiff based reaction. This was further corroborated by the appearance of two separate vinylic protons H-1' and H-6 of the hydrazone bond as characteristic singlets at *ca*  $\delta$  8.79 and 8.65 ppm. The aromatic protons H-4 and H-3 of the furan ring gave a pair of doublets (d) with a *J* value of 3.9 Hz at  $\delta$  7.82 and  $\delta$  7.41 ppm, respectively. The deshielding of H-4 was due to the electronic effect of the nitro group that withdraws electron density away from H-4, while the unsaturated hydrazone bond pushes electron density towards H-3. The phenyl ring showed two aromatic resonances that appeared as two pairs of coupled doublets at  $\delta$  7.84 and  $\delta$  7.75 ppm (<sup>3</sup>*J* = 8.5 Hz), which were attributed to the aromatic protons H-3' and H-4', respectively. For the fluorine containing compounds (**2**, **2a**, **2b**), proton-fluorine coupling (*J*<sub>H-F</sub>) was determined, and it was found that these atoms underwent mostly ortho couplings with *J*<sub>H-F</sub> ~8-9 Hz. In summary, all the protons of each analogue were accounted for.

The <sup>13</sup>C spectra of all analogues showed the two hydrazone moiety peaks of C-1' at  $\delta$  162.7 and C-6 at  $\delta$  149.8, whereas the four aromatic carbons of the furan ring appeared as singlets at  $\delta$  152.5 (C-5),  $\delta$  150.6 (C-2),  $\delta$  118.7 (C-3) and  $\delta$  114.1 (C-4). The aromatic protons of the phenyl ring appeared as singlets at  $\delta$  132.5 (C-5'),  $\delta$  132.1 (C-3'),  $\delta$  130.5 (C-4') and  $\delta$  125.6 (C-2'). Carbon-fluorine coupling (*J*<sub>C-F</sub>) also occurred in compounds **2**, **2a** and **2b** with the <sup>1</sup>*J*, <sup>2</sup>*J*, <sup>3</sup>*J* and <sup>4</sup>*J* values within the expected ranges. IR further confirmed the success of the Schiff base reactions (**1a-9a**, **1b-9b**) by the appearance of its commonly showed characteristic absorption of C=N stretching (1600 – 1560 cm<sup>-1</sup>) and N-O stretching (1550-1500 and 1350-1300 cm<sup>-1</sup>).

#### 4.2. Predicted physicochemical and pharmacokinetic properties

The most used method in drug discovery for quantifying the lipophilicity of compounds is logP<sub>o/w</sub>. The logP<sub>o/w</sub> value of a molecule is the partition coefficient between octanol (lipophilic phase) and water (hydrophilic phase) (Lindsley, 2014). LogP<sub>o/w</sub> is used to examine important biological properties for drug action, which include lipid solubility, distribution in tissue, binding to receptors, cellular uptake, metabolism, and drug bioavailability (Ghose et al., 1998). According to Lipinski et al. (2001), logP<sub>o/w</sub> values are targeted to be between 1 and 5, with values between 1 and 3 being ideal.

**Table 1:** Physicochemical and ADME data of synthesised nitrofuran-based azine derivatives and standard nitrofuran drugs as predicted by SwissADME web tool, <http://www.swissadme.ch>

| Cpd.      | MW<br>(g/mol) | Log P <sub>o/w</sub><br><sup>a</sup> | RA<br><sup>b</sup> | Log S <sup>c</sup><br>ESOL <sup>d</sup> | TPSA<br>Å <sup>2</sup> <sup>f</sup> | HBD <sup>g</sup> | HBA <sup>h</sup> | Lipinski's<br>violation | GI<br>absorption | Leadlike-<br>ness <sup>i</sup> | Druglike-<br>ness <sup>j</sup> |
|-----------|---------------|--------------------------------------|--------------------|---|-------------------------------------|------------------|------------------|-------------------------|------------------|--------------------------------|--------------------------------|
| NFA       | 141.08        | 0.36                                 | 2                  | -1.59                                   | -2.20                               | 76.03            | 4                | 2                       | 0                | High                           | No                             |
| <b>1a</b> | 243.22        | 2.01                                 | 4                  | -2.77                                   | -3.34                               | 83.68            | 0                | 5                       | 0                | High                           | No                             |

|            |        |       |   |       |       |        |   |   |   |      |     |     |
|------------|--------|-------|---|-------|-------|--------|---|---|---|------|-----|-----|
| <b>2a</b>  | 261.21 | 2.32  | 4 | -2.92 | -3.45 | 83.68  | 0 | 6 | 0 | High | Yes | Yes |
| <b>3a</b>  | 277.66 | 2.52  | 4 | -3.35 | -3.99 | 83.68  | 0 | 5 | 0 | High | Yes | Yes |
| <b>4a</b>  | 322.11 | 2.64  | 4 | -3.67 | -4.06 | 83.68  | 0 | 5 | 0 | High | Yes | Yes |
| <b>5a</b>  | 257.24 | 2.15  | 4 | -3.06 | -3.72 | 83.68  | 0 | 5 | 0 | High | Yes | Yes |
| <b>6a</b>  | 273.24 | 1.82  | 5 | -2.83 | -3.51 | 92.91  | 0 | 6 | 0 | High | Yes | Yes |
| <b>7a</b>  | 349.34 | 3.43  | 7 | -4.75 | -5.99 | 92.91  | 0 | 6 | 0 | High | No  | Yes |
| <b>8a</b>  | 259.22 | 1.41  | 4 | -2.62 | -3.39 | 103.91 | 1 | 6 | 0 | High | Yes | Yes |
| <b>9a</b>  | 288.22 | 1.26  | 5 | -2.81 | -4.12 | 129.50 | 0 | 7 | 0 | High | Yes | Yes |
| <b>NTA</b> | 157.15 | 0.70  | 2 | -2.04 | -3.08 | 91.13  | 0 | 3 | 0 | High | No  | Yes |
| <b>1b</b>  | 259.28 | 2.66  | 4 | -3.82 | -5.23 | 98.78  | 0 | 4 | 0 | High | Yes | Yes |
| <b>2b</b>  | 277.27 | 2.88  | 4 | -3.97 | -5.33 | 98.78  | 0 | 5 | 0 | High | No  | Yes |
| <b>3b</b>  | 293.73 | 3.11  | 4 | -4.41 | -5.88 | 98.78  | 0 | 4 | 0 | High | No  | Yes |
| <b>4b</b>  | 338.18 | 3.28  | 4 | -4.72 | -5.94 | 98.78  | 0 | 4 | 0 | High | No  | Yes |
| <b>5b</b>  | 273.31 | 2.82  | 4 | -4.12 | -5.61 | 98.78  | 0 | 4 | 0 | High | No  | Yes |
| <b>6b</b>  | 289.31 | 2.51  | 5 | -3.88 | -5.39 | 108.01 | 0 | 5 | 0 | High | Yes | Yes |
| <b>7b</b>  | 365.41 | 3.94  | 7 | -5.24 | -6.95 | 108.01 | 0 | 5 | 0 | High | No  | Yes |
| <b>8b</b>  | 275.28 | 2.08  | 4 | -3.68 | -5.29 | 119.01 | 1 | 5 | 0 | High | Yes | Yes |
| <b>9b</b>  | 304.28 | 1.96  | 5 | -3.86 | -6.01 | 144.60 | 0 | 6 | 0 | Low  | Yes | Yes |
| <b>NFX</b> | 275.22 | 0.90  | 5 | -2.95 | -4.27 | 120.65 | 2 | 6 | 0 | High | Yes | Yes |
| <b>FZD</b> | 225.16 | 0.32  | 3 | -1.24 | -1.62 | 100.86 | 0 | 6 | 0 | High | No  | Yes |
| <b>NFZ</b> | 198.14 | -0.59 | 4 | -1.21 | -2.45 | 126.44 | 2 | 5 | 0 | High | No  | Yes |
| <b>NFT</b> | 238.16 | -0.50 | 3 | -1.04 | -1.60 | 120.73 | 1 | 6 | 0 | High | No  | Yes |

<sup>a</sup>Calculated log P<sub>o/w</sub> (consensus log P<sub>o/w</sub>). <sup>b</sup>Number of rotatable bonds. <sup>c</sup>Predicted aqueous solubility, where log S is the logarithm of the amount of compound (in moles) that can dissolve in a litre of water. <sup>d</sup>ESOL = estimated aqueous solubility, implemented from a topological method (Delaney, 2004). <sup>e</sup>Implemented from a topological method (Ali et al., 2012) with log S scale: insoluble < -10 < poorly soluble < -6 < moderately soluble < -4 < soluble < -2 very soluble < 0 highly soluble. <sup>f</sup>Topological polar surface area (TPSA): Calculated from (Ertl et al., 2000). <sup>g</sup>Number of hydrogen bond donors. <sup>h</sup>Number of hydrogen bond acceptors. <sup>i</sup>Calculated from (Teague et al., 1999), 250 ≤ MW ≤ 350; XLOGP ≤ 3.5; Number of rotatable bonds ≤ 7. <sup>j</sup>Determined by using Lipinski's rule of five: MW ≤ 500 g/mol; LogP ≤ 5; HBD ≤ 5; HBA ≤ 10; no more than one violation allowed (Lipinski et al., 2001). All values in this table were calculated using SwissADME web tool, <http://www.swissadme.ch> (Daina et al., 2017). NFA: 5-nitro-2-furaldehyde; NTA: 5-nitro-2-thiophenecarboxaldehyde; NFX: nifuroxazide; FZD: furazolidone; NFZ: nitrofurazone; NFT: nitrofurantoin.

All synthesised derivatives complied with Lipinski's rules and had physicochemical properties well within the target ranges as set by Lipinski et al. (2001). Lipophilicity, although a crucial physicochemical parameter, is not the only determining characteristic of good drug design, as evident by other successful drugs that do not fall within the target range (Arnott and Planey, 2012). Therefore, other drug parameters, such as topological polar surface area (TPSA), were predicted for these compounds. TPSA is the surface area of a molecule that emerges from nitrogen or oxygen atoms, as well as hydrogen atoms that are attached to oxygen or nitrogen atoms (Caron and Ermondi, 2016). This shows the correlation with passive molecular transport through membranes, which allows the prediction of GI absorption, Caco-2 monolayer permeability, and penetration of the blood-brain barrier. All the derivatives, with the exception of **9b**, were predicted

to be highly absorbed in the GI tract through passive diffusion and were, therefore, expected to be druglike by nature.

#### 4.3. Pharmacology

The synthesised derivatives were assessed for its *in vitro* antileishmanial activity against the promastigotes of three *Leishmania* promastigote strains, *L. donovani* strains (1S and 9515) and *L. major* strain IR-173. These *Leishmania* spp. were selected to determine the specificity of the synthesised derivatives against *L. major* parasites that cause CL, and *L. donovani* parasites that cause the more serious and debilitating VL (Salam et al., 2014). Amphotericin B (AmB) was used as the reference antileishmanial drug alongside 5-nitro-2-furaldehyde, 5-nitrothiophene-2-carboxaldehyde and the cNFs, nifuroxazide, furazolidone, nitrofurazone and nitrofurantoin were tested as reference drugs for comparison. Cytotoxicity profiles of the derivatives were determined using Vero cells with emetine (EM) as a positive control. These results are reported in **Table 2**.

**Table 2:** Biological results of synthesised nitrofuran-based azine derivatives and nitrofuran antibiotics

| Cpd. | <sup>a</sup> Antileishmanial activity, IC <sub>50</sub> ± SD (μM) <sup>b</sup> (n=3) |                              |                         |                              |                        |                              | Cytotoxicity<br>Vero <sup>c</sup> , IC <sub>50</sub> (μM) |
|------|--|------------------------------|-------------------------|------------------------------|------------------------|------------------------------|---|
|      | <i>L. donovani</i> 1S  | SI <sub>2</sub> <sup>d</sup> | <i>L. donovani</i> 9515 | SI <sub>3</sub> <sup>e</sup> | <i>L. major</i> IR-173 | SI <sub>4</sub> <sup>f</sup> |   |
| NFA  | 11.06 ± 1.50   | 2                            | 13.71 ± 2.37            | 1                            | 7.94 ± 0.79            | 2                            | 17.82 ± 0.25  |
| 1a   | 1.19 ± 0.03  | 14                           | 1.45 ± 0.19             | 11                           | 1.19 ± 0.29            | 14                           | 16.29 ± 3.50  |
| 2a   | 0.73 ± 0.17  | 65                           | 1.18 ± 0.14             | 40                           | 0.73 ± 0.09            | 65                           | 47.42 ± 1.12  |
| 3a   | 0.56 ± 0.04  | 179                          | 0.67 ± 0.04             | 149                          | 0.42 ± 0.07            | 238                          | > 100   |
| 4a   | 4.42 ± 0.00  | 23                           | 3.56 ± 0.00             | 28                           | 0.75 ± 0.18            | 133                          | > 100   |
| 5a   | 0.55 ± 0.20  | 182                          | 0.76 ± 0.05             | 132                          | 0.42 ± 0.08            | 238                          | > 100   |
| 6a   | 1.24 ± 0.10  | 41                           | 1.16 ± 0.11             | 44                           | 0.89 ± 0.12            | 57                           | 51.01 ± 1.26  |
| 7a   | 0.49 ± 0.10  | 204                          | 1.61 ± 0.22             | 62                           | 0.45 ± 0.02            | 222                          | > 100   |
| 8a   | 1.20 ± 0.36  | 16                           | 1.02 ± 0.08             | 18                           | 0.64 ± 0.10            | 29                           | 18.66 ± 1.31  |
| 9a   | 0.34 ± 0.20  | 34                           | 0.70 ± 0.06             | 17                           | 0.70 ± 0.10            | 17                           | 11.56 ± 1.12  |
| NTA  | 3.18 ± 0.59  | 4                            | 5.07 ± 0.45             | 2                            | 3.75 ± 0.39            | 3                            | 12.42 ± 0.88  |
| 1b   | 1.94 ± 0.13  | 22                           | 0.93 ± 0.06             | 45                           | 1.43 ± 0.19            | 29                           | 42.00 ± 8.79  |
| 2b   | 2.81 ± 0.00  | 36                           | 2.00 ± 0.15             | 50                           | 3.79 ± 0.70            | 26                           | > 100   |
| 3b   | > 100  | >1                           | > 100                   | >1                           | 42.78 ± 0.00           | 2                            | > 100   |
| 4b   | > 100  | >1                           | 81.1 ± 0.00             | 1                            | > 100                  | >1                           | > 100   |
| 5b   | > 100  | >1                           | > 100                   | >1                           | > 100                  | >1                           | > 100   |
| 6b   | > 100  | >1                           | > 100                   | >1                           | > 100                  | >1                           | > 100   |
| 7b   | > 100  | >1                           | > 100                   | >1                           | 4.39 ± 0.24            | 23                           | > 100   |
| 8b   | 4.14 ± 0.41  | 8                            | 1.66 ± 0.18             | 20                           | 2.22 ± 0.00            | 15                           | 33.12 ± 1.22  |
| 9b   | > 100  | >1                           | > 100                   | >1                           | > 100                  | >1                           | > 100   |
| AMB  | 0.02 ± 0.00  | 2890                         | 0.02 ± 0.009            | 2890                         | 0.03 ± 0.01            | 1927                         | 57.8 ± 3.2  |
| NFX  | 10.73 ± 1.03   | 9                            | 9.43 ± 0.85             | 11                           | 31.10 ± 2.16           | 3                            | > 100   |

|            |             |     |              |     |              |     |              |
|------------|-------------|-----|--------------|-----|--------------|-----|--------------|
| <b>FZD</b> | 0.32 ± 0.00 | 313 | 0.28 ± 0.28  | 357 | 0.34 ± 0.03  | 294 | > 100        |
| <b>NFZ</b> | 6.54 ± 0.93 | 15  | 1.85 ± 0.14  | 54  | 1.85 ± 0.06  | 54  | > 100        |
| <b>NFT</b> | > 100       | >1  | 23.71 ± 5.44 | 4   | 87.00 ± 2.35 | 1   | > 100        |
| <b>EM</b>  | n.d         |     | n.d          |     | n.d          |     | 0.08 ± 0.009 |

<sup>a</sup> Compounds **1a**, **4a**, **7a**, **8a**, **1b** and **7b**, were tested in suspension, therefore this may account for high SD values

<sup>b</sup> Represented as the mean ± SD from the triplicate biological experiments.

<sup>c</sup> African green monkey kidney epithelial cells.

<sup>d</sup> Selectivity index:  $SI_2 = IC_{50} \text{ Vero}/IC_{50} L. donovani 1S$ .

<sup>e</sup> Selectivity index:  $SI_3 = IC_{50} \text{ Vero}/IC_{50} L. donovani 9515$ .

<sup>f</sup> Selectivity index:  $SI_4 = IC_{50} \text{ Vero}/IC_{50} L. major IR-173$

Abbreviations: AMB: Amphotericin B; NFA: 5-nitro-2-furaldehyde; NTA: 5-nitro-2-thiophenecarboxaldehyde; NFX: nifuroxazole; FZD: furazolidone; NFZ: nitrofuranzone; NFT, nitrofurantoin; EM, emetine; n.d, not determined.

The synthesised intermediates **1-9** exhibited no activity thus were not reported in **Table 2**. The cytotoxicity data indicated that most of the nitrofuran derivatives (**1a-9a**) were moderately toxic to the mammalian cells, except compounds **3a**, **4a**, **5a** and **7a** which were non-toxic ( $IC_{50} > 100 \mu\text{M}$ ). On contrary, all the nitrothiophene derivatives (**1b-9b**), with the exceptions of **1b** and **8b**, were not toxic to the mammalian cells, inferring that their observed antileishmanial activities were intrinsic.

The most active derivatives of the series were compounds **3a**, **5a** and **7a** that demonstrated significant nanomolar activities against promastigotes of the three tested *Leishmania* strains, with high selectivity indexes (SI 179-238). These analogues possessed up to 32-fold superior activity in comparison to its parent compound NFA and the reference nitrofuran drugs.

Compounds **3a** and **5a** were active against all three strains showing nanomolar activity of  $0.56 \pm 0.04 \mu\text{M}$  and  $0.55 \pm 0.20 \mu\text{M}$  for *L. donovani* (1S),  $0.67 \pm 0.04 \mu\text{M}$ ,  $0.76 \pm 0.05 \mu\text{M}$  for *L. donovani* (9515) and  $0.42 \pm 0.07 \mu\text{M}$ ,  $0.42 \pm 0.08 \mu\text{M}$  for *L. major* promastigotes, respectively. The most active of these three compounds was **7a** with nanomolar activity of  $0.49 \pm 0.01 \mu\text{M}$  for *L. donovani* (1S),  $1.61 \pm 0.22 \mu\text{M}$  for *L. donovani* (9515) and  $0.45 \pm 0.02 \mu\text{M}$  for *L. major* promastigotes. Compound **9a** showed excellent activity against all three tested strains with  $IC_{50} 0.34 \pm 0.20 \mu\text{M}$ ,  $0.70 \pm 0.06 \mu\text{M}$  and  $0.70 \pm 0.10 \mu\text{M}$  against *L. donovani* (1S and 9515) and *L. major* promastigotes. However, this compound was also moderately toxic to the mammalian cells, disqualifying it as a potential hit compound. Compound **7a** was not predicted to be a good lead compound. However, this compound turned out to be the most active compound of the series.

In this series, two moieties (furan and thiophene) were compared with the only difference between them being that one bares an oxygen atom whereas the other had a sulphur. The antileishmanial results indicate that the nitrofuran sub-series (**1a-9a**) exhibited a much more potent activity than the nitrothiophene sub-series (**1b-9b**). Thus, this may be explained by the presence of the nitrofuran extra hydrogen bond acceptor i.e., O atom in compounds **1a-9a**. Hydrogen bonds form non-covalent interactions that are significant in both medicinal and biochemistry. These bonds

are important because a single hydrogen bond interaction can enhance the potency of oxygen-containing drugs towards parasites in comparison to their sulphur containing counterparts (Bauer et al., 2019).

Because of its overall superior activity, the nitrofuran sub-series compounds were analysed for structure activity relationships (SARs). When considering the electronic effect and strength, the R groups of compounds can be divided into electron withdrawing groups (EWG) ( $\text{NO}_2 > \text{F} > \text{Cl} > \text{Br} > \text{OCH}_3$ ), neutral groups (H), and electron donating groups (EDG) ( $\text{OH} > \text{OCH}_2\text{C}_6\text{H}_5 > \text{CH}_3$ ). The general pattern of EWG compounds suggests that the activity of these compounds decrease as the strength of the EWG decreases. This is evident by the activities of **3a** bearing the strongest of these EWGs i.e.,  $\text{NO}_2$ , that exhibits  $\text{IC}_{50}$  values of 0.34, 0.70, and 0.70  $\mu\text{M}$ , against *L. donovani* 1S, *L. donovani* 9515 and *L. major*, respectively, compared to 1.24, 1.16, and 0.89  $\mu\text{M}$  of the weakest EWG (the  $\text{OCH}_3$  group).

On contrary, EDG compounds had increased activity as the strength of the EDG decreased. This is evident by the strongest of these EDGs (the OH group) exhibiting  $\text{IC}_{50}$  values of 1.20, 1.02, and 0.64  $\mu\text{M}$ , respectively, compared to 0.55, 0.76, and 0.42  $\mu\text{M}$  of the weakest EDG (the  $\text{CH}_3$  group). The neutral group (H group) overall exhibited weaker activity, with  $\text{IC}_{50}$  values of 1.19, 1.45, and 1.19  $\mu\text{M}$ , respectively against the same parasites.

## 5. Conclusion

A series of nitrofuran-based azine derivatives were synthesised in poor to excellent yields following a two-step process that included hydrazone formation and Schiff base reaction. Most of the compounds were found to be non-cytotoxic, apart from compounds **1a**, **2a**, **6a**, **8a**, **9a**, **1b** and **8b** which exhibited moderate toxicity. It was observed that the nitrofuran sub-series (**1a-9a**) was more potent than the nitrothiophene sub-series (**1b-9b**), and this might be related to the oxygen atom of the furan ring's ability to form hydrogen bonding. The electronic effect analysis showed strong EWGs has good activity, while weak EDG has the better activity. Derivatives **3a** and **5a** were the most potent against all three *Leishmania* tested strains and **7a** against two *Leishmania* strains (*L. donovani* 1S and *L. major*), and these compounds could be identified as anti-promastigote hits. Further work on the nitrofuran series will be done as a possible route to enhancing its antileishmanial activity. Furthermore, the activities of the current derivatives will also be tested against the intracellular amastigote form to confirm their potential as antileishmanial hits.

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*Leishmania donovani*, Strain 1S (MHOM/SD/62/1S), NR-48821

*Leishmania donovani*, Strain 9515 (MHOM/IN/95/9515), NR-48822

*Leishmania major*, Strain IR173 (MHOM/IR/-173), NR-48816

### **Disclaimer**

Any opinion, findings, conclusions, and recommendations expressed in this manuscript are those of the author(s).

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## CHAPTER 4

### SUMMARY AND CONCLUSION

Leishmaniasis is a vector-borne protozoan parasitic disease that is transmitted to humans during hematophagy of an infected female *Phlebotomus* sandfly (WHO, 2021a). This tropical disease is endemic to more than 98 countries and is amongst the top ten most neglected and devastating diseases in the world (WHO, 2017; Güran, 2018:2). This disease can manifest in three clinical forms; the most common being the CL form, the rarer MCL form, and/or the most lethal VL form (Salam *et al.*, 2014:1). There are more than a billion individuals living in *Leishmania* endemic areas, putting them at higher risk of contracting the disease. According to WHO, about 90 000 new cases of VL and a million new cases of CL are reported annually worldwide (WHO, 2021b). Leishmaniasis is estimated to cause 20 000 to 30 000 deaths every year (PAHO, 2020).

Leishmaniasis can be treated and cured with chemotherapeutic drugs, such as the pentavalent antimonies, amphotericin B, miltefosine, pentamidine, paromomycin and the azoles (Georgiadou *et al.*, 2015:46; Braga, 2019:2). However, most of these available drugs are limited by impractical administration (except for miltefosine), overall toxicity, cost, and the occurrence of pathogenic resistance (Freitas-Junior *et al.*, 2012:12). There have been several reports of parasitic resistance against the pentavalent antimonies, amphotericin B, and miltefosine. This has led to the discontinued use of some of these drugs against leishmaniasis in areas where resistance is prevalent, and patient cure rates have declined. Hence, the emphasis on the urgent need for the discovery of new, safe, and effective drugs to treat this disease (Capela *et al.*, 2019:1).

It has been well established over recent years that clinical nitrofurans (cNFs) can be used to treat infectious diseases (Patterson & Wyllie, 2014:291; Zuma *et al.*, 2019:1). The diverse biological activities of nitrofurans have been attributed to their nitro group and its ability to form ROS upon Fenton reactions with enzymes (Pal & Bandyopadhyay, 2012:556; Patel *et al.*, 2018:8). These enzymes are crucial for the growth and development of parasites. When ROS are produced, the parasite membranes are damaged *via* oxidation of their enzymes, which eventually leads to cell death (Pal & Bandyopadhyay, 2012:556). These drugs are active against various infections, including Chagas disease (Sales Junior *et al.*, 2017:1289) and *Leishmania* (Rando *et al.*, 2008:6724; Sifontes-Rodríguez *et al.*, 2015:166).

In addition to the nitrofuran moiety, cNFs also have a hydrazone moiety that possess anti-infective activity. The combination of hydrazones with other biologically active functional groups have been shown to result in compounds with unique chemical and physical characteristics (Verma *et al.*, 2014:69). This includes compounds that are easily synthesised, highly active against a variety of

conditions, and has low toxicity (Asif & Husain, 2013:2). Therefore, this makes nitrofurans good candidates for drug development for the treatment of infectious diseases such as leishmaniasis.

Furthermore, the use of aromatic rings in drug design is common, as these rings offer numerous unique and strong interaction modes with target proteins and, therefore, introduce chemical stability, diverse reactivity paths and possible enhancement of biological activity (Aldeghi *et al.*, 2014:50).

Thus, the aim of this study was to investigate the antileishmanial activity of nitrofuran-based azine derivatives, synthesised and characterised using routine techniques such as NMR, HRMS, and IR spectroscopy. These compounds were also evaluated for *in vitro* cytotoxicity on mammalian cell lines.

The nitrofuran-based azine derivatives were synthesised *via* a two-step process, starting from commercially available 5-nitro-2-furaldehyde (NFA) or 5-nitrothiophene-2-carboxaldehyde (NTA). Firstly, hydrazone formation occurred by the reaction of substituted aldehydes with excess hydrazine hydrate. This reaction was then followed by a Schiff base reaction of the hydrazones with NFA or NTA, and upon elimination of water, the targeted nitrofuran-based azine derivatives were formed in poor to excellent yields of 13-98% after recrystallisation with ethyl acetate or column chromatography. The structures of all the synthesised compounds were confirmed using routine analysis techniques, namely NMR, HRMS and IR.

The synthesised derivatives were screened for *in vitro* antileishmanial activity against the promastigotes of three *Leishmania* promastigote strains, *L. donovani* (strains 1S and 9515) and *L. major* (strain IR-173). These *Leishmania* spp. were selected to determine the specificity of the synthesised derivatives against *L. donovani* parasites that cause the serious and debilitating VL, and *L. major* parasites that cause CL (Salam *et al.*, 2014:1). Compounds **1a-9a**, **1b-2b** and **8b** all exhibited good activity against the three *Leishmania* promastigote strains, with IC<sub>50</sub> values ranging between 0.34 µM and 4.42 µM. The remaining compounds, **3b-7b** and **9b**, showed little to no activity against the promastigotes.

Most of the compounds were non-toxic to the mammalian cells (IC<sub>50</sub> >100 µM), except compounds **1a**, **2a**, **8a**, **9a**, **1b** and **8b**, which were moderately toxic, and **6a** that had low toxicity. The most active compounds of the series were compounds **3a**, **5a** and **7a** as they demonstrated significant nanomolar activities against the promastigotes of the three tested *Leishmania* strains.

The antileishmanial results indicated that the nitrofuran sub-series (**1a-9a**) exhibited a much more potent activity than the thiophene sub-series (**1b-9b**). This might be related to the oxygen atom of the furan ring's ability to form hydrogen bonding. An electronic effect analysis in relation to

activity, was performed on the nitrofuran sub-series, and it revealed that strong electron withdrawing groups (EWGs) has good activity, while weak electron donating groups (EDGs) has the better activity.

During the course of this study a few challenges were faced. The original method to synthesise the phenylhydrazone compounds did not work. This method consisted of using ethanol as solvent, along with hydrazine hydrate (1 eq.) and the required aldehyde (1 eq.). To resolve this issue, the method was changed to the one described in chapter 3 (2.3.1). When performing the biological assays, several compounds exhibited solubility issues. This resulted in high SD values and can affect the overall accuracy of the results. In future studies, if the solubility of the compounds can be increased, it may result in more accurate assays.

In summary, the synthesised nitrofuran-based azine derivatives were brought about through simple synthetic routes in poor to excellent yields and additionally possessed overall good antileishmanial activity. The nitrofuran sub-series proved to be much more active against the three *Leishmania* promastigote strains in comparison to the nitrothiophene sub-series. Most of the compounds proved to be safe against mammalian cells. For further testing, the activities of the current derivatives will be screened against the intracellular amastigote form to confirm their potential as antileishmanial hits.

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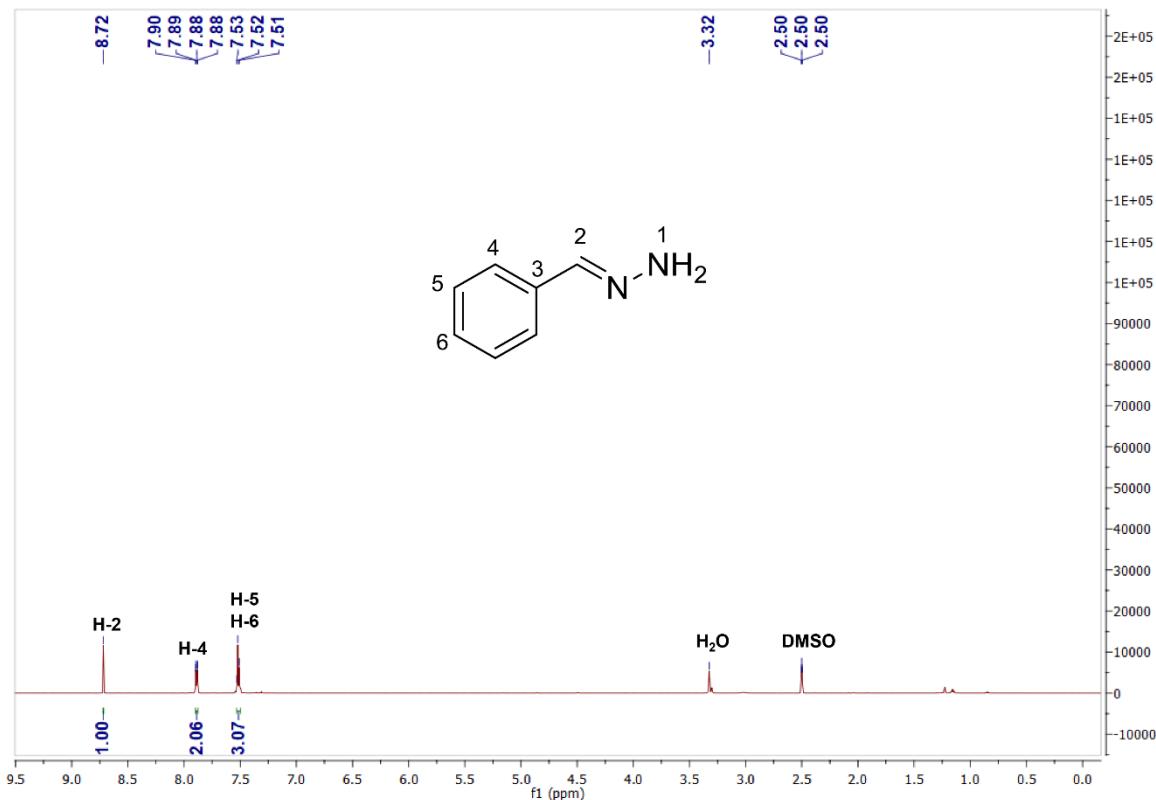
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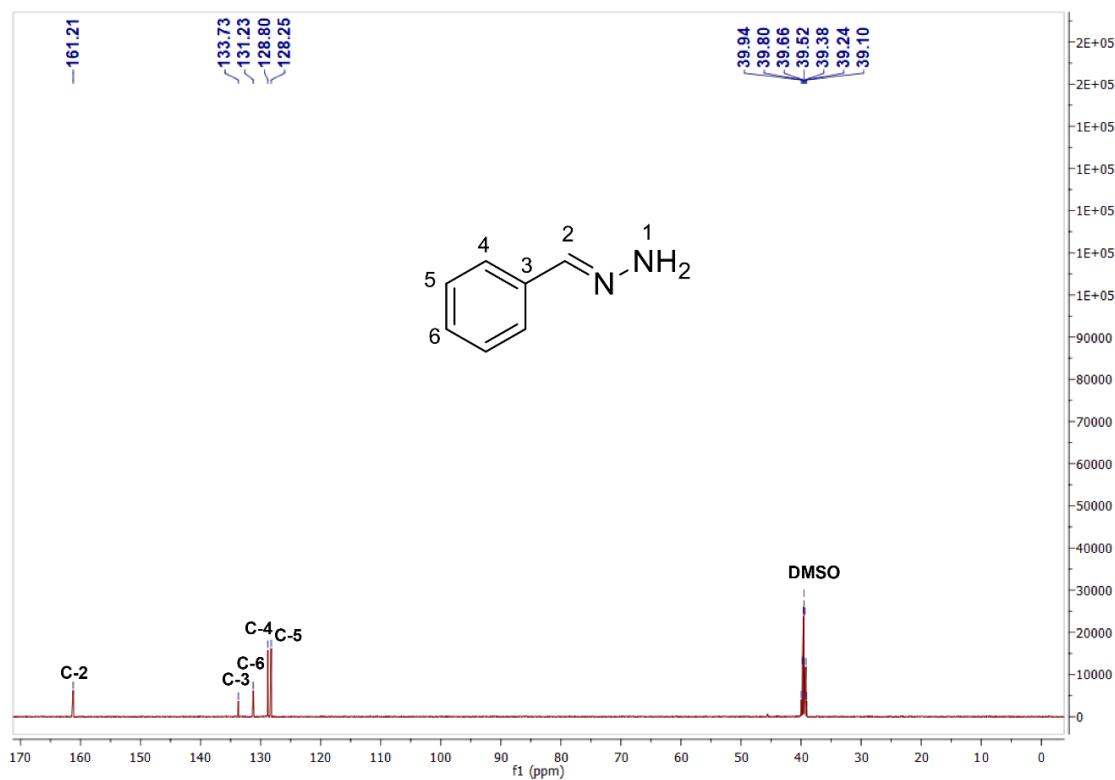
## ANNEXURE A: ANALYTICAL DATA FOR CHAPTER 3

### (E)-benzylidenehydrazine (1)

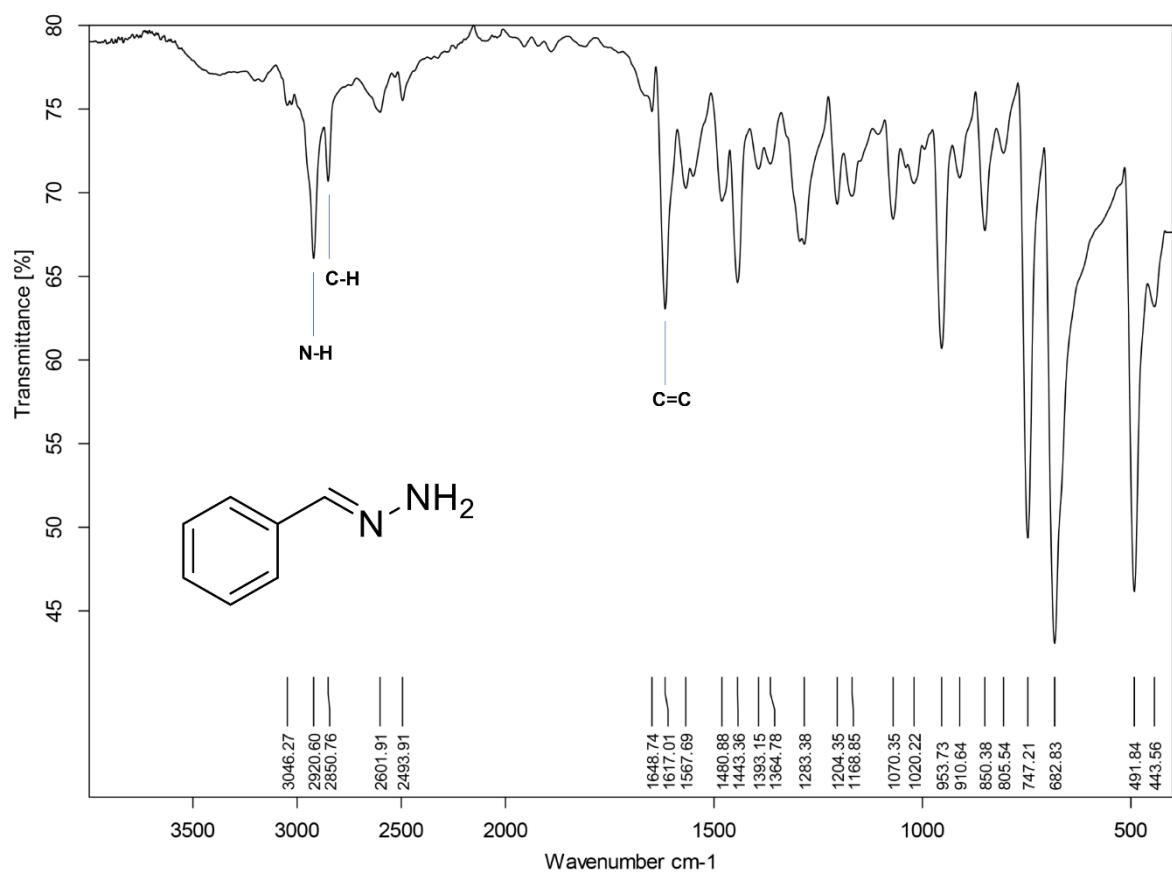
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<sup>13</sup>C NMR in DMSO

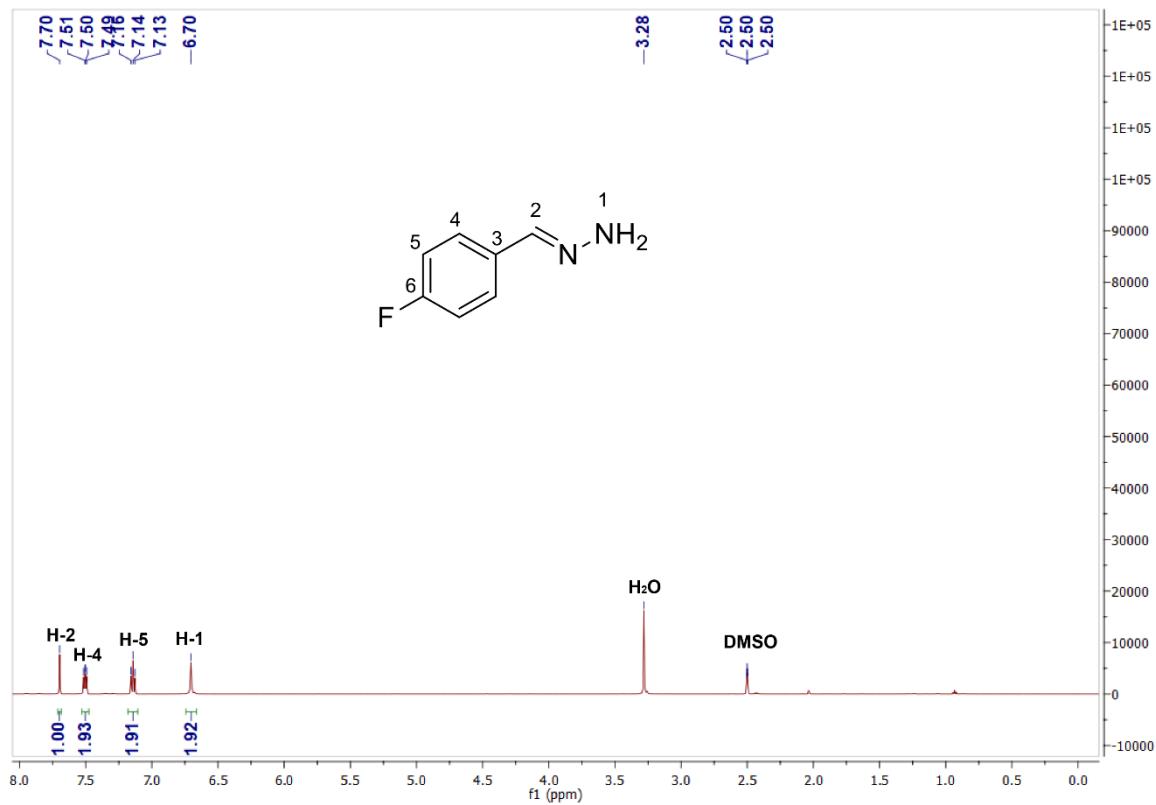


## IR Spectrum

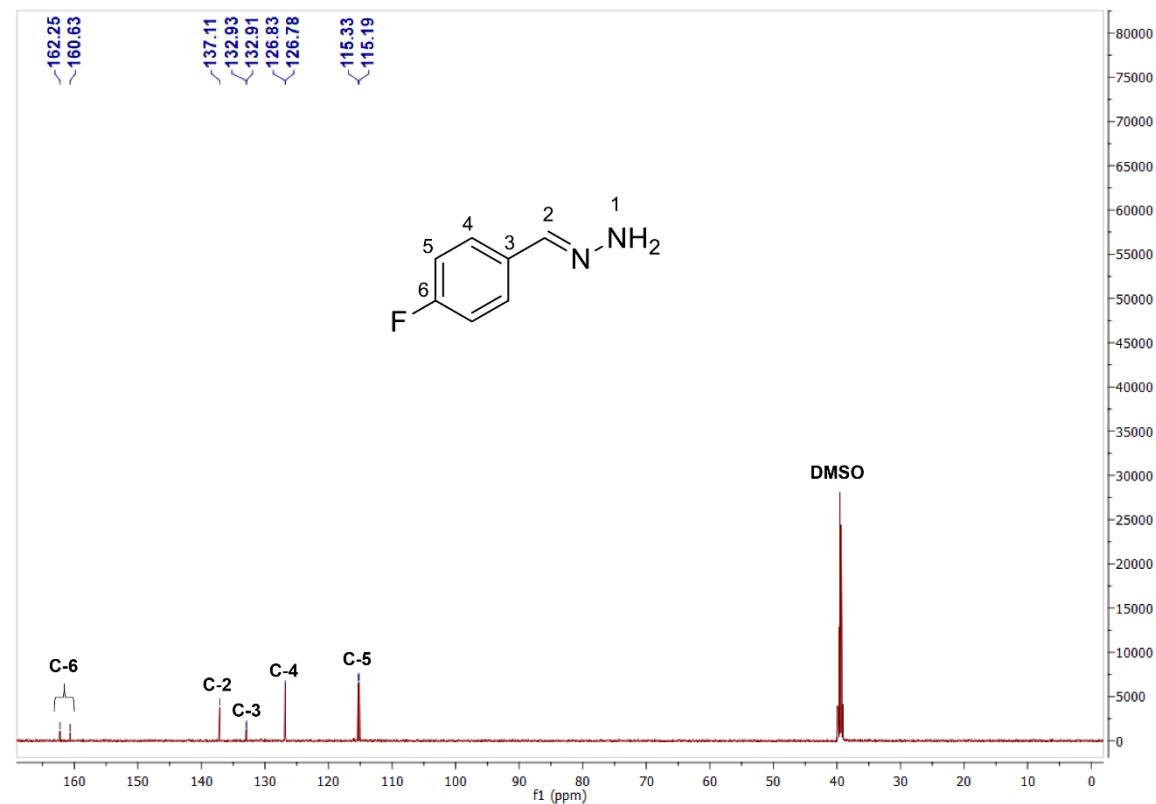


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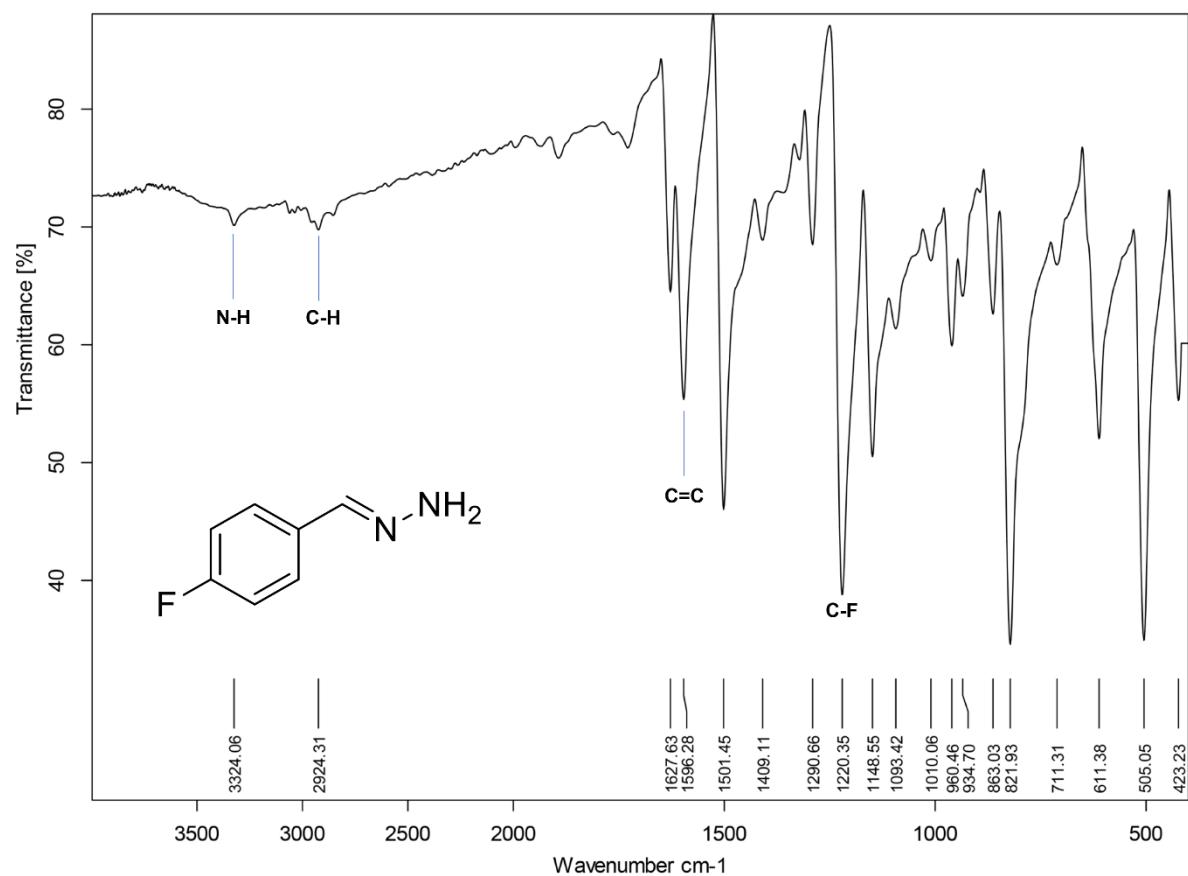
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**$^{13}\text{C}$  NMR in DMSO**

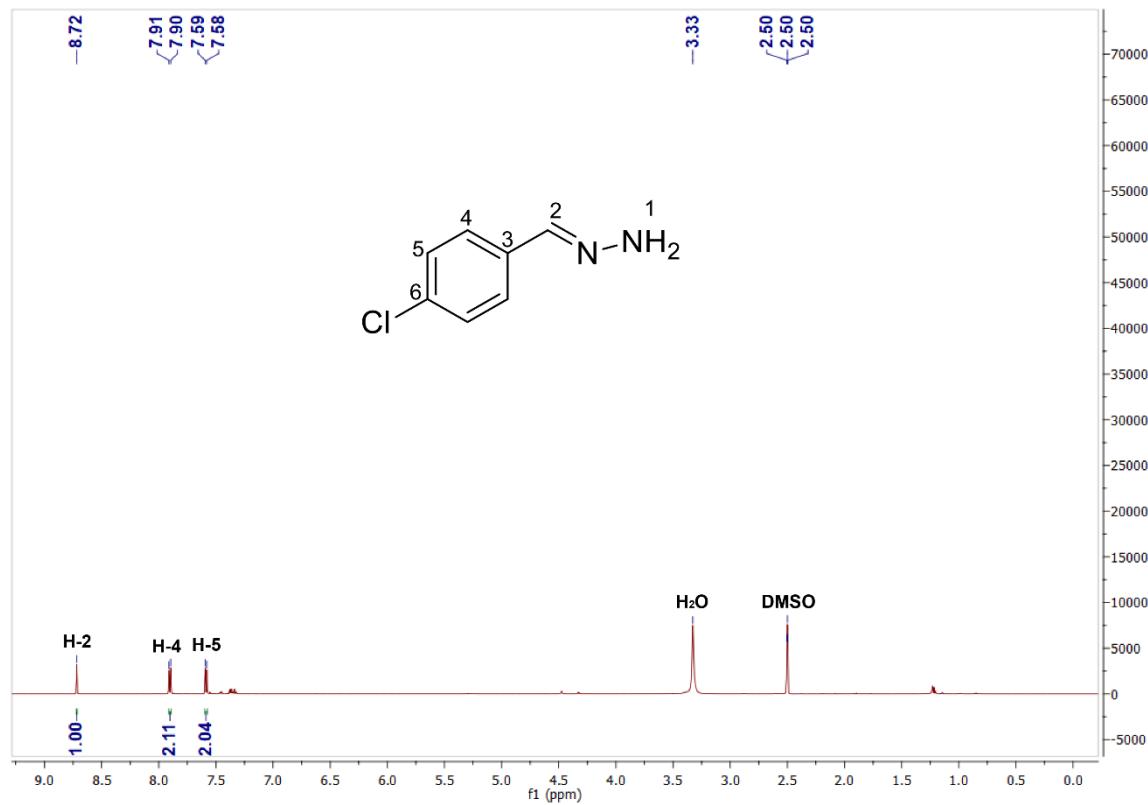


## IR Spectrum

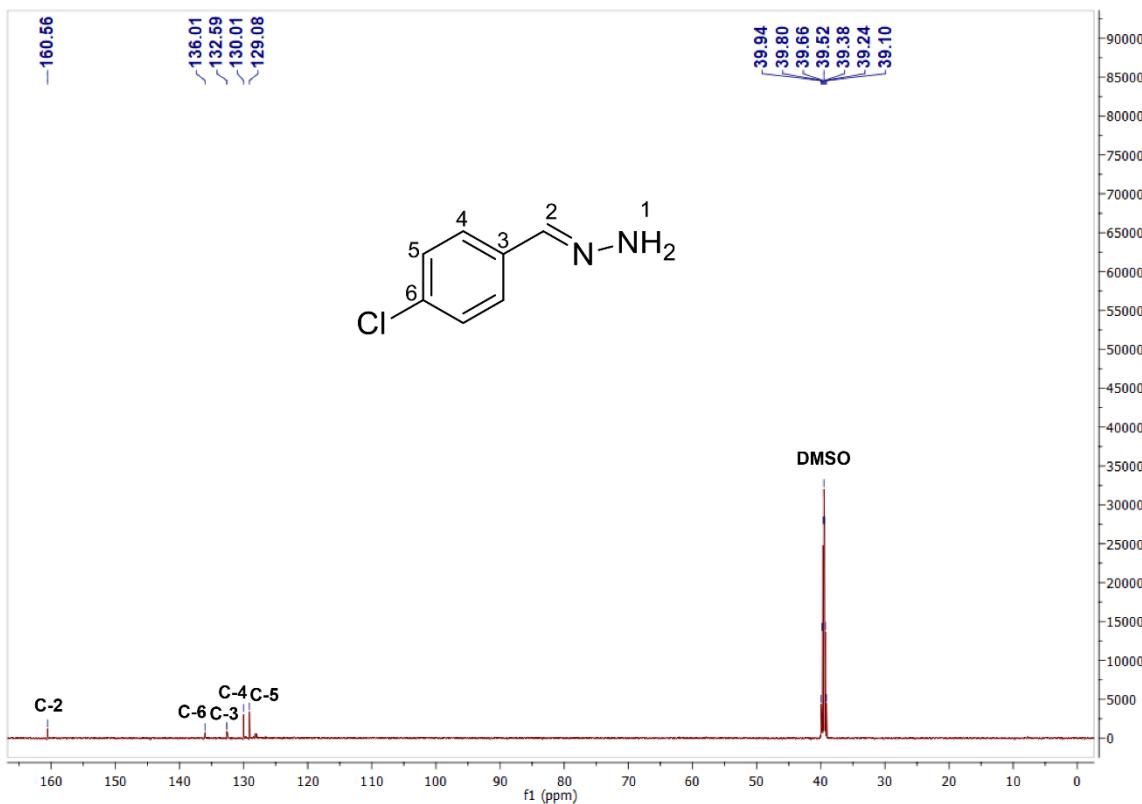


**(E)-(4-chlorobenzylidene)hydrazine (3)**

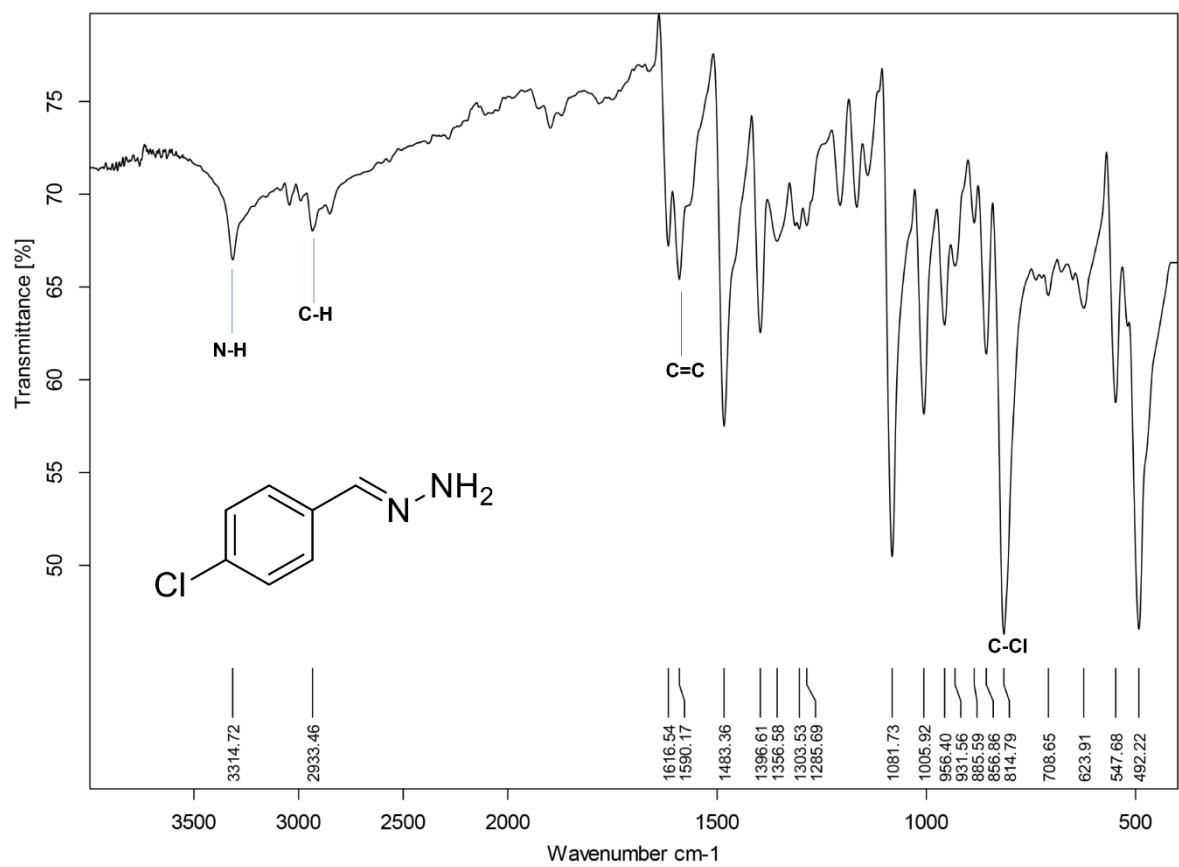
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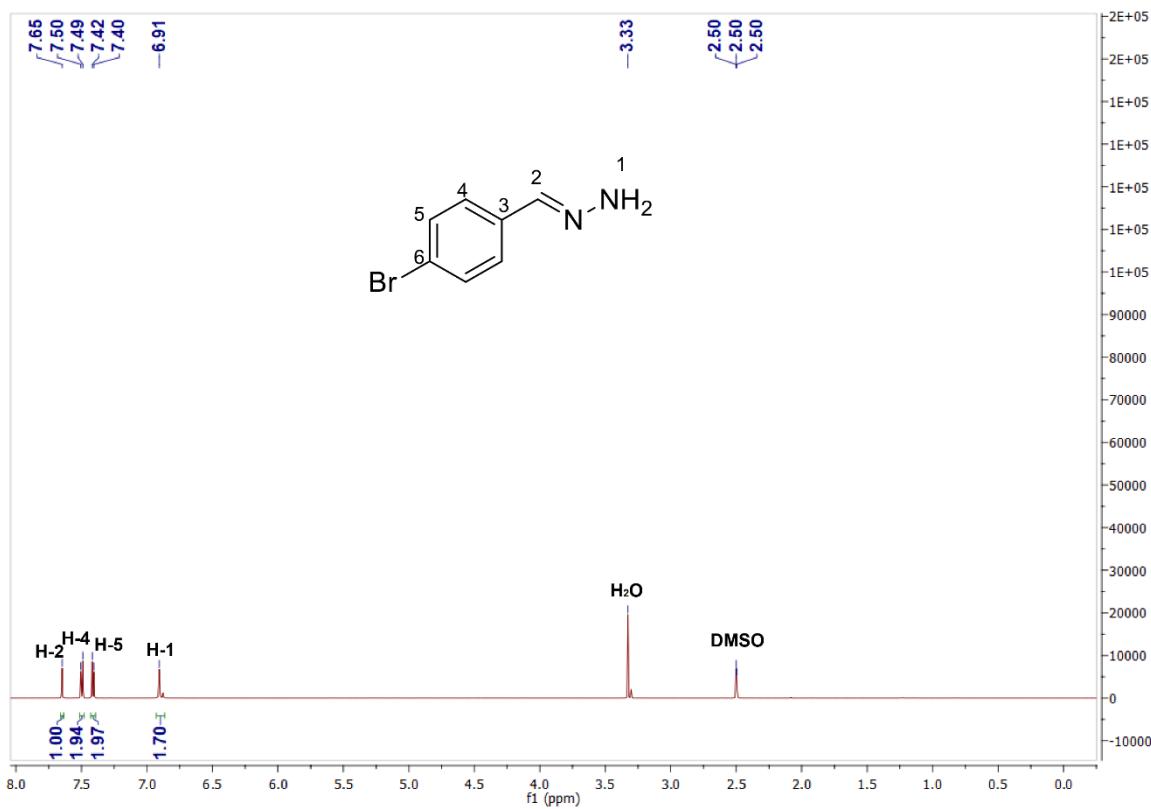


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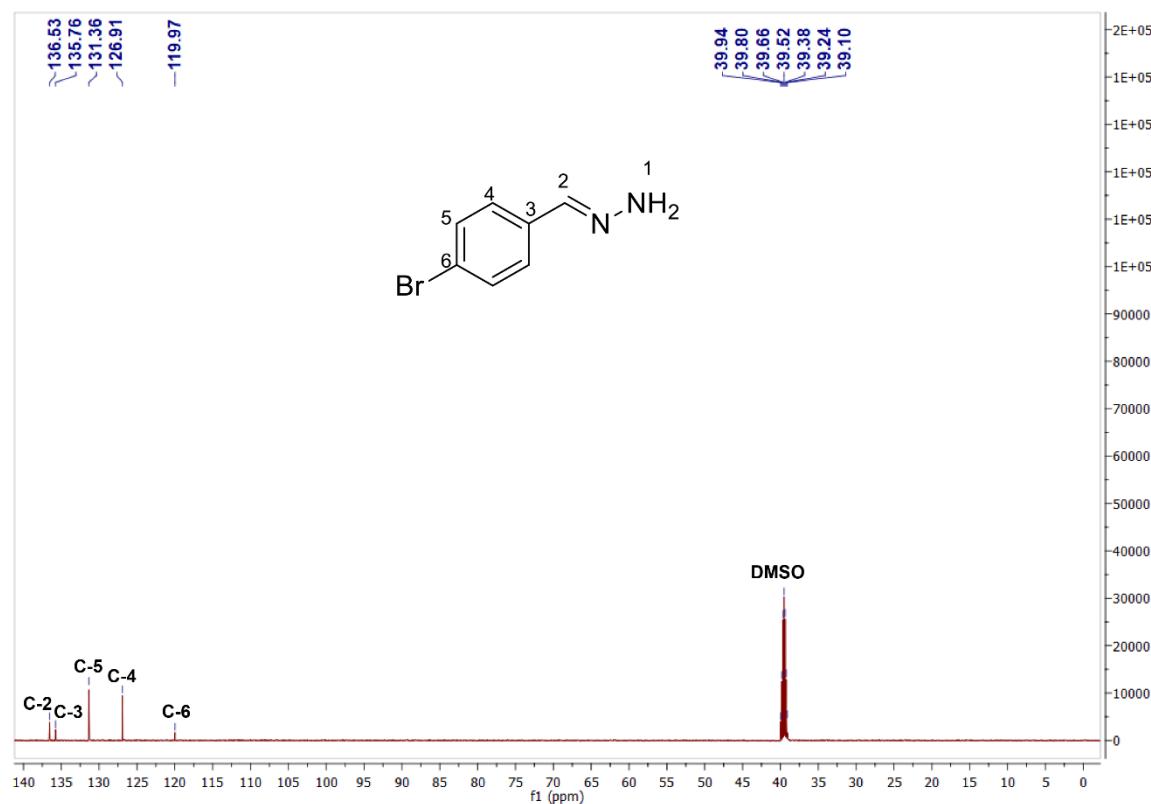


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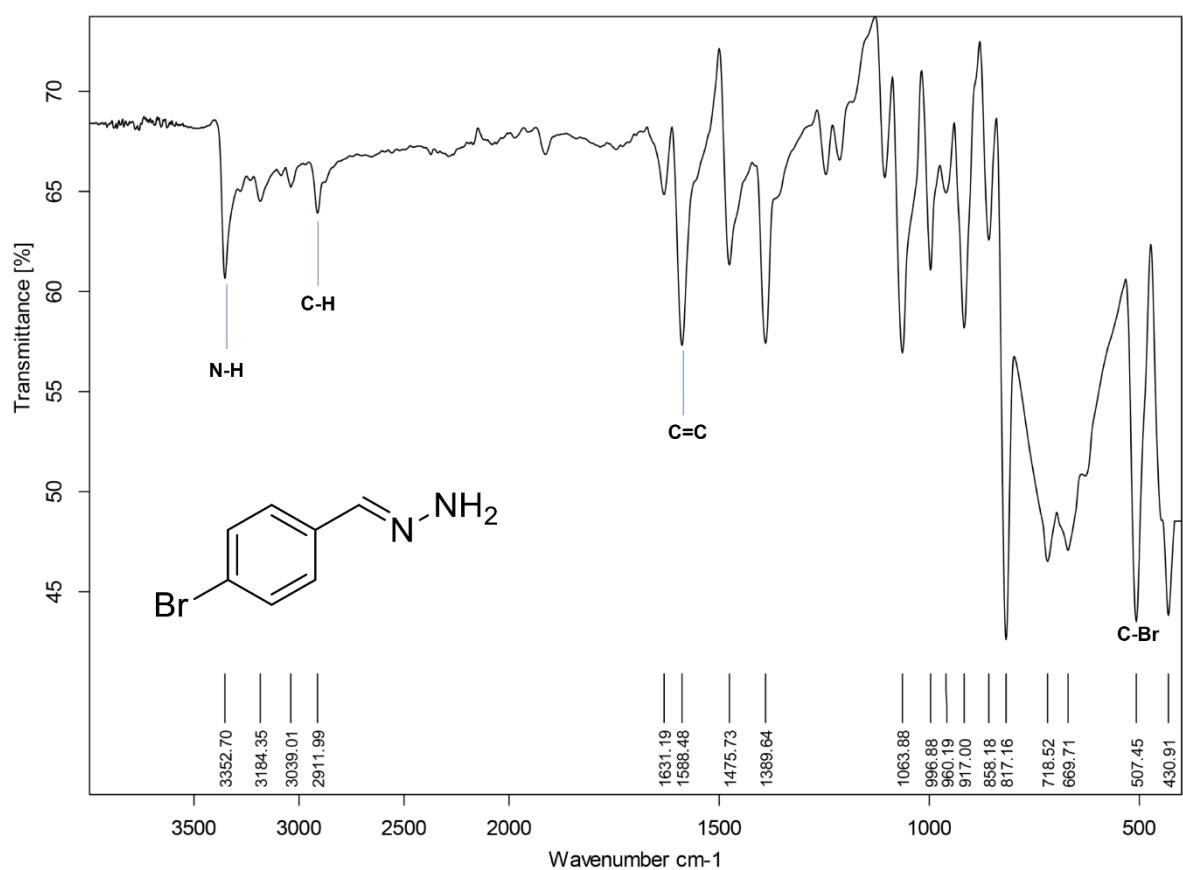
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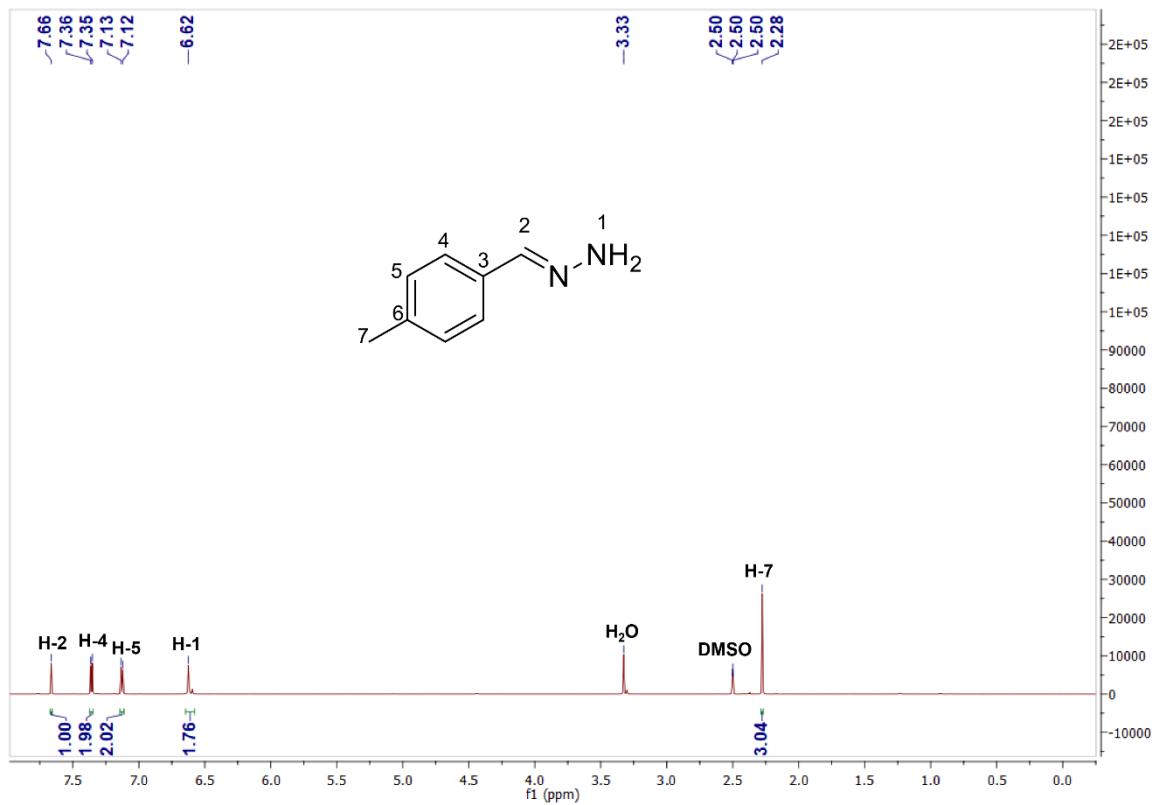


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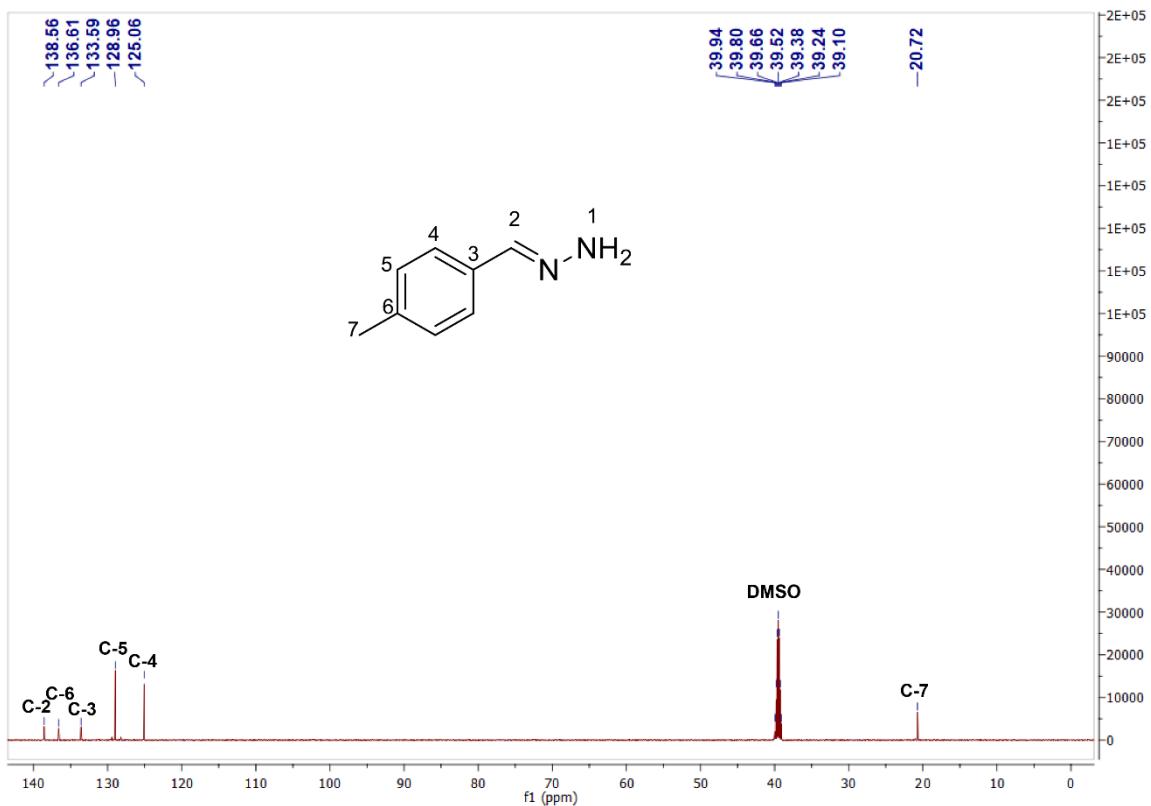


**(E)-(4-methylbenzylidene)hydrazine (5)**

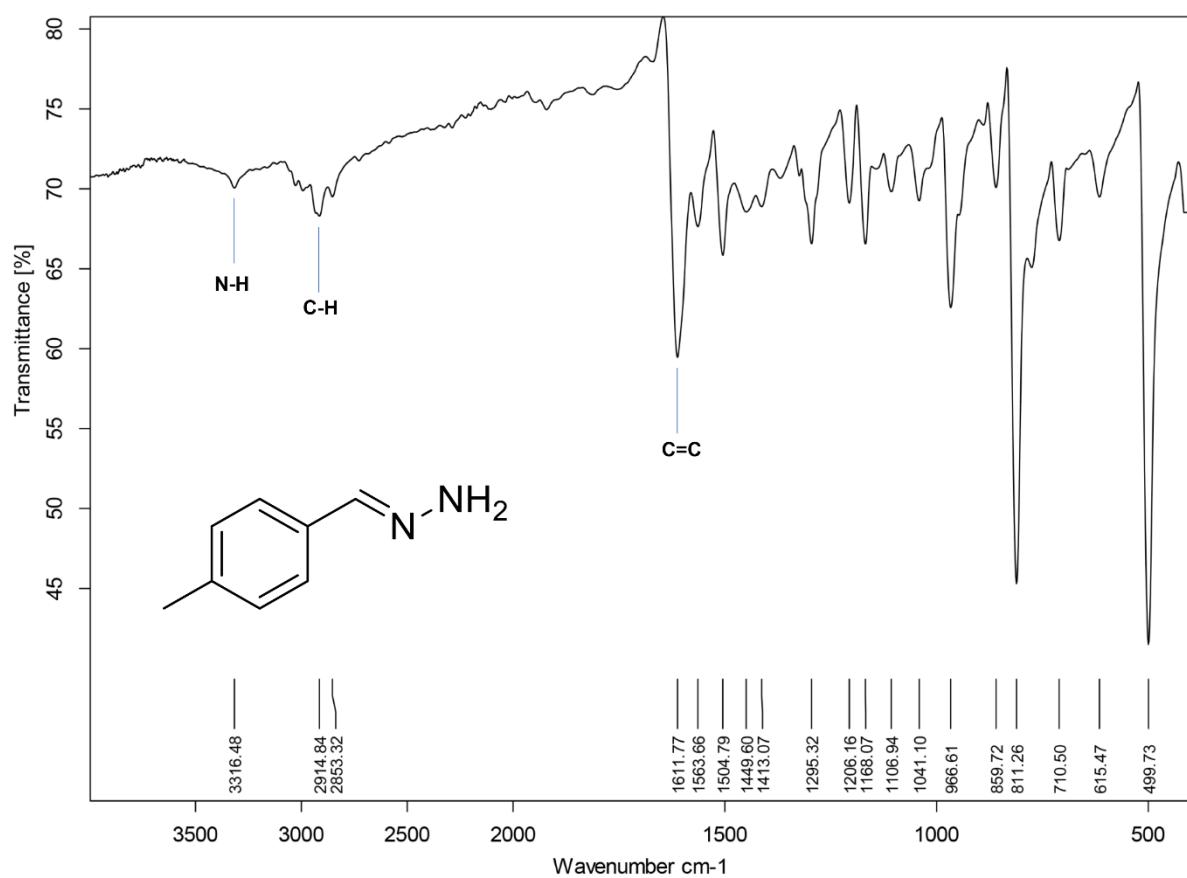
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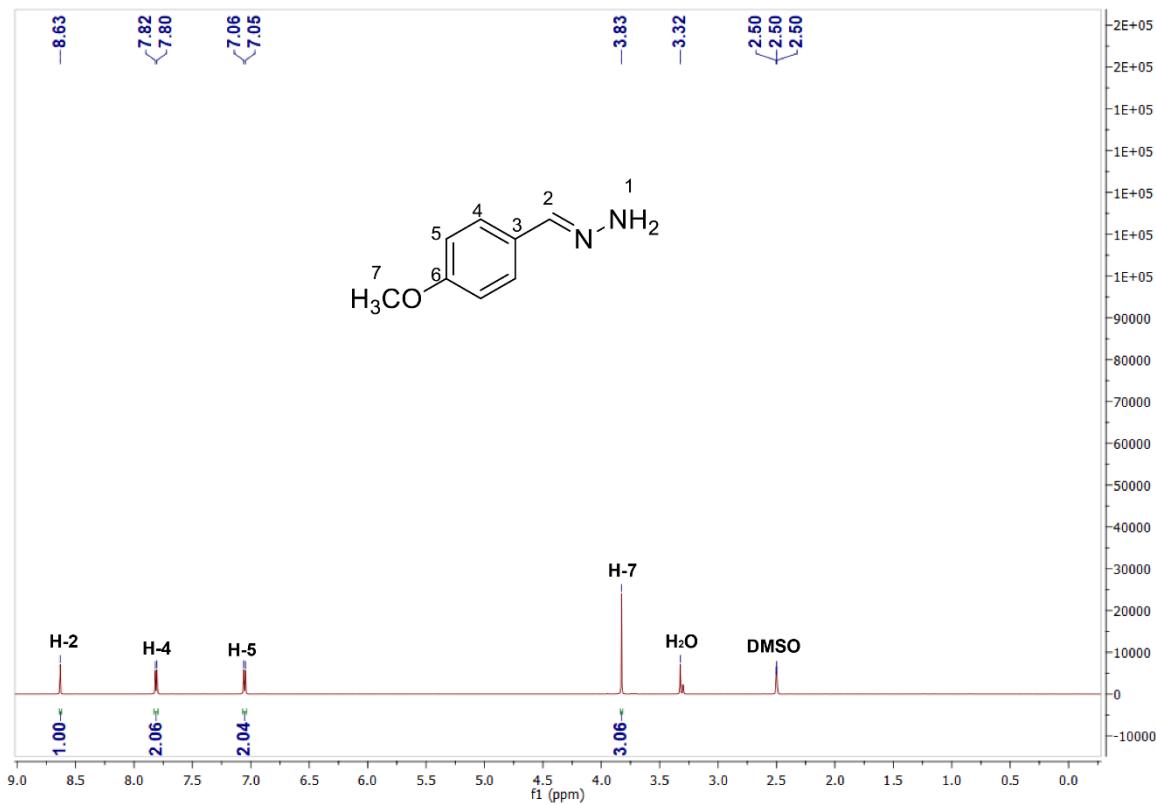


## IR Spectrum

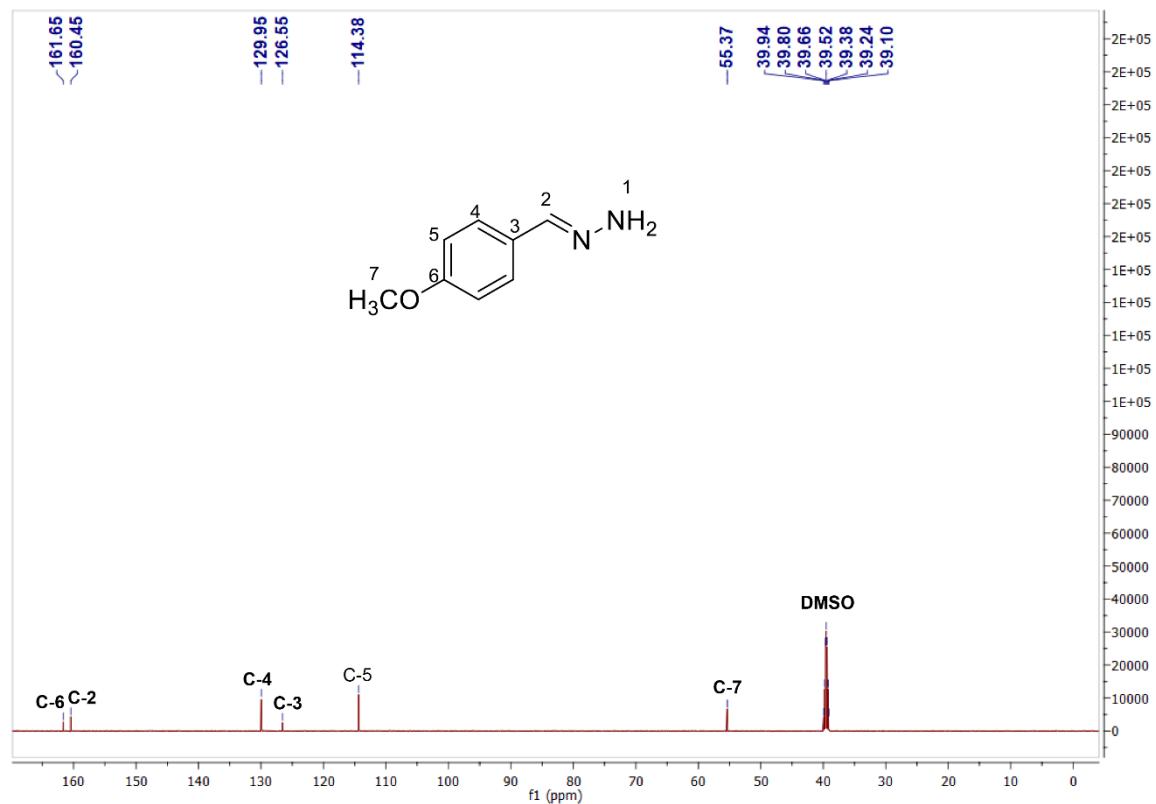


**(E)-(4-methoxybenzylidene)hydrazine (6)**

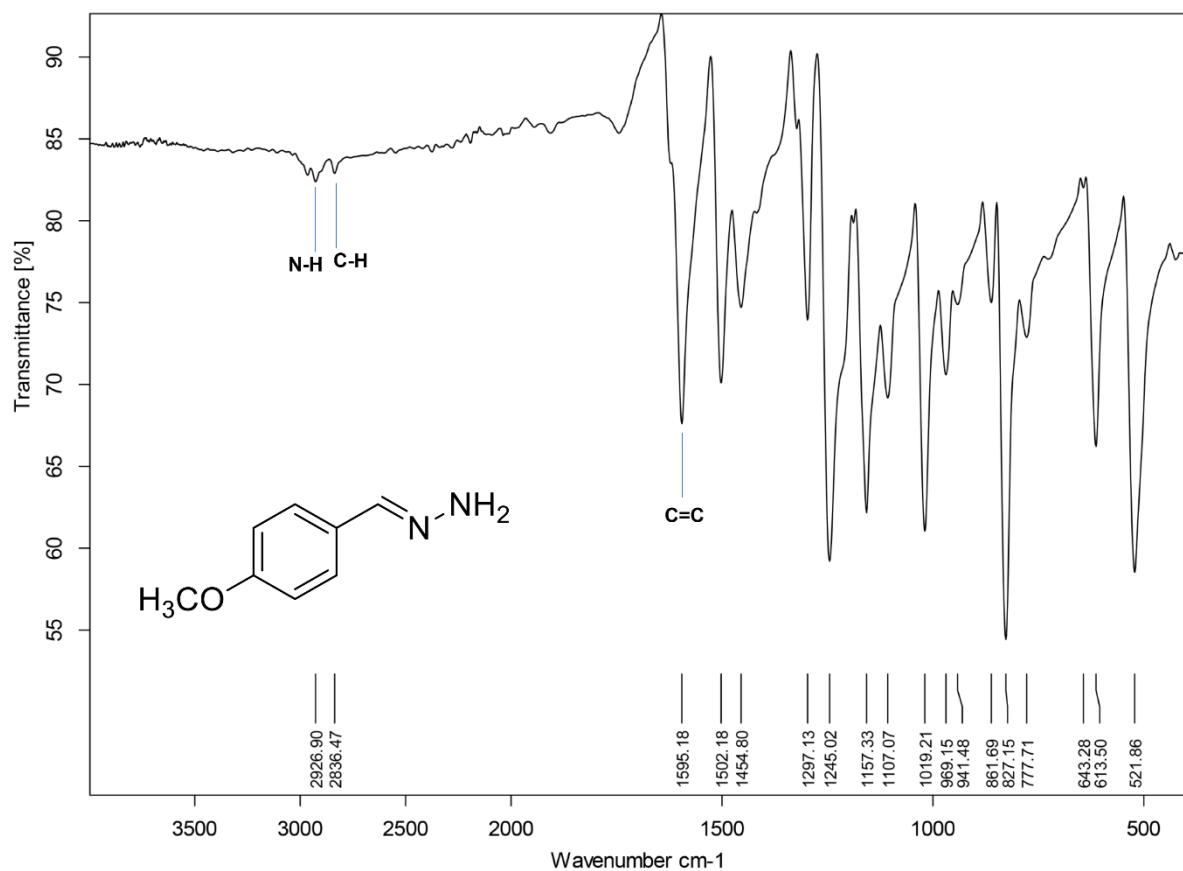
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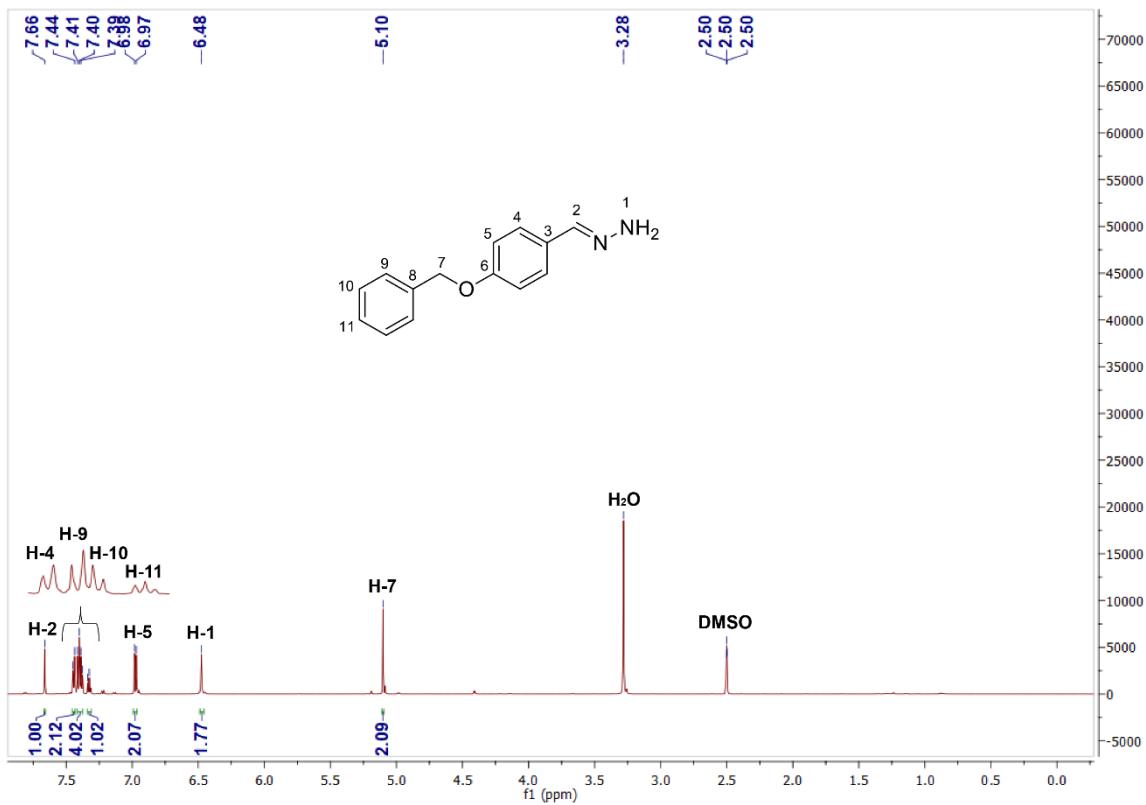


## IR Spectrum

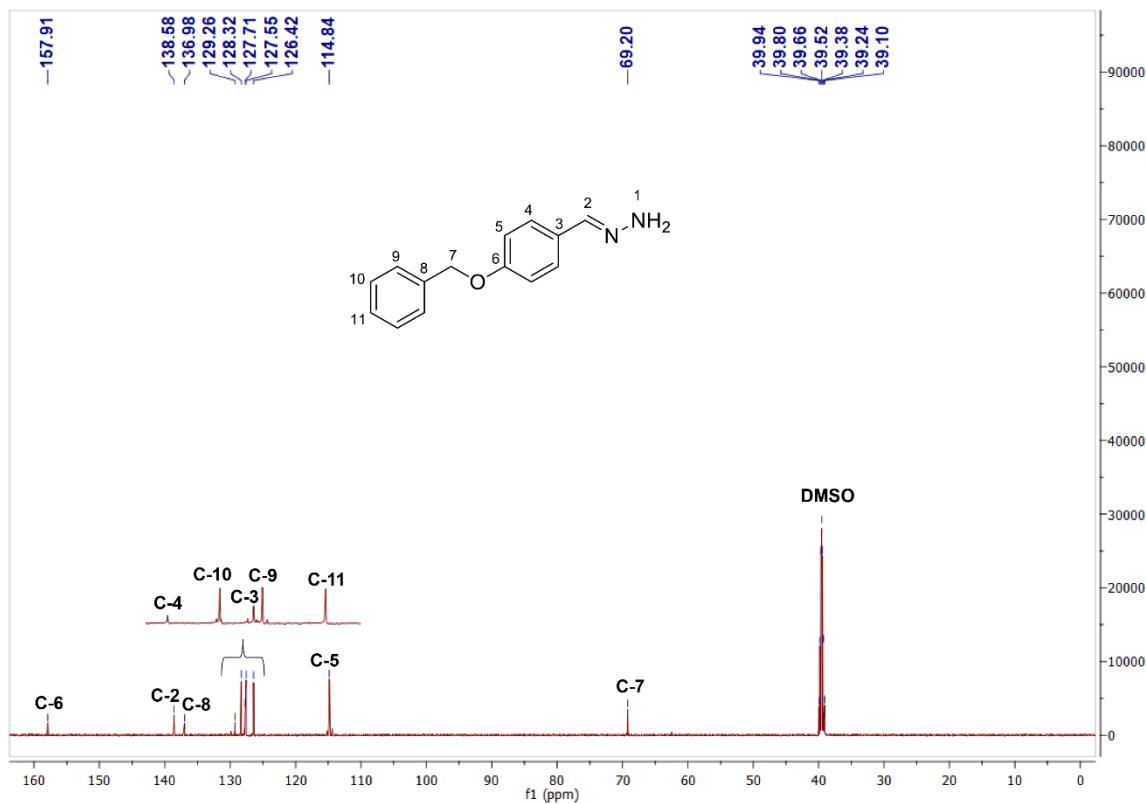


**(E)-(4-(benzyloxy)benzylidene)hydrazine (7)**

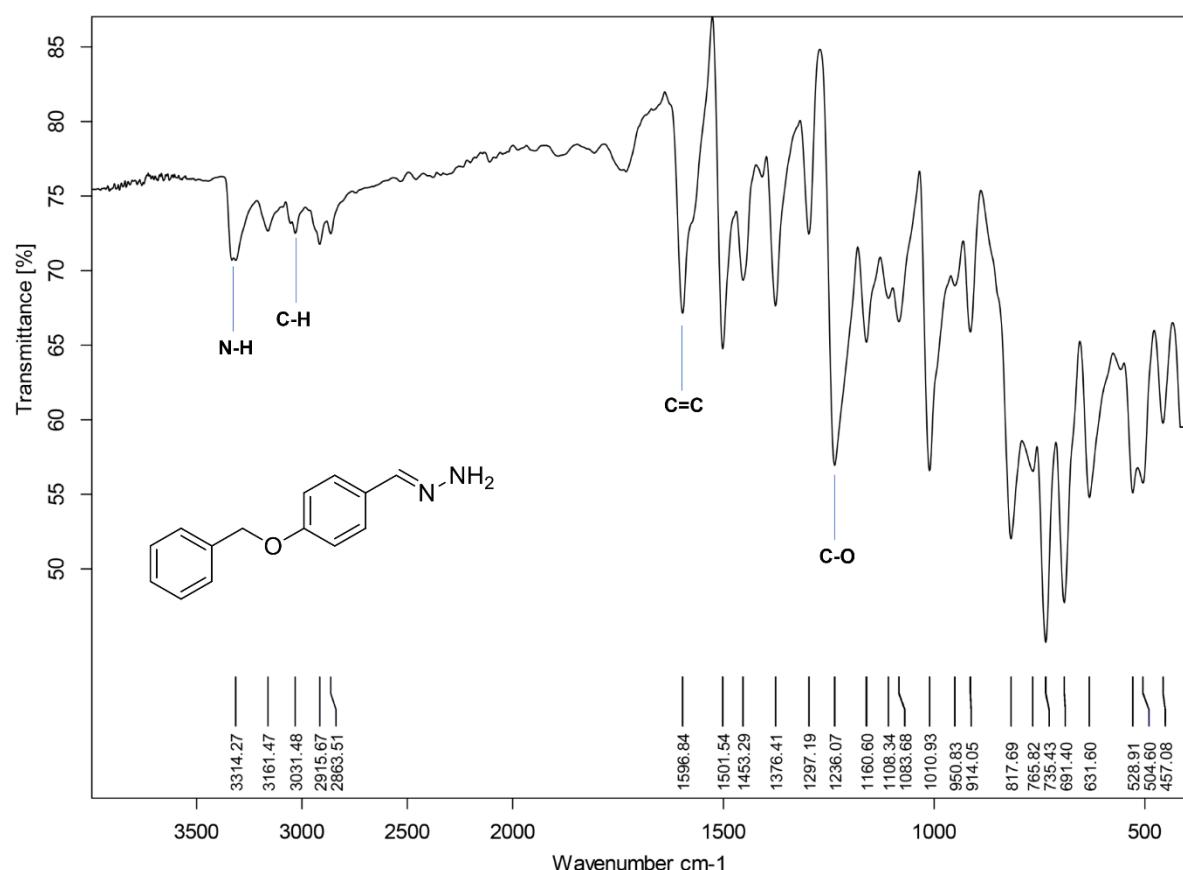
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**$^{13}\text{C}$  NMR in DMSO**

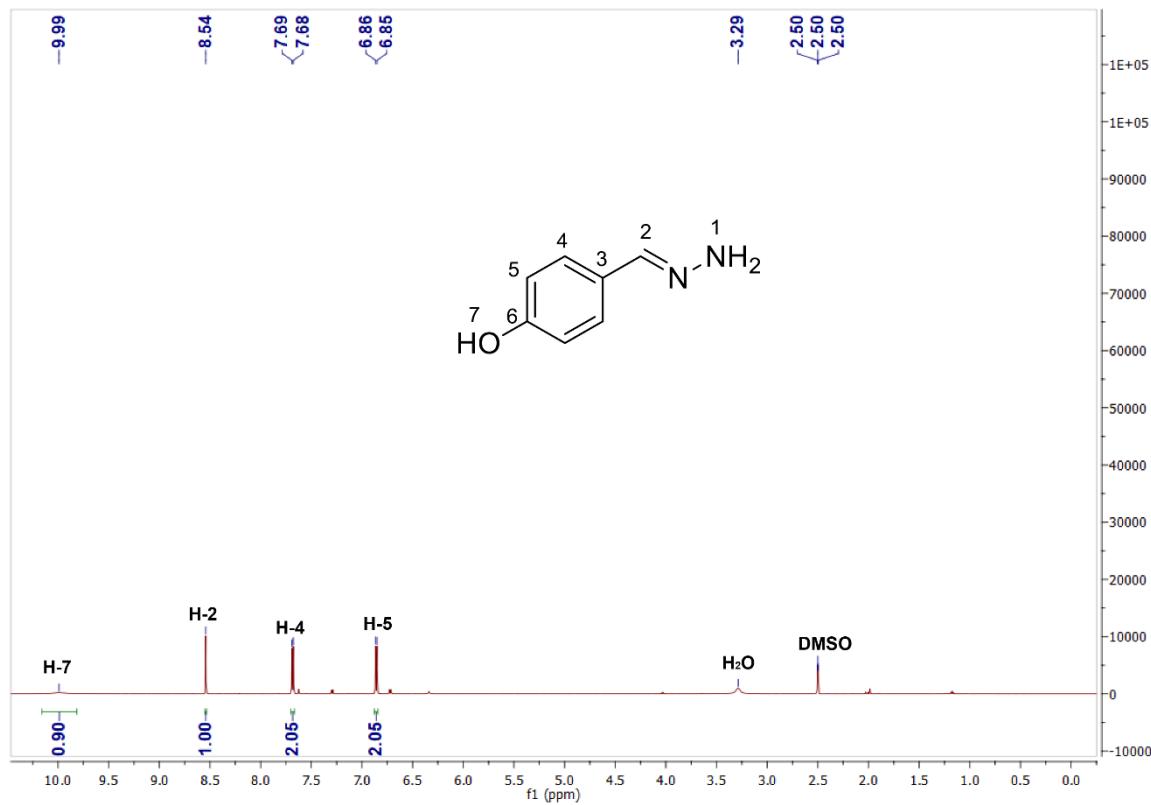


## IR Spectrum

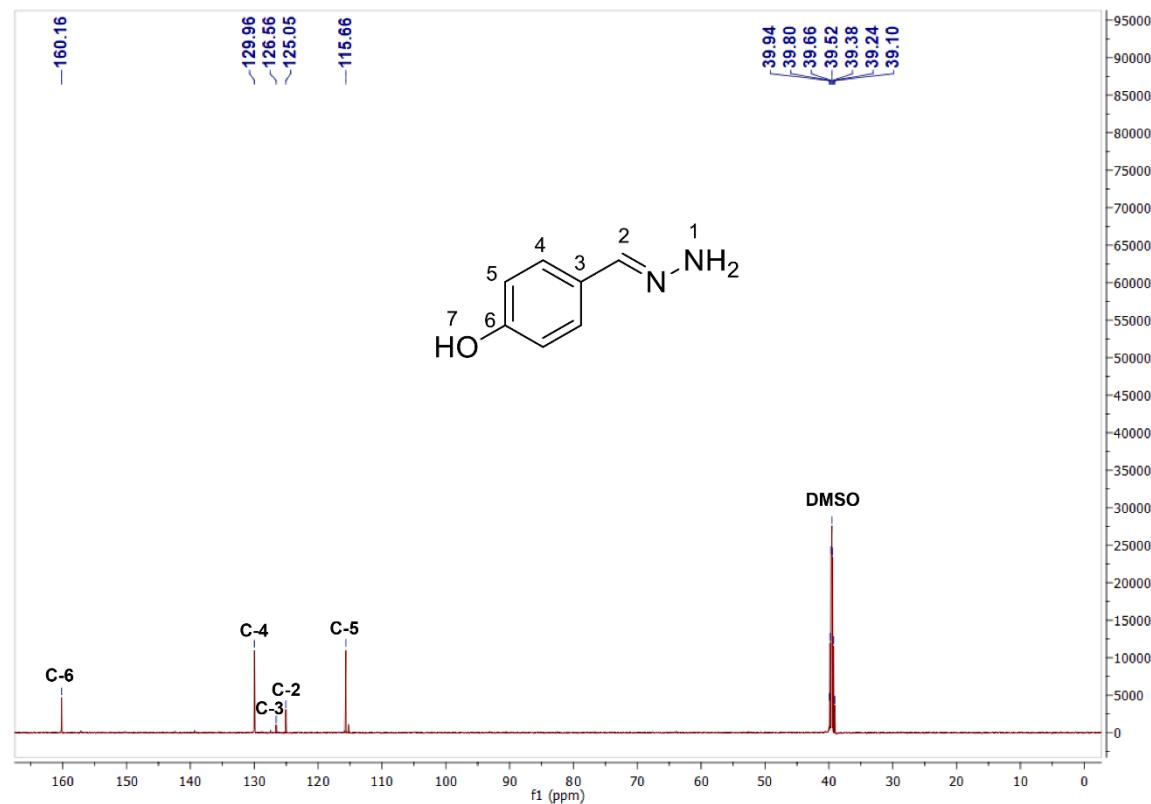


**(E)-4-(hydrazonomethyl)phenol (8)**

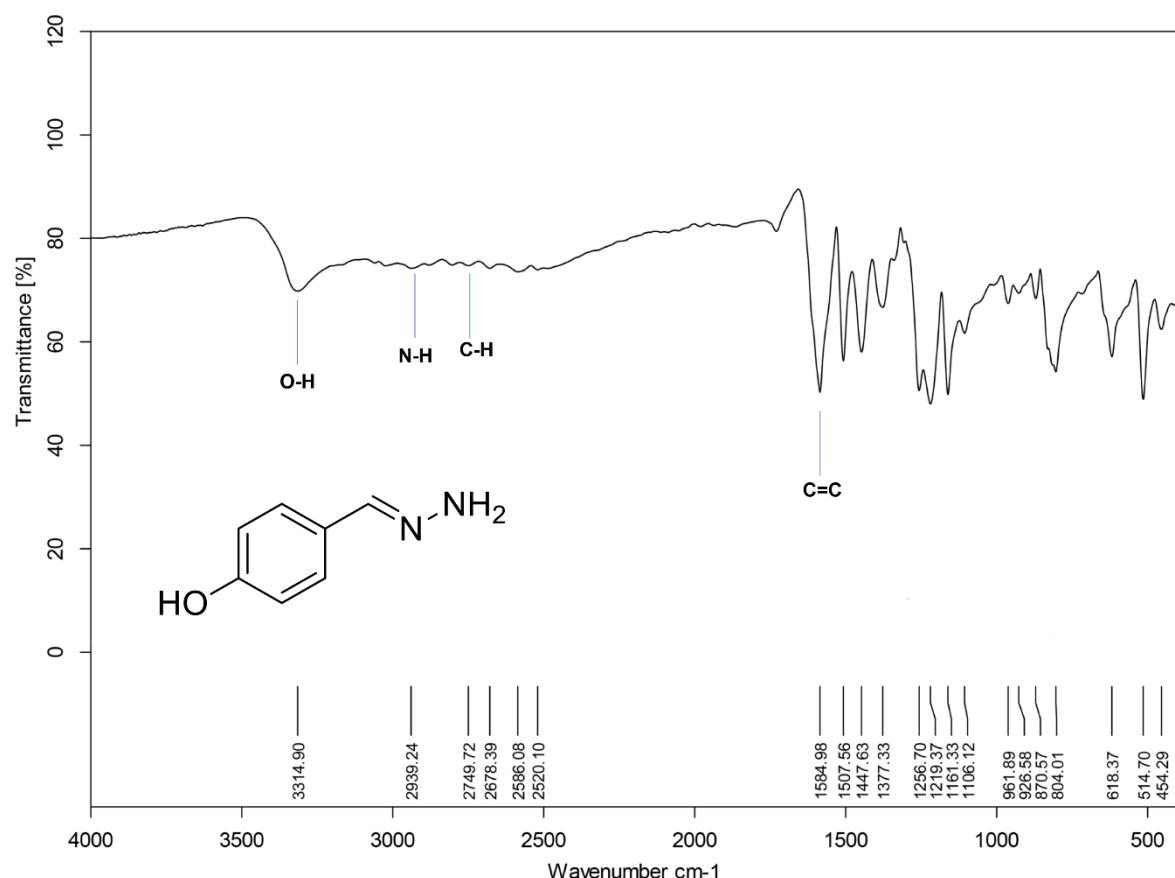
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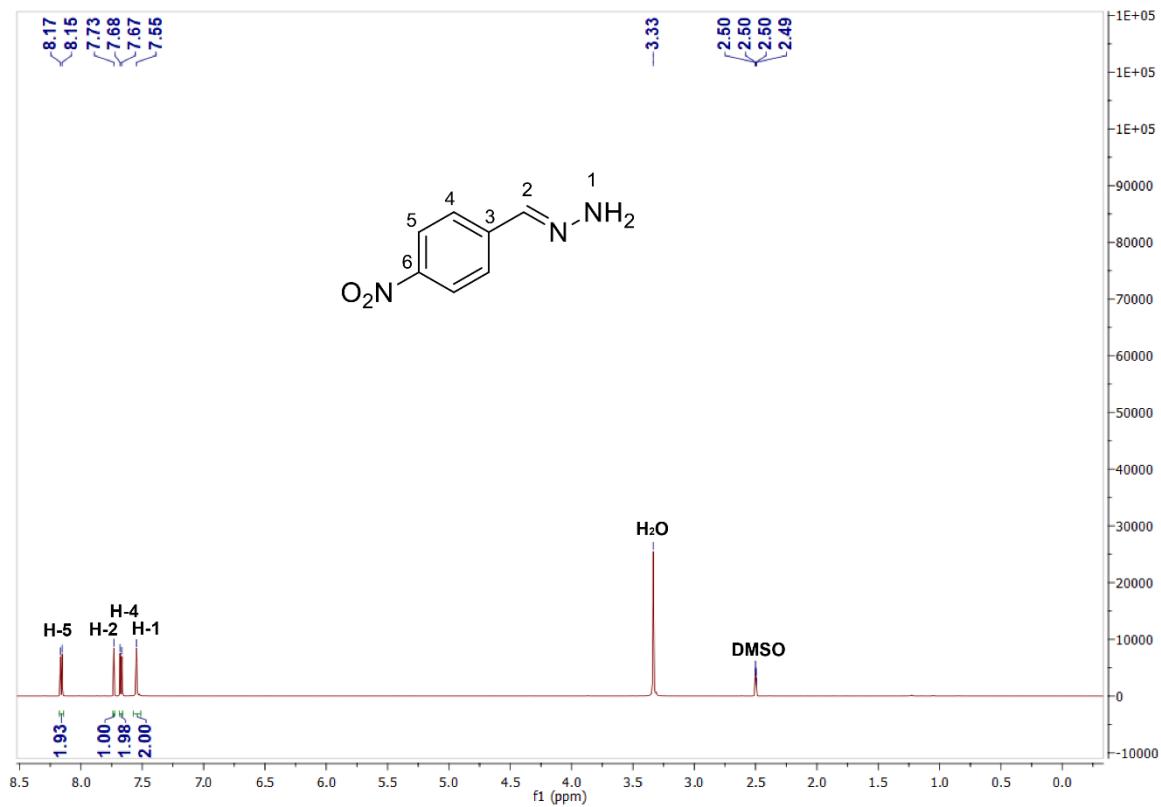


### IR Spectrum

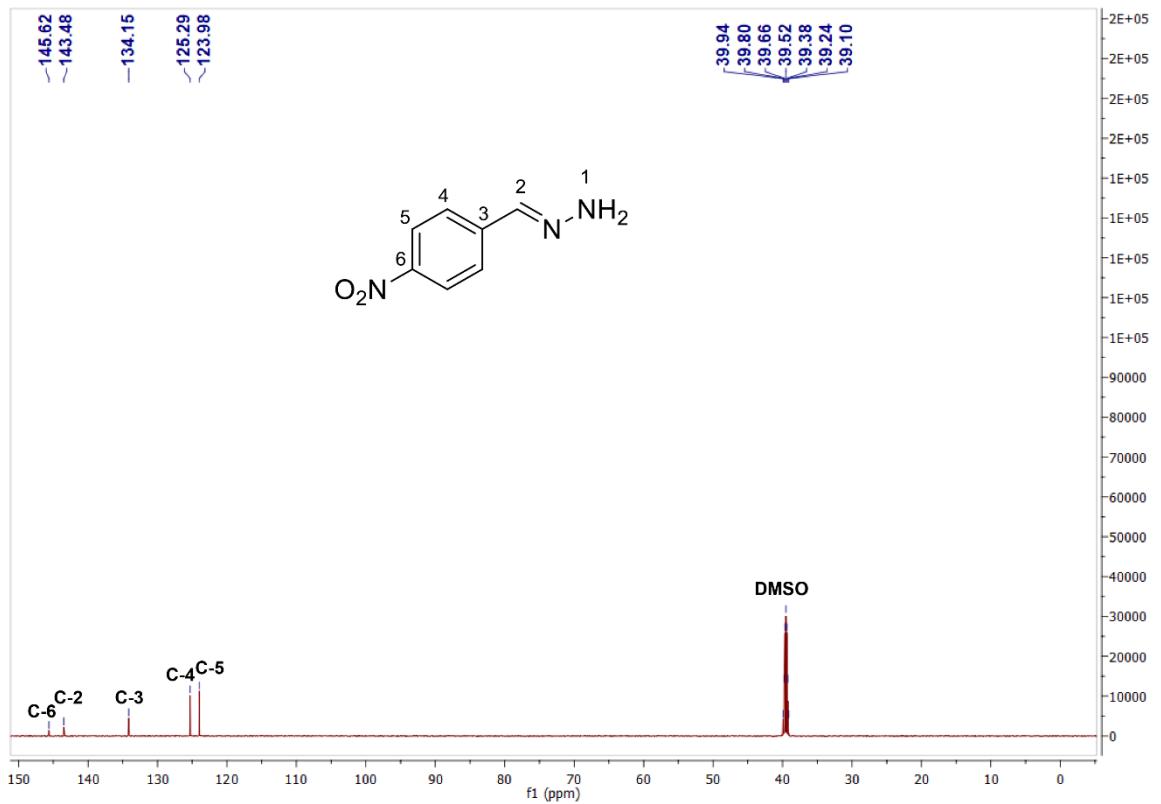


**(E)-(4-nitrobenzylidene)hydrazine (9)**

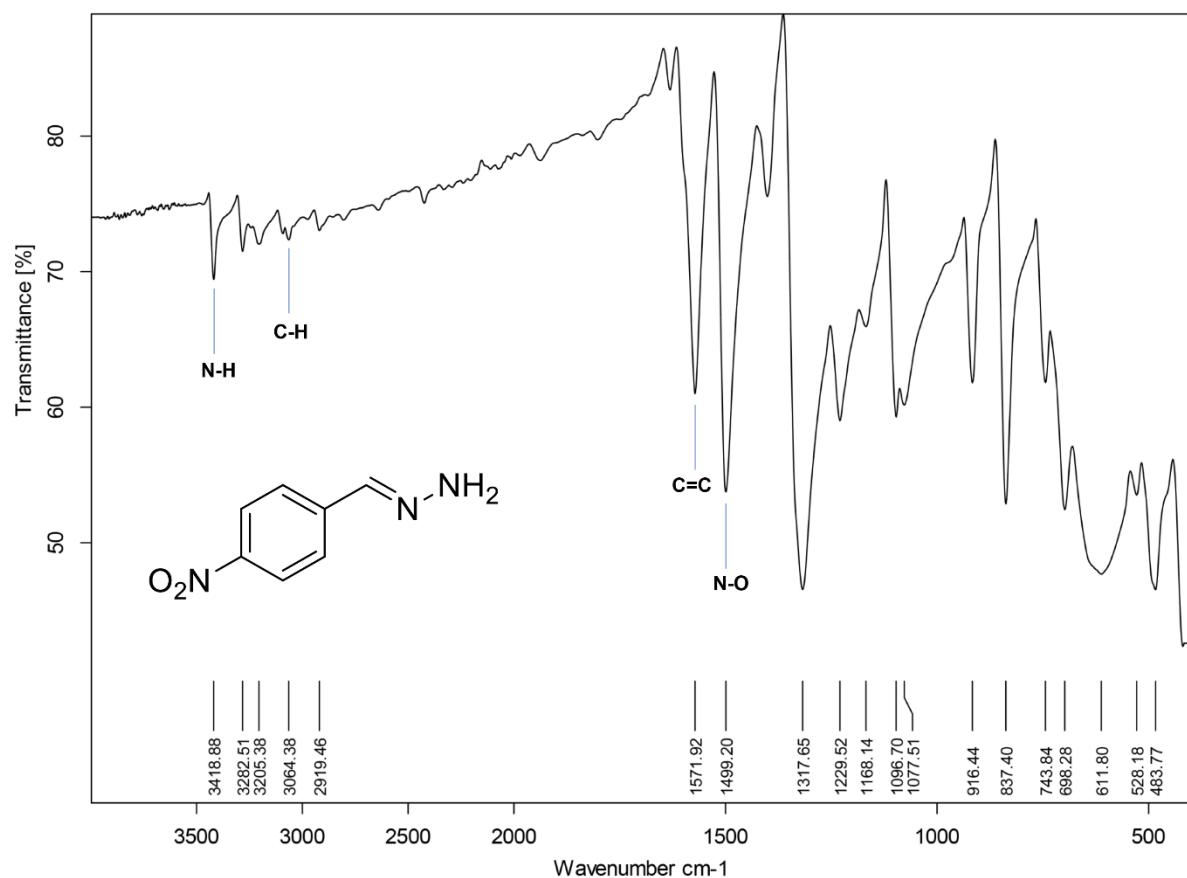
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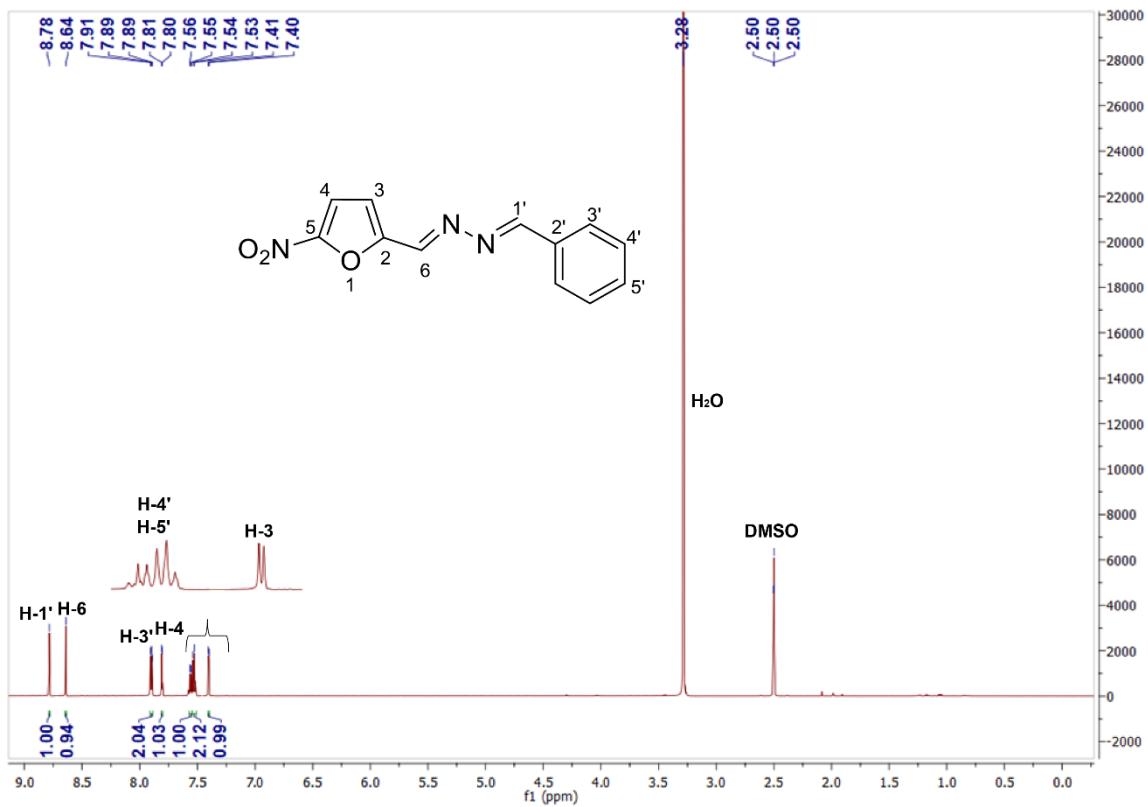


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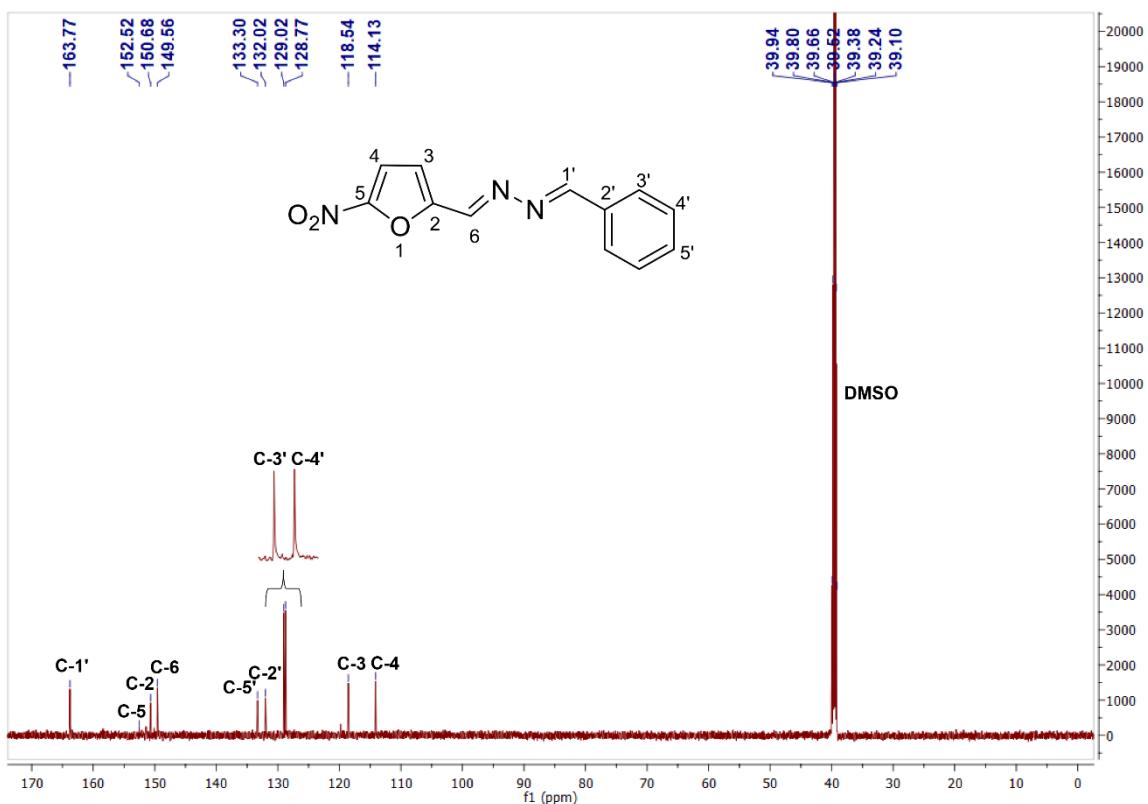


**(1E,2E)-1-benzylidene-2-([5-nitrofuran-2-yl]methylene)hydrazine (1a)**

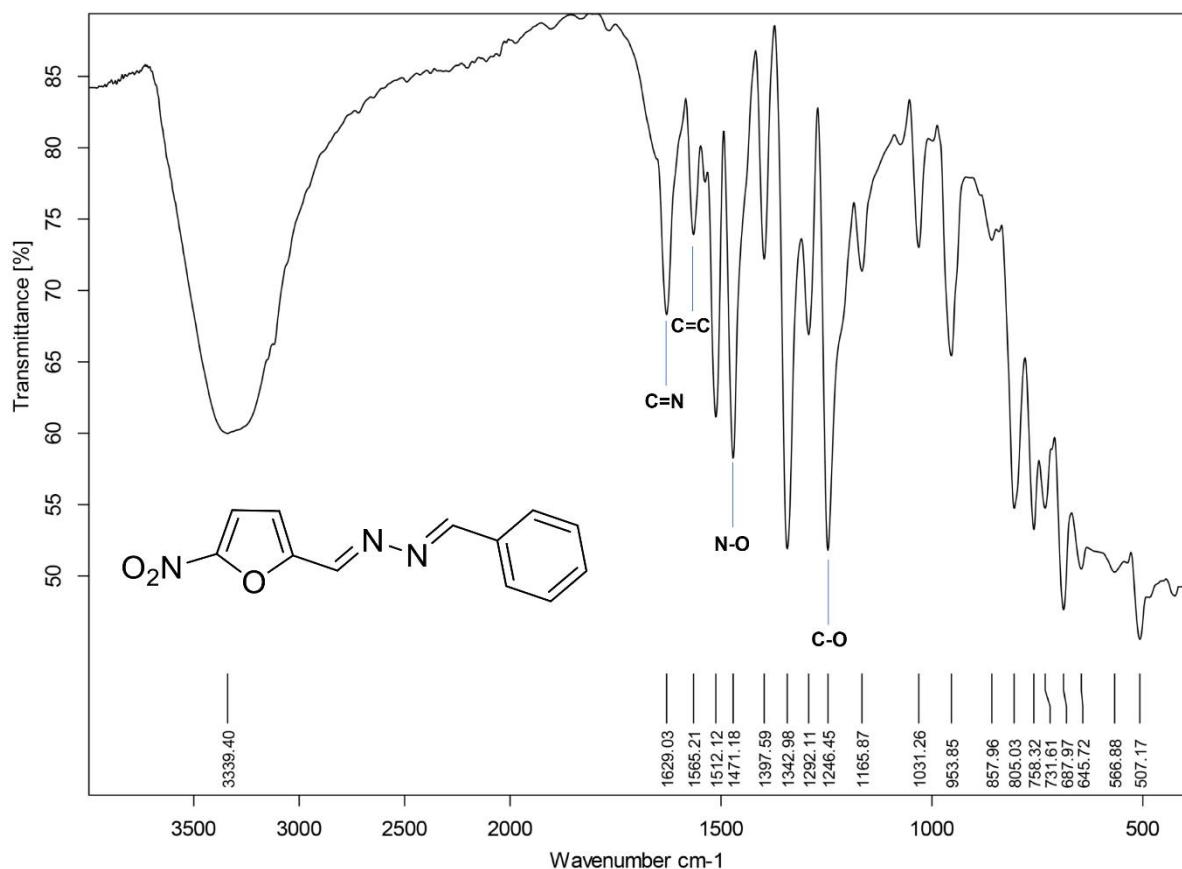
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**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

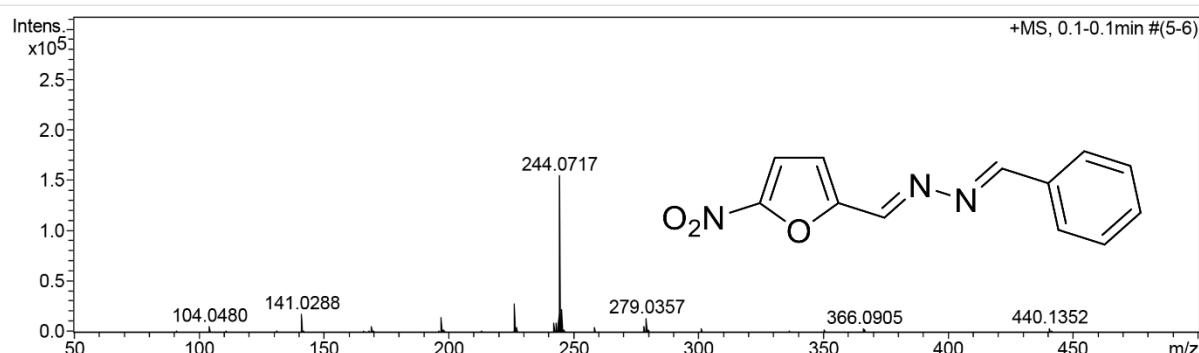
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000033.d       | Acquisition Date  | 10/12/2020 3:33:00 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-1                                 | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

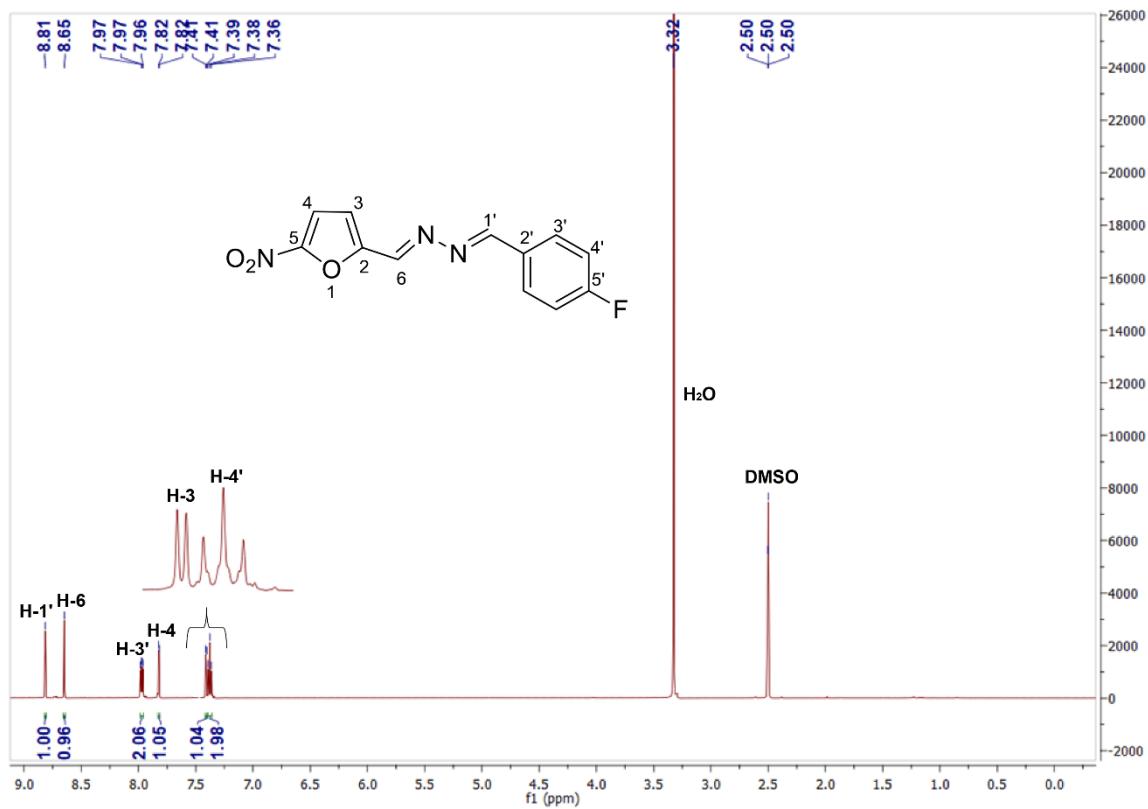
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



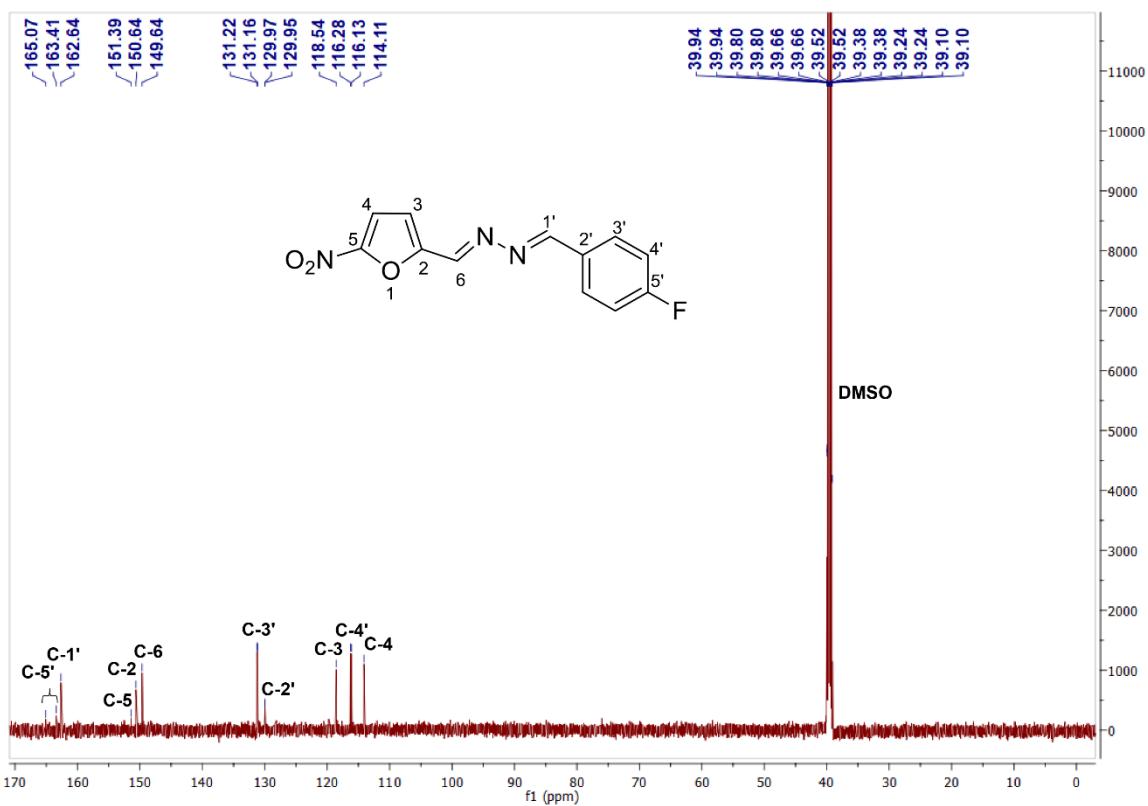
| Meas. m/z | # | Formula           | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|-------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 244.0717  | 1 | C 12 H 10 N 3 O 3 | 100.00 | 244.0717 | -0.1      | -0.3      | 2.1    | 9.5 | even    | ok     |

**(1E,2E)-1-(4-fluorobenzylidene)-2-([5-nitrofuran-2-yl]methylene)hydrazine (2a)**

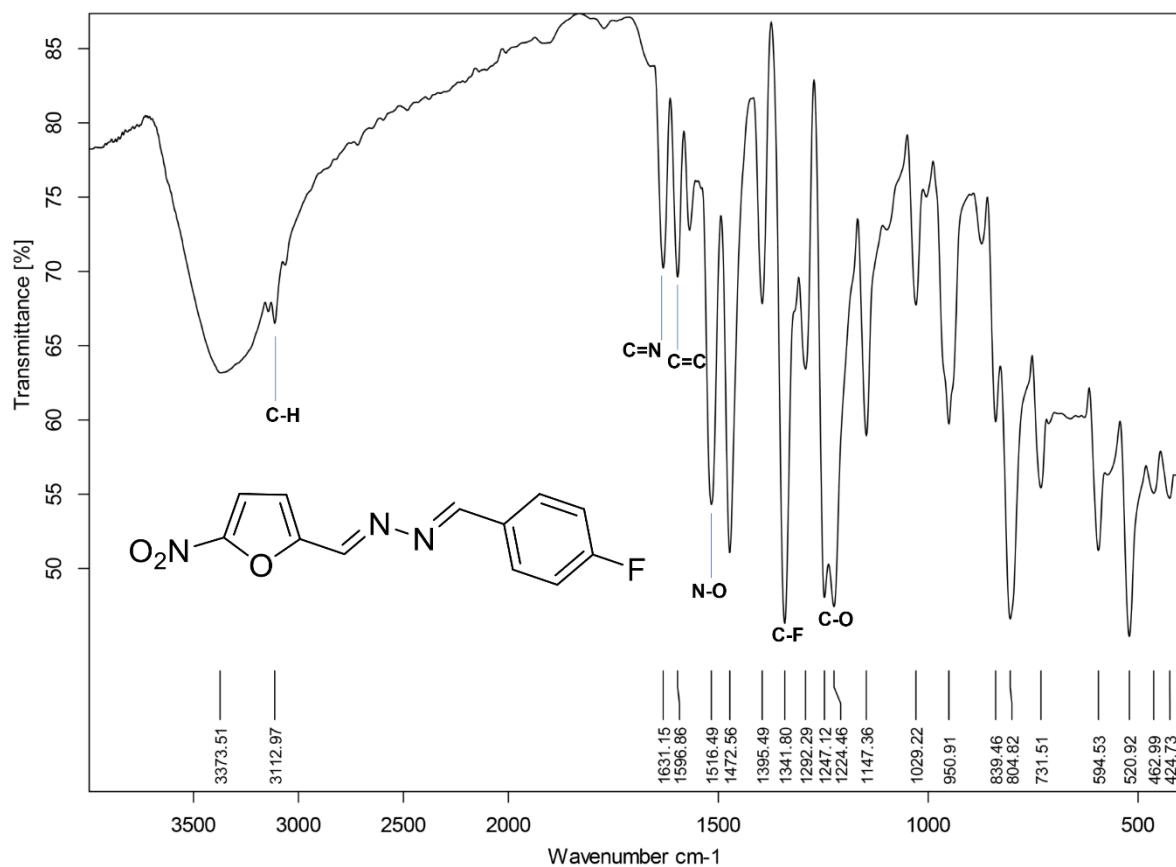
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

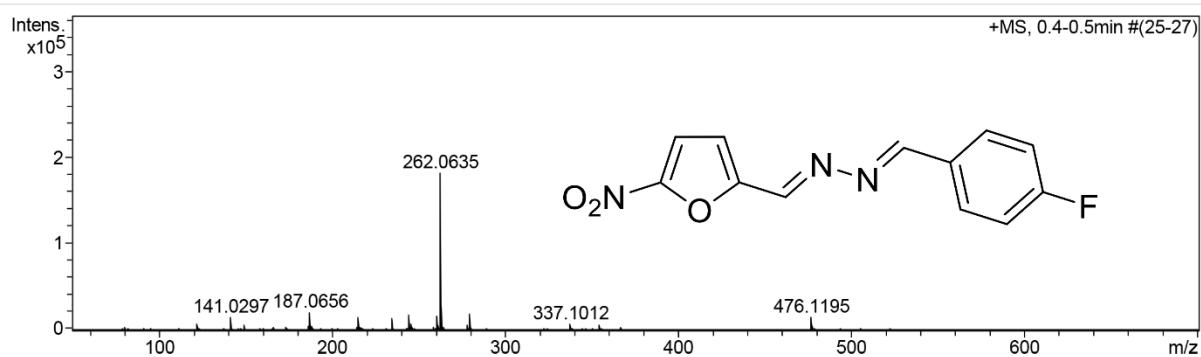
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000037.d       | Acquisition Date  | 10/12/2020 3:37:28 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-7                                 | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

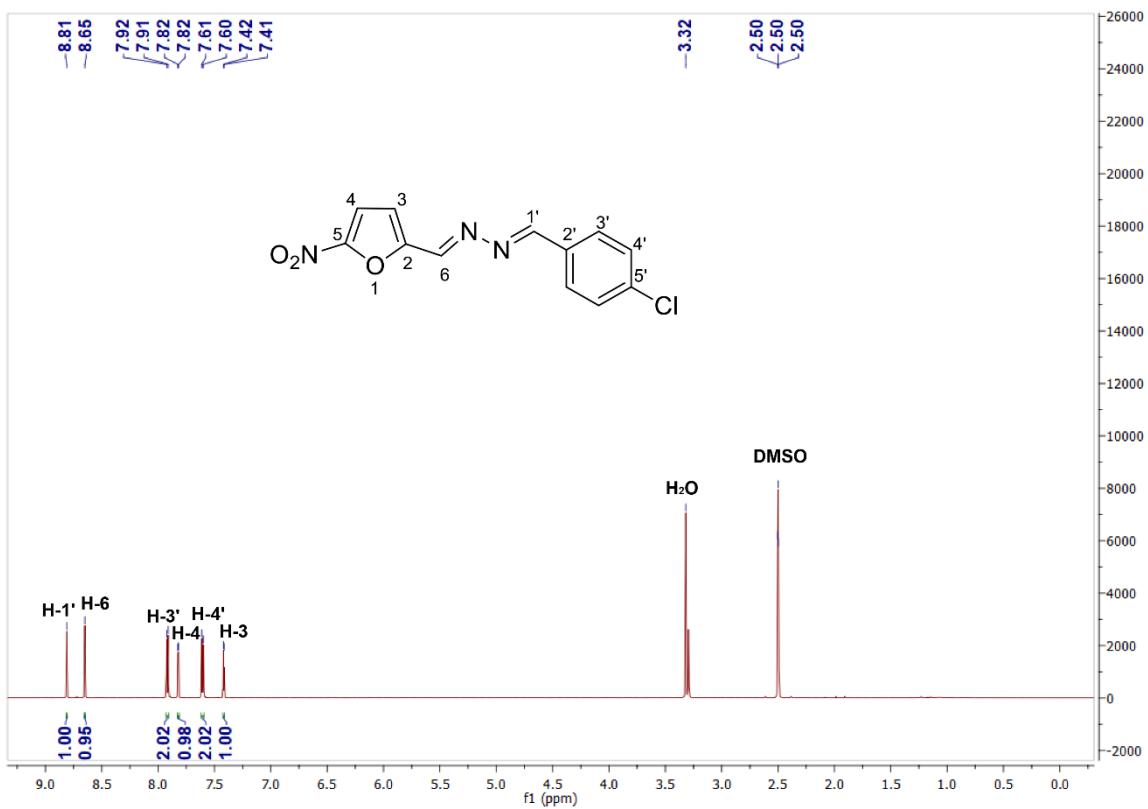
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



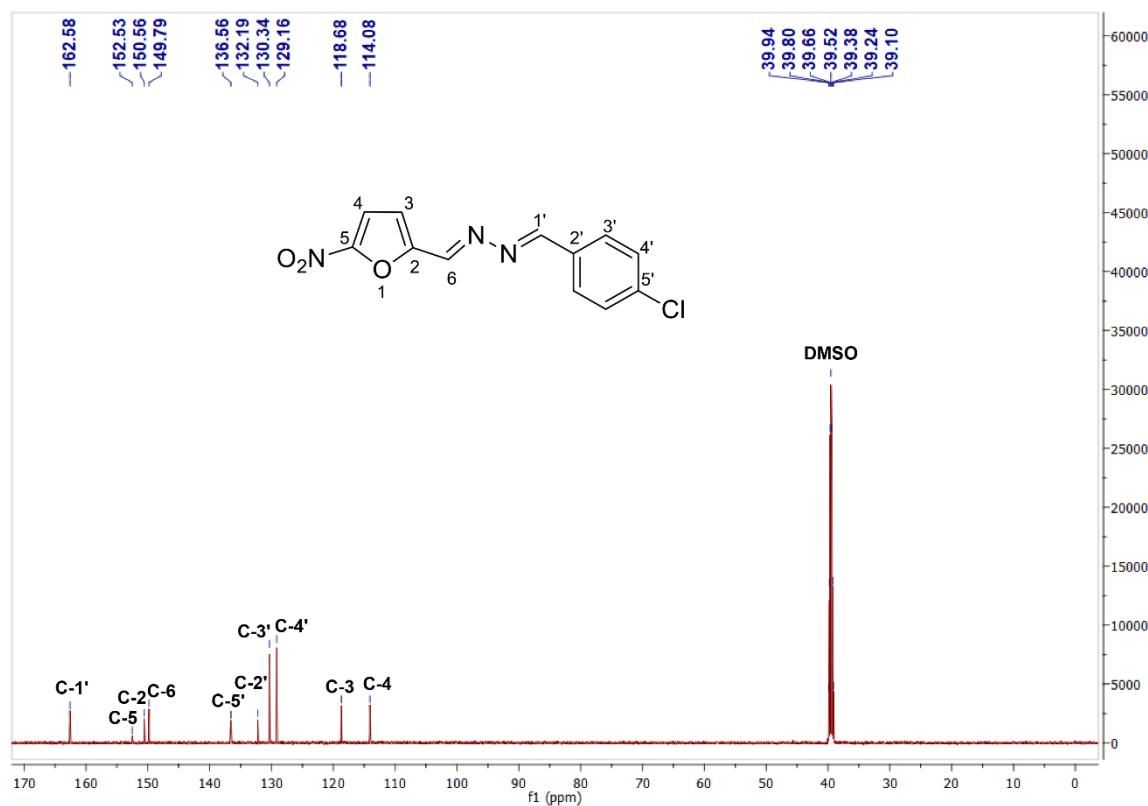
|           |   |   |        |          |           |           |            |     |            |        |
|-----------|---|---|--------|----------|-----------|-----------|------------|-----|------------|--------|
| Meas. m/z | # | Formula   | Score  | $m/z$    | err [mDa] | err [ppm] | $m/\sigma$ | rdb | $e^-$ Conf | N-Rule |
| 262.0635  | 1 | C <sub>12</sub> H <sub>9</sub> FN <sub>3</sub> O <sub>3</sub> | 100.00 | 262.0622 | -1.3      | -5.0      | 1.2        | 9.5 | even       | ok     |

**(1E,2E)-1-(4-chlorobenzylidene)-2-([5-nitrofuran-2-yl]methylene)hydrazine (3a)**

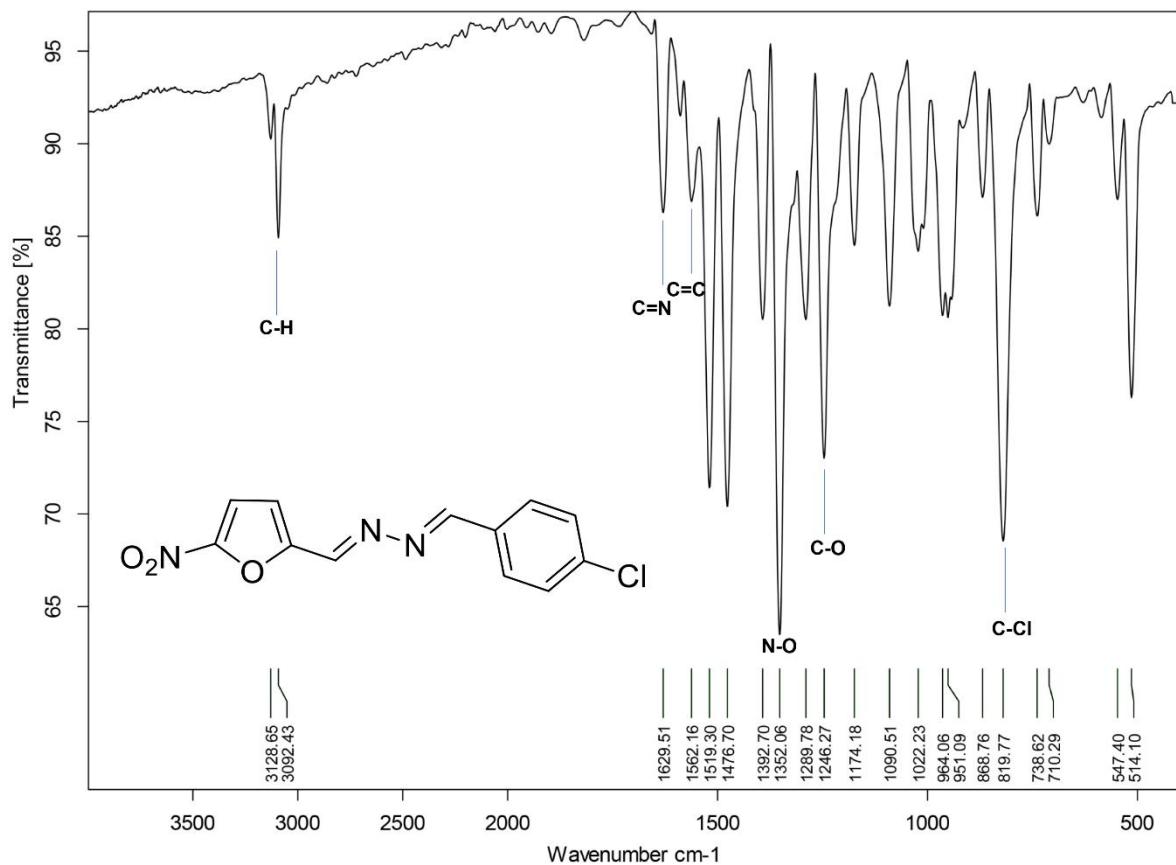
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR spectrum



## HRMS

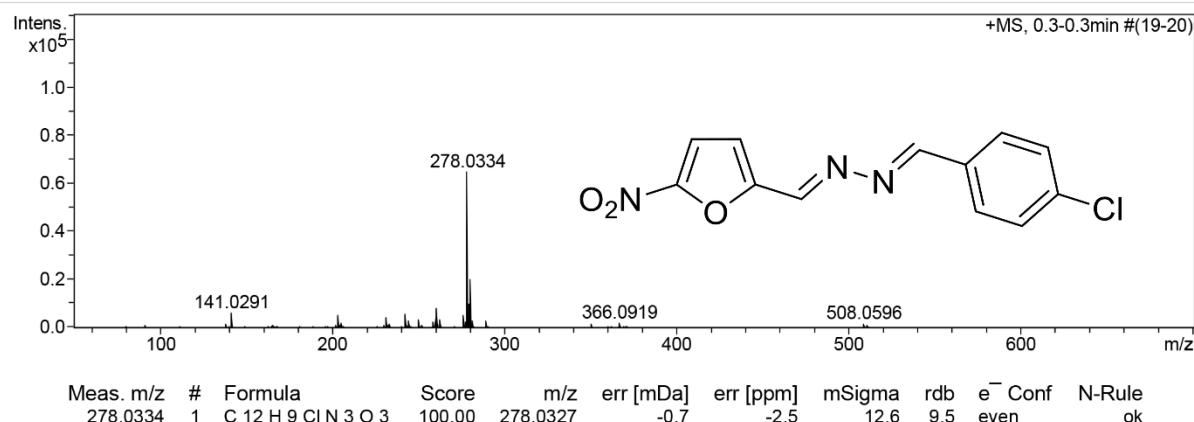
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000035.d       | Acquisition Date  | 10/12/2020 3:35:23 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-3                                 | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

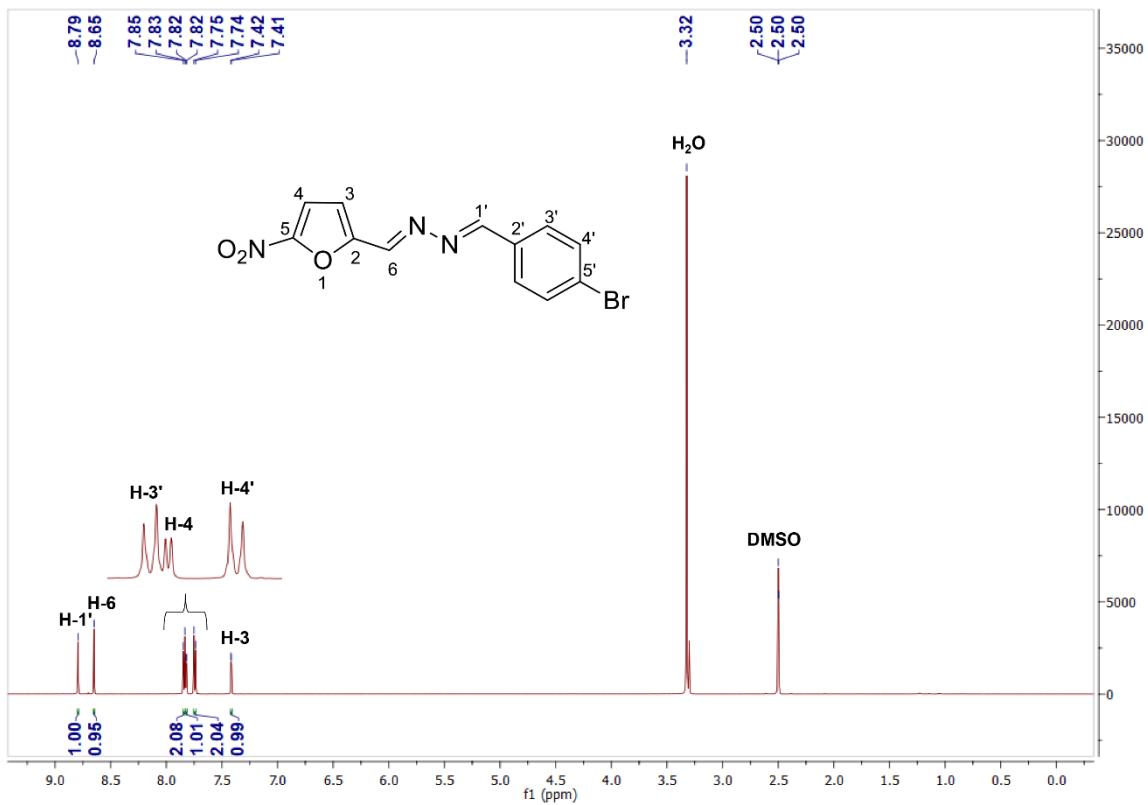
#### Acquisition Parameter

|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |

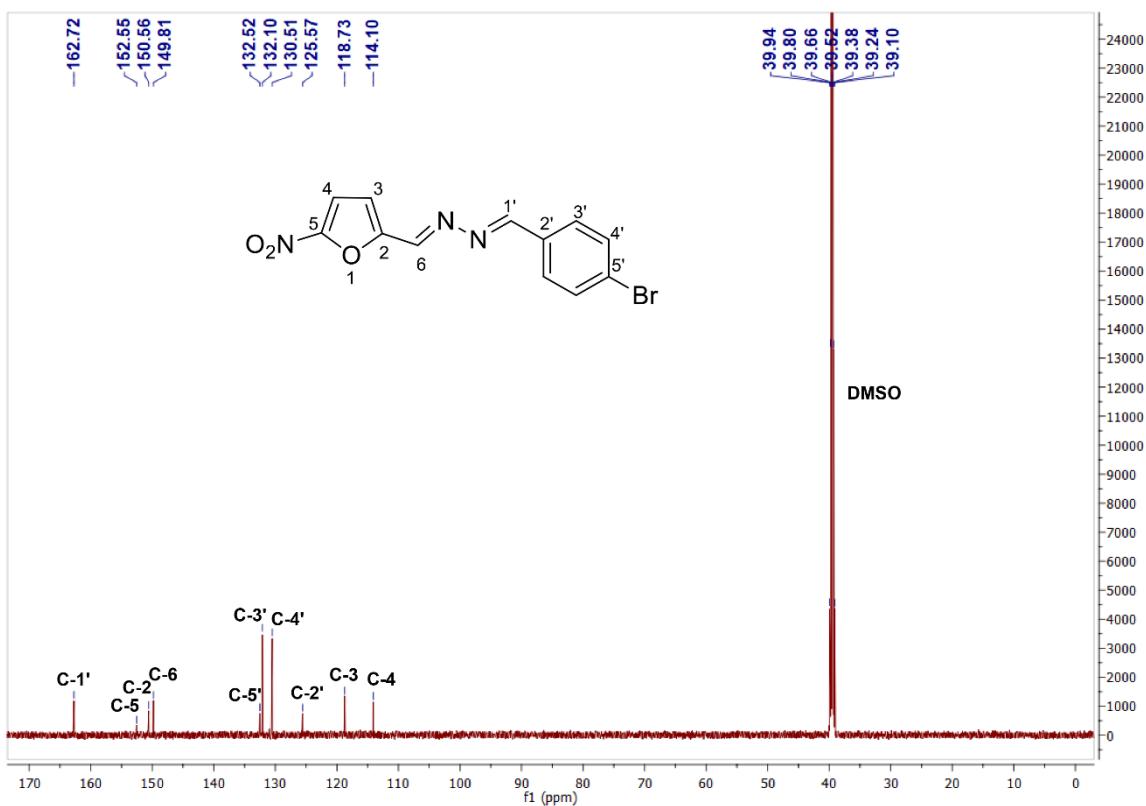


**(1E,2E)-1-(4-bromobenzylidene)-2-([5-nitrofuran-2-yl]methylene)hydrazine (4a)**

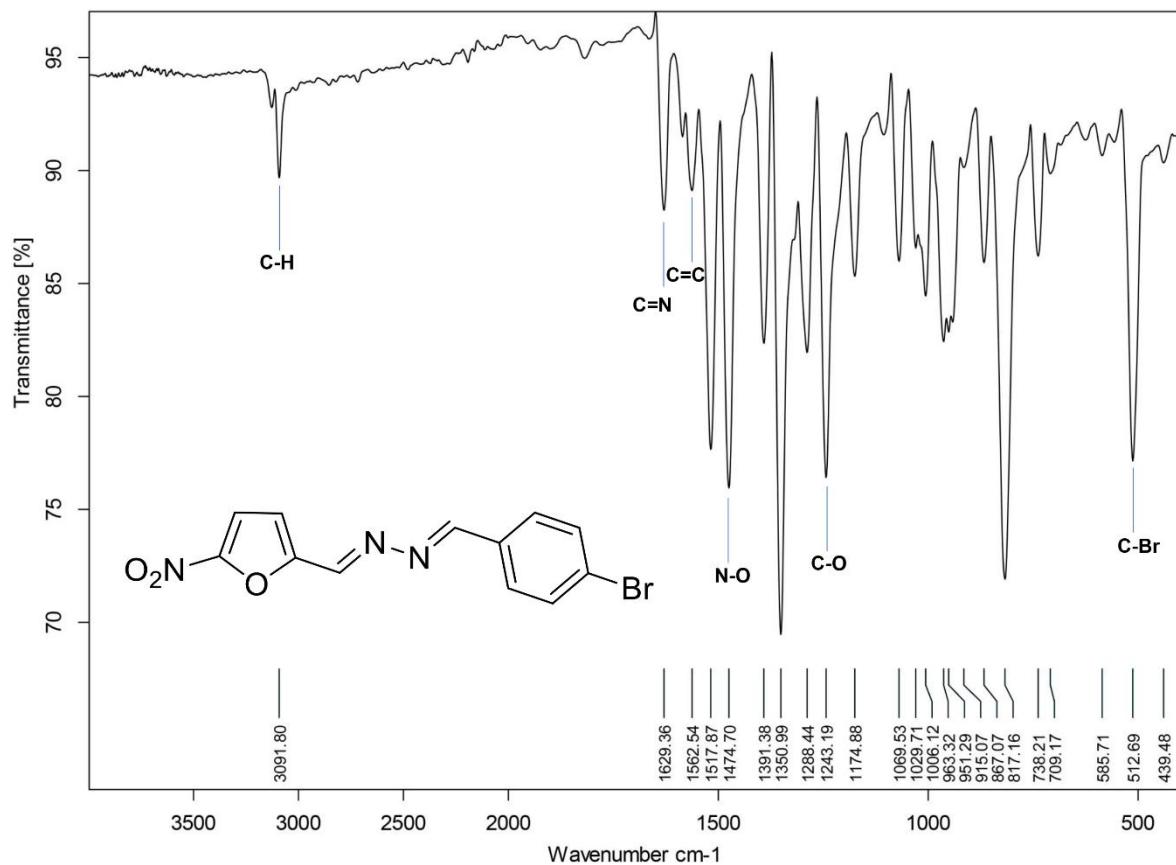
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

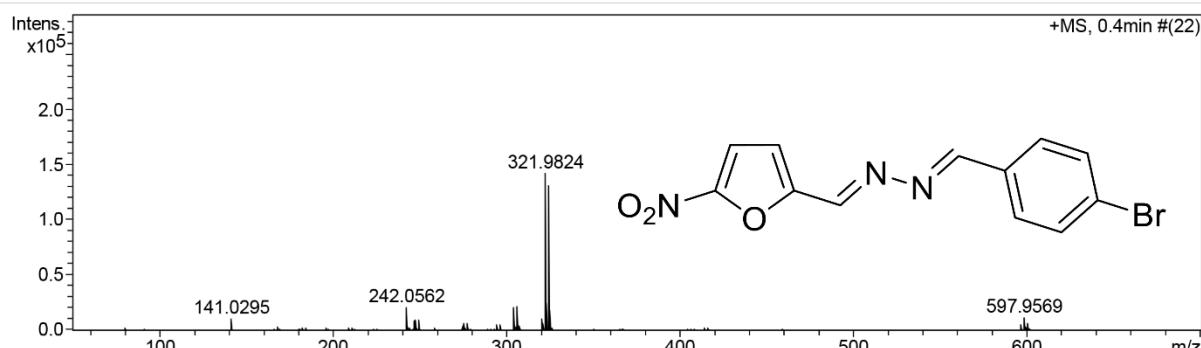
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000036.d       | Acquisition Date  | 10/12/2020 3:36:35 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-4                                 | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

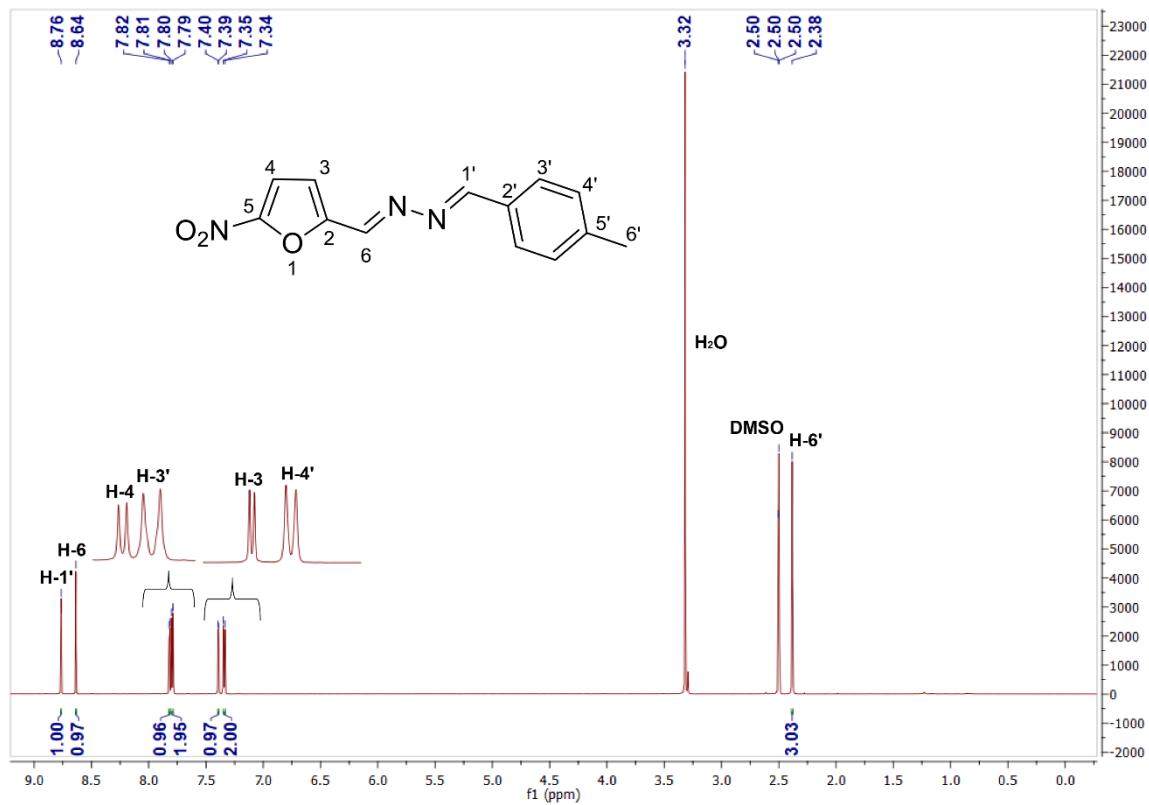
#### Acquisition Parameter

|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |

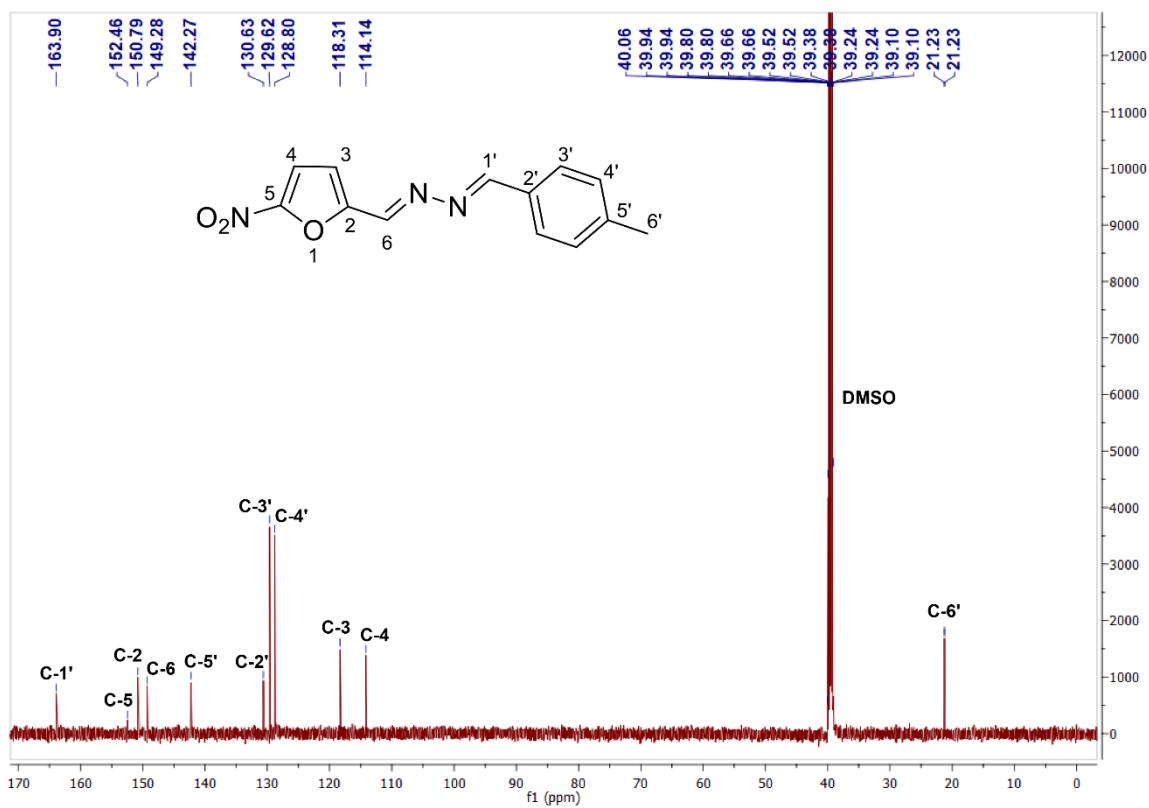


**(1E,2E)-1-(4-methylbenzylidene)-2-([5-nitrofuran-2-yl]methylene)hydrazine (5a)**

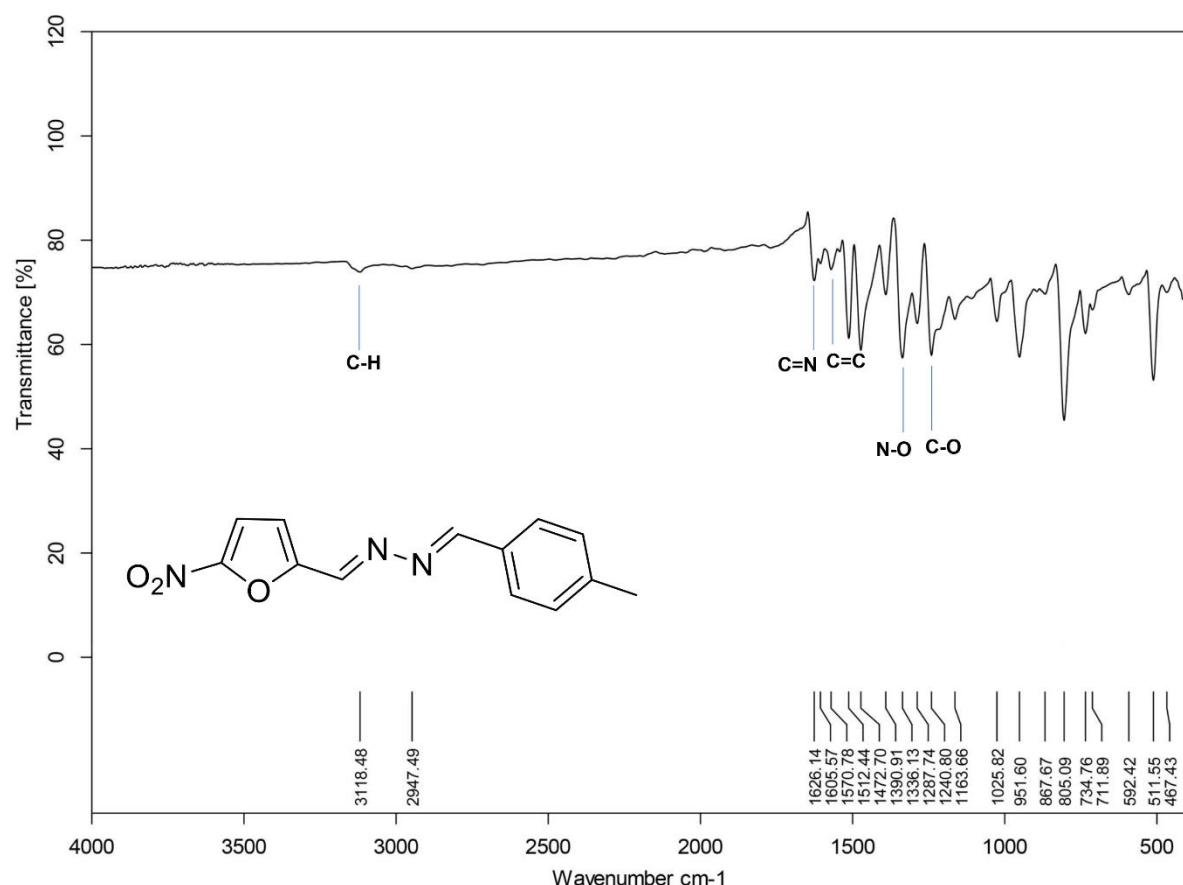
**<sup>1</sup>H NMR in DMSO**



**<sup>13</sup>C NMR in DMSO**



## IR Spectrum



## HRMS

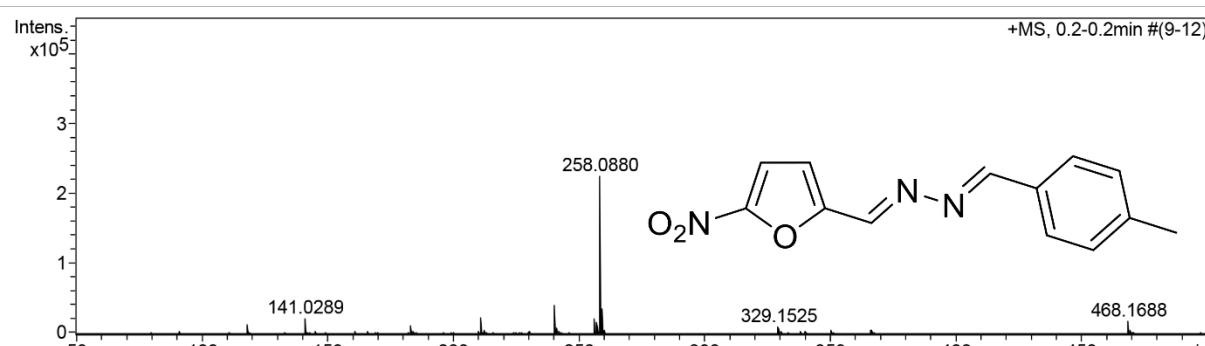
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000031.d       | Acquisition Date  | 10/12/2020 3:30:31 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-5                                 | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

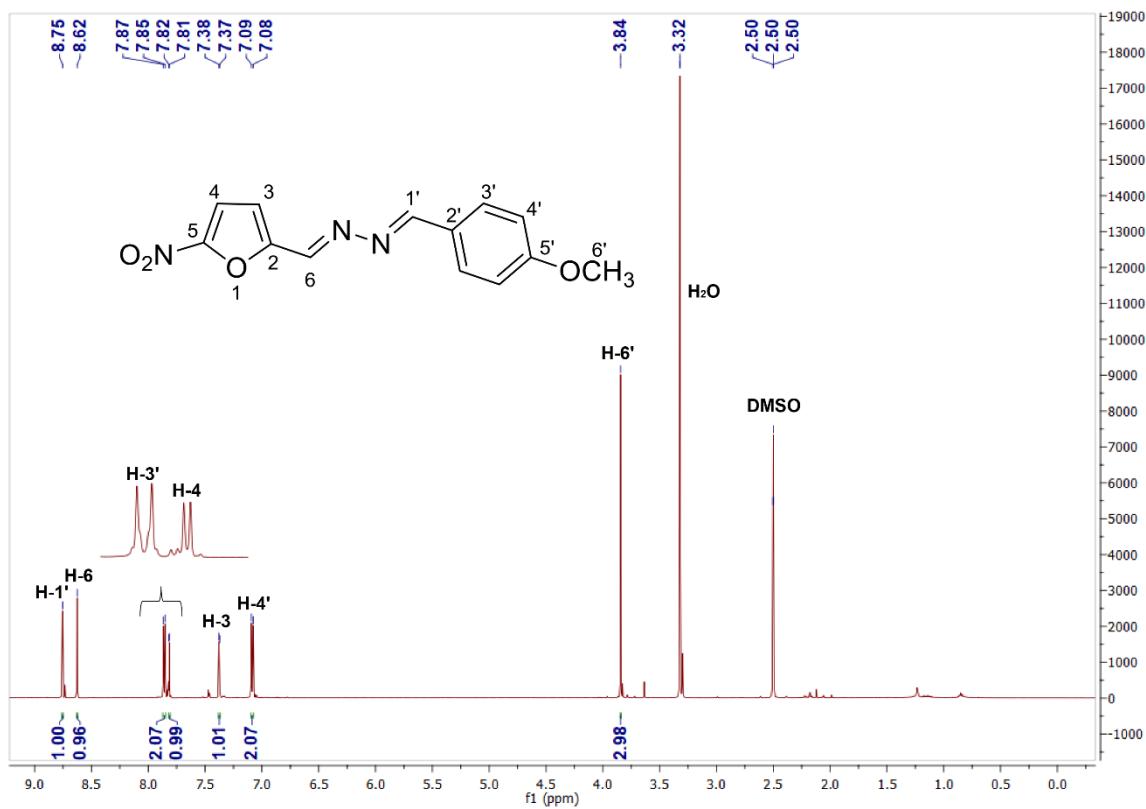
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



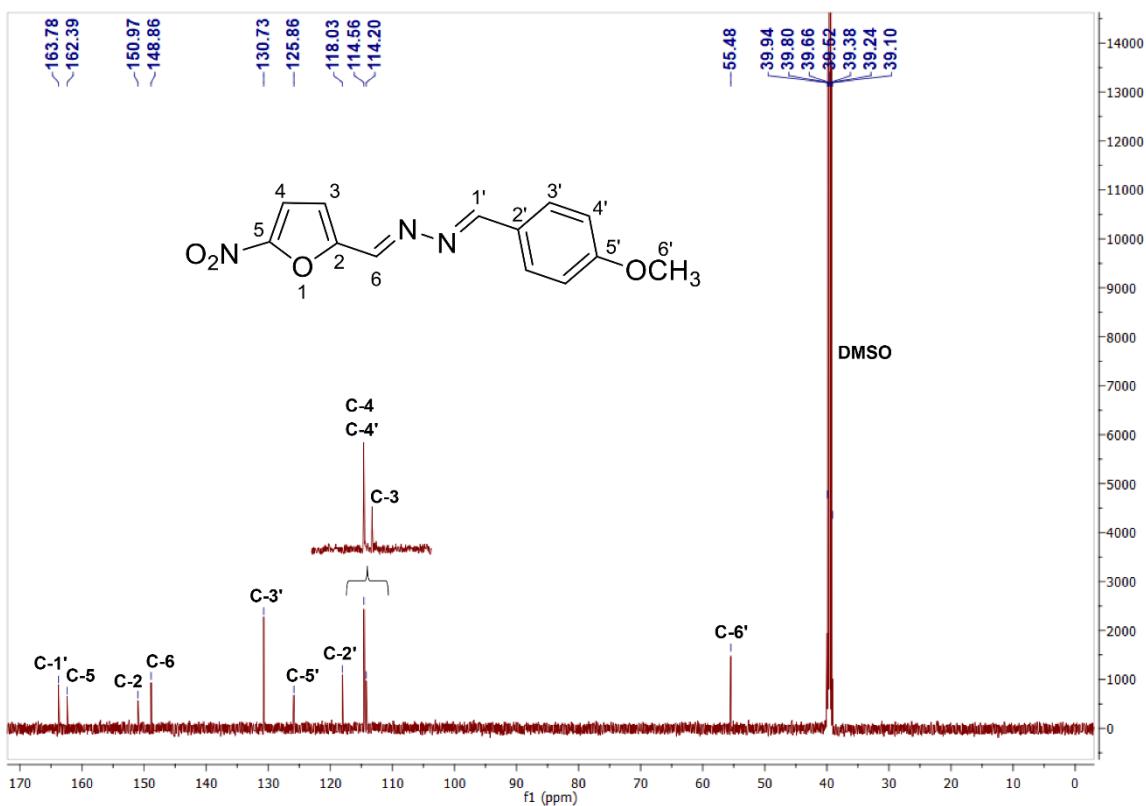
|           |   |                   |        |          |           |           |        |     |                     |        |
|-----------|---|-------------------|--------|----------|-----------|-----------|--------|-----|---------------------|--------|
| Meas. m/z | # | Formula           | Score  | m/z      | err [mDa] | err [ppm] | mSigma | ldb | e <sup>-</sup> Conf | N-Rule |
| 258.0880  | 1 | C 13 H 12 N 3 O 3 | 100.00 | 258.0873 | -0.6      | -2.5      | 4.4    | 9.5 | even                | ok     |

**(1E,2E)-1-(4-methoxybenzylidene)-2-([5-nitrofuran-2-yl]methylene)hydrazine (6a)**

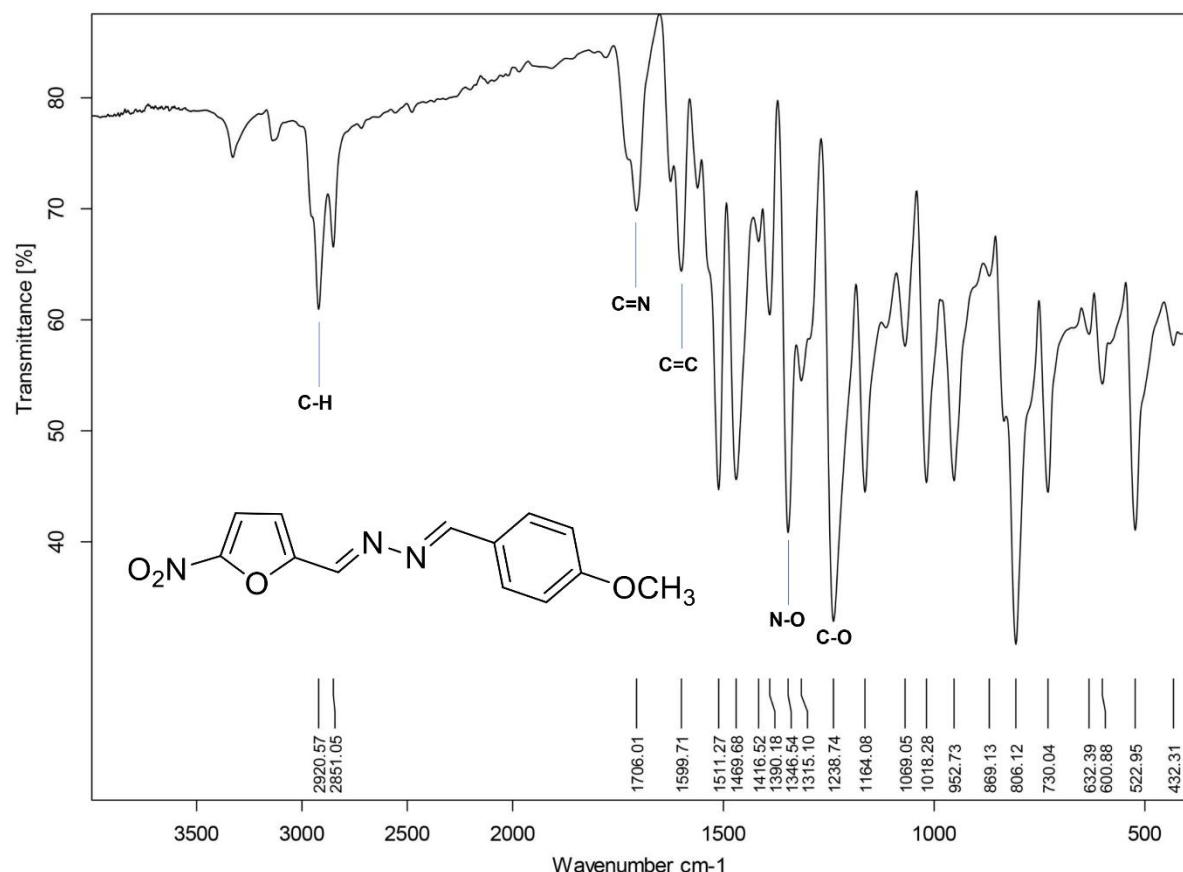
**<sup>1</sup>H NMR in DMSO**



**<sup>13</sup>C NMR in DMSO**



## IR Spectrum



## HRMS

### Mass Spectrum SmartFormula Report

#### Analysis Info

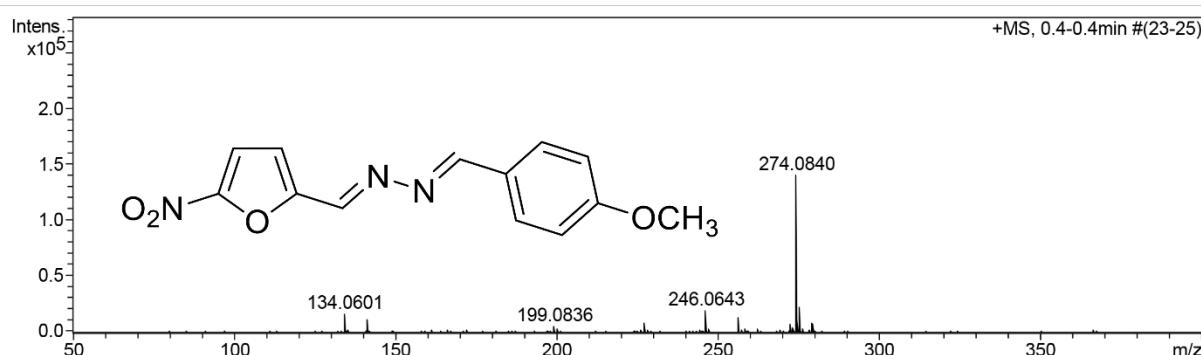
Analysis Name: D:\Data\12102020\ADMS000038.d  
 Method: tune\_low no focus50-1600da12102020.m  
 Sample Name: MV-6  
 Comment:

Acquisition Date: 10/12/2020 3:38:40 PM

Operator: Dr JHL Jordaan  
 Instrument / Ser#: micrOTOF-Q II 2010390

#### Acquisition Parameter

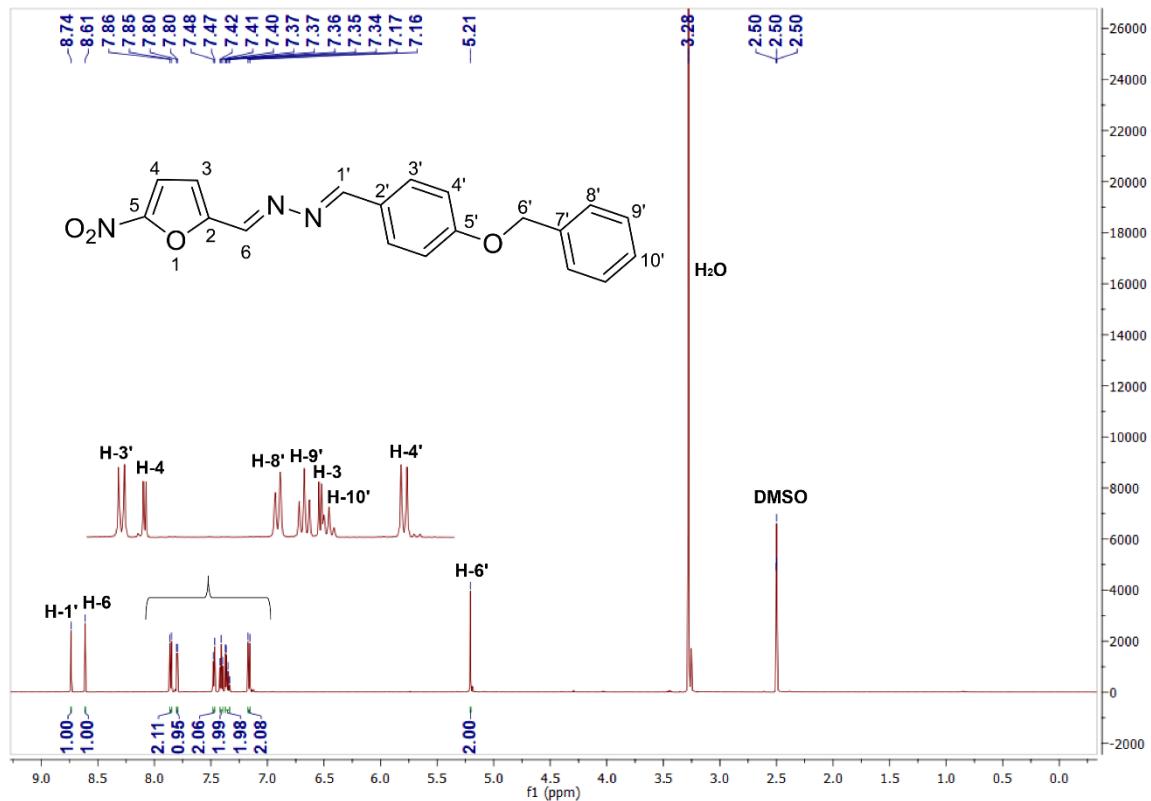
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



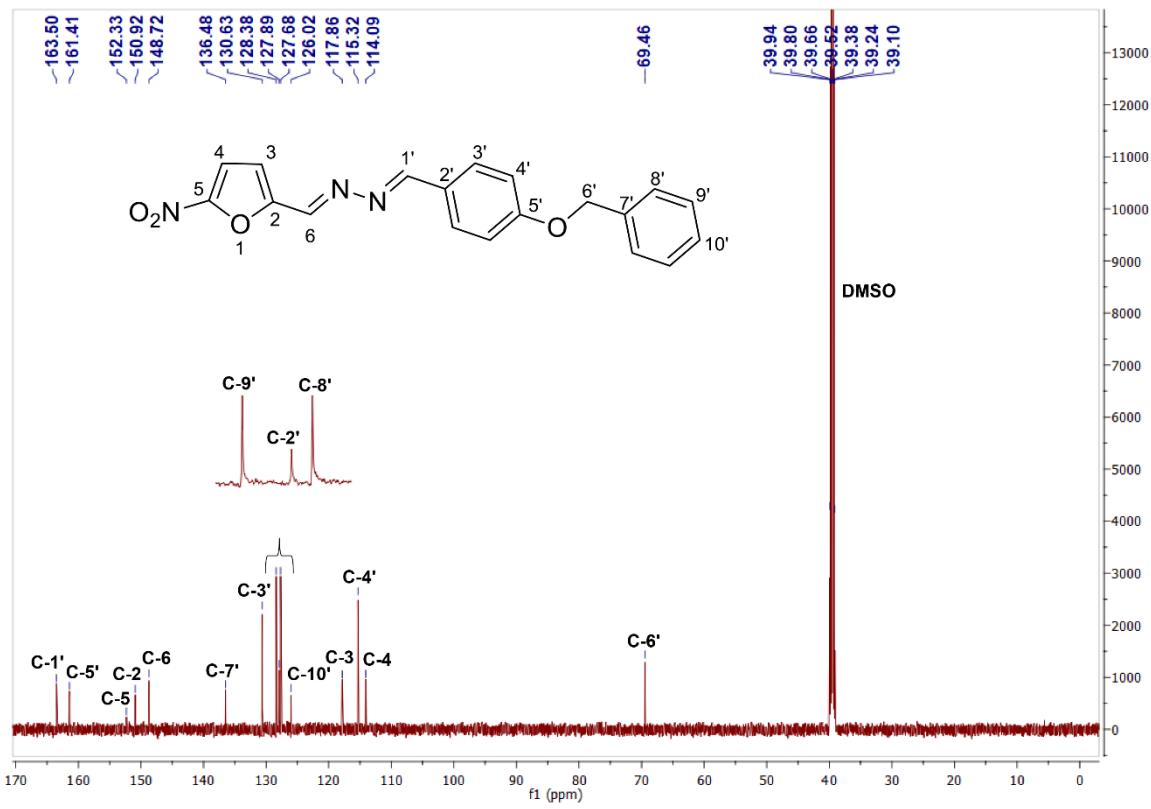
| Meas. m/z | # | Formula           | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|-------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 274.0840  | 1 | C 13 H 12 N 3 O 4 | 100.00 | 274.0822 | -1.8      | -6.5      | 2.0    | 9.5 | even    | ok     |

**(1E,2E)-1-(4-[benzyloxy]benzylidene)-2-([5-nitrofuran-2-yl]methylene)hydrazine  
(7a)**

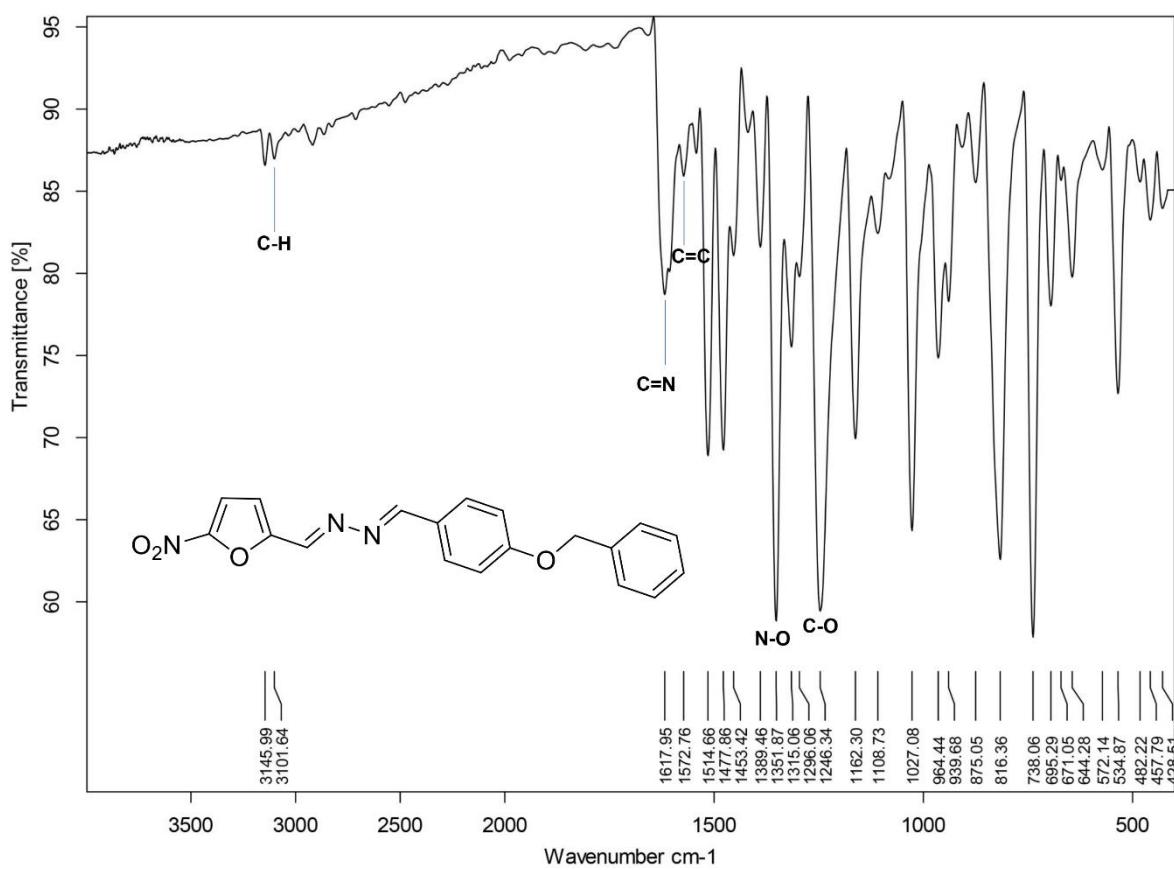
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

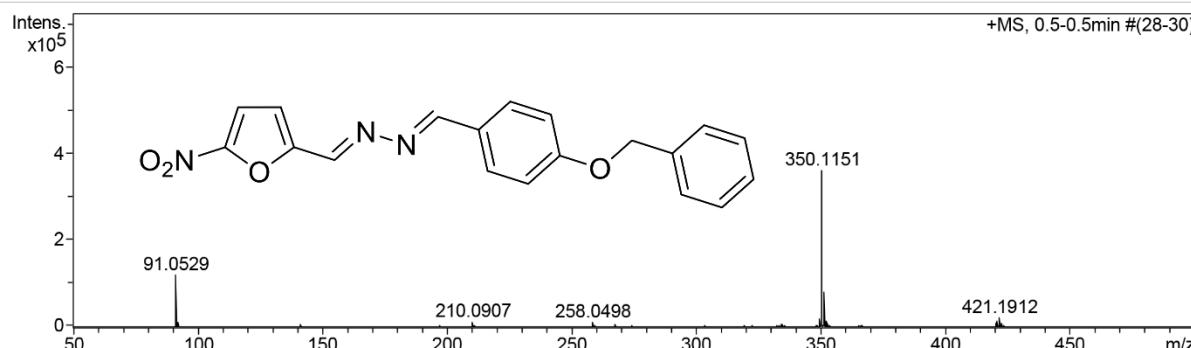
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000028.d       | Acquisition Date  | 10/12/2020 3:26:16 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-09                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

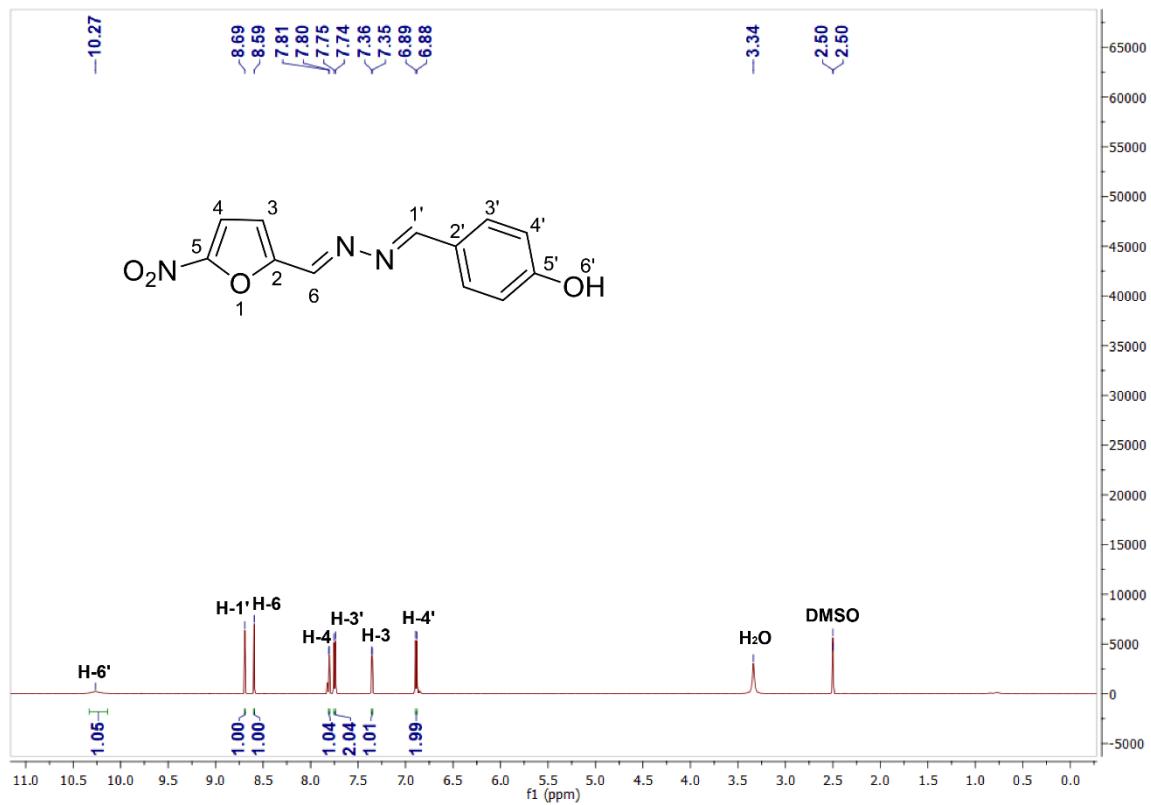
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



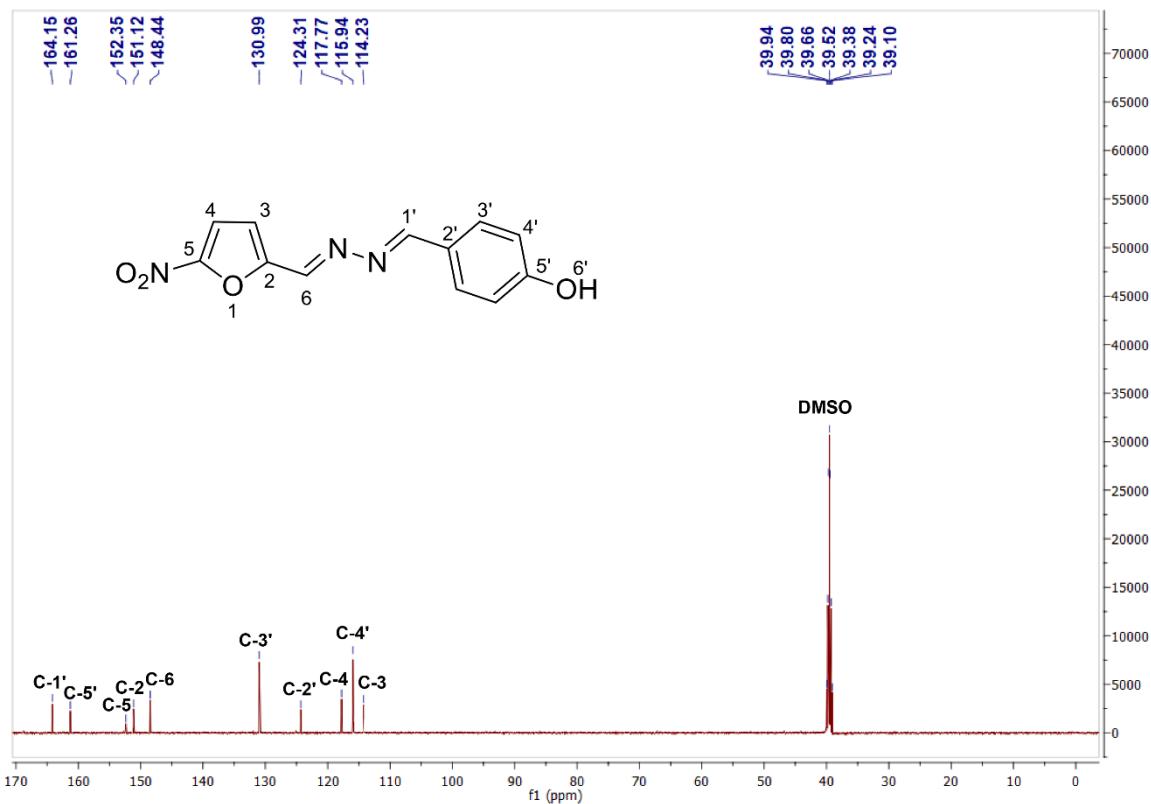
| Meas. m/z | # | Formula           | Score  | $m/z$    | err [mDa] | err [ppm] | $m/\sigma$ | rdb  | $e^-$ Conf | N-Rule |
|-----------|---|-------------------|--------|----------|-----------|-----------|------------|------|------------|--------|
| 91.0529   | 1 | C 7 H 7           | 100.00 | 91.0542  | 1.4       | 15.1      | 1.9        | 4.5  | even       | ok     |
| 350.1151  | 1 | C 19 H 16 N 3 O 4 | 100.00 | 350.1135 | -1.6      | -4.5      | 2.3        | 13.5 | even       | ok     |

**4-(E)-(E)-[(5-nitrofuran-2-yl)methylene]hydrazonomethylphenol (8a)**

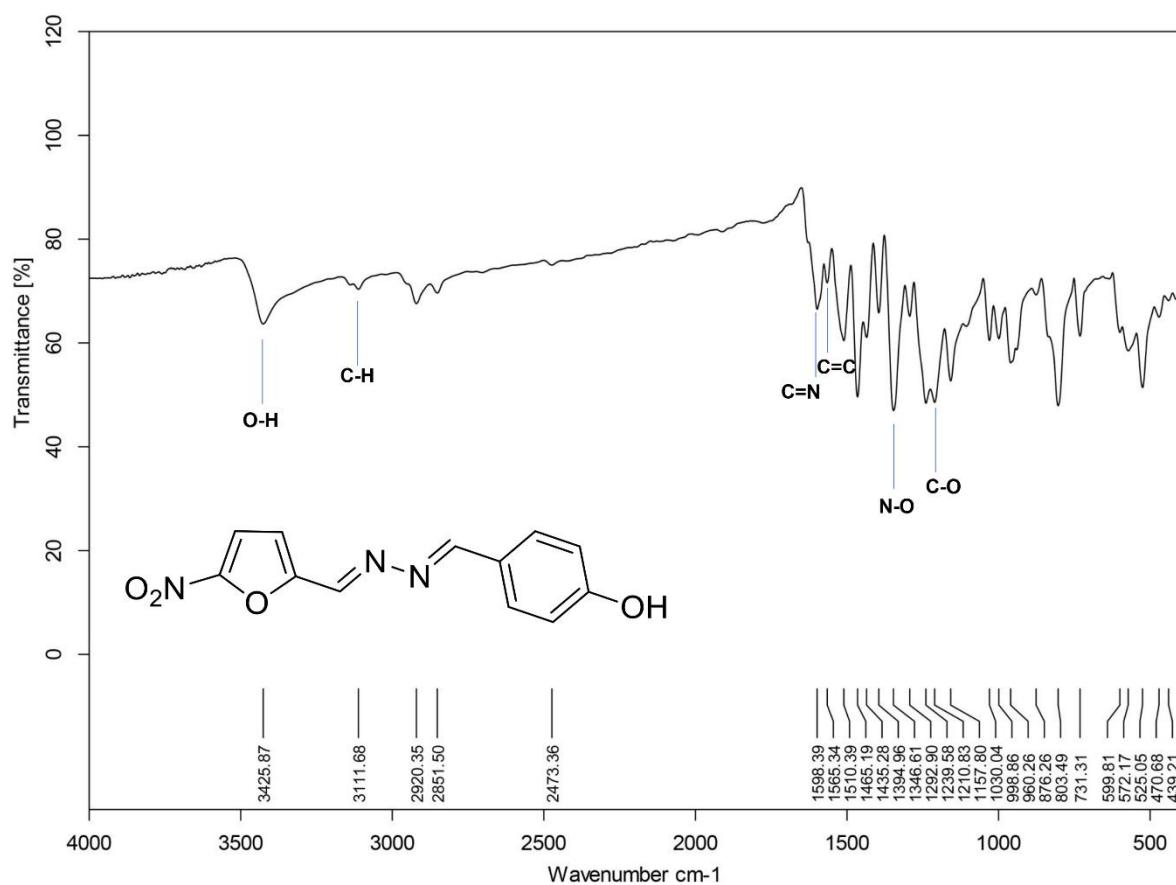
**<sup>1</sup>H NMR in DMSO**



**<sup>13</sup>C NMR in DMSO**



## IR Spectrum



## HRMS

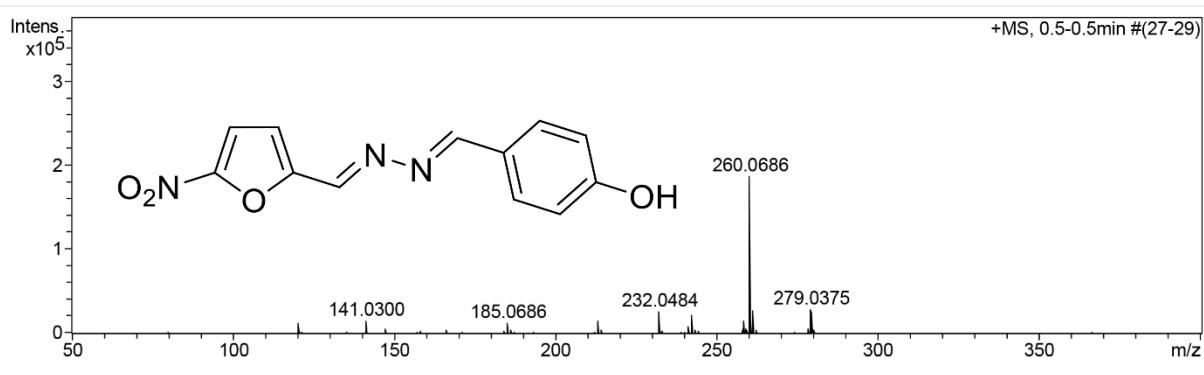
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000039.d       | Acquisition Date  | 10/12/2020 3:39:46 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-8                                 | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

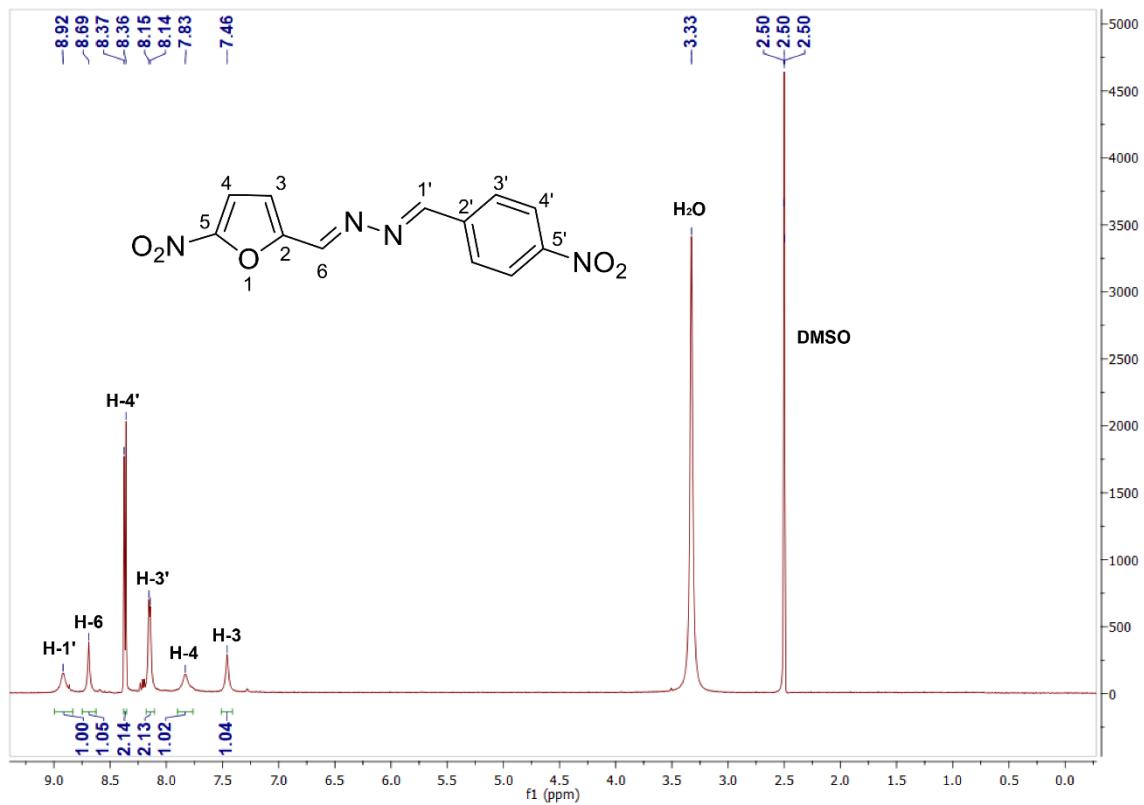
#### Acquisition Parameter

|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |

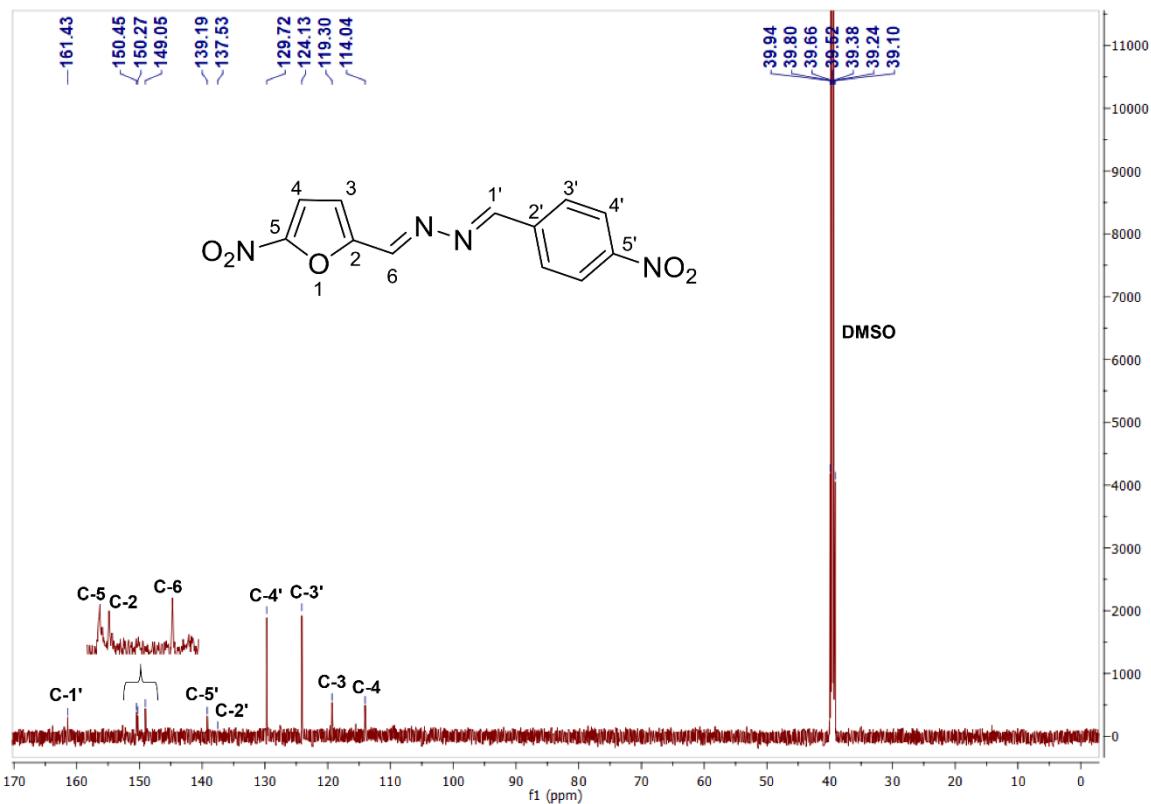


**(1E,2E)-1-(4-nitrobenzylidene)-2-([5-nitrofuran-2-yl]methylene)hydrazine (9a)**

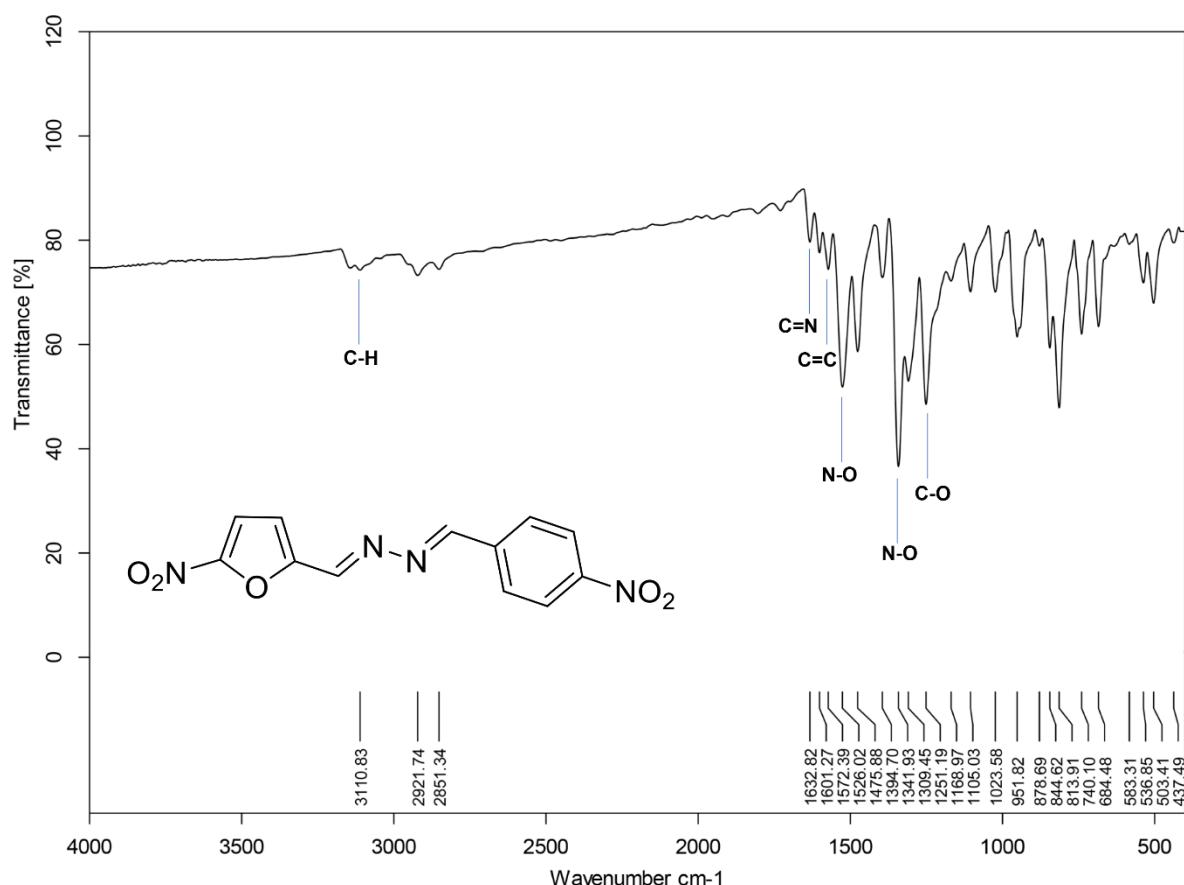
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

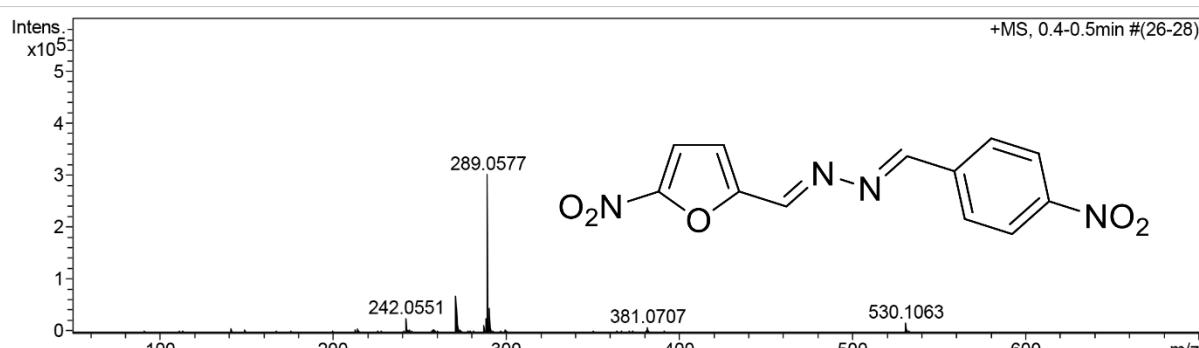
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000034.d       | Acquisition Date  | 10/12/2020 3:34:00 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-2                                 | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

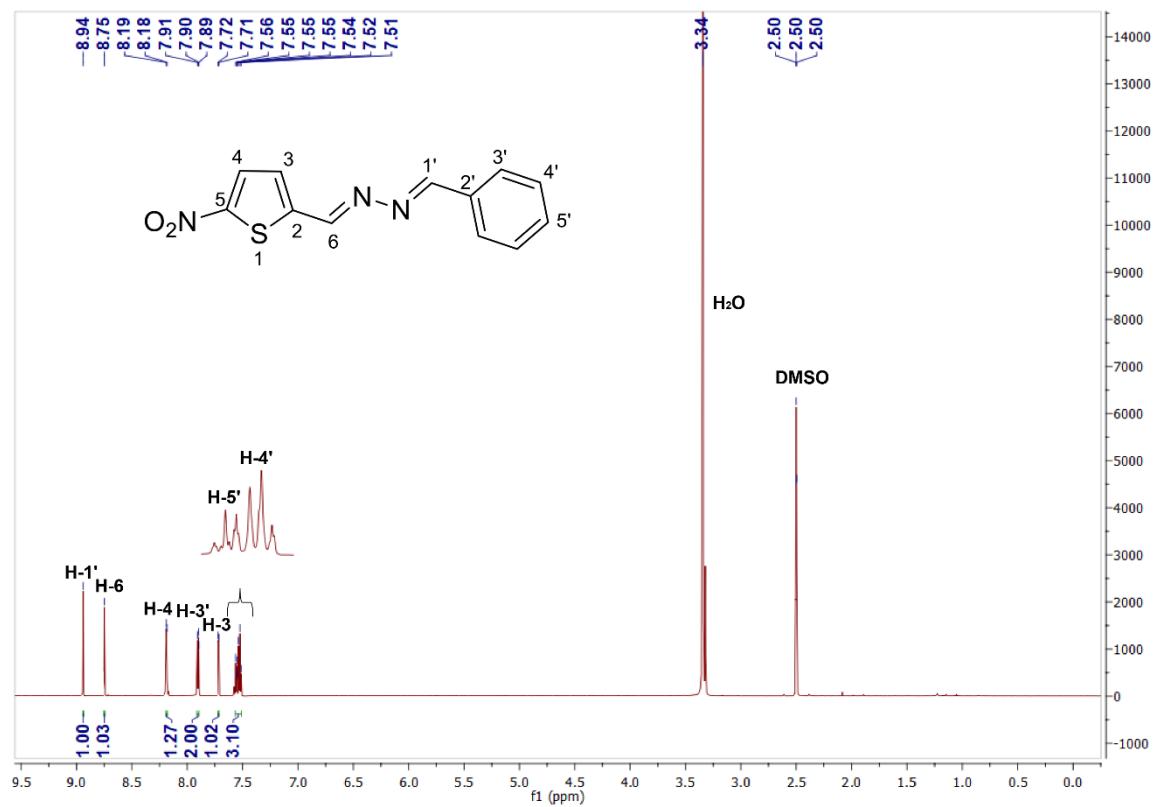
#### Acquisition Parameter

|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |

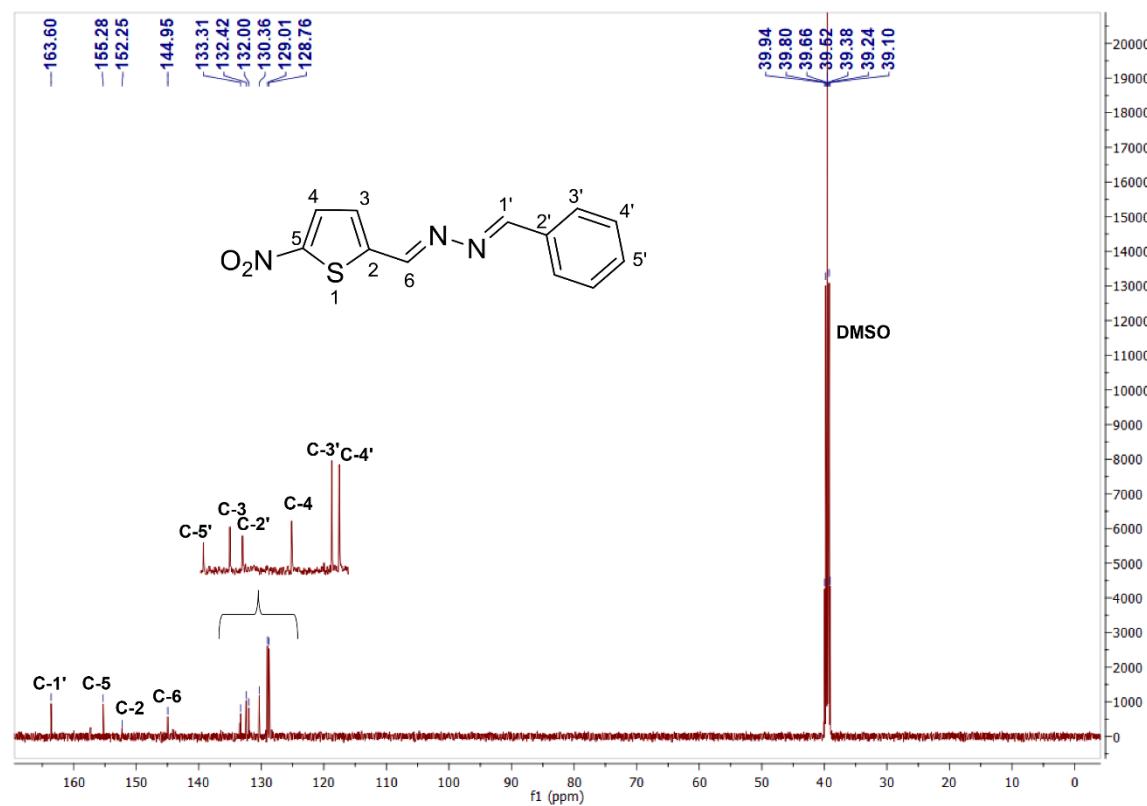


**(1E,2E)-1-benzylidene-2-([5-nitrothiophen-2-yl]methylene)hydrazine (1b)**

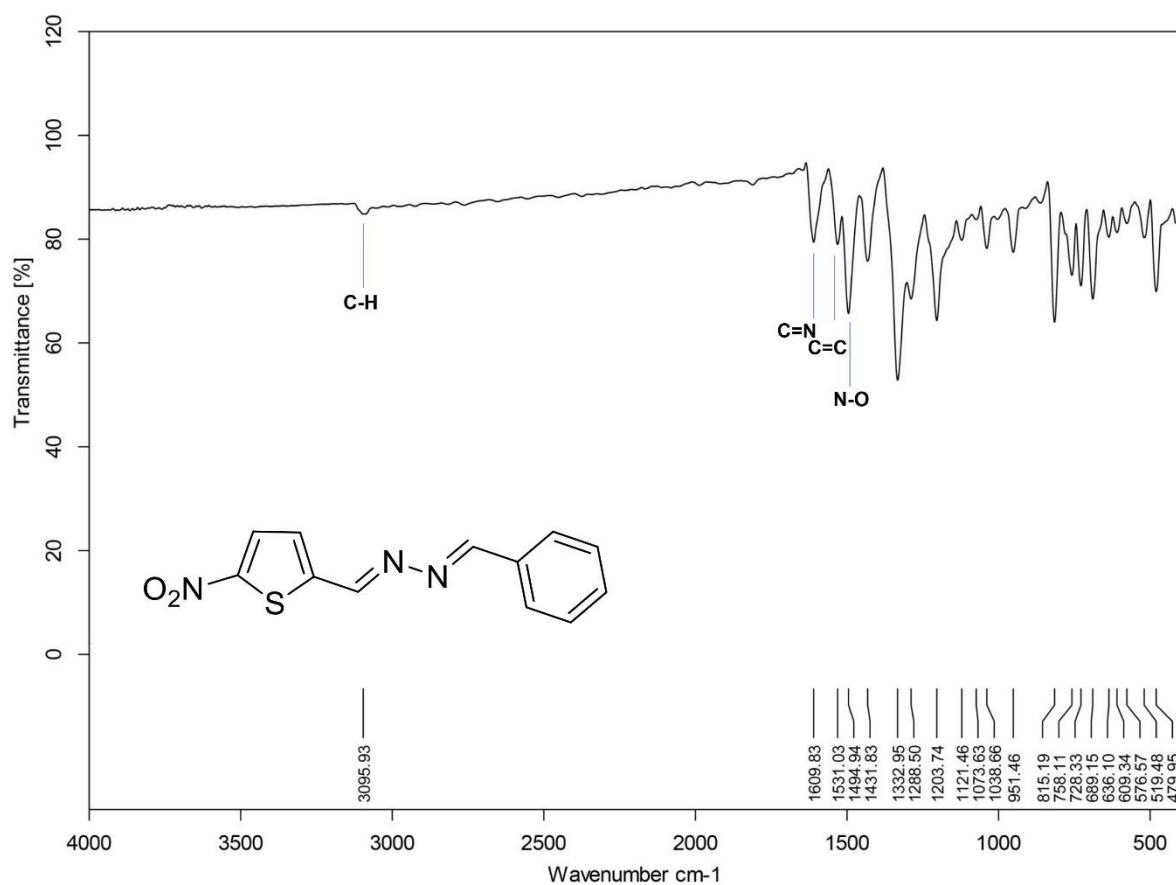
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

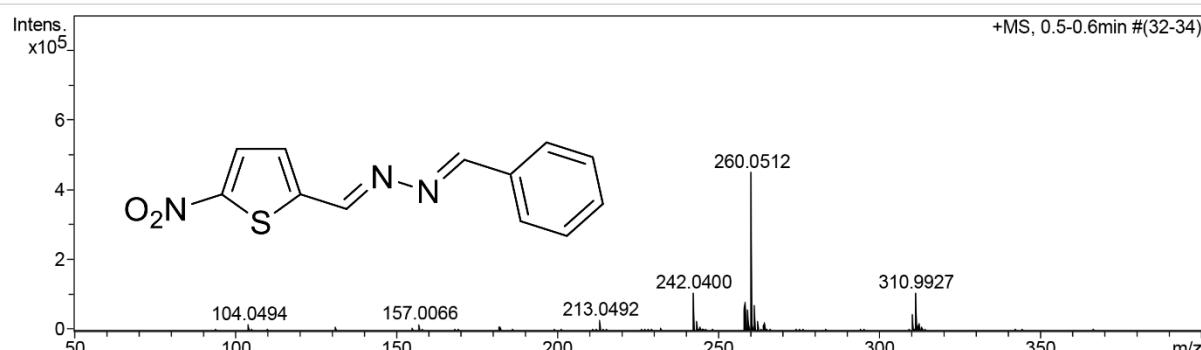
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000040.d       | Acquisition Date  | 10/12/2020 3:41:12 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-10                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

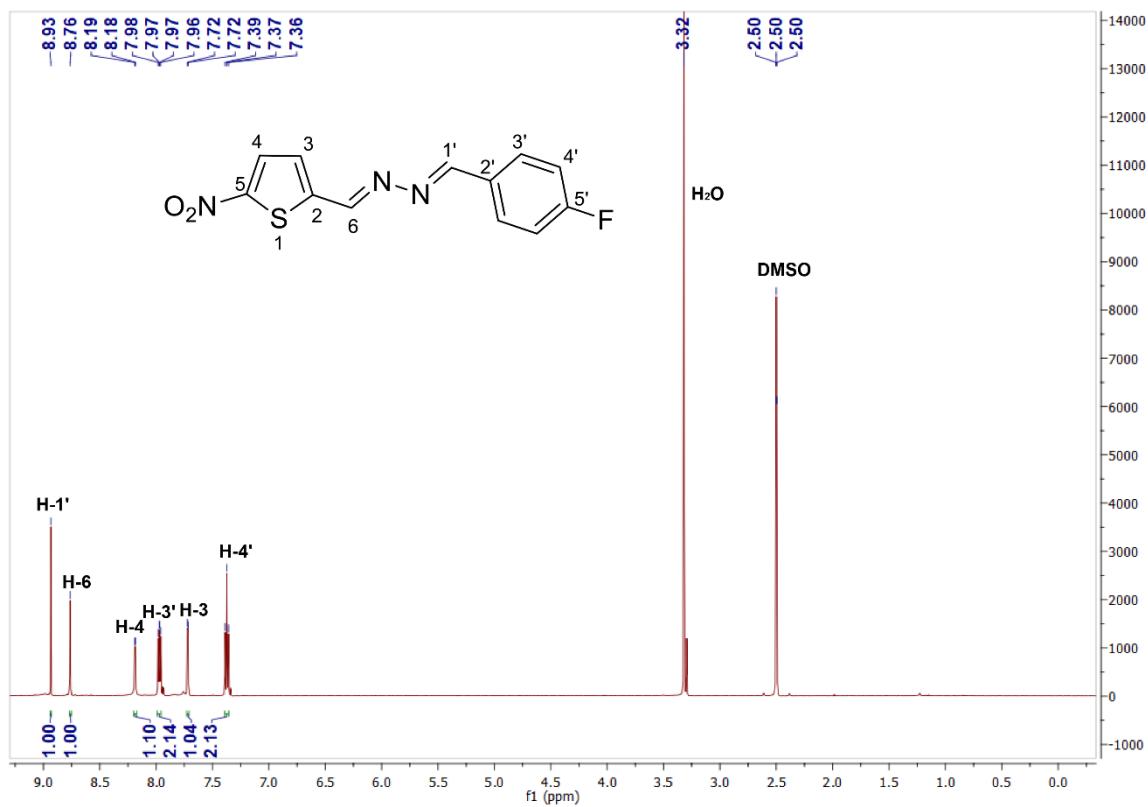
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



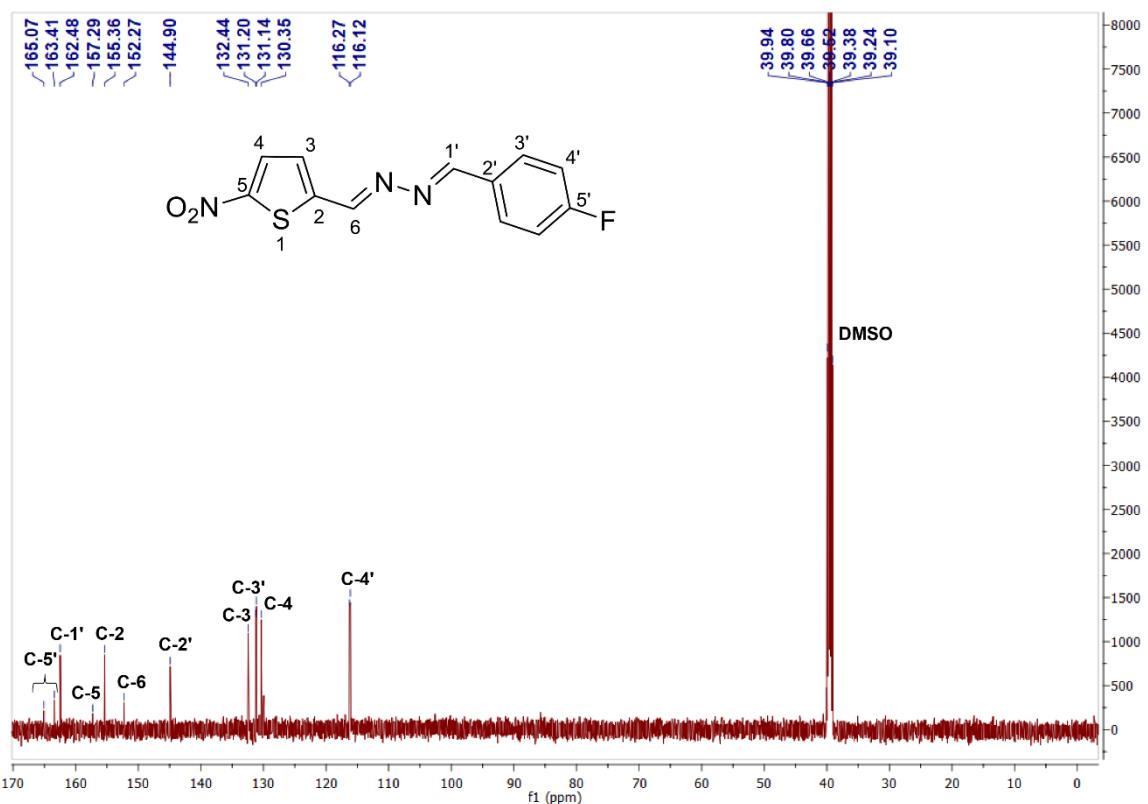
| Meas. m/z | # | Formula             | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|---------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 260.0512  | 1 | C 12 H 10 N 3 O 2 S | 100.00 | 260.0488 | -2.4      | -9.3      | 3.1    | 9.5 | even    | ok     |

**(1E,2E)-1-(4-fluorobenzylidene)-2-([5-nitrothiophen-2-yl]methylene)hydrazine (2b)**

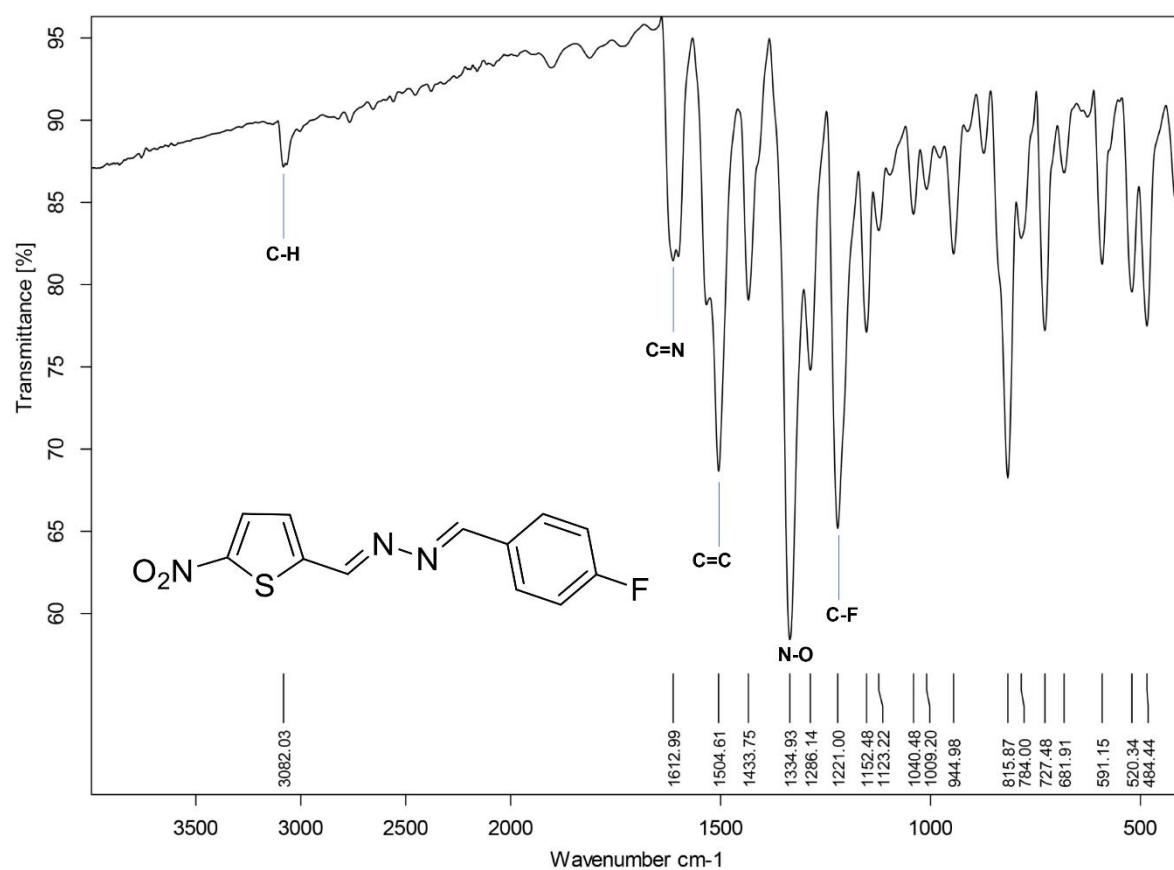
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



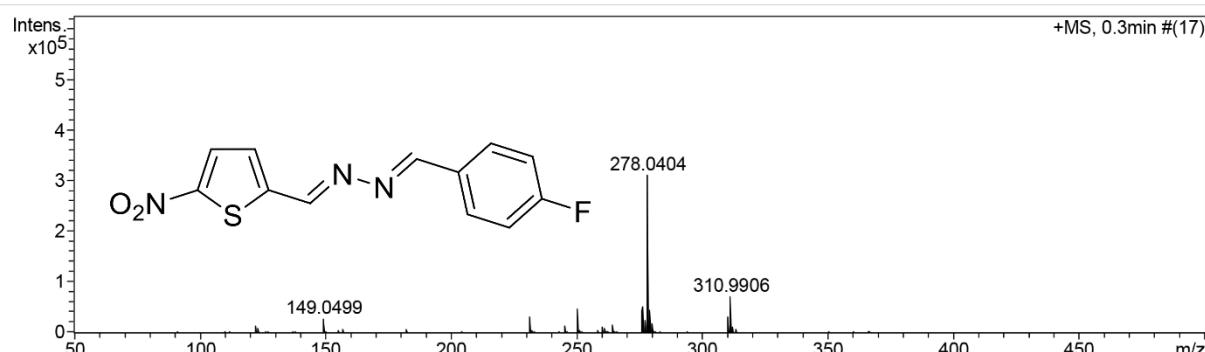
## HRMS

### Mass Spectrum SmartFormula Report

| Analysis Info |                                      | Acquisition Date  | 10/12/2020 3:31:41 PM |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000032.d       |                   |                       |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-16                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

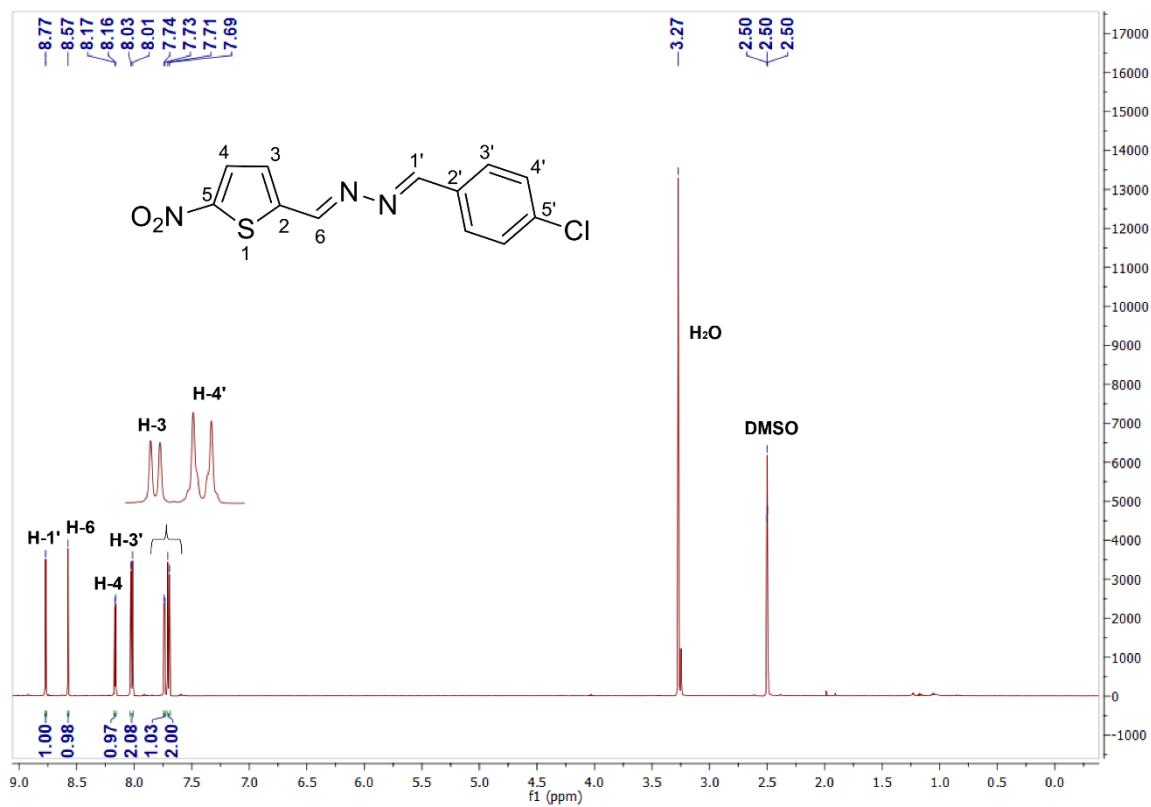
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



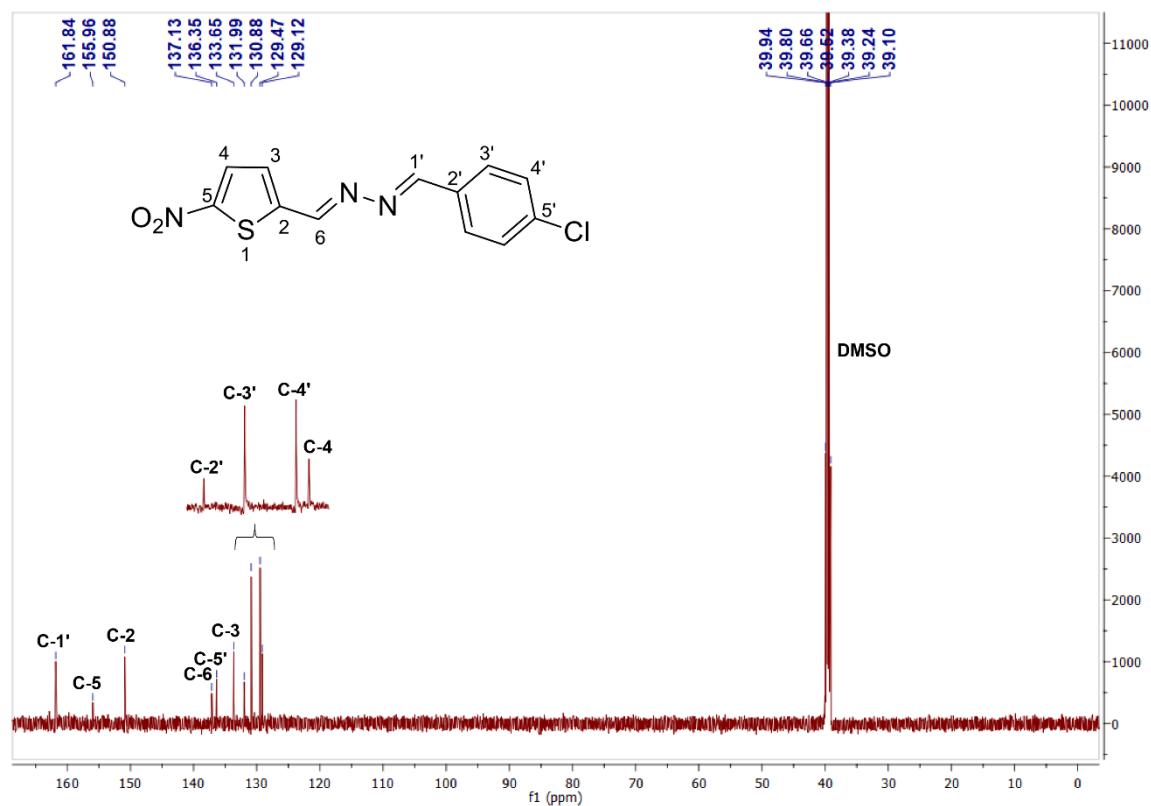
| Meas. m/z | # | Formula              | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|----------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 278.0404  | 1 | C 12 H 9 F N 3 O 2 S | 100.00 | 278.0394 | -1.0      | -3.5      | 0.9    | 9.5 | even    | ok     |

**(1E,2E)-1-(4-chlorobenzylidene)-2-([5-nitrophen-2-yl]methylene)hydrazine (3b)**

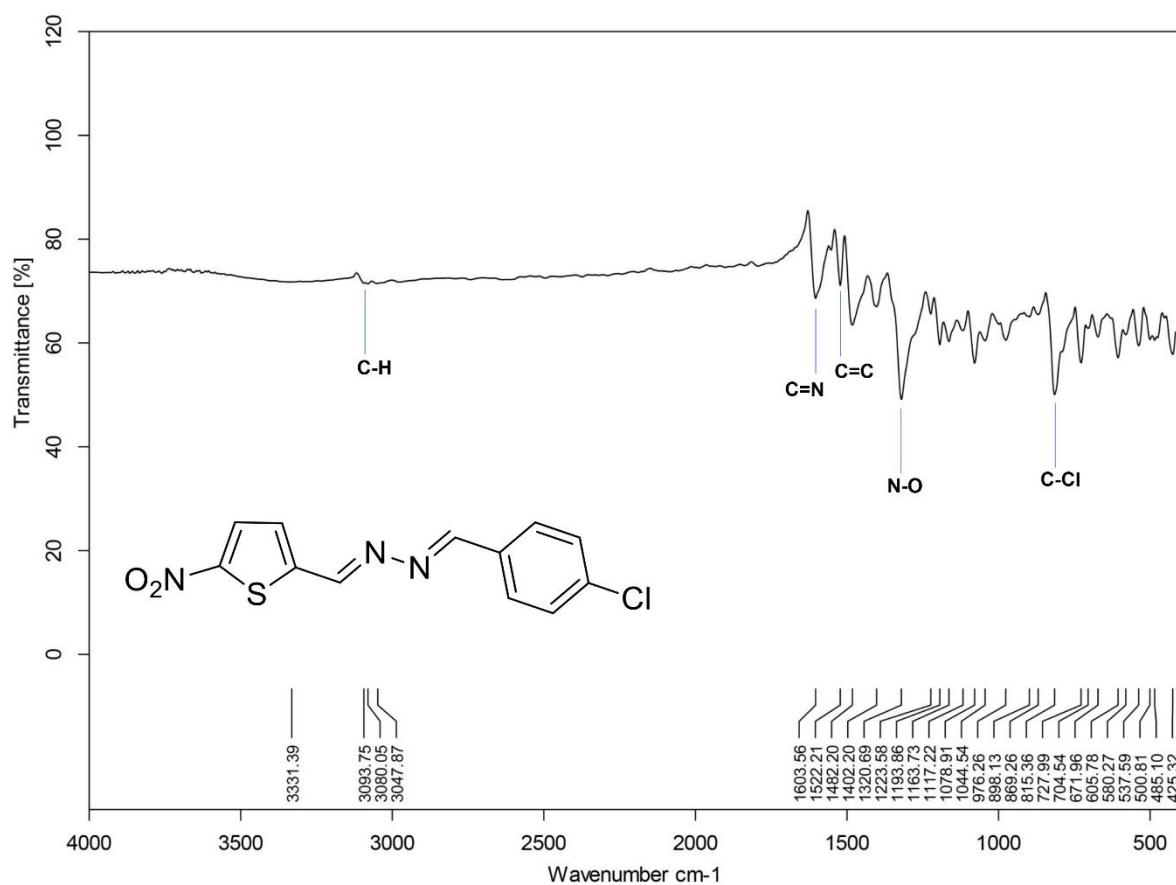
**<sup>1</sup>H NMR in DMSO**



**<sup>13</sup>C NMR in DMSO**



## IR Spectrum



## HRMS

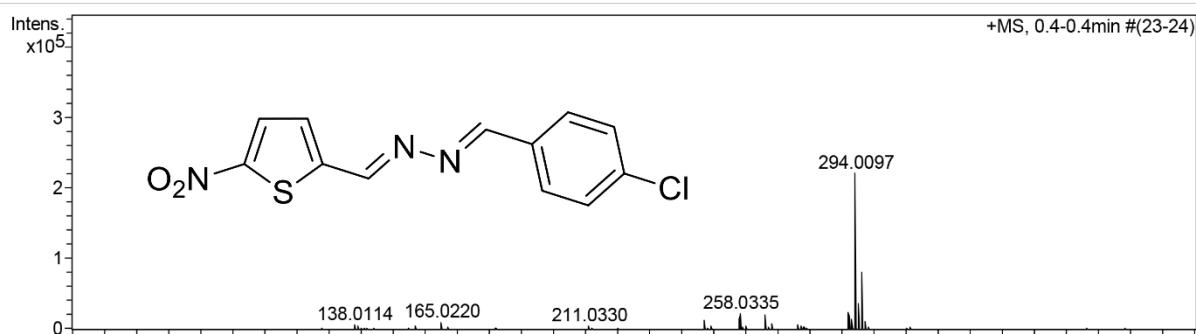
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000042.d       | Acquisition Date  | 10/12/2020 3:44:39 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-12                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

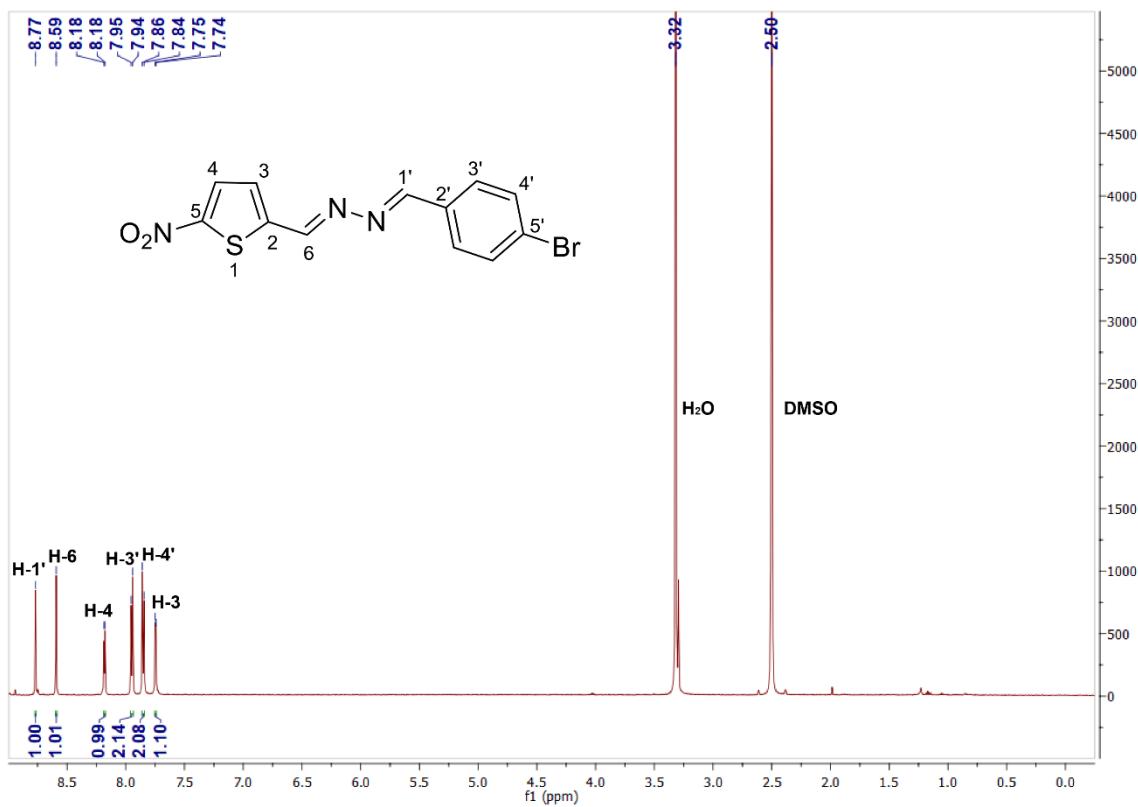
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



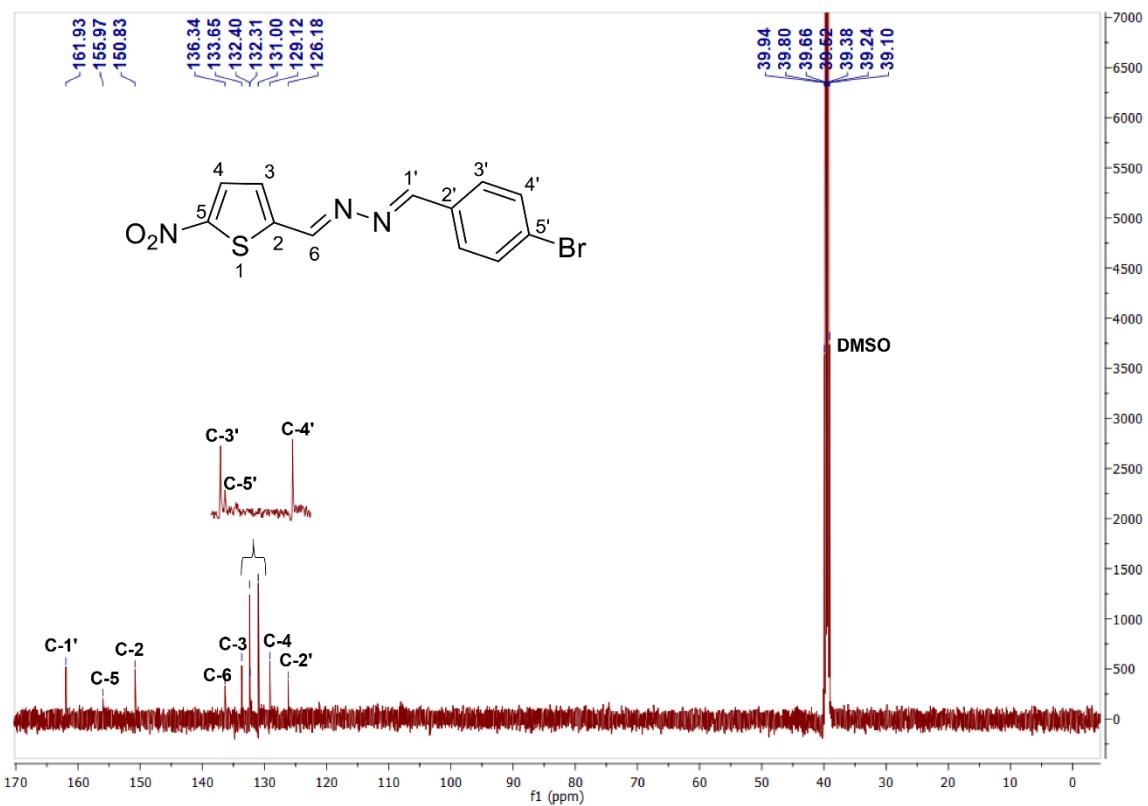
| Meas. m/z | # | Formula               | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|-----------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 294.0097  | 1 | C 12 H 9 Cl N 3 O 2 S | 100.00 | 294.0099 | 0.2       | 0.6       | 8.0    | 9.5 | even    | ok     |

**(1E,2E)-1-(4-bromobenzylidene)-2-([5-nitrophen-2-yl)methylene]hydrazine (4b)**

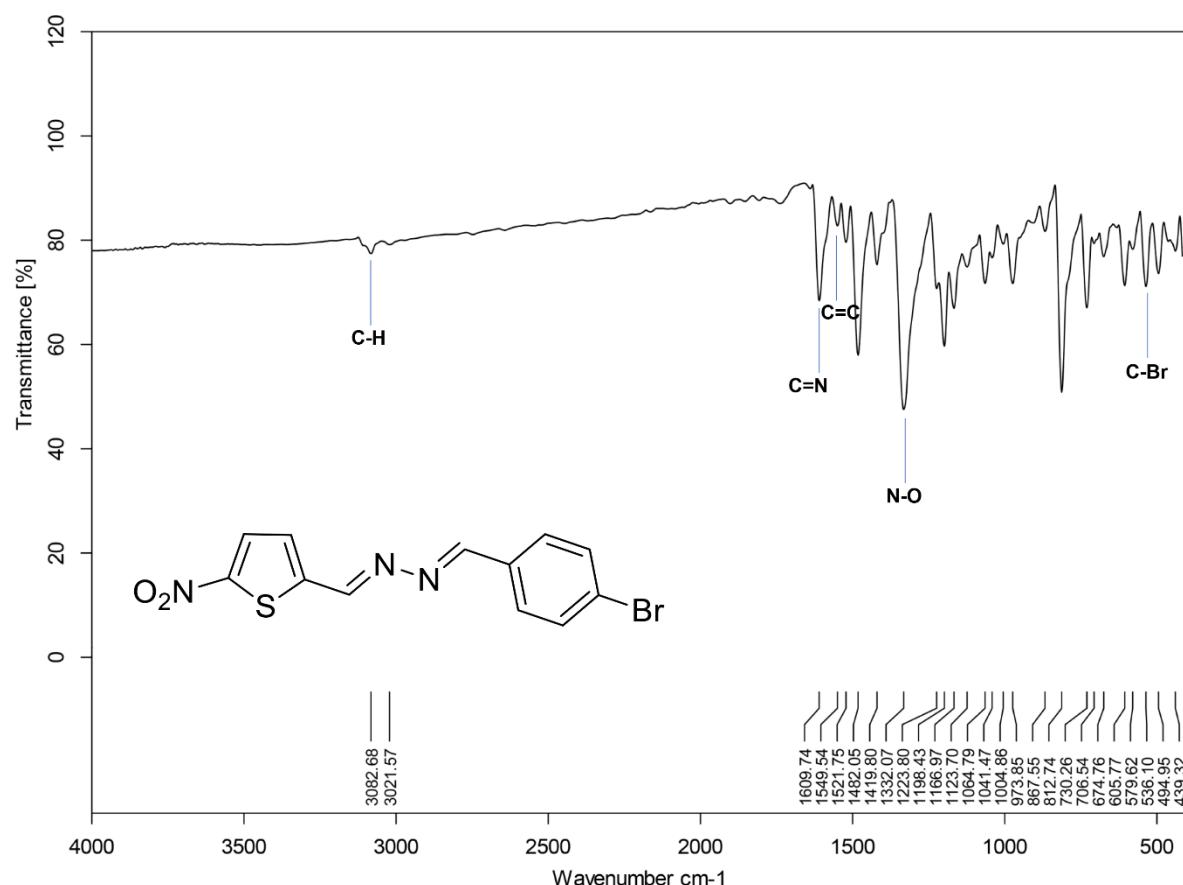
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

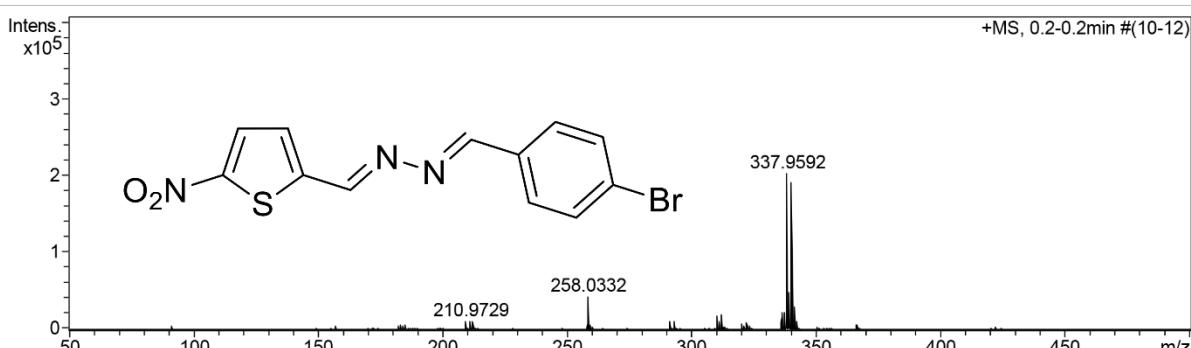
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000030.d       | Acquisition Date  | 10/12/2020 3:29:29 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-13                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

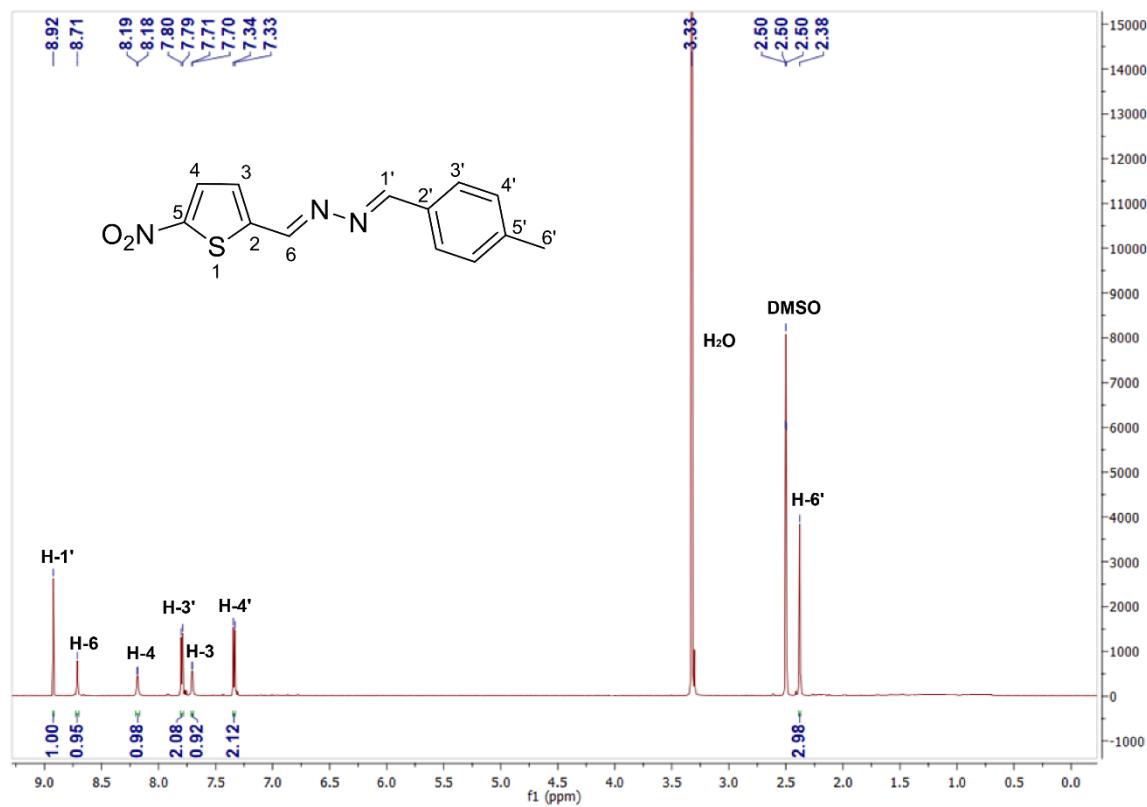
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



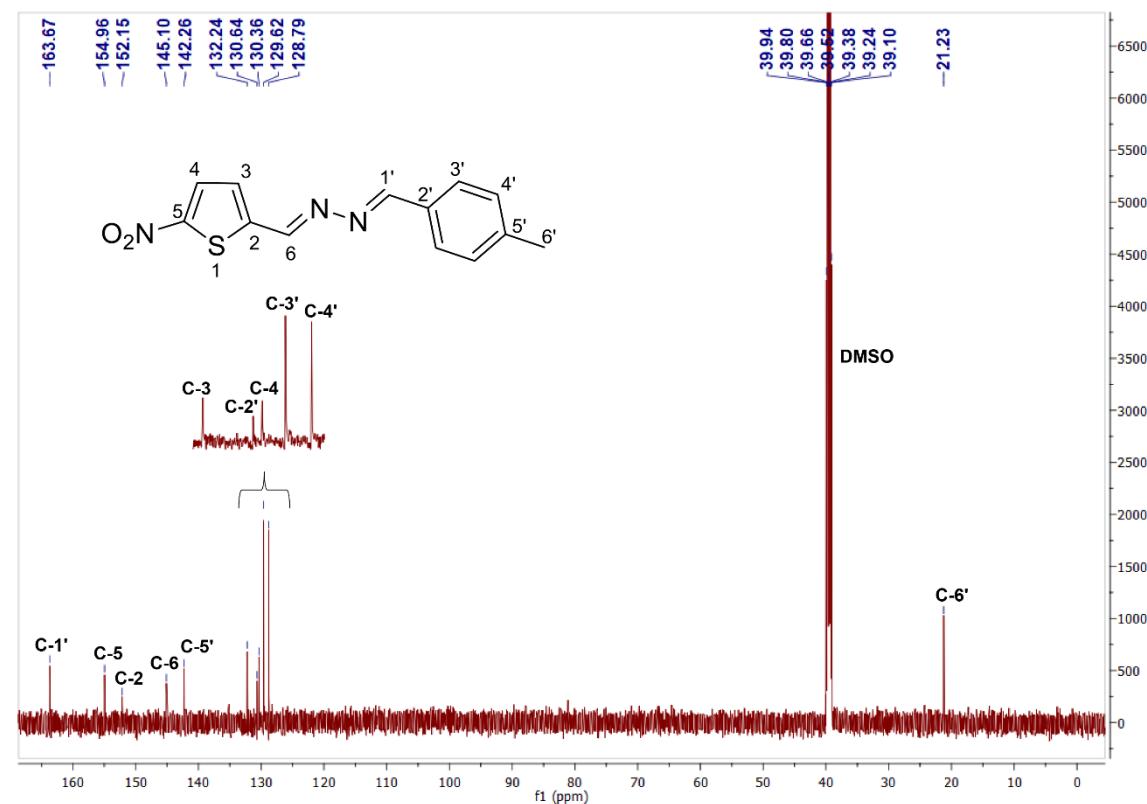
| Meas. m/z | # | Formula               | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|-----------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 337.9592  | 1 | C 14 H 11 Br O 3 S    | 51.79  | 337.9607 | 1.5       | 4.3       | 44.4   | 9.0 | odd     | ok     |
|           | 2 | C 12 H 9 Br N 3 O 2 S | 100.00 | 337.9593 | 0.1       | 0.4       | 46.5   | 9.5 | even    | ok     |

**(1E,2E)-1-(4-methylbenzylidene)-2-([5-nitrothiophen-2-yl]methylene)hydrazine (5b)**

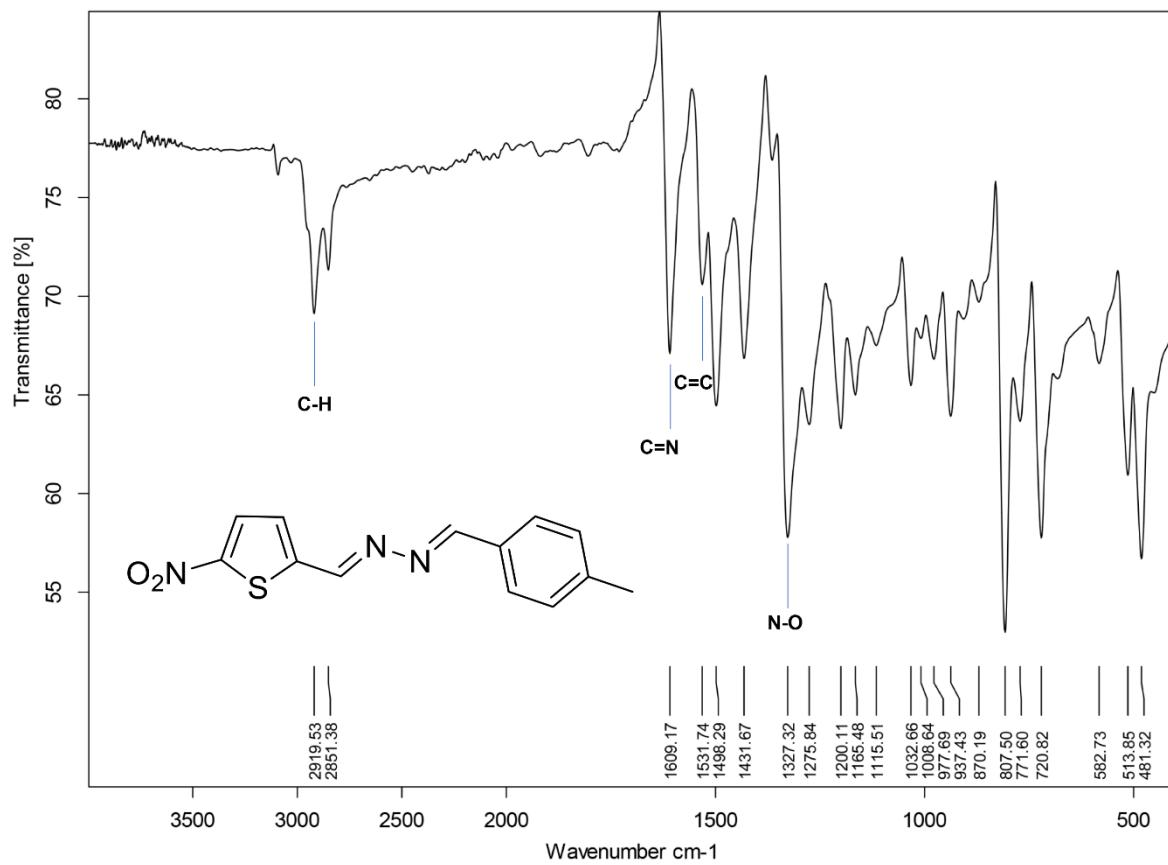
**<sup>1</sup>H NMR in DMSO**



**<sup>13</sup>C NMR in DMSO**



## IR Spectrum



## HRMS

### Mass Spectrum SmartFormula Report

#### Analysis Info

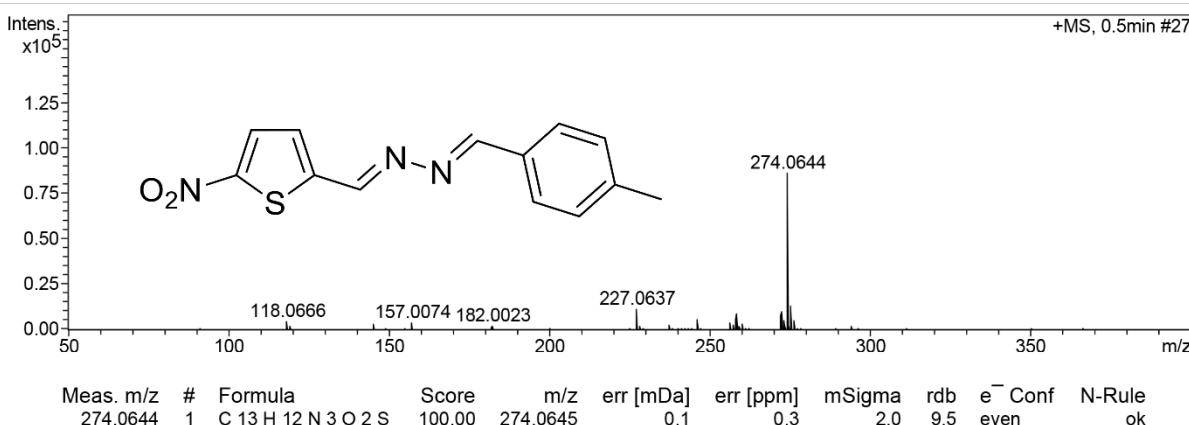
Analysis Name: D:\Data\12102020\LADMS000043.d  
 Method: tune\_low no focus50-1600da12102020.m  
 Sample Name: MV-14  
 Comment:

Acquisition Date: 10/12/2020 3:45:37 PM

Operator: Dr JHL Jordaan  
 Instrument / Ser#: micrOTOF-Q II 2010390

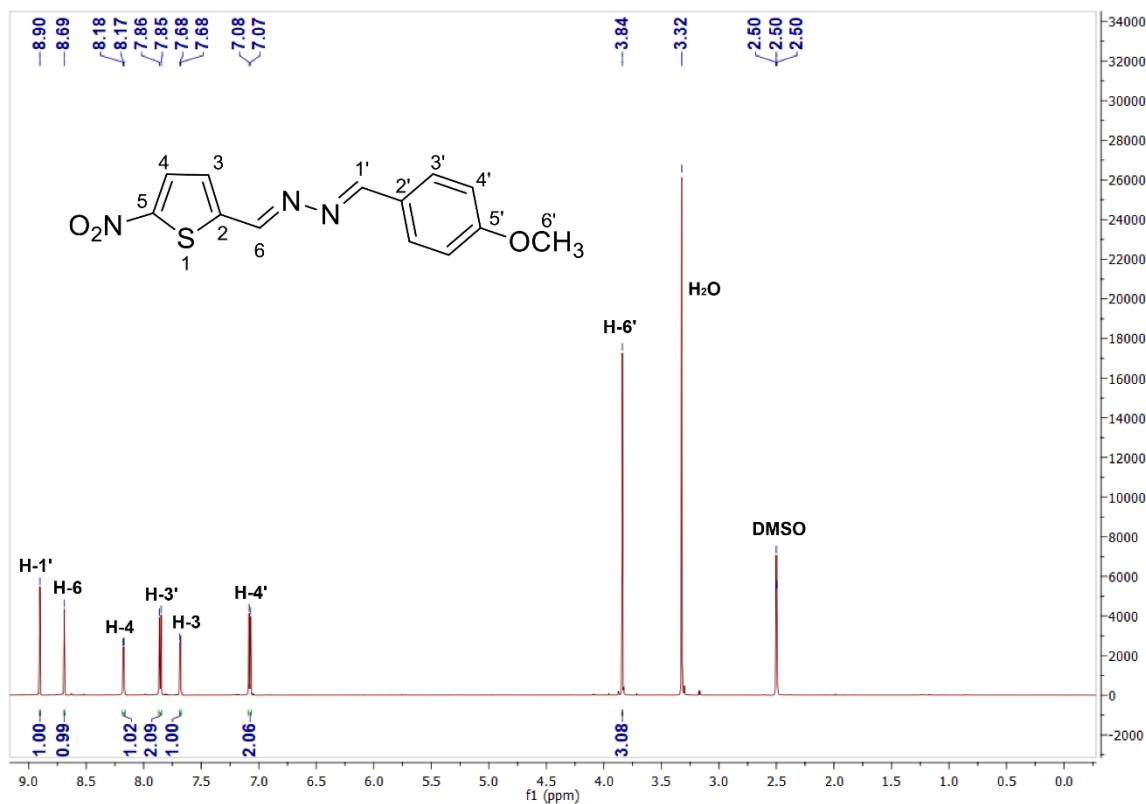
#### Acquisition Parameter

|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |

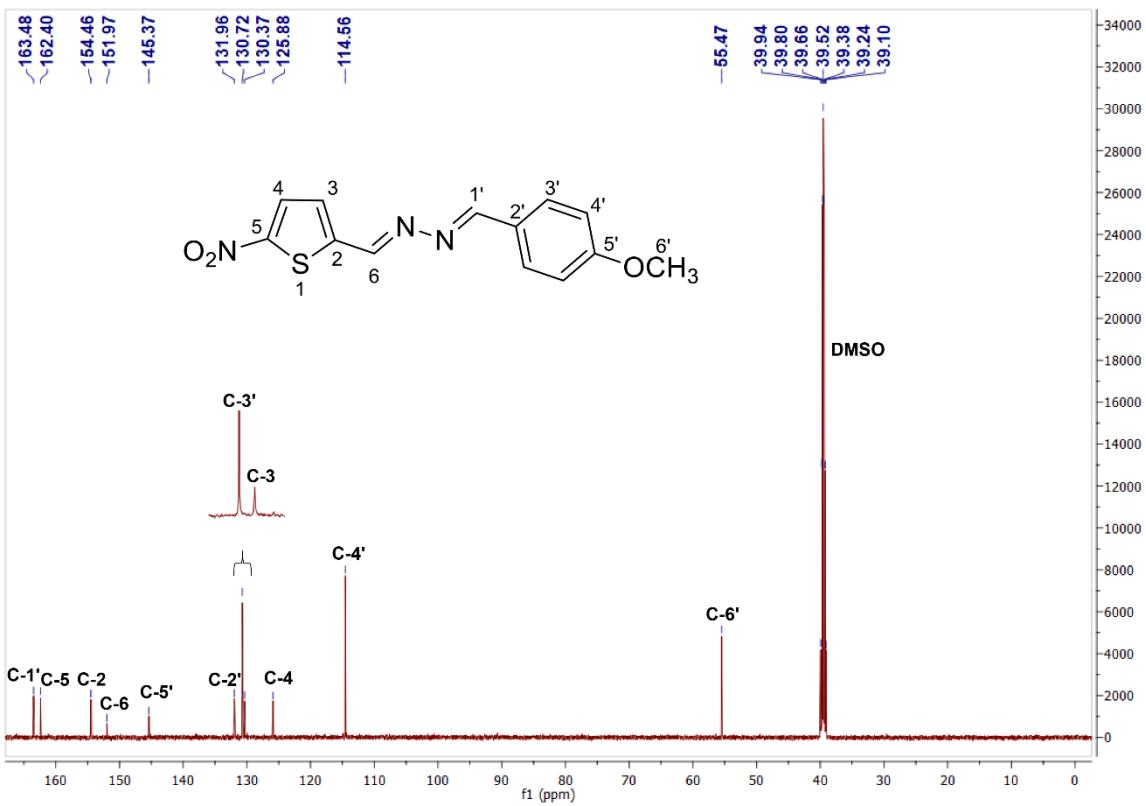


**(1E,2E)-1-(4-methoxybenzylidene)-2-([5-nitrothiophen-2-yl]methylene)hydrazine  
(6b)**

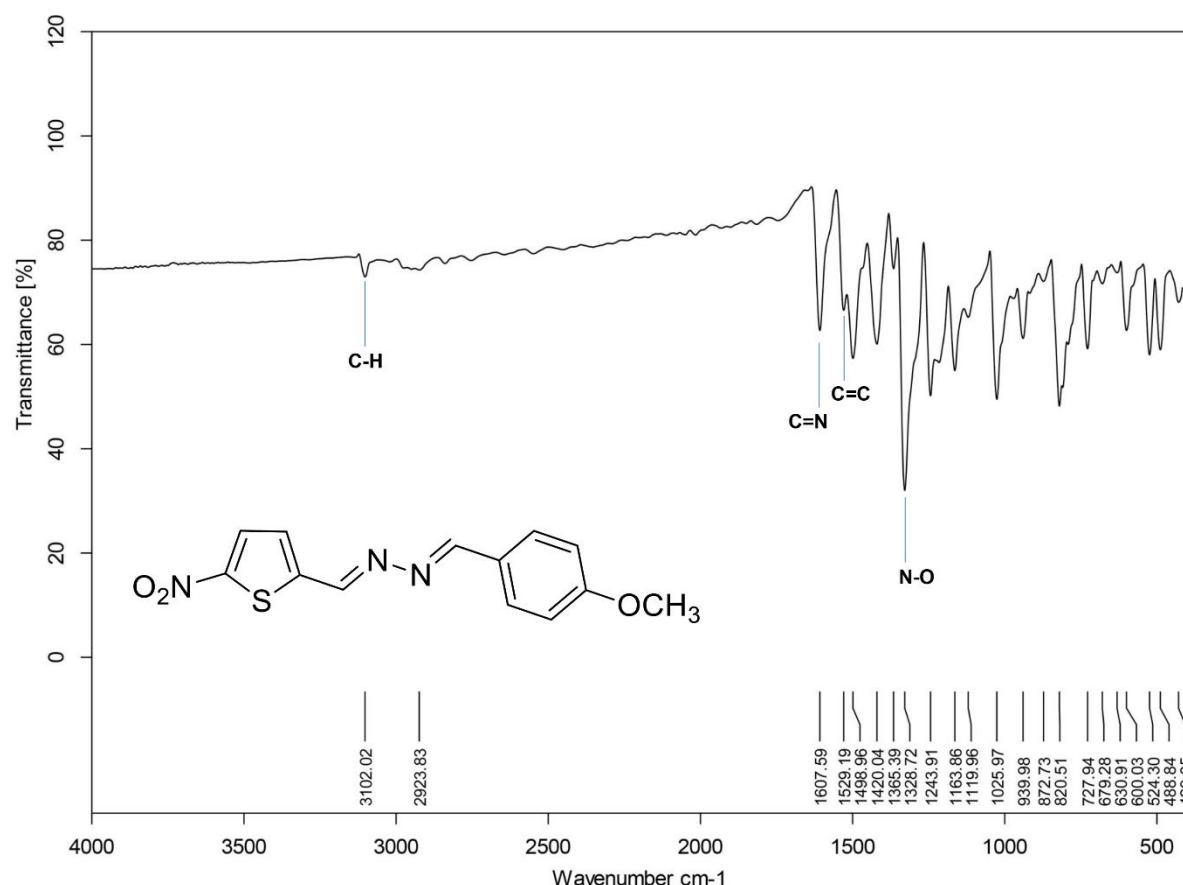
**$^1\text{H}$  NMR in DMSO**



**$^{13}\text{C}$  NMR in DMSO**



## IR Spectrum



## HRMS

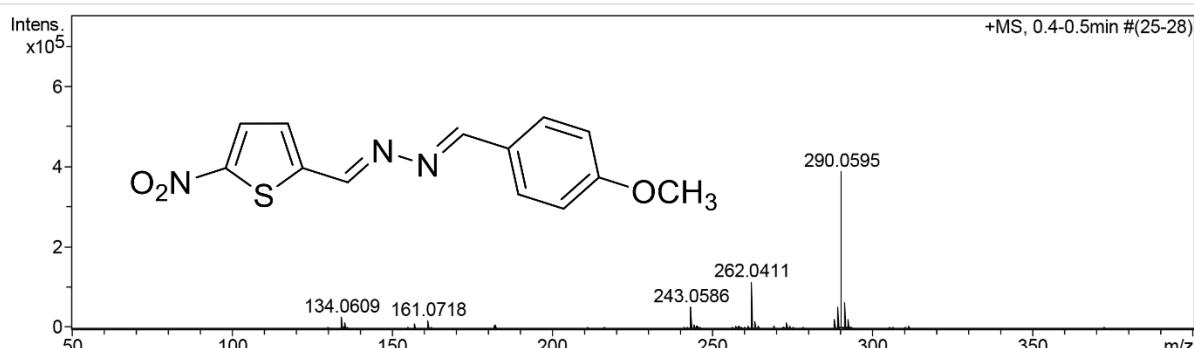
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000044.d       | Acquisition Date  | 10/12/2020 3:46:50 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-15                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

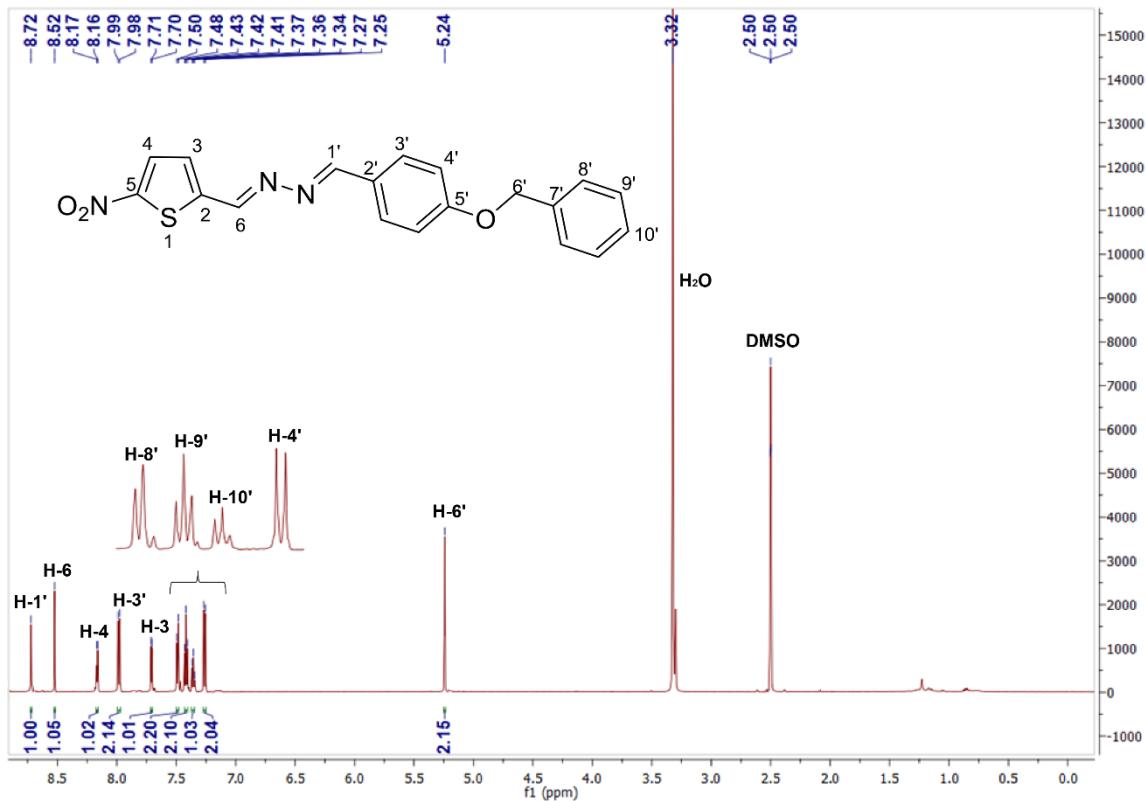
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



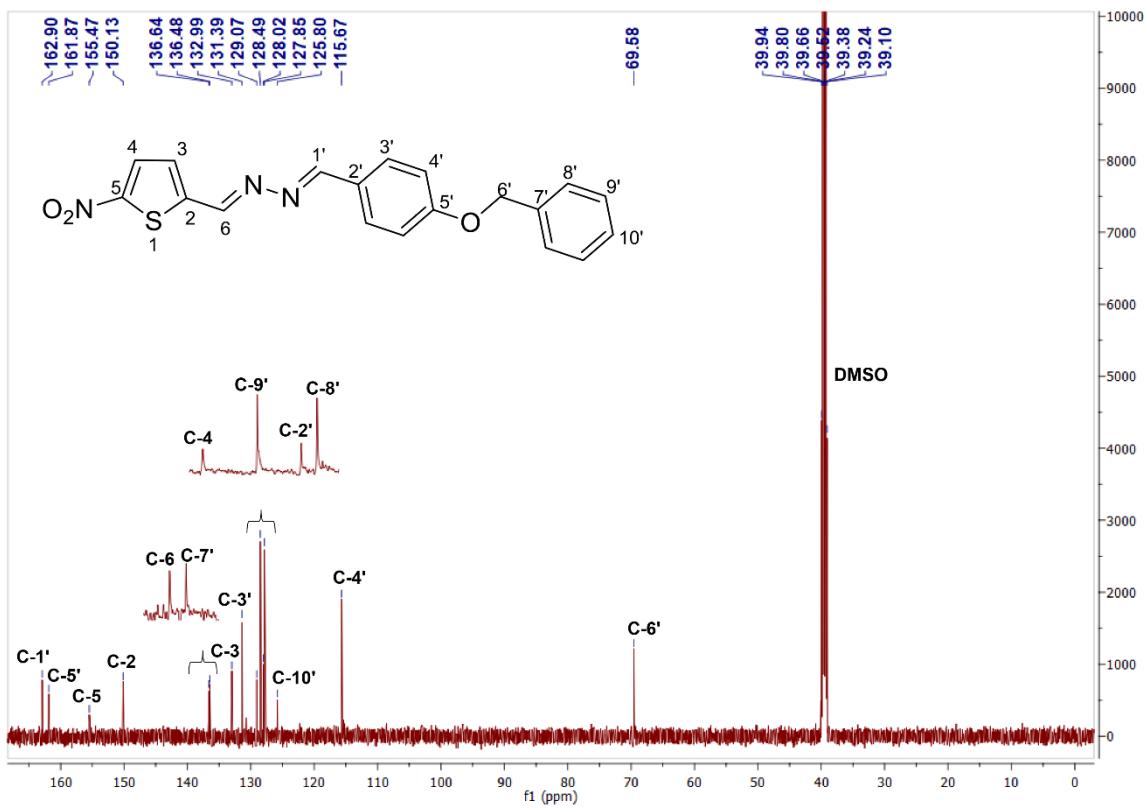
| Meas. m/z | # | Formula             | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|---------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 290.0595  | 1 | C 13 H 12 N 3 O 3 S | 100.00 | 290.0594 | -0.1      | -0.3      | 2.6    | 9.5 | even    | ok     |

**(1E,2E)-1-(4-(benzyloxy)benzylidene)-2-((5-nitrothiophen-2-yl)methylene)  
hydrazine (7b)**

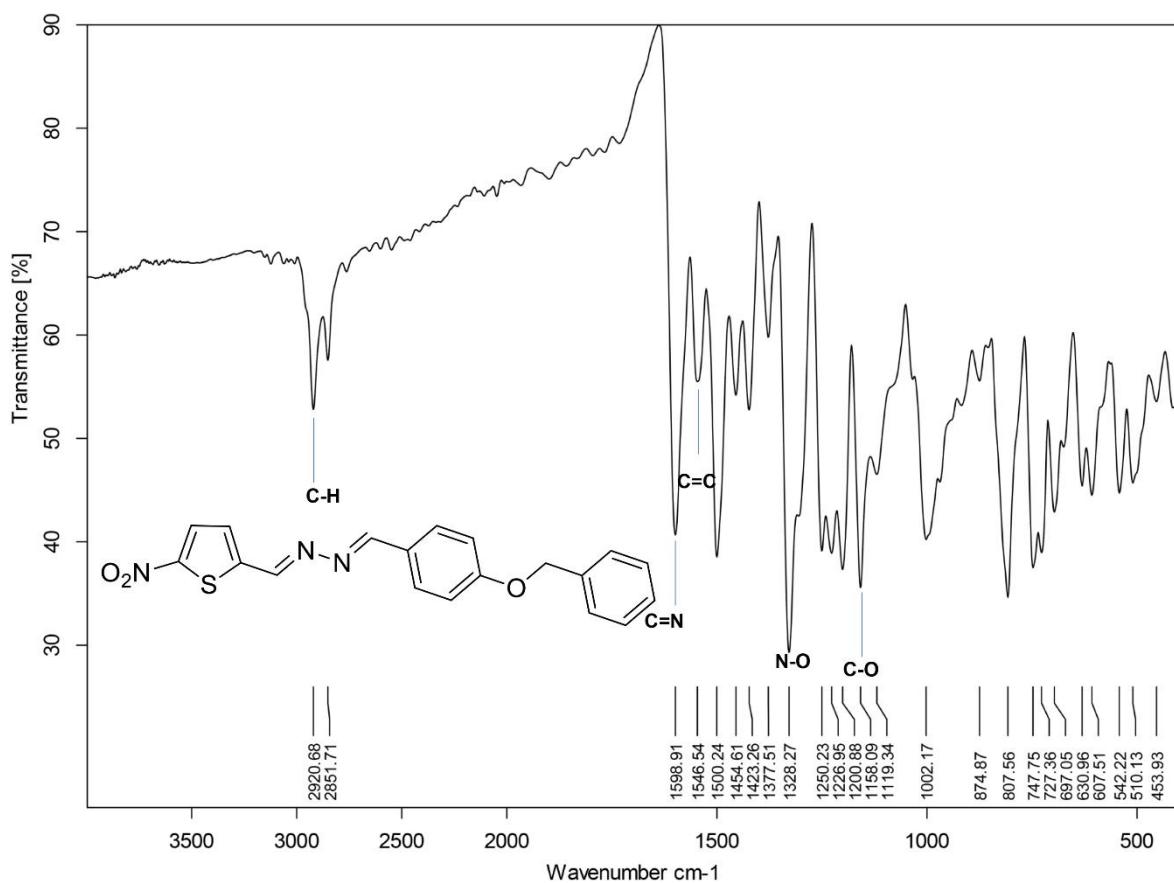
## **<sup>1</sup>H NMR in DMSO**



## **<sup>13</sup>C NMR in DMSO**



## IR Spectrum



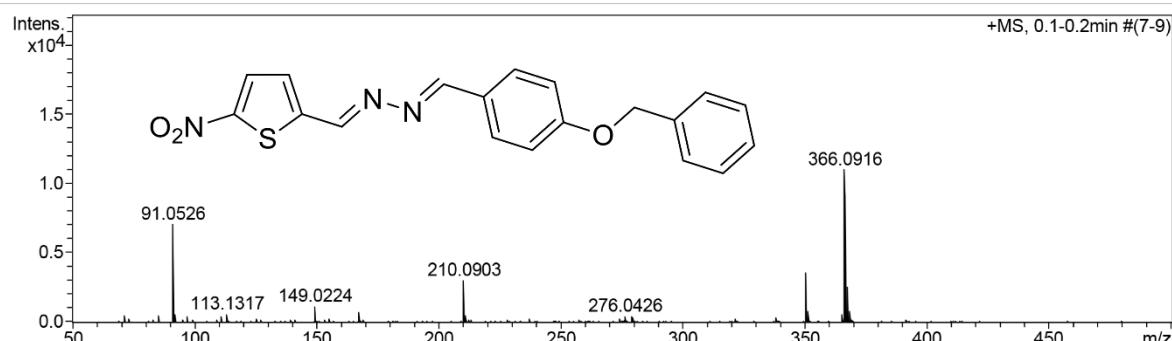
HRMS

## Mass Spectrum SmartFormula Report

| Analysis Info |                                      | Acquisition Date  | 10/12/2020 3:28:30 PM |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000029.d       |                   |                       |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-18                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

## Acquisition Parameter

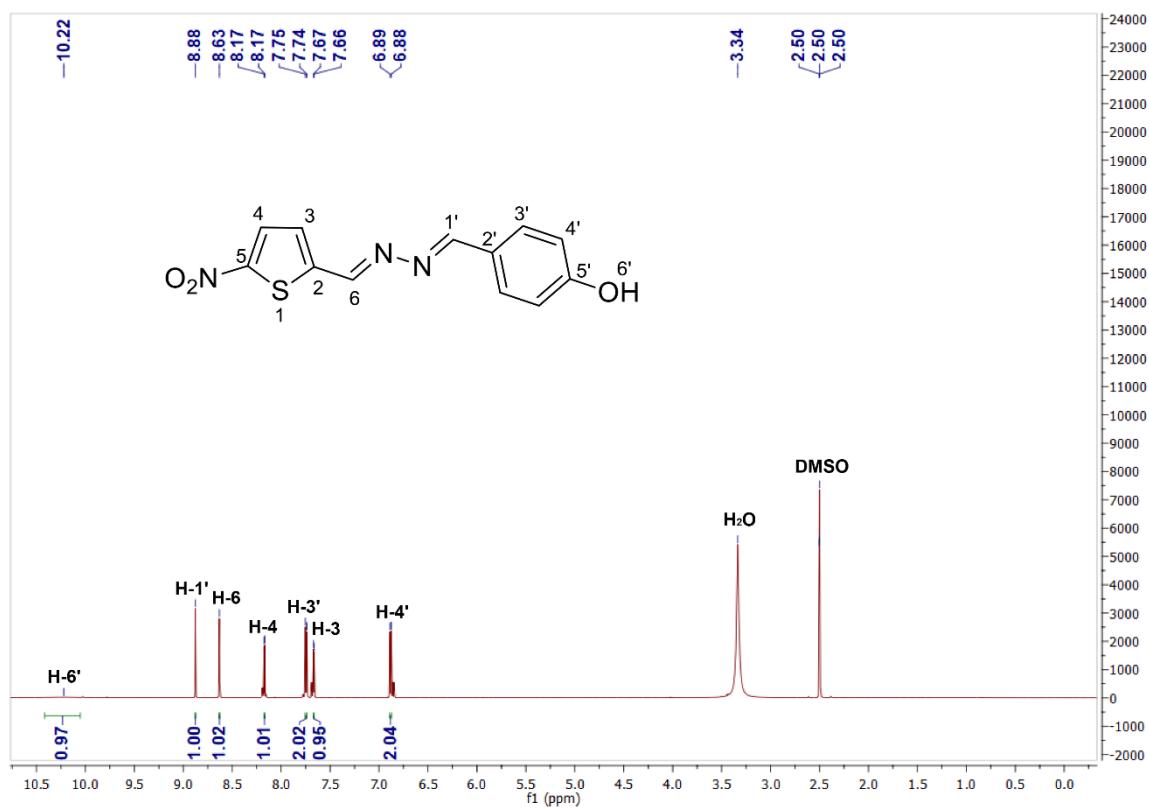
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



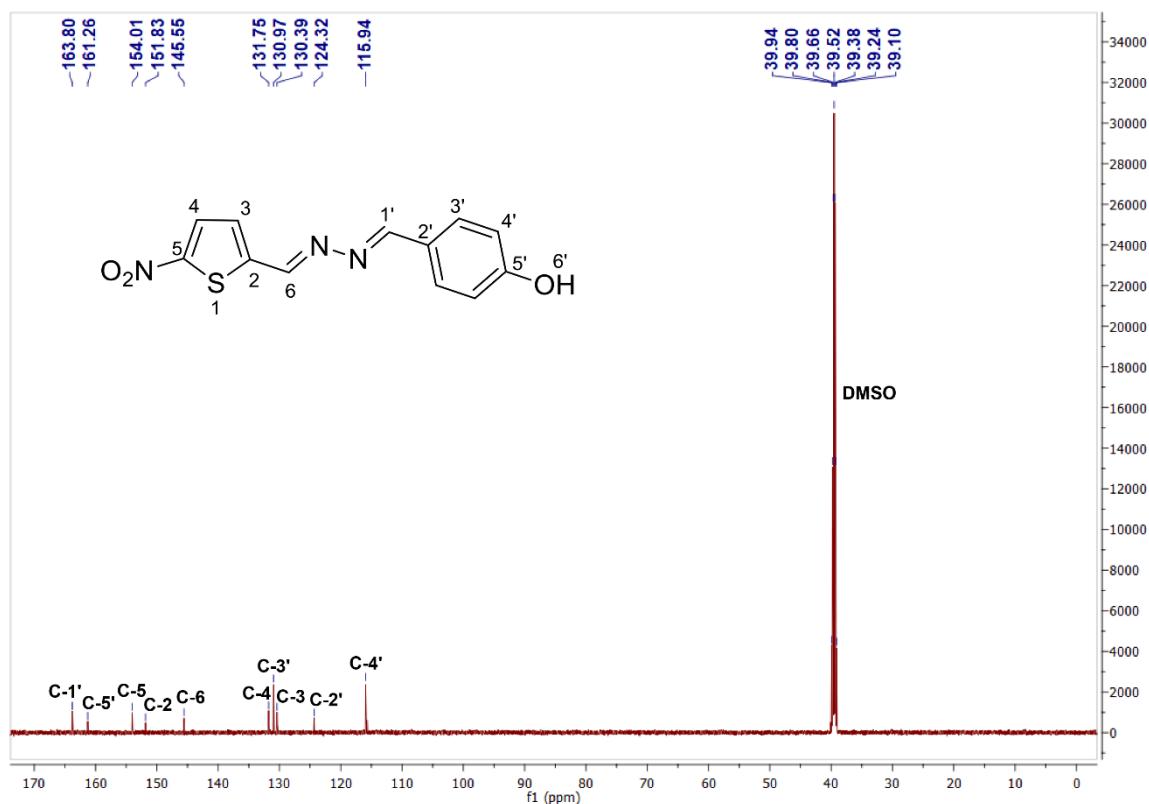
| Meas. m/z | # | Formula             | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb  | e <sup>-</sup> Conf | N-Rule |
|-----------|---|---------------------|--------|----------|-----------|-----------|--------|------|---------------------|--------|
| 91.0526   | 1 | C 7 H 7             | 100.00 | 91.0542  | 1.7       | 18.2      | 5.0    | 4.5  | even                | ok     |
| 210.0903  | 1 | C 14 H 12 N O       | 100.00 | 210.0913 | 1.0       | 4.8       | 14.1   | 9.5  | even                | ok     |
|           | 2 | C 11 H 14 O 4       | 41.62  | 210.0887 | -1.7      | -7.9      | 34.4   | 5.0  | odd                 | ok     |
| 350.1128  | 1 | C 19 H 16 N 3 O 4   | 100.00 | 350.1135 | 0.7       | 2.1       | 17.7   | 13.5 | even                | ok     |
| 366.0916  | 1 | C 19 H 16 N 3 O 3 S | 100.00 | 366.0907 | -0.9      | -2.4      | 7.2    | 13.5 | even                | ok     |

**4-(E)-(E)-([5-nitrothiophen-2-yl)methylene)hydrazonomethylphenol (8b)**

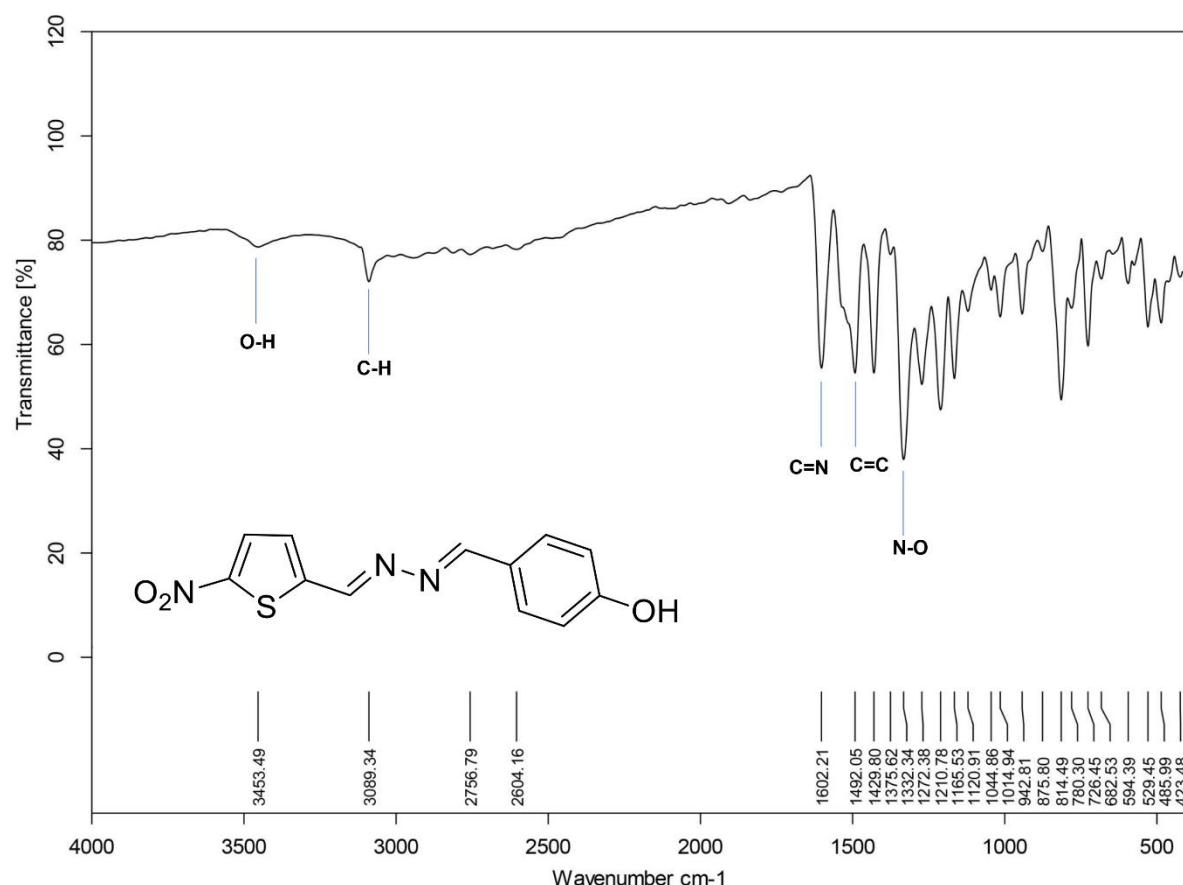
**<sup>1</sup>H NMR in DMSO**



**<sup>13</sup>C NMR in DMSO**



## IR Spectrum



## HRMS

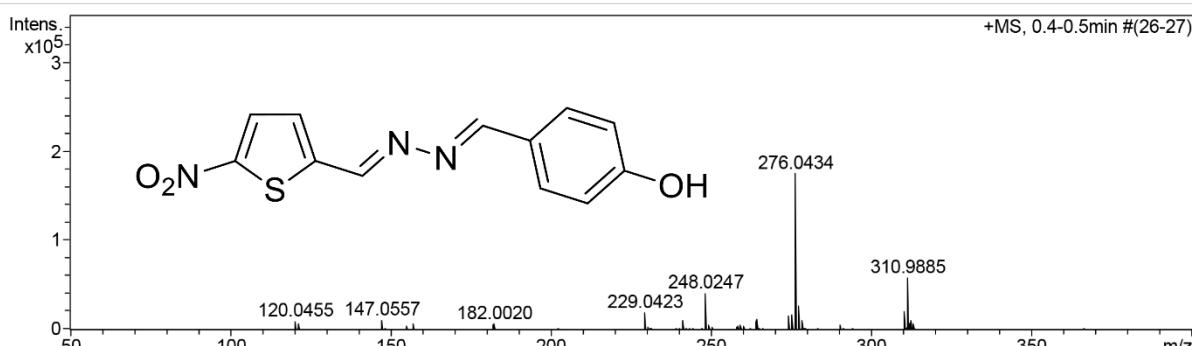
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000045.d       | Acquisition Date  | 10/12/2020 3:48:48 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-17                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

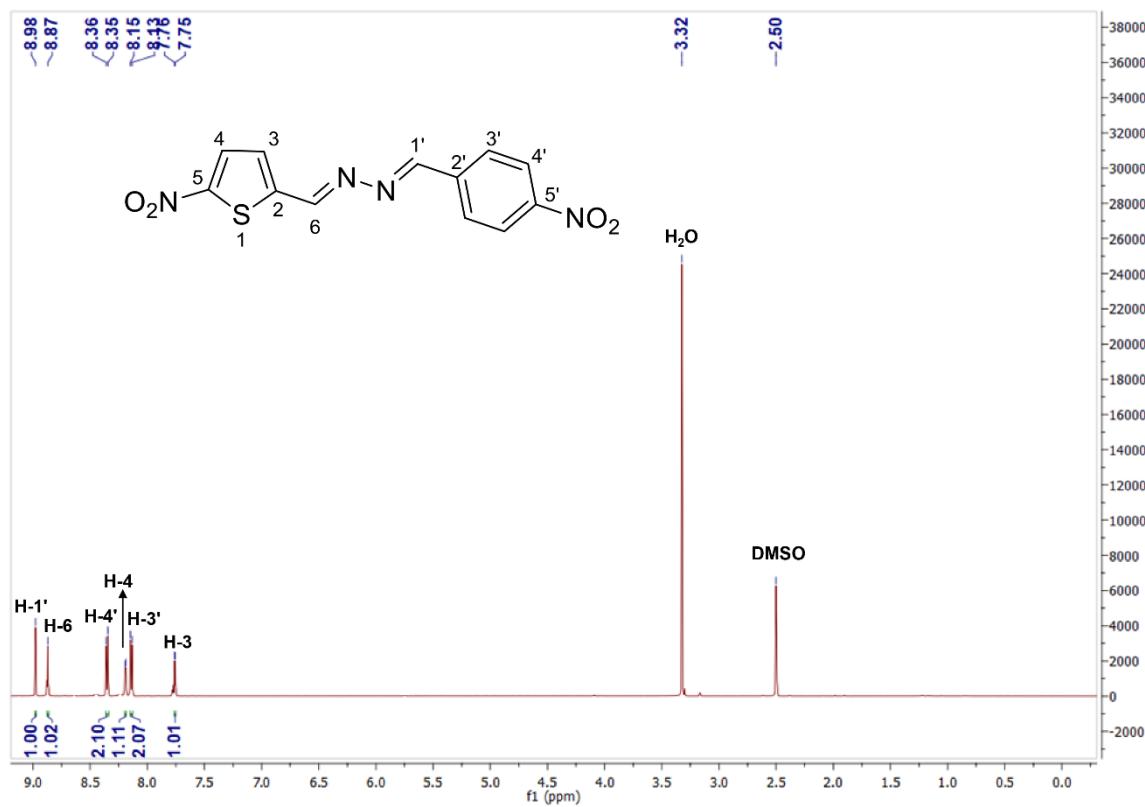
|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



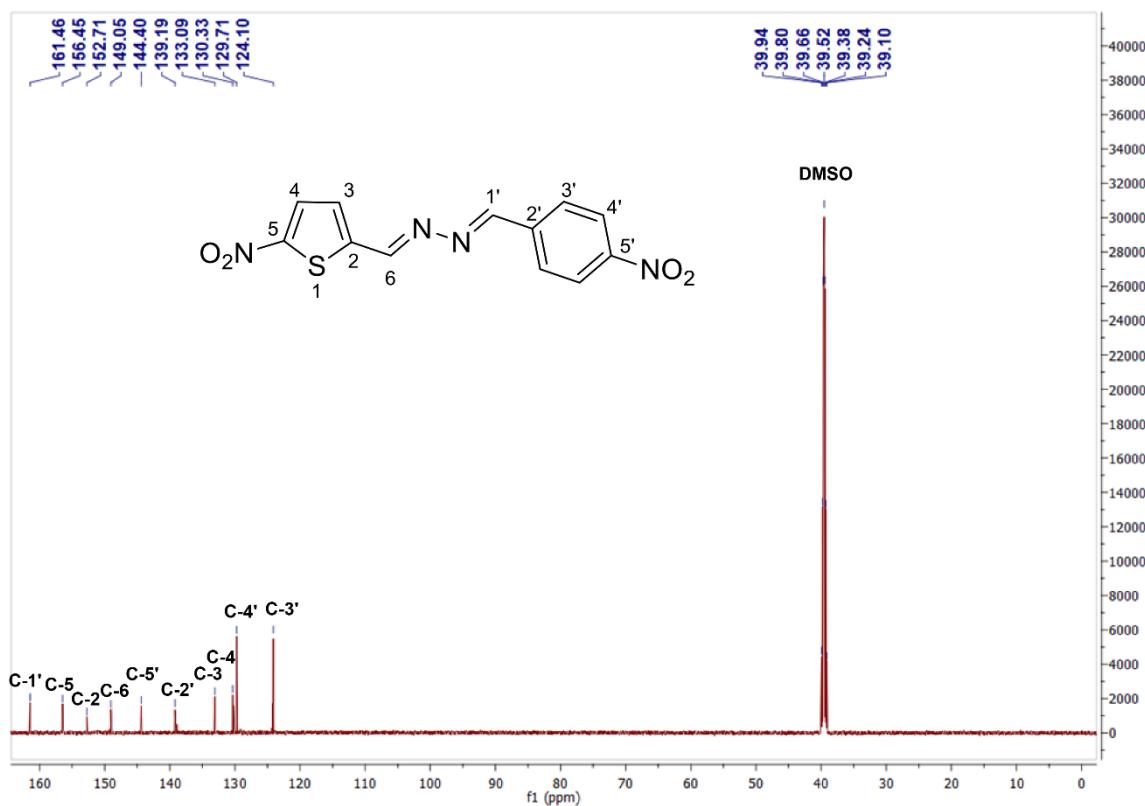
| Meas. m/z | # | Formula             | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb | e⁻ Conf | N-Rule |
|-----------|---|---------------------|--------|----------|-----------|-----------|--------|-----|---------|--------|
| 276.0434  | 1 | C 12 H 10 N 3 O 3 S | 100.00 | 276.0437 | 0.3       | 1.2       | 0.8    | 9.5 | even    | ok     |

**(1E,2E)-1-(4-nitrobenzylidene)-2-([5-nitrothiophen-2-yl]methylene)hydrazine (9b)**

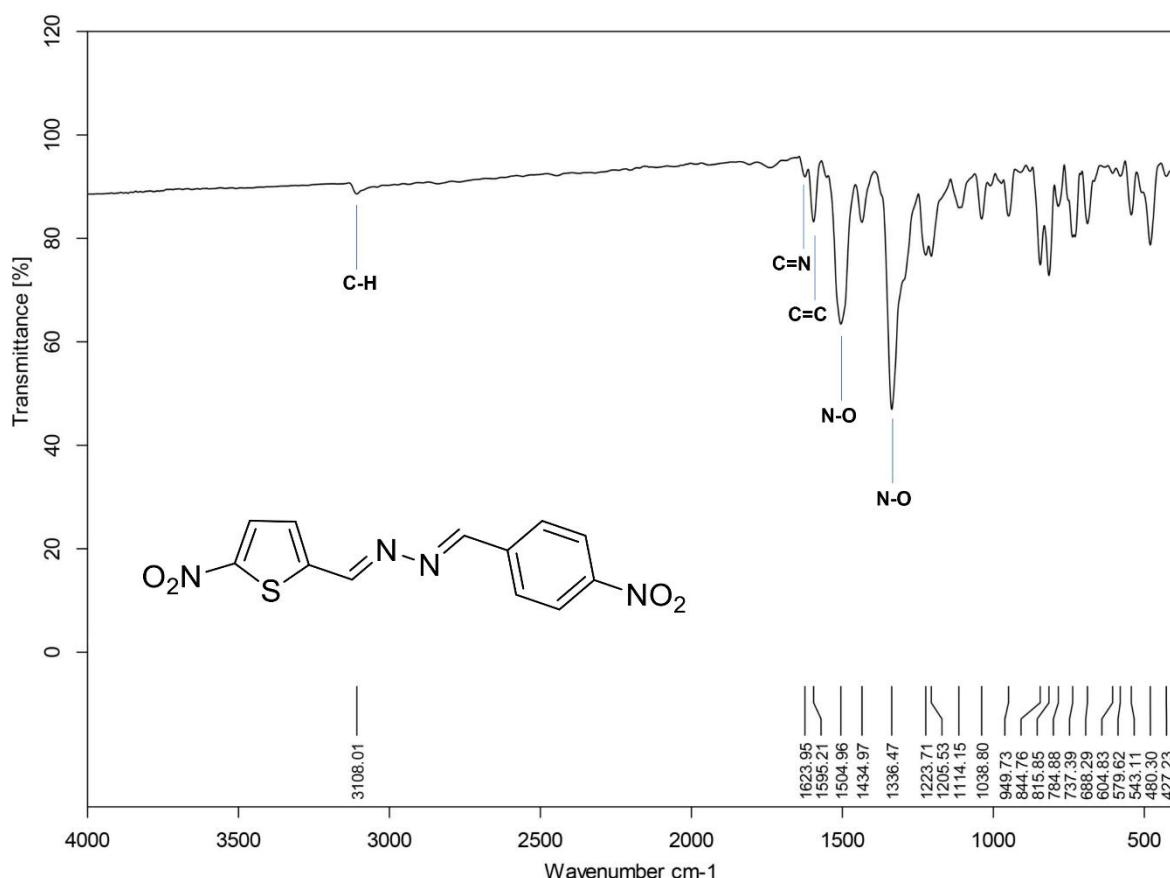
**<sup>1</sup>H NMR in DMSO**



**<sup>13</sup>C NMR in DMSO**



## IR Spectrum



## HRMS

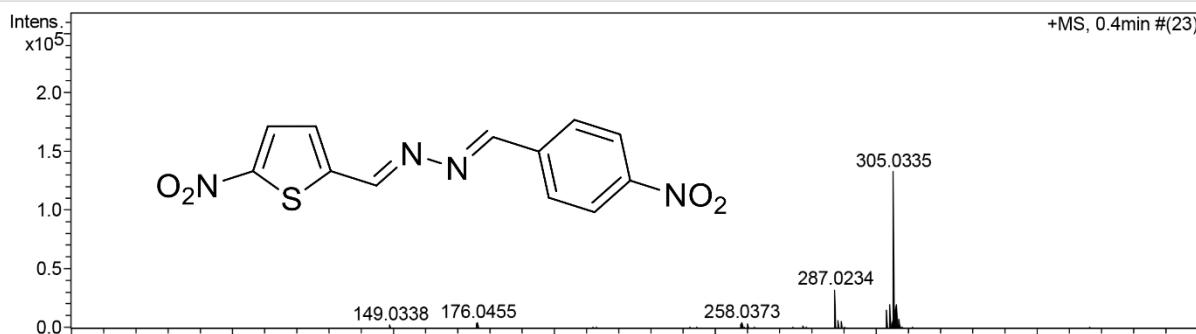
### Mass Spectrum SmartFormula Report

#### Analysis Info

|               |                                      |                   |                       |
|---------------|--------------------------------------|-------------------|-----------------------|
| Analysis Name | D:\Data\12102020\LADMS000041.d       | Acquisition Date  | 10/12/2020 3:43:42 PM |
| Method        | tune_low no focus50-1600da12102020.m | Operator          | Dr JHL Jordaan        |
| Sample Name   | MV-11                                | Instrument / Ser# | micrOTOF-Q II 2010390 |
| Comment       |                                      |                   |                       |

#### Acquisition Parameter

|             |            |                       |           |                  |           |
|-------------|------------|-----------------------|-----------|------------------|-----------|
| Source Type | APCI       | Ion Polarity          | Positive  | Set Nebulizer    | 1.6 Bar   |
| Focus       | Not active | Set Capillary         | 4500 V    | Set Dry Heater   | 200 °C    |
| Scan Begin  | 50 m/z     | Set End Plate Offset  | -500 V    | Set Dry Gas      | 8.0 l/min |
| Scan End    | 1600 m/z   | Set Collision Cell RF | 100.0 Vpp | Set Divert Valve | Waste     |



| Meas. m/z | # | Formula            | Score  | m/z      | err [mDa] | err [ppm] | mSigma | rdb  | e⁻ Conf | N-Rule |
|-----------|---|--------------------|--------|----------|-----------|-----------|--------|------|---------|--------|
| 305.0335  | 1 | C 12 H 9 N 4 O 4 S | 100.00 | 305.0339 | 0.4       | 1.4       | 2.3    | 10.5 | even    | ok     |

## **ANNEXURE B:**

Ethics approval certificate (NWU-HREC).



Private Bag X1290, Potchefstroom  
South Africa 2520

Tel: 086 016 9698  
Web: <http://www.nwu.ac.za>

North-West University Health Research Ethics Committee (NWU-HREC)

Tel: 018 299-1206  
Email: [Ethics-HRECApply@nwu.ac.za](mailto:Ethics-HRECApply@nwu.ac.za) (for human studies)

8 July 2020

## RESEARCH ETHICS COMMITTEE LETTER OF DECISION: NO RISK

Based on the review by the North-West University Health Research Ethics Committee (NWU-HREC) on 08/07/2020, the NWU-HREC hereby clears your study as a no risk study. This implies that the NWU-HREC grants its permission that, provided the general conditions specified below are met, the study may be initiated, using the ethics number below.

**Study title:** Synthesis and *in vitro* anti-protozoan activities of nitrofuran-based azine derivatives

**Principal Investigator/Study Supervisor/Researcher:** Prof DD N'Da

**Student:** M Viljoen - 27352935

**Ethics number:**

|   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|
| N | W | U | - | 0 | 0 | 3 | 8 | 5 | - | 2 | 0 | - | A | 1 |
|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|

Institution      Study Number      Year      Status

**Status:** S = Submission; R = Re-Submission; P = Provisional Authorisation;  
A = Authorisation

**Application Type:** Single study

**Commencement date:** 08/07/2020

**Risk:**

|         |
|---------|
| No Risk |
|---------|

### General conditions:

The following general terms and conditions will apply:

- The commencement date indicates the first date that the study may be started.
- In the interest of ethical responsibility, the NWU-HREC reserves the right to:
  - request access to any information or data at any time during the course or after completion of the study;
  - to ask further questions, seek additional information, require further modification or monitor the conduct of your research;
  - withdraw or postpone clearance if:
    - any unethical principles or practices of the study are revealed or suspected;
    - it becomes apparent that any relevant information was withheld from the NWU-HREC or that information has been false or misrepresented;
    - submission of the required amendments, or reporting of adverse events or incidents was not done in a timely manner and accurately; and/or
    - new institutional rules, national legislation or international conventions deem it necessary.
- NWU-HREC can be contacted for further information via [Ethics-HRECApply@nwu.ac.za](mailto:Ethics-HRECApply@nwu.ac.za) or 018 299 1206

**Special conditions of the ethical clearance due to the COVID-19 pandemic:**

Please note: Due to the nature of the study i.e. laboratory work involving the synthesis of medicinal compounds and *in vitro* testing, this study will be able to proceed during the current alert level, following receipt of this decision letter. No additional COVID-19 restrictions have been placed on the study except that the researcher must ensure that before proceeding with the study that all research team members have reviewed the North-West University COVID-19 Occupational Health and Safety Standard Operating Procedure.

The NWU-HREC would like to remain at your service and wishes you well with your study. Please do not hesitate to contact the NWU-HREC for any further enquiries or requests for assistance.

Yours sincerely,

Digitally signed by Prof.  
Petra Seiter  
Date: 2020.07.09  
15:28:18 +02'00'

---

NWU-HREC Chairperson

Digitally signed by Wayne  
Towers  
Date: 2020.07.08  
10:53:45 +02'00'

---

Head of the Faculty of Health Sciences Ethics Office for Research, Training and Support

Current details: (13210572) Q:\My Drive\My Documents\20190227\NWU-HREC\NWU-HREC\_Approval Letters\9.1.5.4.3\_LoD\_NWU-00000-20-A1\_2020mmdd.docm  
13 February 2020

File reference: 9.1.5.4.3

## **ANNEXURE C:**

The European Journal of Pharmaceutical Sciences Author Guidelines.



# EUROPEAN JOURNAL OF PHARMACEUTICAL SCIENCES

Official Journal of the European Federation for Pharmaceutical Sciences (EUFEPS)

## AUTHOR INFORMATION PACK

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| ● Editorial Board          | p.2 |
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ISSN: 0928-0987

### DESCRIPTION

The journal publishes research articles, review articles and scientific commentaries on all aspects of the pharmaceutical sciences with emphasis on conceptual novelty and scientific quality. The Editors welcome articles in this multidisciplinary field, with a focus on topics relevant for drug discovery and development.

More specifically, the Journal publishes reports on medicinal chemistry, pharmacology, drug absorption and metabolism, pharmacokinetics and pharmacodynamics, pharmaceutical and biomedical analysis, drug delivery (including gene delivery), drug targeting, pharmaceutical technology, pharmaceutical biotechnology and clinical drug evaluation. The journal will typically not give priority to manuscripts focusing primarily on organic synthesis, natural products, adaptation of analytical approaches, or discussions pertaining to drug policy making.

Scientific commentaries and review articles are generally by invitation only or by consent of the Editors. Proceedings of scientific meetings may be published as special issues or supplements to the Journal.

Manuscripts submitted to the Journal are only accepted on the understanding that (a) they are subject to editorial review (generally by two independent reviewers); (b) they have not been, and will not be, published in whole or in part in any other journal; (c) the recommendations of the most recent version of the Declaration of Helsinki, for humans, and the European Community guidelines as accepted principles for the use of experimental animals have been adhered to.

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#### *Types of Paper*

##### *Research articles*

*The European Journal of Pharmaceutical Sciences* publishes research articles in the multidisciplinary field of pharmaceutical sciences, with a focus on topics relevant for drug discovery and development.

More specifically, the Journal publishes reports on medicinal chemistry, pharmacology, drug absorption and metabolism, pharmacokinetics and pharmacodynamics, pharmaceutical and biomedical analysis, drug delivery (including gene delivery), drug targeting, pharmaceutical technology, pharmaceutical biotechnology and clinical drug evaluation.

The journal will typically not give priority to manuscripts focusing primarily on organic synthesis, natural products, adaptation of analytical approaches, or discussions pertaining to drug policy making.

Important other criteria for manuscript acceptance are conceptual novelty, scientific rigorously of the experiments, relevance for a broad readership beyond the specific topic of the manuscript, and adherence to high ethics standards of experimentation. Research articles should comply with the format requirements set forth in the section "Article Structure below".

##### *Review articles*

The manuscript of a review article should be arranged as described for research articles but according to the following sections: title page, abstract and keywords (indexing terms, normally 3-6 items), Introduction, Specific sections determined by the author, Conclusions, Acknowledgements, References, Figure legends and Figures, Tables. Sections ranging from the Introduction to the Conclusions should be numbered. Subdivisions within a section should also be numbered within that section: 2.1., 2.2., 2.3. etc. All pages should be numbered consecutively, the title page being p.1.

##### *Commentaries and Mini-reviews*

One page suggestions for comprehensive reviews, commentaries or mini-reviews should be sent to the Editor-in-Chief at ejps@sdu.dk for consideration. Please see detailed information on commentaries and mini-reviews below.

##### *Commentaries (Guidance)*

The definition of a Commentary for EJPS is three-fold. Firstly, it can be an argued piece of provocative scientific writing purporting to take a balanced position on a controversial pharmaceutical science topic. A second option is for the author to approach the topic from a particular viewpoint on one side of an argument. A third option is to provide a topical update on a hot topic in Pharmaceutical Sciences and this can be more informative than controversial.

Commentaries will be commissioned by the editors in advance or invited from non-commissioned authors if they wish to initially submit a one page summary of the intended Commentary to the editors in advance. All manuscripts will be assessed by 2-3 independent referees.

The journal is looking for a stimulating and provoking essays, with referenced material, but without an extensive reference list. Commentaries can contain one summary figure and/or table and should have no more than 30 references to preferably recent peer-reviewed material. The word count should be approximately 2,000 words maximum.

The commentary should have a short abstract summary of 150 to 200 words and 4-5 key words should be included. The text should be broken down into 4-5 numbered sections beginning with an Introduction and ending with a Conclusions section. A model of the structures is to be found in Eur. J. Pharm. Sci. 19, 1-11 by R.D. Combes.

#### *Mini-review (Guidance)*

Mini-reviews are thought provoking reviews of contemporary pharmaceutical research. Themes are as described in the Scope of the Journal section.

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The structure of the mini-review is as follows: a title page followed by a 200-300 word abstract with 4-5 key words. The text is then divided into numbered sections finishing with a Summary section. References should be kept to a maximum of 60 and should be mostly to recent peer-reviewed material. There is a combined maximum of 5 figures / tables. Authors are encouraged to submit their original unpublished work as part of the review if appropriate. The total length of the review should be a maximum of 4,000 words.

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Provide sufficient details to allow the work to be reproduced by an independent researcher. Methods that are already published should be summarized, and indicated by a reference. If quoting directly from a previously published method, use quotation marks and also cite the source. Any modifications to existing methods should also be described.

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Results should be clear and concise.

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The main conclusions of the study may be presented in a short Conclusions section, which may stand alone or form a subsection of a Discussion or Results and Discussion section.

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If there is more than one appendix, they should be identified as A, B, etc. Formulae and equations in appendices should be given separate numbering: Eq. (A.1), Eq. (A.2), etc.; in a subsequent appendix, Eq. (B.1) and so on. Similarly for tables and figures: Table A.1; Fig. A.1, etc.

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A concise and factual abstract is required. The abstract should state briefly the purpose of the research, the principal results and major conclusions. An abstract is often presented separately from the article, so it must be able to stand alone. For this reason, References should be avoided, but if essential, then cite the author(s) and year(s). Also, non-standard or uncommon abbreviations should be avoided, but if essential they must be defined at their first mention in the abstract itself.

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### **Keywords**

Immediately after the abstract, provide a maximum of 6 keywords, using American spelling and avoiding general and plural terms and multiple concepts (avoid, for example, 'and', 'of'). Be sparing with abbreviations; only abbreviations firmly established in the field may be eligible. These keywords will be used for indexing purposes.

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Abbreviations are a hindrance for the reader. Use as few abbreviations as possible and write out names of compounds, receptors, etc., in full throughout the text of the manuscript, with the exceptions given below. Unnecessary and nonsense abbreviations are not allowed. Generic names should not be abbreviated. As an example, AMP, HAL, HIST, RAMH, TAM, SST, for amphetamine, haloperidol, histamine, (R)-o-methylhistamine, tamoxifen, somatostatin, are not accepted. Abbreviations which have come to replace the full term (e.g., GABA, DOPA, PDGF, 5-HT, for Y-aminobutyric acid, 3,4-dihydroxyphenylalanine, PDGF, 5-hydroxytryptamine) may be used, provided the term is spelled out in the abstract and in the body of the manuscript the first time the abbreviation is used. Unwieldy chemical names may be abbreviated. As an example, 8-OH-DPAT, DOI, DTG, BAPTA, for 8-hydroxy-2-(di-n-propylamino)tetralin, 1-(2,5-dimethoxy-4-iodophenyl)-2-aminopropane, 1,3-di(2-tolyl)-guanidine, 1,2-bis(o-aminophenoxy)ethane-N,N,N',N'-tetraacetic acid, are acceptable; however, the full chemical name should be given once in the body of the manuscript and in the abstract, followed in both cases by the abbreviation. Code names may be used, but the full chemical name should be given in the text and in the abstract. *Authors not conforming to these demands may have their manuscripts returned for correction with delayed publication as a result.*

Some abbreviations may be used without definition:

1 ADP, CDP, GDP, IDP5'-pyrophosphates of adenosine UDPcytidine, guanosine, inosine, uridine AMP etc. adenosine 5'-monophosphate etc. ADP etc. adenosine 5'-diphosphate etc. ATP etc. adenosine 5'-triphosphate etc. CM-cellulosecarboxymethylcellulose CoA and acetyl-CoAcoenzyme A and its acyl derivatives DEAE-celluloseO-(diethylaminoethyl)-cellulose DNAdeoxyribonucleic acid EGTAethylene glycol-bis(β-aminoethyl ether)N,N,N',N'-tetraacetic acid FADflavin-adenine dinucleotide FMNflavin mononucleotide GSH, GSSGglutathione, reduced and oxidized Hepes4-(2-hydroxyethyl)-1-

piperazine-ethanesulphonic acid NADnicotinamide-adenine dinucleotide NADPnicotinamide-adenine dinucleotide phosphate NMNnicotinamide mononucleotide P<sub>i</sub>, PP<sub>j</sub>orthophosphate, pyrophosphate RNAribonucleic acid Tris2-amino-2-hydroxymethylpropane-1,3-diol

Two alternative conventions are currently in use in some cases. For example, for the phosphoinositides there are both the abbreviations recommended by the IUPAC-IUB and those of the Chilton Convention (e.g., PtdIns(4,5)P<sub>2</sub> vs. PIP<sub>2</sub>for phosphatidylinositol 4,5-biphosphate). The journal will accept either of these forms but not their combination.

Abbreviations of units of measurements and other terms are as follows:

*Units of mass*

1 kilogramkg gramg milligrammg microgramμg nanogramng mole (gram-molecule)mol millimolemmol micromoleμmol nanomolenmol picomolepmol femtomolefmol equivalenteq

*Units of time*

1 hour minutemin seconds millisecondms microsecondμs

*Units of volume*

1 litrel millilitreml microlitreμl

*Units of length*

1 metrem centimetremcm millimetremmm micrometreμm nanometrenm

*Units of concentration*

1 molar (mol/l)M millimolar mM micromolarμM nanomolarnM picomolarpM

*Units of heat, energy, electricity*

1 jouleJ degree Celsius (centigrade)<sup>o</sup>C coulombC ampereA voltV ohmΩ siemensS

*Units of radiation*

1 curieCi counts per minutecpm disintegrations per minutecdpm becquerelBq

*Miscellaneous*

1 gravityg dissociation constantK<sub>d</sub> median dosesLD<sub>50</sub>, ED<sub>50</sub> probabilityP routes of drug administrationi.v., i.p., s.c., i.m. square centimetremcm<sup>2</sup> standard deviationS.D. standard error of the meanS.E.M. Svedberg unit of sedimentation coefficientS Hill coefficientn<sub>H</sub>

The isotope mass number should appear before the atomic symbol, e.g., [<sup>3</sup>H]noradrenaline, [<sup>14</sup>C]choline. Ions should be written: Fe<sup>3+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>. The term absorbance (A) is preferred to extinction or optical density. For abbreviations not included in this list consult: *Units, Symbols and Abbreviations, A Guide for Biological and Medical Authors and Editors*, 1994 (The Royal Society of Medicine, London), ISBN 0-905958-78-0, or *Scientific Style and Format. The CBE Manual for Authors, Editors, and Publishers*, 6th edn. (Cambridge University Press, Cambridge), ISBN 0-521-47154-0.

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## Synthesis and *in vitro* anti-protozoan activities of nitrofuran-based azine derivatives

M Viljoen



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Dissertation submitted in partial fulfilment of the requirements for the  
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